# X-RAY AND OPTICAL STUDY OF ALKALI FELDSPAR: II. AN X-RAY METHOD FOR DETERMINING THE COMPOSITION AND STRUCTURAL STATE FROM MEASUREMENT OF 20 VALUES FOR THREE REFLECTIONS.<sup>1</sup>

## THOMAS L. WRIGHT, U. S. Geological Survey, Hawaii National Park, Hawaii

#### ABSTRACT

Measured  $2\theta$  values of the 201, 060, and 204 diffraction peaks of natural and synthetic alkali feldspars empirically can be linearly related to the *a*, *b*, and *c* cell parameters, respectively. The structural state of many alkali feldspars may be estimated directly from a plot of  $2\theta$  (060) against  $2\theta$  (204) prepared for feldspars of known structural state. Feldspars having anomalous cell dimensions may also be recognized on such a plot. Composition of feldspars having normal cell dimensions may be determined from  $2\theta$  (201). Starting parameters for a computer refinement of the cell may be obtained from equations relating cell dimension to  $2\theta$  value of a single diffraction peak for a wide variety of natural, synthetic, and cation-exchanged alkali-feldspar phases. The 'three-peak' method permits any geologist who has access to an X-ray diffractometer to describe and name alkali feldspar phases commonly encountered in petrologic studies.

### INTRODUCTION

In a preceding paper (Wright and Stewart, 1968; Part I of this study), a method was described by which composition and structural state of alkali feldspar could be determined from cell dimensions refined by computer using powder X-ray diffraction data. While processing the computed results, certain distinctive and ubiquitous feldspar reflections were found whose  $2\theta$  values depended principally on a single unit-cell dimension. Discovery of such reflections suggested the possibility of using the  $2\theta$  values instead of cell parameters to estimate structural state and composition of natural alkali feldspars. The three most useful reflections and the cell parameter on which their  $2\theta$  value depends are  $\overline{2}01-a$ , 060-b, and  $\overline{2}04-c$ . The reflections are commonly of sufficient intensity to be accurately measured and indexed from diffraction patterns of alkali feldspars collected from a wide variety of geological environments. Best fit curves relating the  $2\theta$  values to the appropriate cell dimension are essentially linear and permit easy estimation of starting parameters for a complete refinement of the unit cell. The sections following describe this 'three-peak' method of studying alkali feldspars in some detail.

### METHODS OF STUDY

The  $2\theta$  values and cell dimensions that form the basis for the 'three-

<sup>1</sup> Publication authorized by the Director, U. S. Geological Survey

peak' method are abstracted from the data for synthetic, base-exchanged, and natural alkali feldspars compiled in Part I.

Methods of measurement are described in the appendix of that paper. Cell dimensions were obtained using the computer program of Evans, Appleman, and Handwerker (1963) with subsequent modifications by Appleman and Handwerker. Equations relating cell dimension to  $2\theta$ value were computed using a standard least-squares program.

Identification of reflections. The approximate ranges of  $2\theta$  (CuK $\alpha_1$ ) for five reflections that are used in the 'three-peak method' are listed in Table 1. (See also Tables 11–15, Part I). The 201 and 060 reflections may be indexed unambiguously in all natural alkali feldspars. The 204 is a strong, single reflection in most orthoclase, microcline, anorthoclase,

M.:II	A	Approx	imate Range of $2\theta$ (	CuKa1)
Indices, Intensi hkl	Intensity	Potassium-rich phases	Anorthoclases	Albite
(201)	40	20.8°-21.2°	21.6°-21.9°	21.9°-22.1°
(002)	>100	27.4°-27.8°	27.8°-28.0°	27.9°-28.1°
(113)	8	38.6°-39.0°	not present	not present
(060)	25	41.6°-42.0°	41.7°-42.0°	42.2°-42.6°
(204)	30	50.5°-51.1°	51.1°-51.3°	51.2°-51.5°

Table 1. Position of Alkali Feldspar Reflections Used in 'Three-Peak' Refinement

and albite samples. In sanidine or orthoclase having anomalous cell dimensions  $\overline{2}04$  is commonly either overlapped by or joined by one or more additional reflections. If reflections in the vicinity of  $\overline{2}04$  are sharp,  $2\theta$  values of the  $\overline{1}13$  and 002 reflections may be used to make an unambiguous identification of the reflection whose index is  $\overline{2}04$ , as shown in Figure 1.

