

ACCESSORY MINERALS IN THE XIHUASHAN Y-ENRICHED GRANITIC COMPLEX, SOUTHERN CHINA: A RECORD OF MAGMATIC AND HYDROTHERMAL STAGES OF EVOLUTION

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ABSTRACT

The Xihuashan granitic complex (southern China) is mainly composed of coarse-grained porphyritic biotite granite (G–a), medium-grained porphyritic biotite granite (G–b), and medium- to fine-grained porphyritic biotite granite (G–c). The G–a granite is characterized by monazite-(Ce) and xenotime-(Y), in addition to commonly observed zircon and uranoan thorite. In the G–b and G–c intrusive units, monazite-(Ce) becomes very scarce, whereas Y-bearing minerals, including Y-bearing spessartine, xenotime-(Y), gadolinite-(Y), fergusonite-(Y), and yttrian fluorite (up to 15 wt% Y₂O₃), are concentrated to various extents. Spessartine exhibits chemical zoning, which is displayed, in the core, by elevated contents of Y and heavy REE (HREE) (up to 2.32 wt% Y₂O₃ and 1.51% HREE₂O₃) and micrometric inclusions of Y-bearing minerals. Similarly, zircon from the G–b and G–c granites contains polyminerale micro-inclusions in the core. Overall, the accessory minerals point to a magmatic environment of crystallization for the G–a granite. In contrast, the crystallization of the G–b and G–c granites involved magmatic and hydrothermal stages. The late- to postmagmatic fluid is enriched in F and CO₂, and appeared as a result of the final oversaturation of the granitic melt in a fluid phase.

Keywords: yttrian fluorite, yttrium-bearing spessartine–almandine, xenotime-(Y), zircon, electron-microprobe data, granite, Xihuashan, China.

SOMMAIRE

Le complexe granitique de Xihuashan, dans le sud de la Chine, est principalement composé de granite porphyrique à biotite à gros grains (G–a), à grains moyens (G–b) et à grains de moyens à fins (G–c). Le granite G–a contient, en plus du zircon et de la thorite uranifère répandus, deux minéraux accessoires: monazite-(Ce) et xénotime-(Y). Dans les granites G–b et G–c, la monazite-(Ce) devient très rare, tandis que divers minéraux à yttrium, dont spessartine, xénotime-(Y), gadolinite-(Y), et fergusonite-(Y), montrent des teneurs variables en yttrium, de même que la fluorite (jusqu'à 15% en poids de Y₂O₃). La spessartine présente une zonation chimique, qui se caractérise au coeur par des concentrations élevées en Y et terres rares lourdes (jusqu'à 2.32% Y₂O₃ et 1.51% des oxydes de terres rares lourdes, en poids) et des inclusions micrométriques de minéraux yttrifères. De même, le centre des cristaux de zircon des granites G–b et G–c contient aussi de petites inclusions de phases yttrifères. Ces

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observations de minéraux accessoires indiquent un milieu magmatique pour le granite G-a, tandis que les granites G-b et G-c ont cristallisé au cours de stades magmatique et hydrothermal. Ces fluides magmatiques tardifs à post-magmatiques étaient enrichis en F et CO₂, qui proviennent d'une sursaturation finale du magma granitique en phase fluide.

Mots-clés: fluorite yttrifère, spessartine-almandin yttrifère, xénotime-(Y), zircon, granite, Xihuashan, Chine.

INTRODUCTION

Accessory minerals are considered the principal hosts of high-field-strength elements (HFSE) and rare-earth elements (REE) in granitic melts. Many investigations have shown that assemblages of accessory minerals are compositionally more variable in granitic rocks than major minerals. Their crystallization is generally sensitive to parameters like temperature, $f(\text{O}_2)$, and melt composition (e.g., Cuney & Friederich 1987, Rapp & Watson 1986), and their compositional evolution typically continues below the solidus. Therefore, accessory minerals in a granitic rock can serve as a sensitive indicator of crystallization histories at the magmatic stage (Robinson & Miller 1999, Dahlquist 2001, Wang *et al.* 2001), and during postmagmatic events.

South China is an important area of peraluminous granitic activity associated with W-Sn-Nb-Ta mineralization (Hu *et al.* 1984). Examples of mineralized granites include the Xihuashan granite, Jiangxi (W deposit, Maruéjol *et al.* 1990), the Xianghualing granite, Hunan (Sn-Ta-Li deposit, Zhu & Liu 1990), the Yichun granite, Jiangxi (Nb-Ta-Li deposit, Yin *et al.* 1995), and the Limu granite, Guangxi (Ta-Sn deposit, Zhu *et al.* 2001). Of these plutons, most are depleted in Y and the REE (<2 ppm Y, <30 ppm ΣREE); however, the Xihuashan granite is exceptionally rich in REE and particularly in Y (98 < ΣREE < 322 ppm; 91 < Y < 170 ppm). Correspondingly, a series of Y-bearing phosphates, silicates, fluorides, and niobotantalates, and also other associated accessory minerals of Zr, U, Th and Nb, are present in the Xihuashan granitic complex, and motivated our study. Apart from results of a few wet-chemical analyses and partial electron-microprobe analyses of rock-forming and accessory minerals, virtually no data exist on the mineralogy of the Xihuashan granites. We here present a detailed description of a complex assemblage of minerals containing Y, REE, Zr, Nb, W, U, and Th minerals, and provide a discussion on their response to magmatic and hydrothermal evolution during the crystallization and cooling of the pluton.

GEOLOGICAL SETTING

The Xihuashan district is an important part of the Gannan (southern Jiangxi) tungsten metallogenic province, and is world-famous for its large vein-type tungsten deposits. They were discovered in the early 1930s

in the endo- and exocontact zones of the Xihuashan granitic complex (Hsu 1943). Mining of quartz-ferberite-type ores started in 1935. The metallogeny of the deposits and the petrology of associated granites were described by Hsu (1943), Le Bel *et al.* (1984), McKee *et al.* (1987), Giuliani *et al.* (1988), Maruéjol *et al.* (1990) and Shen *et al.* (1994).

The Xihuashan granitic complex belongs to the Nanling Yanshanian (Cretaceous-Jurassic) orogenic metallogenic belt; it is located 10 km northeast of Dayu city (Fig. 1). It crops out over an area of about 20 km² and was emplaced in Cambrian sandstones. Exploratory drilling allows us to subdivide this complex into at least three successive pulses of emplacement. The first stage (γ_5^{2a}) consists of coarse-grained porphyritic biotite granite (G-a). It occurs as the outermost part of the complex, and outcrops over an area of 4.8 km². It consists mainly of quartz, perthitic K-feldspar, oligoclase (An₁₀₋₁₂) and biotite. The second stage (γ_5^{2b}) consists of medium-grained porphyritic biotite granite (G-b), which is ore-producing and hosts tungsten mineralization at the Xihuashan mine. The surface exposure is about 5.2 km². The major rock-forming minerals include quartz, perthitic K-feldspar, sodic plagioclase (An₀₋₂), biotite, and spessartine. The third stage (γ_5^{2c}) crops out over an area of 8 km² in the northern part of the Xihuashan complex. It is a medium- to fine-grained porphyritic biotite granite (G-c), and hosts tungsten mineralization, exploited at the Dangping mine. The major minerals are similar to those of the G-b granite, but fluorite is relatively more abundant in the G-b granite. A fine-grained biotite granite (G-d) is considered as a sterile border of the γ_5^{2c} granite, and restricted to NW-SE-trending dykes in the middle part of the pluton.

