

## SABINAITE: A NEW OCCURRENCE AND NEW DATA†

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### ABSTRACT

At Mont St. Hilaire, Quebec, sabinaité occurs with albite, microcline, dawsonite, analcime, sodalite, pectolite, dolomite and aegirine in the cavities in the sodalite syenite. Electron-microprobe analyses gave  $ZrO_2$  40.64,  $HfO_2$  0.45,  $TiO_2$  10.91,  $Na_2O$  19.53,  $CaO$  0.02,  $CO_2$  27.56 (calc.), sum 99.11 wt.%, corresponding to  $Na_{4.02}(Zr_{1.99}Hf_{0.01})(Ti_{0.87}Zr_{0.12})O_4(CO_3)_4$ . The new ideal formula is  $Na_4Zr_2TiO_4(CO_3)_4$ .  $Z = 4$ .  $D_{calc} = 3.48$  g/cm<sup>3</sup>. Precession photographs show the mineral to be monoclinic,  $C2/c$  or  $Cc$ , with  $a$  10.171(5),  $b$  6.623(4),  $c$  17.976(9) Å,  $\beta$  94.32(4)°. Crystals are tabular on (001), with {001}, {010}, {100} and {110} forms. The {001} cleavage is perfect and {010} distinct. Optically, the mineral is colorless, transparent, with vitreous luster; biaxial positive,  $\alpha$  1.720(2),  $\beta$  1.79(1),  $\gamma$  1.90(calc.),  $2V = 82(1)^\circ$ , all in Na light. Orientation:  $Y = b$ ,  $X \wedge c = 13^\circ$  in the obtuse  $\beta$ . Dispersion moderate, with  $r > v$ .

**Keywords:** sabinaité, Mont St. Hilaire, Quebec, new data.

### SOMMAIRE

Au mont St-Hilaire (Quebec), la sabinaité se trouve associée à albite, microcline, dawsonite, analcime, sodalite, pectolite, dolomite et aegyrine dans les cavités de la syénite à sodalite. Les analyses à la microsonde électronique donnent:  $ZrO_2$  40.64,  $HfO_2$  0.45,  $TiO_2$  10.91,  $Na_2O$  19.53,  $CaO$  0.02,  $CO_2$  27.56 (calculé), total 99.11% (en poids), ce qui correspond à la formule  $Na_{4.02}(Zr_{1.99}Hf_{0.01})(Ti_{0.87}Zr_{0.12})O_4(CO_3)_4$ . La formule idéale révisée serait donc  $Na_4Zr_2TiO_4(CO_3)_4$ .  $Z = 4$ .  $D_{calc} = 3.48$ . Les clichés de précession révèlent un minéral monoclinique,  $C2/c$  ou  $Cc$ ,  $a$  10.171(5),  $b$  6.623(4),  $c$  17.976(9) Å,  $\beta$  94.32(4)°. Les cristaux sont tabulaires, aplatis sur (001), montrant les formes {001}, {010}, {100} et {110}. Le clivage est parfait suivant {001}, distinct suivant {010}. Optiquement, la sabinaité est incolore, transparente et d'éclat vitreux; biaxe positive,  $\alpha$  1.720(2),  $\beta$  1.79(1),  $\gamma$  1.90(calc.),  $2V = 82(1)^\circ$ , (mesures à la lumière de Na). Orientation:  $Y = b$ ,  $X \wedge c = 13^\circ$  dans l'angle obtus de  $\beta$ . Dispersion moyenne,  $r > v$ .

(Traduit par la Rédaction)

**Mots-clés:** sabinaité, mont St-Hilaire (Québec), données nouvelles.

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### INTRODUCTION

Sabinaité, an anhydrous carbonate of sodium, zirconium and titanium, was first found at the Francon quarry in Montreal and described by Jambor *et al.* (1980). The mineral has since been found by Elsa and László Horváth in small amounts with albite, microcline, dawsonite, analcime, sodalite, pectolite, dolomite and aegirine in cavities in the sodalite syenite at Mont St. Hilaire, Quebec. Electron-microprobe analyses, X-ray and optical studies of these well-formed crystals of sabinaité yielded new data that characterize the mineral more fully.

