BEUSITE-TRIPHYLITE INTERGROWTHS FROM THE YELLOWKNIFE PEGMATITE FIELD, NORTHWEST TERRITORIES

MICHAEL A. WISE* AND PETR ČERNÝ Department of Geological Sciences, University of Manitoba, Winnipeg, Manitoba R3T 2N2

ABSTRACT

Calcium-rich beusite occurs in a pegmatite from the regionally zoned Peg swarm of the Archean Yellowknife pegmatite field, Northwest Territories. Beusite occurs as pale brown tabular lamellae, up to 5×2.5 cm $\times 1$ mm in size, interlaminated with greenish grey triphylite lamellae. Beusite is monoclinic, space group $P2_1/c$, a 8.818(1), b 11.750(1), c 6.175(1) Å, ß 99.35(1)° and V 631.3(1) Å³. Optical and single-crystal X-ray-diffraction studies confirm the presence of three crystallographic orientation relationships between beusite and triphylite, the most common being [010] of beusite parallel to [110] of triphylite. Beusite is chemically homogeneous, the Mn/(Mn + Fe) ratio varying slightly from 0.54 to 0.57. Calcium contents are higher than in other reported specimens, the maximum CaO content of the Yellowknife beusite reaching 16.1 wt.%. The elevated Ca content results in a larger b cell dimension. Beusite-triphylite intergrowths represent exsolution from a high-temperature, Li,Ca-rich graftonite-like parent phase.

Keywords: beusite, triphylite, exsolution, Yellowknife, Northwest Territories.

SOMMAIRE

Nous avons trouvé de la béusite riche en calcium dans une des pegmatites de l'essaim de Peg, d'âge archéen, dans la région de Yellowknife (Territoires du Nord-Ouest). Elle forme des interlaminations tabulaires brun pâle atteignant 5×2.5 cm $\times 1$ mm avec la triphylite gris verdâtre. La béusite est monoclinique, groupe spatial $P2_1/c$, a 8.818(1), b 11.750(1), c 6.175(1) Å, β 99.35(1)°, et \tilde{V} 631.3(1) Å³. Les données optiques et de diffraction X obtenues sur cristal unique confirment la présence de trois relations d'orientation entre béusite et triphylite, la plus répandue étant [010] de la béusite parallèle à [110] de la triphylite. Les cristaux sont homogènes; le rapport Mn/(Mn + Fe) ne varie que peu, entre 0.54 et 0.57. Les teneurs en calcium sont plus élevées (jusqu'à 16.1% de CaO en poids) que dans les exemples signalés auparavant, ce qui a pour effet un agrandissement de la dimension b. L'intercroissance de béusite et de triphylite résulterait d'une exsolution à partir d'un précurseur semblable à la graftonite, riche en Li et Ca, et stable à température élevée.

(Traduit par la Rédaction)

Mots-clés: béusite, triphylite, exsolution, Yellowknife, Territoires du Nord-Ouest.

INTRODUCTION

Beusite, $(Mn,Fe,Ca)_3(PO_4)_2$, is the rare manganese end-member of the graftonite-beusite series, which occurs exclusively as a primary accessory phase in granitic pegmatites. Beus (1950) was the first to report the existence of a graftonite-like mineral in which Mn exceeds Fe. A decade later, Brooks & Shipway (1960) also presented chemical data on a Mn-dominant graftonite-like phase. Beusite was given species status following the discovery of a Mndominant graftonite from the San Luis Province, Argentina (Hurlbut & Aristarain 1968). Since then, beusite has been found from southern Finland (Lahti 1981), the Cross Lake pegmatite field, Manitoba (T.S. Ercit, pers. comm. 1984), the Tsaobismund pegmatite, Namibia (Fransolet et al. 1986), and recently in the Yellowknife pegmatite field.

OCCURRENCE

Beusite occurs in a small pegmatite dyke of the regionally zoned Peg swarm of the Archean Yellowknife pegmatite field, located 75 km northeast of Yellowknife, between Upper Ross Lake and Redout Lake (Fig. 1). The pegmatite lies 3.5 km to the east of the Redout Granite, is lenticular, strikes N-S, dips 45-70° E and cuts an interlayered sequence of amphibolite and granodiorite.