Chronological studies of the Xihuashan granitic complex (K-Ar, Sm-Nd and Ar-Ar methods) gave ages of about 150 Ma (Le Bel *et al.* 1984, McKee *et al.* 1987). In Table 1, major- and selected trace-element contents of the granites at Xihuashan are shown. Values of A/CNK (defined in Table 1) for the units of the Xihuashan granitic complex range from 1.04 to 1.07, and suggest a weakly peraluminous granite, but less strongly so than other rare-metal-enriched granites in southern China. The granite samples contain between 91 and 170 ppm Y, and 98-322 ppm REE, strikingly higher than other comparable Chinese granites, and demonstrate a marked depletion in LREE from the G-a granite to the G-b and G-c granites. More detailed

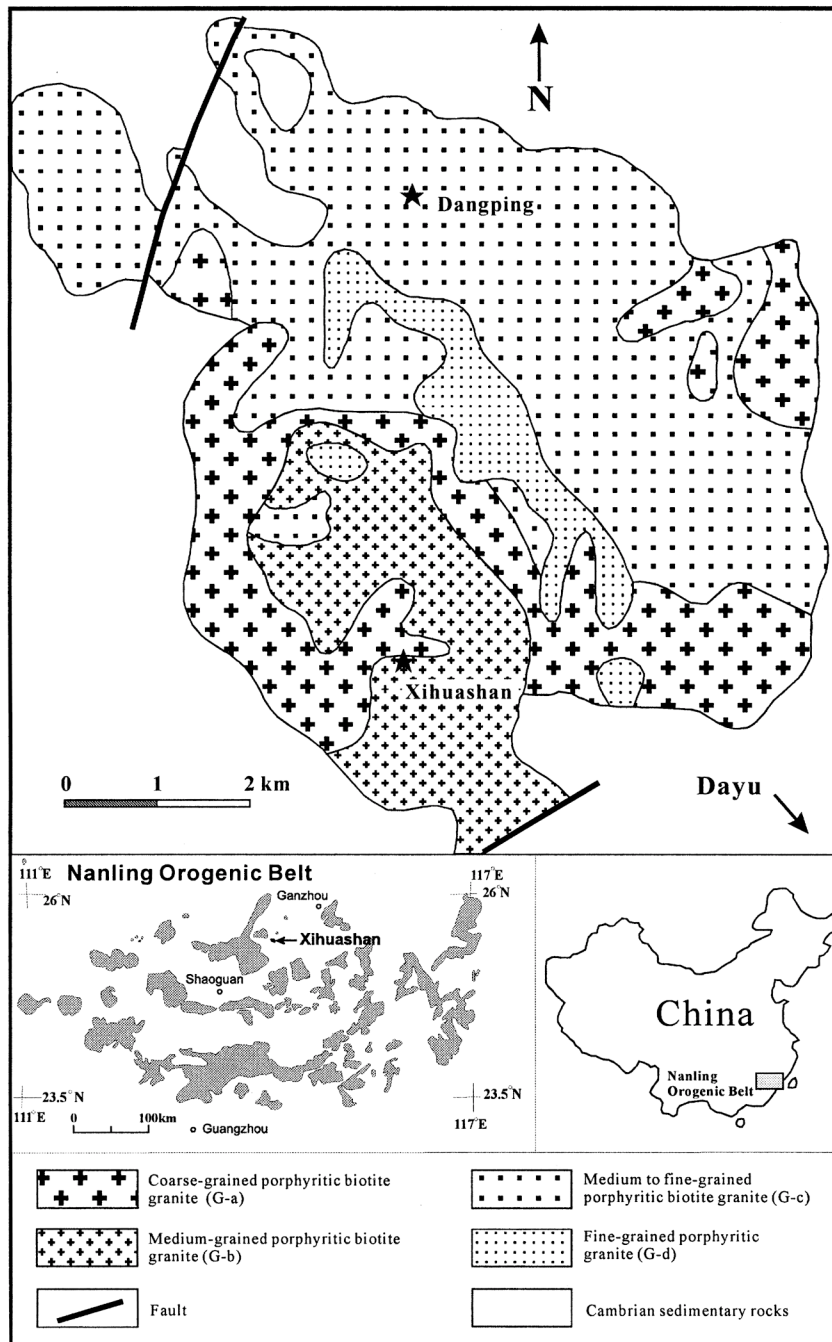


FIG. 1. Geological map of the Xihuashan granitic complex, southern China, modified after Li *et al.* (1986).

petrological and geochemical information on the Xihuashan granites is described in Li *et al.* (1986) and Maru ejol *et al.* (1990).

SAMPLES AND ANALYTICAL METHODS

Polished thin sections of the G-a, G-b and G-c granites were initially examined using the back-scattered electron (BSE) mode with a JEOL JXA8800 electron microprobe at the Department of Earth Sciences, Nanjing University, in order to characterize the textural relationships and to define the mineral paragenesis of accessory minerals. Compositional analyses of minerals were carried out using the same electron microprobe, except for spessartine, xenotime-(Y) and monazite-(Ce), which were analyzed using a Cameca SX50 electron microprobe at Universit  Paul Sabatier in Toulouse. The operating conditions for the JEOL electron-microprobe analysis were as follows for all accessory minerals except the REE minerals: acceleration voltage 20 kV, beam current 20 nA, and diameter of electron beam 1  m, ZAF correction program. The method outlined by Williams (1996) was used for analyses of the minerals of rare-earth elements. Count time of 30 seconds were used for Y, La and Ce, 50 seconds for Pr, Nd and Sm, and 20 seconds for Eu, Gd, Er and Yb. Synthetic Y and REE pentaphosphates (YP₅O₁₄ and REEP₅O₁₄) were used as standards. The procedures of  mli & Griffin (1975) and Roeder (1985) were used to correct peak overlaps between some elements. Europium contents of REE-bearing minerals are generally below detection limits (about at 0.02 wt% Eu₂O₃). The operating conditions for the Cameca electron-microprobe included an acceleration voltage of 15 kV, a beam current of 20 nA, a beam diameter of 1  m, and PAP correction program. Synthetic phosphates were used for measurements of REE and Y (Jarosewich & Boatner 1991).