## PRESENTATION OF DATA

 $2\theta$  values for 060 and  $\overline{2}04$  are plotted against one another in Figures 2 and 3 for the three complete series of feldspars and for most of the other cation-exchanged feldspars reported on by Orville, (1967) and in Part I. The presentation is analogous to the b-c plots of the latter paper (Figs. 2a and 2b). The data from which Figures 2 and 3 are constructed are given in Table 3.  $2\theta$  values for SynSanShaw, P50—90 KF, and their exchange equivalents are given in Table 3 but not plotted on Figure 3 as

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the former was shown to have anomalous cell dimensions and the latter has considerable Ba and Sr in its bulk composition (see Part I). For the three complete feldspar series  $2\theta$  values of  $\overline{2}01$  are plotted against Or content in Figure 4, which can be compared with the plot of *a* against Or content for these same series (Part I, Fig. 1a). Equations relating  $2\theta$ value of  $\overline{2}01$  to Or content are given in Table 4.

Equations relating  $2\theta$  value of a single diffraction peak to the appropri-



FIG. 1. Plotted lines that enable identification of  $\overline{2}04$  where more than one reflection occurs between 50° and 51° 2 $\theta$ , CuK $\alpha_1$ . The upper line relates the position of the  $\overline{1}13$  and  $\overline{2}04$  reflections, according to the relationship  $2\theta$  ( $\overline{2}04$ ) = 1.1780 [ $2\theta$  ( $\overline{1}13$ )]+5.1048±0.0202° 2 $\theta$  ( $\overline{2}04$ ). The lower curve relates the position of the 002 and  $\overline{2}04$  reflections, according to the relationship  $2\theta$  ( $\overline{2}04$ )=1.6886 [ $2\theta$  (002)]+4.1690±0.0317° 2 $\theta$  ( $\overline{2}04$ ). Light boundary lines are drawn at points corresponding to  $\pm 2$  standard deviations of  $2\theta$  ( $\overline{2}04$ ).

ate cell parameter are tabulated and discussed in an Appendix. These equations demonstrate the linear relationship between  $2\theta$  value and cell dimension and may be used to obtain starting parameters for computer refinement of the unit cell. Cell dimensions computed from the equations are not, of course, as accurate as those obtained from a good cell refinement and should not be quoted in comparison with feldspars whose dimensions have been so refined.

In the course of tabulating the values of  $2\theta$  for 204, 060, and  $\overline{2}01$  it was discovered that the values calculated by the computer in a cell refine-

ment differed by as much as  $\pm 0.03^{\circ} 2\theta$  from the observed values for these reflections. In Table 2 the deviations of calculated from observed  $2\theta$  values are summarized for each of the three reflections measured for each complete series. In all but two instances the average absolute deviation



FIG. 2. Observed 20 values of 060 plotted against observed 20 values of 204 for members of three complete series of alkali feldspars reported in Part I. Solid Squares: Maximum microcline—low albite series, data of Orville (1967). Open Squares: P50-56F (orthoclase) series, data of Wright and Stewart (1968). Solid Circles: High sanidine-high albite series, data of Orville (1967). Open Circles: High sanidine-high albite series, data of Donnay and Donnay (1952 and

written communication, 1963).

Data points are drawn to  $\pm 0.02^{\circ} 2\theta$  which exceeds the range observed in measurement of three X-ray patterns for each sample.

is less than  $\pm 0.01^{\circ} 2\theta$ , well within the error inherent in measurement of the reflection. The average deviations using the signed values of  $2\theta$  (observed) minus  $2\theta$  (calculated) are also shown in Table 2. Theoretically these average deviations should all be zero for a random deviation of the calculated values around the observed ones. In the actual cases there is always a small bias. For the sake of consistency and maximum applic-



FIG. 3. Alkali exchange paths (see Part I, Fig. 2) on a plot of  $2\theta(060)$  against  $2\theta(204)$ . Solid lines are drawn by eye from the data of Figure 2. Solid squares define the approximate positions of maximum high and low structural states at NaAlSi<sub>3</sub>O<sub>8</sub> and KAlSi<sub>3</sub>O<sub>8</sub> bulk composition. Dashed lines connect data for exchanged feldspars studied in Part I. Crosscontours are  $2\theta(201)$ , interpolated as straight lines between points for the three complete series. Data for natural maximum microclines (Wright and Stewart, unpub. data), and for natural anorthoclases (Carmichael and Mackenzie, 1964, Stewart, unpub. data) are plotted as small circles, for comparison with Part I, Fig. 2.

ability to observations of natural alkali feldspars, the equations of Tables 5 and 6 and the data of Figures 2–4 relate the *observed 20* to the appropriate computed cell parameter.