The pegmatite is of the beryl-columbitephosphate subtype of rare-element pegmatites, as classified by Cerný (1989), and exhibits welldeveloped, although not continuous, internal zonation. Much of the primary zonation is obscured by albitic units. The border zone consists of a finegrained muscovite ± quartz assemblage; it is followed by a fine-grained microcline perthite + quartz + muscovite + cleavelandite wall zone. Most of the pegmatite is composed of a coarse-grained microcline perthite (graphic) + quartz + muscovite + cleavelandite + beryl zone that hosts most of the accessory minerals. These minerals include, in addition to the beusite-triphylite intergrowths, biotite, vellow-green beryl, ferrocolumbite-ferrotantalite, ferrotapiolite, almandine and pyrite. The core is discontinuous and consists of coarse-grained quartz + microcline + accessory beryl. Replacement muscovite units are present locally.

The beusite-triphylite intergrowths occur as a 6

^{*}Present address: Department of Mineral Sciences, Smithsonian Institution, Washington, D.C. 20560, U.S.A.



FIG. 1. Location map of the regionally zoned Peg swarm (modified after Meintzer 1987 from Hutchinson 1955), showing westward regional zoning (1-5) from the Redout Granite. 1 Pegmatites hosted by interlayered granodiorite and amphibolite, 2 Redout granite, 3 Burwash Formation and 4 Cameron River mafic volcanic suite. Star denotes location of beusite-bearing pegmatite.

 \times 5 \times 3 cm nodule in blocky, pink microcline that appears to have undergone a moderate degree of metasomatic alteration. The microcline has been



FIG. 2. Sketch of beusite-triphylite paragenesis. Mc microcline, Ms fine-grained muscovite, Bt-b biotiteberthierine intergrowths, Be-tp beusite-triphylite intergrowths, Cal calcite, All alluaudite-group mineral.

partly altered to a fine-grained, greenish muscovite that grades into a dark green band, 0.3-0.5 cm wide, of microscopically intermixed biotite and a berthierine-like mineral (R.E. Meintzer, pers. comm.; Fig. 2). This aggregate is in sharp contact with the beusite-triphylite intergrowths and hosts tiny veinlets of calcite. Incipient Na-metasomatism is evident by the presence of a small patch of alluaudite-group minerals near the beusite-triphylite intergrowth. Minor ferrisicklerite is present as a weathering product of triphylite.

PHYSICAL AND OPTICAL PROPERTIES

Beusite is pale brown, displays well-developed cleavage on {010} and {100}, vitreous luster and a pale brown streak. The density, measured on a Berman balance, is 3.600(5) g/cm³. The indices of refraction for beusite were measured by the Becke line method on crushed fragments in white light. The 2V angle was determined from extinction curves refined from spindle-stage data using the program EXCALIBR (Bloss 1981). The optical properties and density of the Yellowknife beusite are compared with those of beusite from three other localities in Table 1. Using the Gladstone-Dale constants of Mandarino (1981), a compatibility index of 0.004 was obtained, indicating superior agreement between optical and chemical data and density.

BEUSITE-TRIPHYLITE INTERGROWTHS

TABLE 1. OPTICAL PROPERTIES FOR BEUSITE FROM YELLOWKNIFE AND OTHER LOCALITIES

	1*	2	3	4
α	1.685	1.695	1.708	1.702
β	1.688	1.696	1.711	1.703
¥	1.700	1.715	1.723	1.722
2V(°) _{obs}	45	25	25	25
2V(°) _{calc}	53	26	53	26
Dispersion	x>x	<u>x>v</u>	<u>r>v</u>	<u>r>v</u>
D(g/cm ³)	3.600	*****	3.698	3.202

1 Yellowknife, this study: 2 Amanda, 3 San Salvador, 4 Los Aleros. Samples 2-4 from San Lais Province, Argenina (Hurlbut 4 Aristarain 1968). Indices of refraction measured in white light. Samples 2-4 measured in Na light.



FIG. 3. Intergrowth of beusite (dark) and triphylite (light) viewed down their c axes.

TABLE	2.	COMPARISON	OF	UNIT-CELL	DATA	FOR	BEUSITE
					+		

Sample No.	a (Å)	b(Å)	c (Å)	β(*)	V(Å ³)
1	8.818(1)	11.750(1)	6.175(1)	99.35(1)	631.3(1)
2	8.797(3)	11.758(4)	6.170(2)	99.31(2)	629.8(3)
3	8.78(1)	11.52(1)	6.15(1)	99.42(35)	613.7(1)
4	8.788(3)	11.434(2)	6.255(1)	98.98(2)	620.8(4)

l Powder data (this study), 2 single-crystal data, Yellowknife (Wise et al. 1989), 3 Los Alercs, San Lais, Argentina (Hurlbut & Aristarain 1968), 4 synthetic beusite, (Nord & Annersten 1967).