THE COMPOSITIONS OF THE ACCESSORY MINERALS

The accessory minerals from the Xihuashan granites were systematically studied and described as follows in the order of fluoride, silicate, phosphate, carbonate, oxide and tungstate.

Fluorite

Fluorite is restricted to the G-b and G-c granites. We recognize three generations. The first (Fluorite I) is rare, but is invariably associated with spessartine (Fig. 2a); the second (Fluorite II) is associated with crystals of zircon and thorite (Fig. 2b), and located predominantly as intergranular minerals among rock-forming minerals such as feldspar, quartz, biotite and muscovite. The third (Fluorite III) coexists with synchysite-(Y) (Fig. 2c), and characteristically surround grains of fluorite I (Fig. 2a).

Representative results of electron-microprobe analyses are shown in Table 2. Fluorite I is compositionally unusual in that it contains more than 8 wt% Y₂O₃, and up to 15 wt% (Sample B, Table 2). Concentrations of REE₂O₃ vary from 2.42 to 7.87 wt%. Calculation of structural formulae gives up to 16 mole % of a (Y,REE)F₃ component in solid solution in fluorite, suggestive of an yttrian fluorite (Ca_{1-x}Y_x)F_{2+x}. Fluorite II has a nearly pure composition, with <0.1 wt% Y₂O₃, although it is commonly associated with Y-bearing minerals including xenotime-(Y) and fergusonite-(Y). Fluorite III has variable Y contents, but invariably lower than 4 wt%. The REE contents in fluorite II and fluorite III are generally below the detection limits.

Back-scattered electron images show clearly that fluorite I is typically rimmed by fluorite III and synchysite-(Y) (Fig. 2a). Textural evidence suggests that the primary yttrian fluorite gave way to Y-poor fluorite (<4 wt% Y₂O₃) and synchysite-(Y) as crystallization progressed, the latter being distributed along the cleavage in the fluorite III (Fig. 2c).

TABLE 1. MAJOR- AND TRACE-ELEMENT CONTENTS OF THE XIHUASHAN GRANITIC COMPLEX AND COMPARISON WITH OTHER PERALUMINOUS RARE-ELEMENT-ENRICHED GRANITES IN SOUTHERN CHINA

Rcf.	XHS						YS	LM	XHL	DFX
	G-a (1)	G-b (1)	G-c (1)	G-a (2)	G-b (2)	G-c (2)	(3)	(4)	(5)	(6)
SiO ₂	74.61	75.74	74.98	73.00	76.43	75.65	71.12	73.55	72.46	79.90
TiO ₂	0.10	0.03	0.05	0.14	0.01	0.06	0.01	0.02	0.00	0.02
Al ₂ O ₃	13.32	13.17	13.55	13.95	12.67	12.82	16.68	14.82	15.76	13.73
Fe ₂ O ₃				1.67	0.91	0.73	0.16			
FeO	0.35	0.26	0.22		0.00	0.00		0.17	0.21	0.76
MnO	1.73	1.23	1.00		0.00	0.00		0.65	0.63	1.03
MgO	0.07	0.10	0.10	0.07	0.11	0.09	0.14	0.13	0.10	0.09
CaO	0.45	0.19	0.26	0.27	0.10	0.03	0.10	0.10	0.12	0.19
Na ₂ O	1.19	0.60	0.79	1.16	0.46	0.48	0.19	0.31	0.14	0.30
K ₂ O	3.39	4.11	3.84	3.32	3.79	4.09	4.88	5.02	5.15	3.84
P ₂ O ₅	4.71	4.35	4.70	5.00	4.61	4.17	3.13	3.93	3.97	2.37
LOI	0.05	0.05	0.05	0.03	<0.01	<0.01	0.36	0.20	0.01	0.27
F				1.13	0.14	0.17	1.61	0.52	0.73	
F=O				0.37	0.75	0.71	1.62	0.66	1.00	
Total	99.97	99.83	99.54	99.95	99.90	98.91	99.32	99.79	99.86	
A/CNK	1.04	1.05	1.05	1.07	1.05	1.06	1.42	1.13	1.21	1.45
Y	96.8	142.1	125.8	91.4	148.3	169.6	0.49	0.72	1.15	1.25
LREE	166.7	45.0	38.6	128.2	62.9	38.9	0.44	2.74	21.9	3.2
HREE	155.2	157.5	189.9	38.3	59.4	57.8	0.29	1.08	5.2	1.7
U				22.1	36.9	31.4		9.9		
Th				34.7	28.3	21.1	2.6	11.5	0.0	
Nb	23.5	52	21	14	31.5	25.5	48.8	82.6		100
Ta	5	20	30.5	6	18	18	78.2	75.3		87
Zr	200	165	10.5	107	66	65	25.4	32.6	37	153
Hf	7.1	11.2	86				4.3	4.0	2	15

Granite plutons: XHS: Xihuashan, YS: Yashan, LM: Limu, XHL: Xianghualing, DFX: Dengfuxian. References: (1) Xu (1986), (2) Maru ejol *et al.* (1990), (3) Huang *et al.* (2002), (4) Zhu *et al.* (2001), (5) Liu (1990), (6) Xu (1986). Fe₂O₃: total iron expressed as Fe₂O₃, A/CNK = Al₂O₃/(2CaO + Na₂O + K₂O) (molar proportions). The major-element composition is reported in wt%, and trace-element concentrations are expressed in ppm. LREE and HREE represent the total concentrations of the light and heavy rare-earth elements, respectively.