# DETERMINATION OF COMPOSITION AND STRUCTURAL STATE OF ALKALI FELDSPAR AND IDENTIFICATION OF 'ANOMALOUS' FELDSPARS USING THE 20 VALUES FOR 201, 060, AND 204

The methods of this section are exactly analogous to those described in Part I (p. 45ff) using the cell dimensions a, b, and c, and the feldspar can also be named and described using terminology given in Part I (p. 33ff). The procedure following the 'three-peak' method is as follows:

1. Measure carefully  $2\theta$  values for 201, 060, 204, 113 (if present), and



FIG. 4. Or content plotted against  $2\theta(\overline{2}01)$  for three complete series of exchanged feldspars. Data sources and symbols as given for Figure 2. Equations relating  $2\theta(\overline{2}01)$  to Or content for each series are given in Table 4.

002 using methods described in Part I (Appendix). Refer to Figure 1 to identify  $\overline{2}04$  when more than one sharp diffraction peak appears between 50° and 51° 2 $\theta$  (CuK $\alpha_1$ ). In the rare cases where reflections near to and including  $\overline{2}04$  are fuzzy the 'three-peak' method is inapplicable.

2. Plot the  $2\theta$  values of 060 and  $\overline{2}04$  on Figure 3 and read off the approximate  $2\theta$  value of  $\overline{2}01$  from the contours. If this value agrees within

 $0.1^{\circ} 2\theta$  with the  $2\theta$  value directly measured for  $\overline{2}01$  the feldspar may be assumed to have normal cell dimensions. In cases where the  $2\theta$  value of  $\overline{2}01$  derived from Figure 3 exceeds the measured value by  $0.1^{\circ} 2\theta$  or more then the feldspar may be termed anomalous.

3. For normal feldspars structural state is obtained directly from

	Pa		$(2\theta \text{ obset})$	erved)– $(2\theta$ of	calculated) <sup>1</sup>	Number
Series	flec- tion	Absolute Devia- tion	Average Absolute Deviation	Signed Deviation	Average Signed Deviations	of Observa- tions
High sanidine—	(201)	.148	.007	+.014	+0.001	20
high albite	(060)	.090	.0045	060	-0.003	20
Orville (1967)	$(\overline{2}04)$	.156	.008	138	-0.007	20
High sanidine—	(201)	.054	.006	006	-0.0005	9
high albite	(060)	.123	.014	+.117	+0.012	9
Donnay and Donnay (1952)	(204)	.047	.010	+.047	+0.010	5
P50–56 KF	(201)	.072	.007	010	-0.001	10
Orthoclase	(060)	.051	.005	005	-0.0005	10
Wright and Stewart (1968)	(204)	.065	.0065	011	-0.001	10
Maximum microcline	(201)	.070	.0065	010	-0.001	11
low albite	(060)	.051	.005	040	-0.004	10
Orville (1967)	$(\bar{2}04)$	.041	.004	+.011	+0.001	11

TABLE 2	. Deviations	OF CALCULATED	VALUES	FROM	OBSERVED
	VALUES OF	$2\theta(\bar{2}01), 2\theta(060)$	, AND $2\theta$	(204)	

<sup>1</sup> Differences are calculated as absolute (unsigned) quantities in column 3 and 4. In columns 5 and 6 the signs of the differences are retained. The values in columns 5 and 6 should be zero for perfectly random variation.

Figure 3 in terms of the closest plotted feldspar of known structural state (see example below). Because of the difference in ordering scheme for Na-rich compared with K-rich feldspars, discussed in detail in Part I (p. 27ff), the structural state should be described only in terms of the cation-exchanged feldspar series in which the starting material was nearly the same composition as the unknown. The structural state of natural *albite* and *anorthoclase* should be compared to the equivalents of high albite (including S63-30), S62-34, Albite III, and low albite. The structural state of *potassium-rich feldspars* should be compared to the equivalents for High Sanidine, Puye (=low sanidine), P50-56F, Benson, SH 1070, Spencer B, Spencer U, and maximum microcline.