### X-RAY CRYSTALLOGRAPHY

Precession photographs of sabinaité from Mont St. Hilaire show the mineral to be monoclinic, space group  $C2/c$  or  $Cc$ . The cell parameters measured from these photographs and then refined by a least-squares method using powder-diffraction data obtained with a Gandolfi camera (Table 1) are  $a$  10.171(5),  $b$  6.623(4),  $c$  17.976(9) Å,  $\beta$  94.32(4)°. The powder pattern was indexed on the basis of the precession photographs, using only the strong single-crystal reflections.

Jambor *et al.* (1980) derived a monoclinic cell for sabinaité; they found  $a$  6.605(3),  $b$  10.186(5),  $c$  37.94(5) Å,  $\beta$  90°, where  $a$  and  $b$  were based on electron-diffraction data and  $c$  and  $\beta$  were calculated from X-ray powder-diffraction data. Their  $a$  and  $b$  apparently correspond respectively to  $b$  and  $a$  of the present cell but a simple rational relationship between their  $c$  and the present  $c$  cannot be established. Their  $c$  is dimensionally close to  $2c-a$  (38.09 Å) in the present cell but this leads to a  $\beta$  angle of 109.74°.

### PHYSICAL AND OPTICAL PROPERTIES

Sabinaité from Mont St. Hilaire is colorless, transparent and vitreous. Solid inclusions of irregular shape and well-formed rhombs with strong negative relief are common. The crystals, averaging  $0.4 \times 0.3 \times 0.05$  mm, are tabular, flattened on  $c$  and slightly elongate along  $a$ . Forms present are {001}, {010}, {100} and {110} (Fig. 1). The crystal morphology indicates the presence of a centre of symmetry; the most probable space-group is, therefore,  $C2/c$ . The presence of the perfect {001} and distinct {010}

TABLE 1. X-RAY POWDER-DIFFRACTION DATA FOR SABINAITE

hkl	Mont St. Hilaire <sup>1</sup>			Francon <sup>2</sup>		
	$d_{\text{cal}} (\text{\AA})$	$d_{\text{obs}} (\text{\AA})$	$I$	$d_{\text{cal}} (\text{\AA})$	$d_{\text{obs}} (\text{\AA})$	$I$
002	8.962	8.96	100	8.945	8.97	10
111	5.238	5.23	5			
200	5.071	5.08	10	5.076	5.10	1
202	4.563	4.57	5	4.567	4.57	< 1/2
004	4.481	4.48	25	4.472	4.48	2
202	4.278	4.28	20	4.276	4.30	1
204	3.491	3.488	5			
020	3.312	3.305	10	3.306	3.307	< 1/2
021	3.258	3.251	50	3.251	3.277	2
022	3.107	3.107	10		3.252	
310	3.011			3.012	3.009	1
006	2.989	2.990	50	2.982	2.991	6
311	2.938	2.945	20	2.938	2.942	< 1/2
023	2.897	2.901	3			
312	2.798	2.805	5			
313	2.765			2.766	2.768	< 1/2
221	2.757	2.763	10		2.742	< 1/2
221	2.723	2.727	25	2.720	2.724	1
205	2.663	2.662	10	2.661	2.665	< 1/2
313	2.619			2.618		
222	2.619	2.624	25	2.616	2.628	< 1/2
402	2.489	2.492	5	2.492	2.494	1
402	2.393	2.390	10			
008	2.241			2.236		
225	2.238	2.239	40	2.236	2.243	1
130	2.157	2.155	10	2.154	2.150	1
208	2.109	2.106	20	2.107	2.126	< 1/2
317	2.019			2.018	2.109	1
421	2.014	2.017	45	2.014	2.017	5
420	2.013			2.013		
208	1.995					
226	1.992			1.852		
422	1.990	1.992	10	1.847		
421	1.986			1.845	1.847	6
511	1.943	1.947	10	1.841		
119	1.900			1.829	1.824	2
317	1.889	1.893	15	1.824		
028	1.856			1.794		
330	1.849			1.789	1.795	3
331	1.847	1.850	25	1.733		
227	1.844			1.735	1.730	< 1/2
331	1.831			1.733		
332	1.825	1.827	15			
513	1.807	1.811	2			
332	1.796			1.692	1.693	< 1/2
00.10	1.793	1.795	35	1.682		
136	1.762	1.761	2	1.684	1.687	3
514	1.735	1.735	10	1.648		
334	1.734			1.648	1.646	5
319	1.715			1.646		
425	1.712			1.624	1.624	< 1/2
228	1.709	1.711	15			
029	1.707			1.603		
600	1.690			1.600	1.601	< 1/2
334	1.685	1.687	15	1.569	1.569	< 1/2
516	1.683					
335	1.671	1.668	2			
2.0.10	1.651			1.648		
229	1.650	1.648	10	1.648	1.646	5
041	1.649			1.646		
604	1.622			1.624	1.624	< 1/2
408	1.620					
319	1.612	1.615	5			
517	1.603			1.603		
335	1.602	1.601	10	1.600	1.601	< 1/2
241	1.571	1.571	5	1.569	1.569	< 1/2
336	1.546					
604	1.546	1.546	15			
243	1.513	1.512	2			
00.12	1.494					
621	1.492	1.495	10			
622	1.469	1.463	2			
624	1.457					
338	1.456	1.457	10			