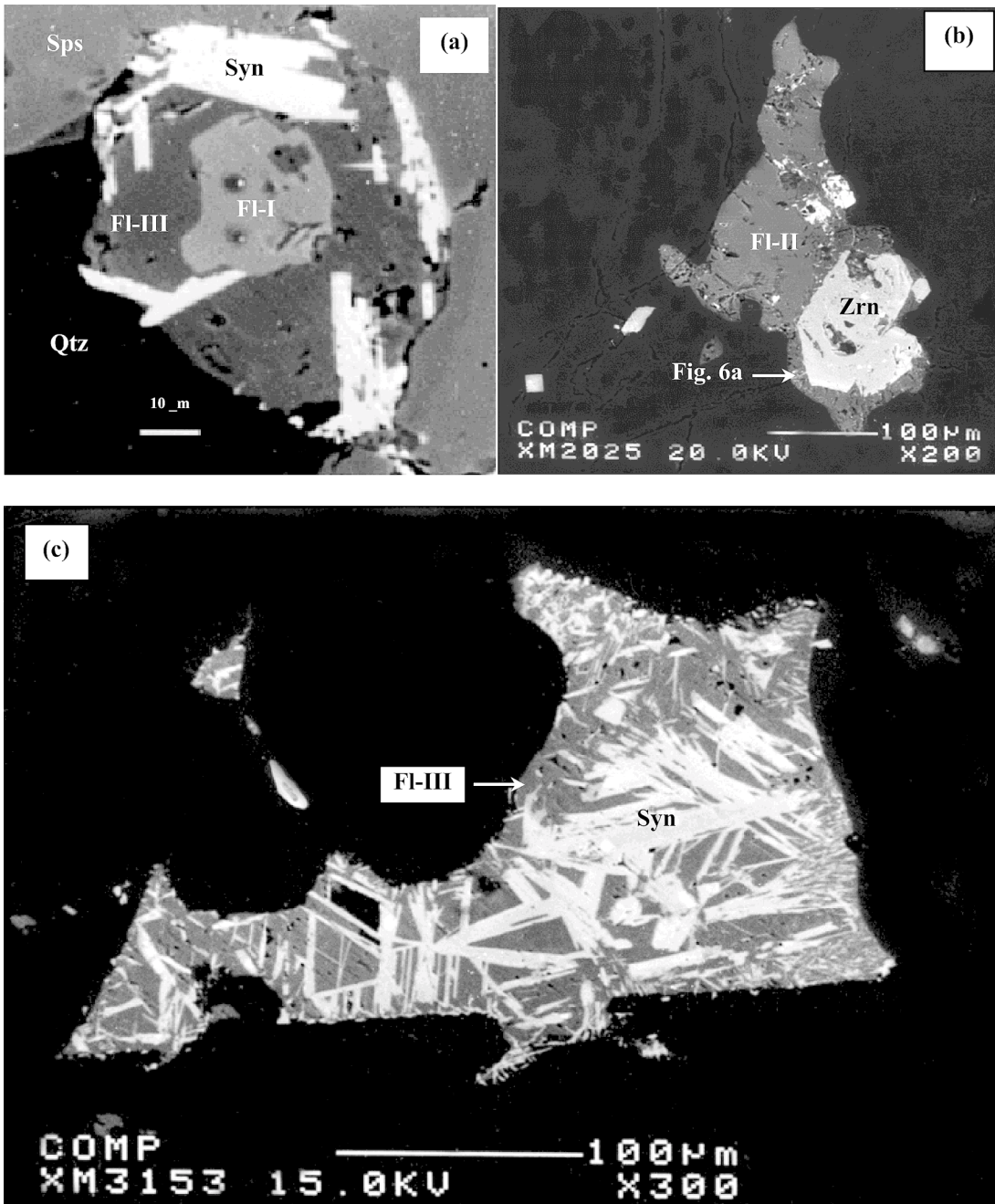


FIG. 2. Back-scattered electron images of fluorite. (a) Relic of early yttrian fluorite (FI-I) surrounded by later fluorite III (FI-III) and synchysite (Y) (Syn). (b) Association of fluorite II (FI-II) and hafnian zircon (Zrn). (c) Association of fluorite III and synchysite (Y). Sps: spessartine, Qtz: quartz.

TABLE 2. REPRESENTATIVE COMPOSITION OF FLUORITE AND ASSOCIATED SYNCHYSITE-(Y) FROM THE INTRUSIVE UNITS OF THE XIHUASHAN GRANITE

	A		B		C		D							
	Fl-I	Fl-III Syn-(Y)	Fl-I	Fl-III Syn-(Y)	Fl-I	Fl-III Syn-(Y)	Fl-I	Fl-III Syn-(Y)	Fl-I	Fl-II	Fl-II	Fl-II		
Y ₂ O ₃ wt%	12.98	3.90	27.95	14.97	3.75	28.45	13.20	5.54	27.74	14.77	2.66	25.61	0.06	0.61
La ₂ O ₃	0.48	0.03	1.64	0.29	0.02	1.46	0.41	0.01	1.32	0.59	0.00	3.32	-	0.08
Ce ₂ O ₃	1.65	0.11	4.65	1.16	0.08	3.62	1.52	0.15	3.18	1.63	0.18	7.85	-	0.04
Nd ₂ O ₃	0.84	-	2.12	0.71	-	1.85	0.57	0.02	1.75	1.19	0.01	1.91	-	-
Sm ₂ O ₃	0.09	-	1.01	0.12	-	0.89	0.15	-	1.17	0.91	-	0.91	-	-
Gd ₂ O ₃	0.49	-	1.96	0.31	-	1.35	0.10	-	2.16	0.92	-	1.76	-	-
Tb ₂ O ₃	0.24	-	0.09	0.07	-	0.05	0.02	-	0.16	0.02	-	0.08	-	-
Dy ₂ O ₃	0.64	-	1.86	0.12	-	1.79	0.16	-	2.17	1.18	-	1.67	-	-
Ho ₂ O ₃	-	-	0.34	-	-	0.35	-	-	0.34	0.11	-	0.31	-	-
Er ₂ O ₃	-	-	1.61	-	-	1.68	-	-	1.26	1.18	-	1.45	-	-
Yb ₂ O ₃	0.30	-	1.15	0.15	-	1.06	0.24	-	0.99	0.14	-	1.04	-	-
CaO	54.81	67.87	19.08	54.97	67.26	19.59	56.09	66.16	19.61	51.10	68.89	18.91	71.00	70.92
Total	72.52	71.90	63.45	72.86	71.11	62.14	72.45	71.87	61.85	73.72	71.74	64.81	71.07	71.64
F	46.59	47.90	4.70	45.22	47.94	4.70	46.13	47.95	5.14	44.92	48.17	4.09	48.45	48.23
F=O	-19.57	-20.12	-1.97	-18.99	-20.13	-1.97	-19.38	-20.14	-2.16	-18.87	-20.23	-1.72	-20.40	20.31
Total	99.55	99.68	66.18	99.09	98.91	64.86	99.21	99.68	64.83	99.77	99.68	67.18	99.12	99.56
CO ₂ (cal)		32.73		32.73		32.53		32.53		32.75				
Total	99.55	99.68	98.91	99.09	98.91	97.59	99.21	99.68	97.36	99.77	99.68	99.94	99.12	99.56
Y at. %	0.103	0.028	0.726	0.117	0.027	0.739	0.103	0.040	0.725	0.120	0.019	0.665	0.000	0.004
La	0.003	0.000	0.030	0.002	0.000	0.026	0.002	0.000	0.024	0.003	0.000	0.060	-	-
Ce	0.009	0.001	0.083	0.006	0.000	0.065	0.008	0.001	0.057	0.009	0.001	0.140	-	-
Nd	0.004	-	0.037	0.004	-	0.032	0.003	0.000	0.031	0.006	0.000	0.033	-	-
Sm	0.000	-	0.017	0.001	-	0.015	0.001	-	0.020	0.005	-	0.015	-	-
Gd	0.002	-	0.032	0.002	-	0.022	0.000	-	0.035	0.005	-	0.029	-	-
Tb	0.001	-	0.001	0.000	-	0.001	0.000	-	0.003	0.000	-	0.001	-	-
Dy	0.003	-	0.029	0.001	-	0.028	0.001	-	0.034	0.006	-	0.026	-	-
Ho	-	-	0.005	-	-	0.005	-	-	0.005	0.001	-	0.005	-	-
Er	-	-	0.025	-	-	0.026	-	-	0.019	0.006	-	0.022	-	-
Yb	0.001	-	0.017	0.001	-	0.016	0.001	-	0.015	0.001	-	0.015	-	-
Ca	0.873	0.972	0.998	0.867	0.973	1.025	0.881	0.959	1.032	0.838	0.980	0.988	1.000	0.995
Total	0.999	1.001	2.000	1.001	1.000	2.000	1.000	2.000	1.000	1.000	1.999	1.000	0.999	
F	2.190	2.024	0.725	2.106	2.046	0.725	2.138	2.052	0.798	2.175	2.023	0.631		1.998