Foldenar Sories	Bulk Composition	Obse	erved 20 (CuK	$(\alpha_1)^a$
reluspar series	(Wt. % Or)	(201)	(060)	(204)
Synthetic high sanidine-	100	20,959	41.577	50.845
high albite	95	20,995	41.558	50.836
(Orville, 1967)	90	21.059	41.577	50.850
	85	21.085	41.575	50.852
	75	21,197	41.598	50.892
	70	21.254	41.611	50.888
	65	21 300	41.605	50.904
	60	21 369	41.652	50,920
	55	21 404	41 659	50.930
	50	21 480	41,680	50.932
	45	21 525	41 696	50.945
	40	21 591	41 722	50.984
	35	21 657	41.764	51.034
	30	21 703	41 828	51.082
	25	21 757	41 896	51 130
	20	21 807	41.954	51,194
	15	21 856	42 023	51,261
	10	21 918	42,103	51.322
	5	21 977	42.171	51.390
	Ő	22.019	42.222	51.433
Synthetic high sanidine-	100	20.945	41,570	50.870
high albite	80	21.160	41.590	50.890
(Donnay and Donnay 1952)	60	21.340	41,650	50.920
	50	21,470	41.690	50.945
	40	21.580	41.745	50.960
	30	21.715	41.835	51.068*
	20	21.805	41.955	51.175*
	10	21.915	42.080	51.281*
	0	22,000	42.195	51.468*
Orthoclase (P50-56KF)	100 (extrapolated)	20,905*	41.655*	50.700*
(Wright and Stewart 1968)	94.5	20.960	41.653	50.723
	86.8	21.018	41,653	50.738
	85.9	21.051	41.660	50.706
	76.7	21.146	41.651	50.730
	68.3	21.225	41.671	50.766
	60.4	21.301	41.693	50.766
	51.6	21.431	41.716	50.783
	42.8	21.541	41.763	50.835
	34.0	21.655	41.801	50.870
	26.3	21.723	41.916	50.955
	16.8	21.820	42.045	51.090
	8.4	21.918	41.171	51.203
	0 (extrapolated)	22.043*	42.291*	51.320*
Maximum microcline-low	99.5	20.993	41.809	50.527
albite	89.7	21.091	41.829	50.565
(Orville 1967)	79.8	21.176	41.841	50.576
	70.0	21.254	41,853	50.589
	60.1	21.387	41.900	50.642
	50.3	21.465	41.938	50.680
	40.4	21.570	41.999	50.723
	30.6	21.684	42.094	50.810
	20.7	21.815	42.245	50.948
	10.9	21.945	42.441*	51.082
	1.0	22.057	42.500	51.142

### TABLE 3. DATA USED TO PREPARE FIGURES 2 AND 3.

Feldspar	Series	Bulk Compo (Wt. % C	osition Dr)	(201) (Ob	served 20 Cul (060)	ζ <sub>α1</sub> ) <sup>a</sup> (204)
SynSanShaw ( Stewart, 1968,	Wright and Table 2)	100 0		20.950 22.011	41.583 42.206	50.900* 51.435
S 63-30		0 100		$\begin{array}{c} 22.024\\ 20.978\end{array}$	$\begin{array}{r} 42.231\\ 41.546\end{array}$	$\begin{array}{c} 51.504 \\ 50.838 \end{array}$
Puye		42 95 2		21.565 20.988 21.990	41.739 41.590 42.215	$50.964 \\ 50.796 \\ 51.411$
P50-90		62 92 12		21.316 20.985 21.876	41.618 41.571 42.076	$50.863 \\ 50.773 \\ 51.260$
Benson		94 7 96		20.955 21.938 20.931	$\begin{array}{r} 41.653 \\ 42.237 \\ 41.659 \end{array}$	$50.670 \\ 51.230 \\ 50.678$
SH 1070		$87\\94.5\\4$		$21.048 \\ 20.961 \\ 21.960$	$\begin{array}{r} 41.690 \\ 41.686 \\ 42.255 \end{array}$	$50.690 \\ 50.651 \\ 51.226$
Spencer B		91 ~6		21.058 21.927	$\begin{array}{r} 41.743\\ 42.216\end{array}$	$50.625 \\ 51.047$
Spencer U		$\overset{94}{\sim}_{0}$		20.998 20.986 22.020	41.792 41.766 42.482	$50.583 \\ 50.585 \\ 51.159$
S62-34		$\sim_{100}^{\sim_{0}}$		$22.031 \\ 20.946$	42.282 41.610	$51.392 \\ 50.772$
Low albite III		0 100		22.013 20.928	$\begin{array}{r} 42.360\\ 41.726\end{array}$	51.303 50.657
Natural maxin Amicr (Stewar SH 22500 (Wr MP-3 Mi (Wri P63-1 Mi (Do. Natural anorth	num microcli t, unpub.) ight 1967) ght, unpub.) ) noclases	ines 96 96 94.2 93.3		21.003 21.006 21.020 21.040	$\begin{array}{r} 41.797 \\ 41.803 \\ 41.800 \\ 41.803 \end{array}$	50.512 50.546 50.543 50.560
AHR 48° AHR 35° AHR 50° AHR 42° BTB 8 FR-1° #1° #2 #3 #4 #5 #6 #7	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Other <sup>b</sup>  0.7 0.8 1.2 1.2 1.0 1.0 1.5	21.795 21.800 21.795 21.766 21.850 21.830 21.899 21.866* 21.801* 21.797* 21.808 21.771* 21.791*	$\begin{array}{c} 41.947\\ 41.938\\ 41.955\\ 41.955\\ 41.900\\ 42.000\\ 42.030\\ 42.010^*\\ 41.972^*\\ 41.938^*\\ 41.947\\ 41.880^*\\ 41.936^* \end{array}$	$\begin{array}{c} 51.197\\ 51.240\\ 51.212\\ 51.218\\ 51.270\\ 51.280\\ 51.304\\ 51.290^*\\ 51.217^*\\ 51.23^*\\ 51.165\\ 51.179^*\\ 51.185^* \end{array}$