Yello	Beu wknife,	site ¹ N.W.T.,	Canada	Beusi Los Aleros, San	te ^z Luis,	Argentina
hkl	đ	đ	r	d	đ	I
	(calc)	(obs)		(calc)	(obs)	
111				4.293	4.29	20
121				3.607		
220				3.510	3.49	100
T 31				3,130	3.13	40
T02	3.033	3.031	100	3.021	3.01	35
131	2.995	3.000	44	2,950		
012				2.934	2.94	25
300	2 939	2 042	74	2.887	2 90	20
230	2,911	2,908	69	2.873	2.863	100
221	2.875	2.876	70	2.850	2.846	20
231	2.745	2.747	12		-	-
311				2.712		
212	2.642	2.642	9	2.701	2.708	60
321			-	2.511		
231				2.494	2.502	5
311	2.415	2.412	34	2.402		
132				2.398	2.402	30
330	2.331	2.330	11	2.375	2.3/2	-
241				2.303	2.301	5
302	2.295	2.297	18	2.286		
212	2.280	2.282	45	2.268	2.273	5
132				2.227		
222				2.147	4.444	5
411	2.127	2.130	39		_	-
410				2.128	2.137	5
151	2.097	2.099	14	2.063	2.064	5
340	2.064	2.064	18		-	-
250				2.039	2.038	10
251	2.006	2.006	13			
242				1.970		
332	1.980	1.979	23	1.965	1.967	5
T23	1.938	1.937	24	1 932	1 026	40
060				1.920	1.720	40
023	1.920	1.919	9	<u>-</u>	-	-
412				1.891	1.891	5
113				1.879		•7
431	1.893	1.894	14	1.870	1.8/4	5
T52	1.004	1.004	25	1.832	1.832	10
350				1.801		
242				1.803	1.802	5
133	1 796	1.785	14	1.755		
252	1.776	1.775	14	1.753	1.754	5
431	1.746	1.747	31	1.733		
500				1.732	1.732	5
440	1 717	1 716	16	1.731	-	_
510	1.717	1.110	10	1,713	1.712	5
351				1.680	1.680	5
043	1.671	1.671	43	1.655		
223				1.654	1,654	5
402	1.048	1.048	11	1,640	~	
252				1.632	1.635	20
333				1.630		
170				1.617	1.616	10
360				1.596	1.598	5
412	1.632	1.632	24	11050		•
T71	1.604	1.603	13			
511	1.591	1.591	11			
361	1.534	1.534	22			
352	1.500	1.500	36	Plus 12	addit	ional
153	1.482	1.483	7	lines.		
304	1.449	1.449	19			
181	1.418	1.418	12			
461	1.382	1.382	18			
281	1.372	1.371	31			
533	1.354	1.354	12			
182	1.322	1.322	23			
T15	1.228	1.228	14			

TABLE 3. POWDER-DIFFRACTION DATA FOR BEUSITE

¹This study. ²Hurlbut & Aristarain (1968); FeKα radiation Mn filter.

Optical studies of the Yellowknife beusite have revealed crystallographic characteristics similar to those observed by Hurlbut & Aristarain (1968). A thin-section examination shows uniform and continuous extinction for all beusite lamellae (0.1-1.5 mm thick), thus indicating a single common crystallographic orientation of this phase within its tabu-



TABLE 4. CHEMICAL COMPOSITION FOR BEUSITE AND TRIPHYLITE

	1*	2	3	4	5	6	7	8	9	10
P205	41.40	41.6	42.1	41.8	41.0	46.2	46.1	45.7	45.6	45.6
MgO	2.64	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
CaO	4.78	16.1	15.9	14.3	15.5	0.2	0.0	0.1	0.2	0.0
MnO	36.56	23.1	23.1	25.1	23.2	18.1	19.0	18.3	17.7	18.4
Fe0	14.62	19.8	19.8	19.1	19.0	27.6	26.8	26.8	27.1	26.5
Total	100.00	100.6	100.9	100.3	98.7	92.1	91.9	90.9	90.0	90.5
Beusi based	te form on 2(P)	la bas per i	sed on formula	8 oxyç a unit.	ren ato	oms and	triph	ylite	formul	a
P2+	2.005	1.99	2.00	2.01	2.00	2.00	2.00	2.00	2.00	2.00
Mg ²⁺	0.225	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ca ²⁺	0.293	0.98	0.96	0.87	0.96	0.01	0.00	0.01	0,01	0.00
Mn ²⁺	1.771	1.10	1.10	1.21	1.13	0.79	0.82	0.80	0.78	0.81
Fe ²⁺	0.699	0.94	0.93	0.90	0.91	1.18	1.15	1.16	1.17	1.15
Total	4.993	5.01	4.99	4.99	5.00	3.98	3.97	3.97	3.99	3.96