<sup>1</sup> 114.6 mm Gandolfi camera,  $\text{CuK}\alpha$  radiation,  $\lambda = 1.5418 \text{\AA}$ , Si internal standard, visual intensities.

<sup>2</sup> Data from Jambor *et al.* (1980) reindexed on  $a = 10.181(6)$ ,  $b = 6.612(3)$ ,  $c = 17.94(1)$ ,  $\beta = 94.39(5)^\circ$ .

cleavages confirms the observations of Jambor *et al.* (1980) on sabinaitite from the Francon quarry.

The optical properties of sabinaitite from Mont St. Hilaire were determined in sodium light using a spindle stage and crystals previously oriented by the X-ray-precession method. The mineral is biaxial positive, in contrast to sabinaitite from Francon, which was reported to be biaxial negative (Jambor *et al.* 1980). The indices of refraction are  $\alpha$  1.720(2),  $\beta$  1.79(1). The  $\gamma$  index could not be determined as in the high-index oil a film of minute particles rapidly formed on the surface of the crystal. The calculated value for  $\gamma$  is 1.90. The  $2V$  angle is  $82(1)^\circ$ , determined directly by measuring the angle between the optic axes. The orientation is  $Y = b$ ,  $X\Delta c = 13^\circ$  in the obtuse angle of  $\beta$ . The mineral shows moderate inclined dispersion with  $r > v$ .

#### CHEMICAL FORMULA

Sabinaitite from Mont St. Hilaire was analyzed by means of an electron microprobe, using a defocused beam. An initial energy-dispersion scan confirmed the presence of Na, Zr, Ti and Hf. Subsequent quantitative analyses were carried out using the following standards: synthetic  $\text{ZrO}_2$  (Zr), synthetic  $(\text{Zr,Hf})\text{O}_2$  (Hf), albite (Na) and hornblende (Ti,Ca). F was sought but not detected.  $\text{CO}_2$  was not determined because of the small amount of sample available. The data given in Table 2 represent the average results of five analyses on different spots of the same crystal. The averaged analytical data compare well with those for sabinaitite from Francon (Jambor *et al.* 1980).

In the light of the new cell-dimension data for sabinaitite, the chemical formula proposed by Jambor *et al.* (1980) presents two serious problems. Firstly, the calculated density ( $3.61 \text{ g/cm}^3$ ) based on the new cell is much too high in comparison with the measured density of  $3.36 \text{ g/cm}^3$ . Secondly, the cell content of 18 Na, 9 Zr and 18 non-carbonate oxygen atoms is incompatible with the space group  $C2/c$  or  $Cc$ . Hence a new formula must be sought. Calculations of the cell content based on the mass-volume-density relationship shows that the Na:Zr:Ti:C proportions are close to 4:2:1:4 for 16 oxygen atoms. The formula for sabinaitite from Mont St. Hilaire was therefore calculated on the basis of 8 oxygen atoms per formula, assuming a stoichiometric amount of  $\text{CO}_2$ . The calculated formula is  $\text{Na}_{4.02}(\text{Zr}_{1.99}\text{Hf}_{0.01})(\text{Ti}_{0.87}\text{Zr}_{0.12})\text{O}_4(\text{CO}_3)_4$ , with  $Z = 4$ . This formula requires 27.56 wt.%  $\text{CO}_2$ , which compares well with the analytical value of 27.1% for sabinaitite from Francon. The density calculated from this empirical formula is  $3.48 \text{ g/cm}^3$ . Similarly, the analysis of sabinaitite from Francon corresponds to  $(\text{Na}_{4.18}\text{Ca}_{0.02})(\text{Zr}_{1.99}\text{Hf}_{0.01})\text{Ti}_{0.94}\text{O}_{4.14}(\text{CO}_3)_{3.86}$ , with  $Z = 4$ . The density calculated for sabinaitite from Francon, based on the new cell-parameters (Table 1), is  $3.44 \text{ g/cm}^3$ , in good agreement with the meas-