TABLE 3—(continued)

	R 11 G .	В	ulk Com	osition	Obse	erved 20 (CuK	$(\alpha_1)^a$
	Feldspar Series		(Wt. %	Or)	(201)	(060)	(204)
#8	22.5	72.0	4.4	1.1	21.736*	41.913*	51.191*
#9	23.0	69.3	7.4	0.3	21.776*	41.923*	51.227*
#10	23.2	75.2	1.2	0.4	21.741*	41.898*	51.117*
#11	23.3	73.8	1.5	1.3	21.742*	41.871*	51.125*
#12	24.0	65.9	10.1		21.696*	41.848*	51.082*
#13	31.7	66.3	1.8	0.2	21.677*	41.753*	51.018*
#14	32.5	61.4	6.1	100	21.694*	41.830*	51.073*
#15	33.0	66.8	0.2		21.654*	41.742*	51.018*

TABLE 3—(continued)

 $^{\rm a}$  Where  $2\theta$  was not measured the value calculated from the refinement is given and starred, e.g.,  $51.068^*.$ 

<sup>b</sup> Analyses of specimens.

<sup>c</sup> Stewart, unpub.

<sup>d</sup> Carmichael and Mackenzie (1964, Table II).

Composition of normal feldspars may be obtained from Figure 4 or the equations of Table 4 once structural state is established.

4. For anomalous feldspars the 'apparent' structural state may be derived in the same way as for normal feldspars. It is yet to be established to what extent this will depart from the true structural state (see discussion in Part I (p. 71). Composition *cannot* be obtained from the  $2\theta$  value for  $\overline{201}$ . However, the appendix tables may be used to derive starting parameters for a cell refinement from which cell volume may be used to estimate composition (Part I, p. 71).

Three samples of alkali feldspar, previously named and described in

	S	TATE, WHERI	z y = mx + b		
Feldspar series	m	b	No. of Observa- tions	Standard error of y <sup>a</sup>	Reference
High sanidine- high albite	-92.18	+2030.05	29	1.17 (3–0.91)	Orville (1967)
P50-56KF (Orthoclase)	-87.69	+1930.77	12	1.37 (3–1.13)	Wright and Stewart (1968)
Maximum microcline- low albite	-92.19	+2031.77	11	2.40 (2–1.08)	Orville (1967)

TABLE 4. EQUATIONS RELATING  $2\theta$  (201) (x) and Of Content (y) for Complete Series of Alkali Feldspars of Constant Structural State Where x = mx + b

<sup>a</sup> Where the best fit is not linear the degree and standard error of the best fit polynomial are given in parenthesis, *e.g.* (2-0.0042) indicates the best fit is a quadratic for which the standard error of y is 0.0042