A, B, C, D: association of fluorite I, fluorite III and synchysite-(Y) in four different samples. Fl-II: association with zircon. Structural formula calculated based on $(Ca + Y + REE) = 1$ for fluorite and $(Ca + Y + REE) = 2$ for synchysite-(Y). The amount of CO₂ was calculated by stoichiometry; -: below detection limits. Electron-microprobe data.

Spessartine

In the Xihuashan granitic complex, a garnet-group mineral is very rare in the G-a granite, but is widespread in both the G-b and G-c granites. It occurs as euhedral grains from 100 to 200 µm across. Observations in back-scattered electron images indicate that most garnet crystals contain polymineralic inclusions, dominated by Y-bearing minerals, including gadolinite-(Y), xenotime-(Y), fergusonite-(Y), and less commonly, zircon, thorite and uraninite. These inclusions, mostly micrometric, are restricted to the central portion of the crystals (Fig. 3a).

Representative electron-microprobe data (Table 3) define the garnet composition from the Xihuashan as spessartine (spessartine > 50 mole% except for a few cases), similar to what is found in rare-metal granitic pegmatites (Černý & Hawthorne 1982). Analytical data

reveal an enrichment in Y and the heavy REE (HREE), in particular in the core. The maximum concentrations of Y₂O₃, Gd₂O₃, Er₂O₃, Dy₂O₃ and Yb₂O₃ are 2.32, 0.26, 0.35, 0.26 and 0.84 wt%, respectively. Although yttrium and the HREE are particularly compatible with the garnet structure, compositional data for the Y-bearing spessartine, surprisingly, are uncommon and incomplete. Jaffe (1951) demonstrated a correlation between manganese content and yttrium content in garnet. In fact, Y concentrations in excess of 1 wt% Y₂O₃ are reported to occur in spessartine associated with pegmatitic environments (Jaffe 1951, Wakita *et al.* 1969). Jaffe (1951) proposed a coupled substitution of the type $Y^{3+} Al^{3+} \Leftrightarrow Mn^{2+} Si^{4+}$. However, as demonstrated in Figure 4, there is no simple correlation between Y + HREE and Mn, Fe, Ca and Mg in spessartine in the Xihuashan granites. A negative correlation between (Y, REE) and Al sug-

TABLE 3. COMPOSITION OF ZONED SPESSARTINE CRYSTALS FROM THE G-c GRANITE AT XIHUASHAN

	1	2	3	4	5	6	7	8	9	10	11	12	13	14
	Rim	Core	Rim	Core	Rim	Core	Rim	Rim	Core	Rim	Core	Rim	Core	Rim
MgO wt%	-	-	-	0.01	-	-	-	-	0.04	-	-	-	0.01	-
Al ₂ O ₃	20.55	19.38	20.46	19.36	20.58	19.06	20.92	20.61	18.93	20.33	18.75	19.91	19.12	20.87
SiO ₂	36.21	35.43	36.01	35.09	36.33	35.23	35.92	36.22	35.06	36.33	34.65	35.60	35.07	36.01
CaO	0.76	1.37	0.62	1.37	0.77	1.46	0.62	0.66	1.76	0.66	2.22	1.14	1.54	0.63
Cr ₂ O ₃	-	-	-	-	-	0.01	-	-	-	-	0.02	0.03	-	0.03
MnO	22.35	21.01	22.40	22.79	22.78	22.24	22.54	22.25	22.31	23.43	21.99	22.25	23.26	23.24
FeO	20.86	20.40	20.62	17.92	19.95	18.34	20.40	21.12	17.64	20.56	18.23	19.59	17.55	19.73
Y ₂ O ₃	0.29	1.58	0.18	2.14	0.12	1.99	0.05	0.06	2.32	0.06	2.17	0.91	1.71	0.21
Gd ₂ O ₃	0.14	-	0.12	0.11	0.18	0.18	-	-	-	-	0.17	-	0.26	-
Dy ₂ O ₃	0.22	0.04	0.04	0.35	-	0.08	-	0.09	0.13	0.04	-	0.15	0.25	0.15
Er ₂ O ₃	0.03	0.13	0.09	0.26	-	0.28	-	-	0.19	-	0.19	0.09	0.15	0.06
Yb ₂ O ₃	0.19	0.53	-	0.79	0.03	0.54	0.10	0.18	0.84	0.10	0.48	0.13	0.39	0.20
Total	101.61	99.88	100.54	100.17	100.74	99.41	100.55	101.19	99.21	101.49	98.87	99.81	99.31	101.14
Mg <i>apfu</i>	-	-	-	0.001	-	-	-	-	0.005	-	-	-	0.001	-
Al	1.985	1.917	1.991	1.921	1.993	1.899	2.030	1.993	1.892	1.963	1.883	1.958	1.907	2.020
Si	2.967	2.973	2.973	2.953	2.985	2.977	2.958	2.972	2.972	2.977	2.952	2.971	2.967	2.956
Ca	0.067	0.123	0.055	0.123	0.068	0.132	0.055	0.058	0.160	0.058	0.202	0.102	0.139	0.056
Cr	-	-	-	-	-	0.001	-	-	-	-	0.002	0.002	-	0.002
Mn	1.551	1.493	1.567	1.624	1.586	1.592	1.572	1.546	1.602	1.626	1.587	1.573	1.667	1.616
Fe	1.429	1.431	1.424	1.261	1.371	1.296	1.405	1.449	1.250	1.409	1.299	1.368	1.241	1.355
Y	0.013	0.070	0.008	0.096	0.005	0.089	0.002	0.003	0.104	0.003	0.099	0.040	0.077	0.009
Gd	0.004	-	0.003	0.003	0.005	0.005	-	-	-	-	0.005	-	0.007	-
Dy	0.006	0.001	0.001	0.009	-	0.002	-	0.002	0.003	0.001	-	0.004	0.007	0.004
Er	0.001	0.004	0.002	0.007	-	0.007	-	-	0.005	-	0.005	0.002	0.004	0.002
Yb	0.005	0.013	-	0.020	0.001	0.014	0.002	0.004	0.022	0.002	0.012	0.003	0.010	0.005
Total	8.028	8.025	8.024	8.018	8.014	8.014	8.024	8.027	8.015	8.039	8.046	8.023	8.027	8.025
Sps	50.90	49.00	51.43	53.97	52.43	52.71	51.85	50.65	53.09	52.58	51.40	51.70	54.67	53.40
Alm	46.90	46.96	46.75	41.91	45.34	42.91	46.34	47.46	41.44	45.55	42.05	44.94	40.72	44.76
Pyr	0.00	0.00	0.02	0.03	0.00	0.00	0.00	0.00	0.18	0.00	0.00	0.00	0.05	0.00
Grs	2.20	4.04	1.80	4.09	2.23	4.37	1.80	1.89	5.30	1.87	6.55	3.36	4.57	1.84