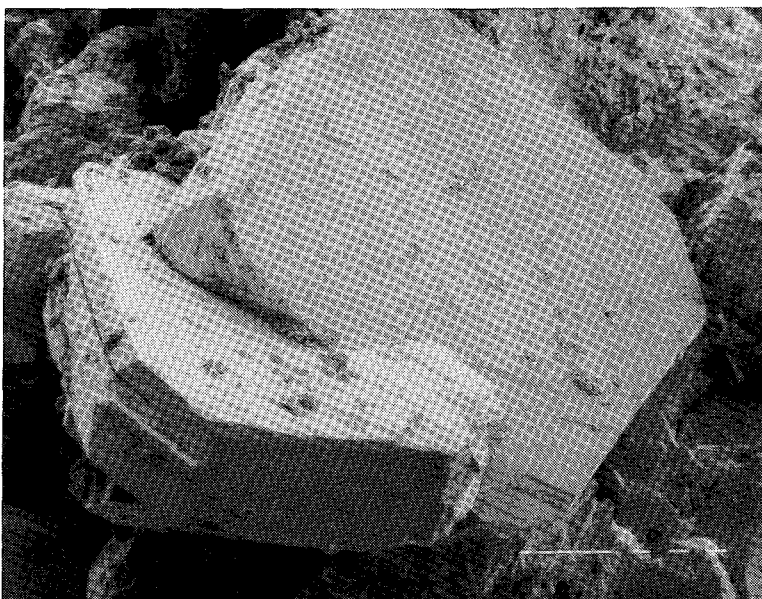


FIG. 1. Sabinaites crystals from Mont St. Hilaire, Quebec. The scale bar is 0.1 mm. SEM photomicrograph.

ured density of  $3.36 \text{ g/cm}^3$  reported by Jambor *et al.* (1980). The ideal formula for sabinaites is, therefore,  $\text{Na}_4\text{Zr}_2\text{TiO}_4(\text{CO}_3)_4$ .

The chemical molar refractivity  $K_C$  (Mandarino 1979), based on the new empirical formula and the constants given by Mandarino (1981), are 0.226 and 0.228, respectively, for sabinaites from Mont St. Hilaire and Francon. The  $K_p/K_C$  value (Mandarino 1979) for sabinaites from Mont St. Hilaire, based on the calculated density, is  $0.230/0.226 = 1.018$  whereas the  $K_p/K_C$  value for sabinaites from Francon, based on the observed ( $3.36 \text{ g/cm}^3$ ) and calculated ( $3.44 \text{ g/cm}^3$ ) densities, is  $0.237/0.228 = 1.039$  and  $0.232/0.228 = 1.017$ , respectively. The compatibility of the chemical and physical data for sabinaites is, therefore, excellent to superior (Mandarino 1979).

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TABLE 2. COMPOSITION OF SABINAITE

	(1)	(2)	(3)
ZrO <sub>2</sub> (Wt.%)	40.64	39.1	39.34
HfO <sub>2</sub>	0.45	0.47	
TiO <sub>2</sub>	10.91	12.05	12.76
Na <sub>2</sub> O	19.53	20.7	19.79
CaO	0.02	0.2	
CO <sub>2</sub>	(27.56)	27.1	28.10
	99.11	99.57	99.99

- (1) Mont St. Hilaire, Quebec, CO<sub>2</sub> calculated.  
 (2) Francon quarry, Montreal, Quebec, Jambor *et al.* (1980).  
 (3) Ideal  $\text{Na}_4\text{Zr}_2\text{TiO}_4(\text{CO}_3)_4$ .

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