	Symmetry	Varia	bles	Equation	v = mx + b	No. of	Standard
Feldspar series	and Composition	8	y	w	9	Obser- vations	error of y <sup>a</sup>
High sanidine-high albite	Monoclinic Or≥40	$2\theta(\overline{2}01)$	a	-0.4420	+17.868	11	.0049
Donnay (1967) and Donnay and Donnay (1952) re-refined by Wright and Stewart,	Triclinic Or <40	$2\theta(\overline{2}01)$	æ	3982	+16.921	12	.0057
(2021)	Monoclinic Or≥40	$2\theta(060)$	q	2996	+25.478	17	.0031
	Triclinic Or<40	$2\theta(060)$	q	2194	+22.127	12	.0034 (2-0.0027)
	Monoclinic Or≥40	$2\theta(\overline{2}04)$	c	1373	+14.159	17	.0014
	Triclinic $Or < 40$	$2\theta(204)$	Ċ	0987	+12.189	<b>00</b> :	.0010
P50-56KF (Orthoclase) Wright and Stewart,	$\begin{array}{c} \text{Monoclinic} \\ \text{Or} \geq 40 \end{array}$	$2\theta(\overline{2}01)$	a	4408	+17.838	90	.0040
(1906)	Triclinic Or < 40	$2\theta(\overline{2}01)$	a	— .4262	+17.527	ŝ	6000.
	Monoclinic Or≥40	$2\theta(060)$	q	2782	+24.585	Ø	.0019
	Triclinic Or<40	$2\theta(060)$	q	1599	+19.624	3	.0067

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redispar series         composition         x         y         m         b           Monoclinic $20(204)$ c        1296         +13.897         13.897           Monoclinic $0r \ge 40$ $0r \ge 40$ $0r \ge 40$ $117.756$ +12.710           Maximum microcline-low albite         Triclinic $2\theta(204)$ c        1088         +17.756           Maximum microcline-low albite         Triclinic $2\theta(201)$ a        4367         +17.756           Maximum microcline-low albite         Triclinic $2\theta(201)$ a        2367         +16.613           Or<40         Triclinic $2\theta(201)$ a        2367         +25.367           Or<240 $0r \le 40$ $0r < 40$ $0r < 40$ +13.644         +13.644           Triclinic $2\theta(204)$ $b$ $2556$ +23.644         +13.644           Triclinic $2\theta(204)$ $c$ $1271$ +13.644         +13.644	:	Symmetry	Varia	bles	Equation	, y = mx + b	No. of Obser-	Standard
	Feldspar series	and Composition	×	s,	ш	q	vations	of ya
Triclinic Or <40 $2\theta(304)$ $c$ $1088$ $+12.710$ Maximum microcline-low albite       Triclinic $2\theta(301)$ $a$ $4367$ $+17.756$ Maximum microcline-low albite       Triclinic $2\theta(301)$ $a$ $4367$ $+17.756$ Orville (1967)       Triclinic $2\theta(301)$ $a$ $4367$ $+17.756$ Triclinic $2\theta(301)$ $a$ $2364$ $+16.613$ Triclinic $2\theta(060)$ $b$ $2967$ $+25.367$ Triclinic $2\theta(060)$ $b$ $2556$ $+23.644$ Triclinic $2\theta(304)$ $c$ $1271$ $+13.644$ Triclinic $2\theta(304)$ $c$ $1271$ $+13.644$		Monoclinic Or≥40	$2\theta(204)$	U	1296	+13.897	8	.0017
Maximum microcline low abite         Triclinic $2\theta(201)$ $a$ $4367$ $+17.756$ Orville (1967)         Triclinic $2\theta(201)$ $a$ $4367$ $+17.756$ Triclinic $0r < 40$ $2r < 40$ $2\theta(201)$ $a$ $3841$ $+16.613$ Triclinic $2\theta(060)$ $b$ $3841$ $+16.613$ Triclinic $2\theta(060)$ $b$ $2967$ $+25.367$ Triclinic $2\theta(060)$ $b$ $2556$ $+23.644$ Triclinic $2\theta(204)$ $c$ $1271$ $+13.644$ Triclinic $2\theta(204)$ $c$ $1271$ $+13.644$		Triclinic Or < 40	$2\theta(\overline{2}04)$	v	1088	+12.710	3	.0012
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Maximum microcline-low albite Orville (1967)	Triclinic Or≥40	$2\theta(\overline{2}01)$	ø	4367	+17.756	7	.0022
$\begin{array}{llllllllllllllllllllllllllllllllllll$		Triclinic Or<40	$2\theta(\overline{2}01)$	a	3841	+16.613	4	.0025
$\begin{array}{llllllllllllllllllllllllllllllllllll$		Triclinic Or≥40	$2\theta(060)$	q	2967	+25.367	2	.0013
$\begin{array}{llllllllllllllllllllllllllllllllllll$		Triclinic Or < 40	$2\theta(060)$	9	2556	+23.644	3	.0015
Triclinic $2\theta(204)$ c $1011$ $+12.324$		Triclinic Or≥40	$2\theta(\overline{2}04)$	S	1271	+13.644	2	.0011 (30010)
0r < 40		Triclinic Or <40	$2\theta(\overline{2}04)$	C	1011	+12.324	4	.0046