Structural formulae are calculated based on O = 12 atoms per formula unit (*apfu*). All Fe is expressed as Fe²⁺. Sps: spessartine; Alm: almandine; Pyr: pyrope; Grs: grossular. -: below detection limits. Electron-microprobe data.

gests an alternative scheme of substitution in the structure, possibly more simply, (Y, REE)³⁺ ⇌ Al³⁺.

The core of the spessartine is (Y,HREE)-rich and contains Y-bearing minerals

Back-scattered electron images generally show a brighter core and a darker rim (Fig. 3a). An inclusion-free, (Y, HREE)-poor rim is clearly indicated by analyses along a line (Fig. 3b). Spot analyses carried out on zoned crystals gives concentrations of Y and HREE up to 2.32 and 1.51 wt% of the respective oxides in the core, but less than 0.3 and 0.6 wt% at the rim (Table 3). In contrast, Al exhibits a core-to-rim increase, although variability is only at a level of 1–2 wt%, in accordance with a (Y, HREE) ⇌ Al substitution. Variations in Fe and Mn are less obvious. Such zoning was documented by several authors, *e.g.*, Hickmott *et al.* (1987), Lanzirotti (1995), Stowell *et al.* (1996), and Pyle & Spear (1999).

The spessartine also contains many micro-inclusions of Y-bearing minerals, such as gadolinite-(Y), fergusonite-(Y) and xenotime-(Y), restricted to the core (Fig. 3a). A combination of Y-mineral inclusions and enrich-

ment in Y in the spessartine core suggests that the early melt that led to the G-b and G-c units was very enriched in Y.

Gadolinite-(Y)

This mineral is very scarce and found exclusively in form of inclusions in the central part of garnet grains, commonly in association with micrograins of xenotime-(Y), fergusonite-(Y) and uraninite, although some discrete grains are also observed. The inclusions of gadolinite-(Y) are irregular in shape, typically 5–20 μm in size, whereas the discrete grains are commonly 0.5 mm across.

Zircon

Zircon is present in all units of the Xihuashan granites. In the G-a granite, small dipyrramids (~20 μm) are included randomly in rock-forming minerals (*e.g.*, feldspar, biotite). In the G-b and G-c granites, zircon occurs mostly as intergranular crystals among quartz, feldspar or biotite. Micro-inclusions are commonly found, but are restricted to the typically inclusion-

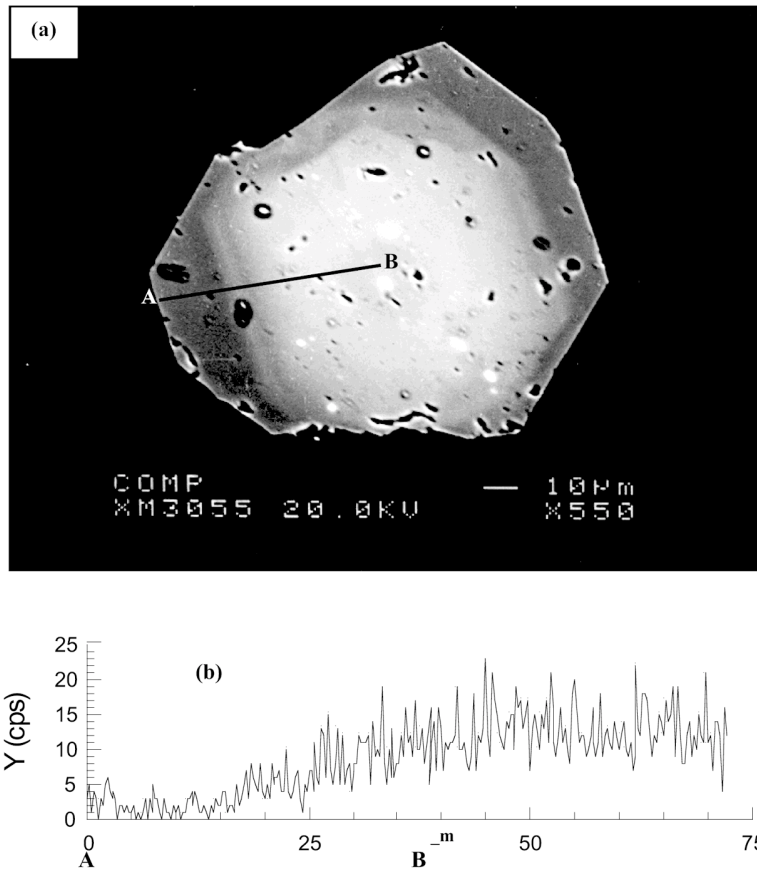


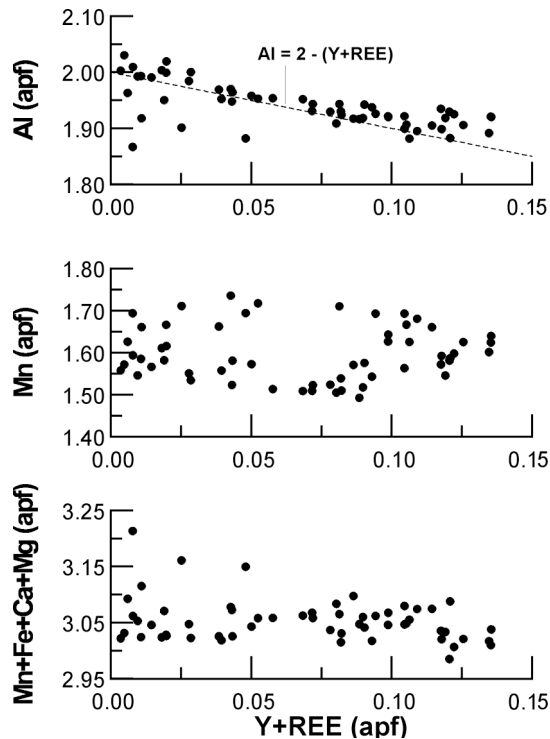
FIG. 3. (a) Back-scattered electron image of zoned grain of spessartine. Small inclusions of gadolinite-(Y) and xenotime-(Y) (bright dots) are restricted to the spessartine core. (b) Line analysis for yttrium along the traverse shown in Figure 3a, illustrating a core enriched in Y relative to the rim.

riddled core, which is mostly enclosed by an outer inclusion-free zone (Figs. 5a, b). Common polymineralic inclusions consist of anhedral xenotime-(Y), thorite and uraninite. Zircon may be also associated with xenotime-(Y), thorite and uraninite (Figs. 5b, c). Such aggregates may be observed in association with fluorite II (Fig. 2b). Zircon containing a microcrystal of thorite is included in spessartine (Fig. 5d), and provides information about the sequence of saturation of these minerals in the melt.