Beusite: 1 Los Aleros, Argentina (Burlbut & Aristarain 1968), (2-5) Yellowknife, this study. Triphylite: (6-10) Yellowknife, this study. "Wet-chemical analysis by Dr. Jun Ito (Burlbut & Aristarain 1968). Composition recalculated to 100% disregarding 0.14 Li₂O, 1.5 SiO₂, 0.87 H₂O. All other analyses were made by the authors on the electron microprobe.

lar morphological units (Fig. 3). In contrast, the lamellae of grey to greenish grey triphylite are present in several optical orientations, two of which could be identified. Most of the triphylite lamellae (0.5-2 mm thick) are elongate along [110], which is parallel to [010] of beusite. Morphologically, the platy lamellae of triphylite are parallel to the {100} cleavage of

beusite. The contact between the two minerals is strongly undulatory. The second orientation is much less common: < 0.1- to 0.5-mm-wide lamellae of triphylite, elongate along [001], which also parallels the [010] of beusite, have straight planar contacts with beusite.

The banded intergrowth grades into a mediumgrained, granular intergrowth of the same phases toward the end of the lamellae. The optical orientations within the granular intergrowth are the same as described for the lamellar intergrowths (*i.e.*, a single, uniform orientation for beusite and several orientations for triphylite). Unlike the triphylite found in the layered intergrowth, much of the granular triphylite has undergone extensive alteration to secondary phosphates.

X-RAY CRYSTALLOGRAPHY

Beusite is monoclinic with space group $P2_1/c$. Table 2 compares the unit-cell dimensions of the Yellowknife beusite with those of other samples. Unitcell dimensions were determined for both a single crystal and powder. X-ray powder-diffraction data were obtained on a Philips PW 1710 automated powder diffractometer. Operating conditions were as follows: graphite-monochromatized $CuK\alpha_1$ radiation (λ 1.54060 Å); operating voltage, 40 kV; operating current, 40 mA; scanning speed 1/2° 20/min. Powderdiffraction data were calibrated using SiO₂ [a 4.9145(1), c 5.4061(3) Å] as an internal standard, and cell dimensions were refined by least-squares using a modified version of the CELREF program of Appleman & Evans (1973). Table 3 compares the diffraction data of the Yellowknife beusite and the type beusite from Argentina.

The cell dimensions of the single crystal were refined from the setting angles of thirteen automatically aligned intense reflections collected on a Nicolet R3m automated four-circle diffractometer. Graphite-monochromatized MoK α (λ 0.7107 Å) radiation was used at an operating voltage and current of 40 kV and 40 mA, respectively.

The unit-cell dimensions of the Yellowknife beusite, particularly the b dimension, are quite different from those of the Argentinian and synthetic samples. The anomalously large b dimension appears to be related to the elevated Ca content (Wise *et al.* 1990).

Precession photographs of a randomly selected fragment of the beusite-triphylite intergrowth, taken with the c axis of triphylite as the precession axis, show a third orientation relationship of the two phosphates, in addition to those established optically. Beusite reflections are oriented along preferred crystallographic directions of triphylite, which indicates that the [100] of beusite nearly parallels [110] of triphylite (Fig. 4).



CHEMICAL COMPOSITION

Chemical analyses were performed on a MAC 5 electron microprobe in the energy-dispersion mode. The following standards were used for analysis: apatite (P,Ca), MnF_2 (Mn), and chromite (Fe,Mg). Specimen current was 0.010 mA at 15 kV. Acquisition time for each spectrum was 200 seconds. Data reduction was made using a modified version of MAGIC V (Colby 1980). Electron-microprobe data







FIG. 6. Graftonite-beusite compositions in terms of Ca, Fe and Mn. Dashed line denotes hypothetical limit of Ca solubility based on the Yellowknife data. Dots: data of Hurlbut & Aristarain (1968) and Fransolet (1977), open circles: this study. Open squares: synthetic data of Nord (1982) and Nord & Ericsson (1982).

of Yellowknife beusite and triphylite are presented in Table 4. Because lithium is not detectable on the microprobe, the results listed for triphylite reflect partial analyses only.