TABLE 5-(continued)

X-RAY AND OPTICAL STUDY OF ALKALI FELDSPAR

Specimen	$2\theta(\overline{2}01)$	$2\theta(060)$	$2\theta(\overline{2}04)$
X	21.180	41.610	50.840
Y	21.026	41.741	50.675
Z	20.940	41.891	50.775

Part I (Table 9 and discussion following) are also used here as examples of the method. The  $2\theta$  values for  $\overline{2}01$ , 060, and  $\overline{2}04$  are as follows:

These feldspars can be described as follows:

1. Specimen X is a sanidine having a structural state between that of high sanidine and *Puye* equivalents. Composition estimated from  $\overline{2}01$  is  $Or_{76}$ .

2. Specimen Y is slightly anomalous and has an approximate structural state intermediate to equivalents of SH-1070 and Spencer B.

3. Specimen Z is highly anomalous;  $\overline{2}01$  (derived from Fig. 3)— $\overline{2}01$  (observed)=0.6° 2 $\theta$ . Apparent structural state is close to that of Spencer B. Composition cannot be estimated.

Type of Feldspar	Variab	les	Equation,	y = mx + b		o. of	Standard
- ) F - of I diapar	æ	у	m	b	ti	ions	of y <sup>a</sup>
Monoclinic	$2\theta(\overline{2}01)$	a	4389	+17.7	799	40	.0039
	$2\theta(060)$	Ь	2908	+25.11	1	40	.0022
	$2\theta(\overline{2}04)$	С	1278	+13.67	8	39	.0011
Maximum microcline	<b>2</b> θ(201)	a	2790	+14.44	40	10	.0048 (20046)
	2 <del>0</del> (060)	Ъ	2687	+24.19	5	7	.0014
	$2\theta(\overline{2}04)$	С	1196	+13.26	52	10	.0011
	2 $\theta(\overline{2}01)$	a	2 <del>0</del> (060)	ь	$2\theta(\overline{2}04)$	С	$2\theta(131)-2\theta(1\overline{31})$
Intermediate microcline	21.053	8.5644	41.745	12.9705	50.615	7.2077	0.293
	21.045	8.5638	41.718	12.9793	50.688	7,1993	0.215
	21.025	8.5770	41.788	12.9628	50.778	7.1897	0.105
	20.998	8.5784	41.792	12.9569	50.583	7.2128	0.240

Table 6. Equations Relating Cell Dimensions to Observed  $2\theta$  Values of Single Diffraction Peaks for Potassium-Rich Alkali Feldspars

\* See footnote, Table 5.

These descriptions do not differ significantly from those given in Part I (p. 71) using the axial cell dimensions a, b, and c.

The 'three-peak' method outlined is sufficient, so far as is known, to enable the identification and description of alkali feldspars encountered routinely in petrologic studies. In studies aimed specifically at identifying variations in the properties of a related group of alkali feldspars it is desirable to compute the unit-cell dimensions more precisely from fully measured X-ray diffraction patterns. For this purpose, the equations given in Tables 5–7 may be used to yield starting parameters for a cell refinement.

The standard errors associated with a, b, and c for the equations relating reflection-position to axial dimension are very close to the average

Variables		Equation, $y = mx + b$		No. of	Standard
x	у	m	ь	tions	of y
$2\theta(\overline{2}01)$	a	4046	+17.058	21	.0008
$2\theta(060)$	b	2311	+22.622	21	.0010
$2\theta(\overline{2}04)$	С	1060	+12.567	21	.0005

Table 7. Equations Relating Cell Dimensions to Calculated  $2\theta$  Values of Single Diffraction Peaks for Natural Anorthoclases

standard error associated with the respective dimensions computed in a complete refinement of the unit cell; *i.e.* 1 part in 2500 for a, 1 part in 6000 for b, and 1 part in 6000 for c. This is not to say that the 'three-peak' method is a substitute for a good computer refinement, but does indicate that in most cases the determination of the composition and structural state of a feldspar and identification of whether or not a feld-spar is anomalous using the 'three-peak' method will agree with that obtained from the methods given in Part I using the cell parameters. In natural suites of alkali feldspars studied by the author (Wright and Stewart, unpublished data) there are no significant differences in interpretation using the two methods.

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I am greatly indebted to my colleague D. B. Stewart for help in refining the unit-cell dimensions of the alkali feldspars and for help in computing the equations relating reflection —position to unit-cell dimension. Stewart also critically read and materially improved an early version of this manuscript. James Papike, Joan Clark, and R. I. Tilling of the Geological Survey suggested a number of changes which have been incorporated into the final manuscript.