We have analyzed forty crystals of zircon. The crystals from the G-a unit are relatively homogeneous (Table 4). The levels of Hf range from 1.66 to 3.68 wt% HfO₂. The proportion of the hafnon end-member (HfSiO₄), indicated by the atomic ratio 100Hf/(Zr + Hf) (Hf# in Table 4), varies from 1.7 to 3.35, indicating that these crystals are normal zircon according to the termi-

nology proposed by Correia Neves *et al.* (1974). Zircon from the G-b and G-c granites is more enriched in Hf. The levels of Hf in zircon vary from 1.6 to 20.3 wt% HfO₂ (Nos. 3, 6a, Table 4). The Hf# is variable, between 1.42 to 19.97. These crystals are hafnian.

The level of U generally is in the range of 0.08 to 4.81 wt% UO₂, notably lower than contents found in zircon from the Beauvoir granite (up to 10.73 wt% UO₂; Cuney & Brouand 1987) or the Yichun granite (up to 9.94 wt% UO₂; Huang *et al.* 2002). Thorite incorporates uranium relative to zircon. Where zoned, the rim of the zircon is distinctly poorer in U than the core. The concentration of Th varies generally in the range from less than the detection limit to 1.89 wt% ThO₂, and that of Y is from 0.13 to 4.56 wt% Y₂O₃.



Zonation in zircon

Zircon generally displays oscillatory zoning, and was studied with BSE images and WDS analyses. As seen in Figure 6, the grains generally exhibit a brighter rim of variable width on BSE images. The map of Hf distribution shows a striking enrichment in Hf at the rim of the crystal. In all cases where multiple analyses could be performed in heterogeneous crystals, guided by BSE images, a systematic core-to-rim increase in Hf was found (Table 4, Fig. 7), in contrast to U, Th and Y, which are depleted at the rim. The phenomenon of outward enrichment in Hf has been described in many granites and granitic pegmatites, *e.g.*, the Beauvoir granite, France (Wang *et al.* 1992), the Suzhou granite, China (Wang *et al.* 1996), the Laoshan granite, China (Wang *et al.* 2000), the Homolka granitic pegmatite, Czech

FIG. 4. Plots of concentrations of Y + REE versus Al, Mn and total of Mn, Fe, Ca and Mg in the spessartine (in atoms per formula unit).

TABLE 4. COMPOSITION OF SELECTED ZIRCON CRYSTALS FROM THE XIHUASHAN GRANITIC COMPLEX

	G-a					G-b				G-c			
	1	2	3	4	5	6a	6b	7a	7b	8a	8c	9a	9b
	R		C			R		C		R		C	
SiO ₂ wt%	32.16	32.03	32.23	30.18	27.64	29.00	31.10	29.06	29.96	30.80	31.89	29.25	30.85
ZrO ₂	63.97	62.13	66.70	56.95	51.10	47.62	61.58	48.44	55.22	56.37	61.69	49.64	61.75
HfO ₂	1.91	3.68	1.64	5.28	1.55	20.29	5.02	19.50	5.06	11.19	4.16	19.64	4.76
UO ₂	0.30	0.81	0.18	0.97	4.81	0.57	0.29	0.45	3.07	0.27	0.37	0.27	0.41
ThO ₂	0.10	0.01	0.14	0.68	1.41	0.16	0.03	0.19	0.41	0.02	-	0.09	0.11
Y ₂ O ₃	0.47	0.34	0.26	3.10	1.51	1.14	0.74	1.19	2.36	0.80	0.67	0.70	0.83
La ₂ O ₃	-	-	-	-	0.00	-	0.02	-	-	-	-	-	-
Ce ₂ O ₃	0.07	0.02	0.03	0.06	0.00	0.07	0.02	0.04	0.04	0.03	0.02	0.03	0.03
P ₂ O ₅	0.23	0.26	0.30	2.10	1.18	1.08	0.64	0.93	1.06	0.54	0.48	0.85	0.75
Al ₂ O ₃	-	-	-	0.04	0.14	-	-	-	-	-	-	0.01	-
Total	99.21	99.29	101.48	99.36	89.34	99.93	99.44	99.80	97.18	100.02	99.29	100.47	99.49
Si <i>apfu</i>	0.998	1.001	0.981	0.953	0.981	0.970	0.978	0.971	0.979	0.986	0.996	0.970	0.970
Zr	0.968	0.947	0.990	0.877	0.884	0.777	0.944	0.789	0.880	0.880	0.940	0.802	0.947
Hf	0.017	0.033	0.014	0.048	0.016	0.194	0.045	0.186	0.047	0.102	0.037	0.186	0.043
U	0.002	0.006	0.001	0.007	0.038	0.004	0.002	0.003	0.022	0.002	0.003	0.002	0.003
Th	0.001	0.000	0.001	0.005	0.011	0.001	0.000	0.001	0.003	0.000	-	0.001	0.001
Y	0.008	0.006	0.004	0.052	0.028	0.020	0.012	0.021	0.041	0.014	0.011	0.012	0.014
La	-	-	-	-	0.000	-	0.000	-	-	-	-	-	-
Ce	0.001	0.000	0.000	0.001	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000
P	0.006	0.007	0.008	0.056	0.035	0.030	0.017	0.026	0.029	0.015	0.013	0.024	0.020
Al	-	-	-	0.002	0.006	-	-	-	-	-	-	0.004	-
Total	2.001	2.000	1.999	2.001	1.999	1.997	1.998	1.997	2.001	1.999	2.000	1.997	1.998
#Hf	1.7	3.4	1.4	5.2	1.7	20.0	4.6	19.1	5.1	10.4	3.8	18.8	4.3

R: rim, C: core. Structural formulae were calculated based on O = 4 atoms per formula unit (*apfu*). -: below detection limits. #Hf = 100 × Hf/(Zr + Hf). Electron-microprobe data.

Republic (Uher *et al.* 1998), and likely is a general feature in rare-metal-enriched granites.