The Yellowknife examples of beusite are homogeneous in composition. The range of Mn/(Mn + Fe) values is very narrow, from 0.54 to 0.57, and somewhat lower than for the Argentinian examples (0.60–0.72; Hurlbut & Aristarain 1968). By contrast, the Ca content found in beusite from Yellowknife is high. In all cases of natural beusite and terrestrial graftonite, some Ca is present, but the beusite from Yellowknife contains the highest amount of Ca recorded to date (16.1 wt.% CaO). Unlike all other known examples of beusite, no Mg was detected in the Yellowknife sample.

DISCUSSION

In nature, graftonite-beusite commonly occurs intergrown with a member of the triphylitelithiophilite series or with sarcopside. Crystallization of graftonite-beusite is regarded as primary, generally occurring during the stages of inner intermediate zone and core consolidation. The interlamination has been examined by previous workers and is considered to represent exsolution from a hightemperature, (Li,Ca)-rich graftonite-like parent phase. Subsequent decrease in temperature resulted in the migration and concentration of Li into triphylite and Ca into beusite. Hurlbut (1965) suggested that the role of Li in the parent phase determines the type of intergrowth that will form. If Li is absent, sarcopside-graftonite intergrowths will occur; where Li is present, graftonite will occur with triphylitelithiophilite. Hurlbut & Aristarain (1968) interpreted the granular intergrowth of beusite-lithiophilite as being caused by rapid cooling from crystal boundaries, with reduced ionic mobility.

According to published compositions of graftonite-beusite, all terrestrial examples of the series contain some Ca, typically between 3 and 13 wt.% CaO. Only graftonite found in fine octahedrite meteorites is known to be Ca-deficient (Olsen & Fredriksson 1966). The most prominent feature of the Yellowknife beusite is its high Ca content. A metasomatic, cation-exchange introduction of Ca into beusite might be envisaged at hydrothermal stages. However, late fluids that have locally deposited calcite with apatite inclusions in the parent pegmatite would have affected triphylite first; triphylite-lithiophilite minerals are sensitive to fluoride-, carbonate- and calcium-bearing fluids, yielding (manganoan) apatite and calcite as alteration products (unpublished observations of P. Cerný and A.-M. Fransolet). The triphylite lamellae in the Yellowknife beusite have not suffered this alteration, which suggests that the levels of Ca observed in the

beusite are a direct reflection of the composition of the parent mineral before exsolution.

A comparison of the Mn/(Mn + Fe) values in triphylite and beusite from four different localities shows that the graftonite-structured mineral tends to incorporate Mn more readily than triphylite (Fig. 5a). Similarly, the graftonite structure is better suited to incorporating Ca than that of triphylite (Fig. 5b).

The compositions of the Yellowknife beusite extend the previously experimentally determined boundaries of $Fe_3(PO_4)_2-Ca_3(PO_4)_2$ and $Mn_3(PO_4)_2$ -Ca₃(PO₄)₂ solid solutions. Nord & Ericsson (1982) and Nord (1982) determined the maximum solubility of Ca to be about 35 atom % in $Fe_3(PO_4)_2$ and 15 atom % in $Mn_3(PO_4)_2$. The low solubility of Ca in $Mn_3(PO_4)_2$ is surprising in light of the flexible nature of the graftonite structure. Ideally, one third of the cation sites are six- or sevencoordinated, depending on the presence of large cations such as Mn or Ca. Calvo (1968) showed that Ca²⁺ occupies and prefers the unique sevencoordinated M(1) site over the M(2) and M(3) sites. However, Nord & Ericsson (1982) showed that in spite of the larger size of the M(1) site, the substitution of Ca for Fe in the graftonite structure takes place at both the M(1) and M(3) sites, with virtually no preference. Since one third of the cation sites are sufficiently large to accomodate the available Ca, there is no obvious reason why the maximum solubility of Ca should be less in beusite than in graftonite. In most graftonite-beusite specimens examined to date, the amount of Ca present is not sufficient to completely fill the M(1) site; the difference is made up by the incorporation of Mn or Fe. Based on cation abundances, the M(1) site of the Yellowknife beusite may be nearly filled with Ca, as Ca closely approaches 33% of the total cations. Based on our present understanding of the crystal chemisty of the graftonite-beusite series and on the Yellowknife data, we suggest that Ca could constitute one third of all cations in both graftonite and beusite (Fig. 6). The substitution may, however, be different in Ca-disordered phases and in those with Ca restricted to the M(1) site.