#### THOMAS L. WRIGHT

## Appendix: Equations Relating Cell Dimension and Observed $2\theta$ Values for Single Diffraction Peaks

Alkali feldspar series of constant structural state. Equations relating  $2\theta$ values ( $CuK\alpha_1$  measured according to the procedure described in Part I, Appendix) to axial dimension are given in Table 5 for three complete series of alkali feldspars described previously (Wright and Stewart, 1968; Orville, 1967). For each series the  $2\theta$  value and computed axial dimension are related in two linear segments intersecting at a composition of Or40Ab60. In the series structurally equivalent to high sanidine and to P50-56F orthoclase this intersection is coincident with the composition at which the symmetry changes from monoclinic to triclinic. An intersection at the same composition was found for the maximum microcline-low albite series although all members of this series are triclinic.<sup>1</sup> Thus in each series the data for compositions more potassic than Or<sub>40</sub> were treated separately from data for compositions more sodic than Or<sub>40</sub>. For each set of data consisting of more than 5 points, 3rd, 2nd, and 1st degree polynomials were computed using a least-squares program. If the data set contained 5 or less points only a linear fit was computed; in larger data sets the differences in standard error for higher order equations were so small that a linear relationship was assumed. These equations can be used to obtain starting parameters for refinement of the cell dimensions of any homogeneous alkali feldspar of essentially binary composition (Or+Ab>95%), but, for reasons given below, are not applicable either to alkali feldspar phases in perthites or to alkali feldspars of ternary composition.

Natural potassium-rich alkali feldspars. Equations relating observed  $2\theta$  value to axial dimension were computed from data for 40 monoclinic potassium-rich alkali feldspars, and 10 maximum microclines. These equations are listed in Table 6 along with  $2\theta$ 's and axial dimensions for 4 intermediate microclines. The data used to derive those equations are mostly obtained from refinement of the cell dimensions of perthitic orthoclases. Plots of  $2\theta$  values against cell dimension for perthitic orthoclases were made and it was found that additional potassium-rich alkali feldspars, including one cryptoperthitic high sanidine, three feldspars with anomalous cell dimensions, possibly low sanidine, and four homogeneous potassic feldspars (Or<sub>90</sub>), fit the lines constructed for perthitic orthoclases. Taken together, the natural monoclinic potassium-rich

 $^{1}$  A change of slope was previously suggested by examination of the variation of cell dimensions with *Or* content across the maximum microcline-low albite series. (See Part I).

feldspars show a very limited range for a (8.48–8.62 Å) but have wide variations in b and c which can be related to the structural state and whether or not the cell dimensions are anomalous. The natural monoclinic feldspars are thus distinguished from the feldspars of the synthetic high-sanidine series and alkali-exchanged orthoclase series for which a, b, and c all vary together with changing composition. The equations of Table 6 must be used to derive starting cell parameters for perthitic potassium-rich feldspars. The starting parameters for homogeneous monoclinic potassium-rich feldspars may be derived either from the equations of Table 6 or, if the structural state is known to be orthoclase or high sanidine, the appropriate equations of Table 1.

Although separate equations have been computed in Table 6 for monoclinic potassium-rich feldspars and for maximum microclines, only the equation relating  $2\theta$  (060) to b differs significantly between the two sets of data. Starting parameters for intermediate microclines must be obtained by interpolation of the values given from the equations for monoclinic potassium-rich feldspars and for maximum microclines.

Natural sodium-rich and ternary feldspars. The unit-cell dimensions of anorthoclases described by Carmichael and Mackenzie (1964) have been used to calculate  $2\theta$  values for  $\overline{2}01,060$ , and  $\overline{2}04$ . The unit-cell dimensions of six anorthoclases refined by D. B. Stewart (unpub. data) have been included with these and equations relating the calculated<sup>1</sup> reflection positions and cell dimensions of the anorthoclases are given in Table7. On the assumption that all natural anorthoclases are comparable to those studied by Carmichael and Mackenzie (1964) and by Stewart the starting parameters for refinement of natural anorthoclases are best derived from the equations of Table 7.

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<sup>1</sup> Because observed values of  $2\theta$  were not published by Carmichael and Mackenzie (1964), calculated values of  $2\theta$  were used. As noted previously, this introduces a small bias when observed  $2\theta$ 's of natural anorthoclases are used to calculate dimensions from the equations of Table 7.

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