ACKNOWLEDGEMENTS

This study was supported by a NSERC operating grant to P. Černý. The samples were collected during a regional study of the Yellowknife pegmatite field that was financially supported by the Tantalum Mining Corporation of Canada, Ltd. and the Geology Division, Indian and Northern Affairs of Canada, Yellowknife. The authors thank Dr. F.C. Hawthorne for his helpful comments. The authors also thank R.F. Martin, E.E. Foord and two anonymous referees for critical reviews of the manuscript.

REFERENCES

- APPLEMAN, D.E. & EVANS, H.T., JR. (1973): Job 9214: Indexing and least-squares refinement of powder diffraction data. U.S. Geol. Surv., Comput. Contrib. 20 (NTIS Doc. PB2-16188).
- BEUS, A.A. (1950): Magniophilite and mangankoninckite, two new minerals from pegmatites. Dokl. Akad. Nauk S.S.S.R. 73, 1267-1279 (in Russ.; Chem. Abstr. 45, 3766).
- BLOSS, F.D. (1981): The Spindle Stage: Principles and Practice. Cambridge University Press, Cambridge, England.
- BROOKS, J.H. & SHIPWAY, C.H. (1960): Mica Creek pegmatites, Mount Isa, North Western Queensland. *Aust. Queensland Gov. Mining J.* 61, 511-522.
- CALVO, C. (1968): The crystal structure of graftonite. Am. Mineral. 53, 742-750.
- ČERNÝ, P. (1989): Exploration strategy and methods for pegmatite deposits of tantalum. *In* Lanthanides, Tantalum and Niobium (P. Möller, P. Černý & F. Saupe, eds.), Springer-Verlag, New York.
- COLBY, J.W. (1980): MAGIC V a computer program for quantitative electron-excited energy dispersive analysis. *In* QUANTEX-Ray Instruction Manual. Kevex Corporation, Foster City, California.
- FRANSOLET, A.-M. (1977): Intercroissances et inclusions dans les associations graftonite-sarcopsidetriphylite. Bull. Soc. fr. Minéral. Cristallogr. 100, 198-207.
 - _____, KELLER, P. & FONTAN, F. (1986): The phosphate mineral associations of the Tsaobismund pegmatite, Namibia. *Contrib. Mineral. Petrol.* 92, 502-517.
- HURLBUT, C.S., JR. (1965): Detailed description of sarcopside from East Alstead, New Hampshire. Am. Mineral. 50, 1698-1707.

<u>& ARISTARAIN, L.F. (1968): Beusite, a new</u> mineral from Argentina, and the graftonite-beusite series. *Am. Mineral.* 53, 1799-1814.

- HUTCHINSON, R.W. (1955): Regional zonation of pegmatites near Ross Lake, District of Mackenzie, Northwest Territories. Geol. Surv. Can. Bull. 34.
- LAHTI, S.I. (1981): On the granitic pegmatites of the Eräjärvi area in Orivesi, southern Finland. Bull. Geol. Surv. Finland 314.
- MANDARINO, J.A. (1981): The Gladstone-Dale relationship. IV. The compatibility concept and its application. Can. Mineral. 19, 441-450.
- MEINTZER, R.E. (1987): The Mineralogy and Geochemistry of the Granitoid Rocks and Related Pegmatites of the Yellowknife Pegmatite Field, Northwest Territories. Ph.D. thesis, Univ. Manitoba, Winnipeg, Manitoba.
- NORD, A.G. (1982): Graftonite-type and graftoniterelated $(Mn_{1-z}Me_z)_3(PO_4)_2$ solid solutions. *Mat. Res. Bull.* 17, 1001-1010.
- & ERICSSON, T. (1982): Cation distributions in (Fe_{1-z}Me_z)₃(PO₄)₂ graftonite-type solid solutions determined by Mössbauer spectroscopy. Z. Kristallogr. 161, 209-224.
- OLSEN, E. & FREDRIKSSON, K. (1966): Phosphates in iron and pallasite meteorites. *Geochim. Cosmochim. Acta* 30, 459-470.
- WISE, M.A., HAWTHORNE, F.C. & ČERNÝ, P. (1990): Crystal structure of Ca-rich beusite from the Yellowknife pegmatite field, Northwest Territories. *Can. Mineral.* 28, 141-146.
- Received November 3, 1988, revised manuscript accepted August 31, 1989.