On other hand, some grains are marked by a highly U-enriched core (Fig. 8), up to 4.81 wt% UO_2 , but also by the presence of numerous micro-inclusions of uraninite. In contrast, the rim contains only 0.79–1.21 wt% UO_2 .

In addition, the composition of the core from the G–b and G–c units almost overlaps that of zircon from the G–a unit (Fig. 7). In fact, zircon from the G–a contains on average 2.5 wt% HfO_2 , 0.38% UO_2 , 0.10% ThO_2 and 0.49% Y_2O_3 . Similarly, concentrations of Hf, U, Th and Y for the zircon core from the G–b granite average 3.81, 0.89, 0.16 and 1.13 wt% of the respective oxides. Overall compositional characteristics suggest that zircon from the G–b and G–c units may be formed initially under conditions similar to those in the G–a granite.

Thorite

Thorite is present in all units of the Xihuashan granitic complex, but differs in its mode of occurrence. It forms inclusions in monazite-(Ce) or xenotime-(Y) in the G–a granite. However, three generations of thorite may be recognized in the G–b and G–c granites: the first generation (thorite I) occurs as micro-inclusions enclosed in zircon commonly in association with xenotime-(Y) (Fig. 5a). Thorite II consists of anhedral grains, generally in contact with zircon in association with xenotime-(Y) or uraninite (or both; Figs. 5b, c). Grains of thorite III are typically small and anhedral, and are associated with apatite and fluorocarbonates of *LREE* (probably bastnäsite), derived from the alteration of monazite-(Ce) (Fig. 9).

Representative electron-microprobe results (Table 5) show that the U content is generally more than 11.48

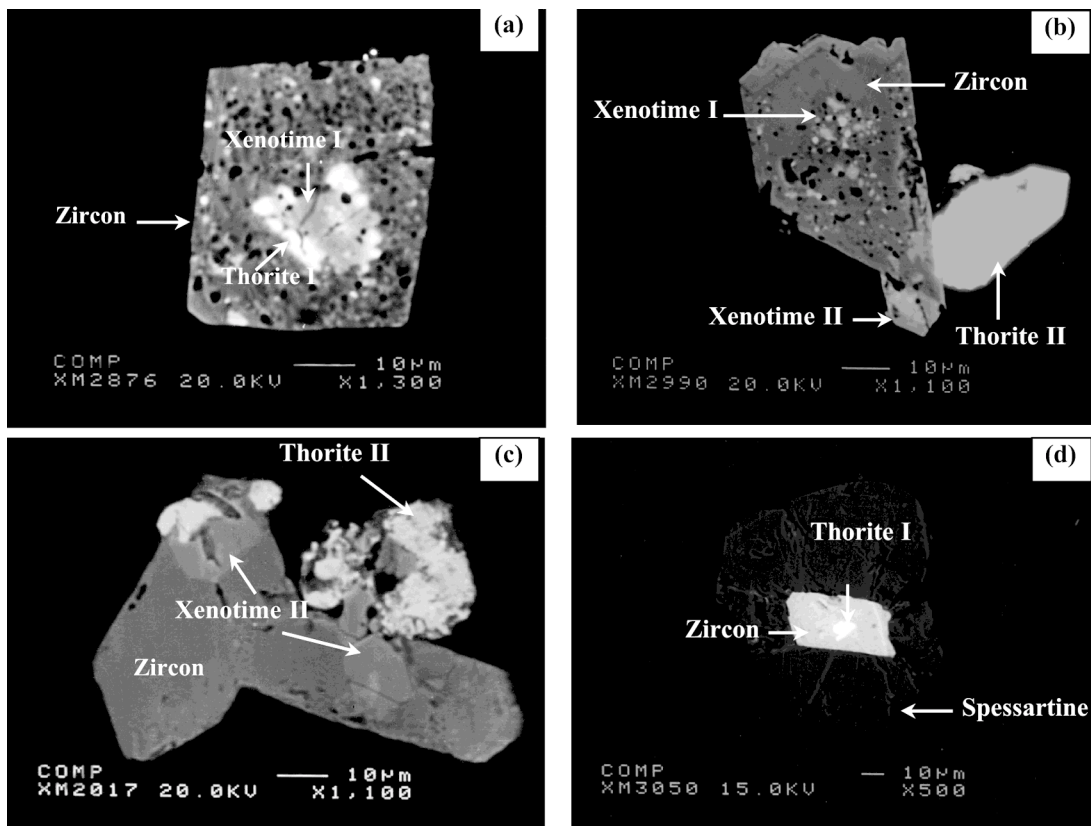


Fig. 5. Back-scattered electron images of zircon and associated minerals. (a) Porous zircon with inclusions of xenotime I and thorite I. (b) Porous zircon with small inclusions of xenotime I. Zircon is outside associated with larger euhedral crystals of xenotime II and thorite II. (c) Association of zircon with xenotime II and thorite II. (d) Zircon with inclusion of thorite I contained within spessartine.

wt% UO_2 (>13 mol. % USiO_4), indicating that the thorite is uranoan (Fig. 10). The maximum content of U (32.8 wt% UO_2 , No. 12, Table 5) approaches its experimentally established maximum, which is on the order of 33.1 wt%, or 40 mol. % USiO_4 (Zimmer 1986). The inclusions of thorite in zircon commonly incorporate the highest amounts of uranium, whereas the lowest contents of U (<17 wt% UO_2) typify the discrete grains (e.g., Nos. 4, 9, Table 5).

Thorite contains also Y, but generally below 5 wt% Y_2O_3 . Contents as high as 16 wt% Y_2O_3 invariably pertain to small grains coexisting with xenotime-(Y), and may well be due to overlap problems owing to grain size.

Thorite derived from the alteration of monazite is compositionally characterized by 24.5–26.5 wt% ThO_2 and 1–4 wt% Y_2O_3 , and exhibits extensive metamictization, as demonstrated by low analytical totals (<93 wt%), probably due to the presence of H_2O .

Xenotime-(Y)

Xenotime-(Y) is particularly abundant in Ca-poor peraluminous granites accounting for a large fraction of the Y and heavy REE (HREE) contents in the whole rocks (Wark & Miller 1993, Bea 1996, Förster 1998b).

Xenotime-(Y) is observed throughout the Xihuashan complex, and represents the most important carrier of yttrium. In the G-a unit, it occurs as the only mineral of yttrium, in association with monazite-(Ce) containing thorite inclusions. Two generations are distinguished in the G-b and G-c granites. Xenotime I consists of micro-inclusions enclosed in the core of zircon, commonly accompanied by anhedral thorite, suggesting a primary mineral having the composition $(\text{Zr,Th,Y})(\text{Si,P})\text{O}_4$ (Figs. 5a, b). Anhedral crystals also are observed as micro-inclusions in the Y-rich core of garnet crystals. Xenotime II forms larger euhedral crystals, and generally surrounds zircon (Figs. 5b, c).

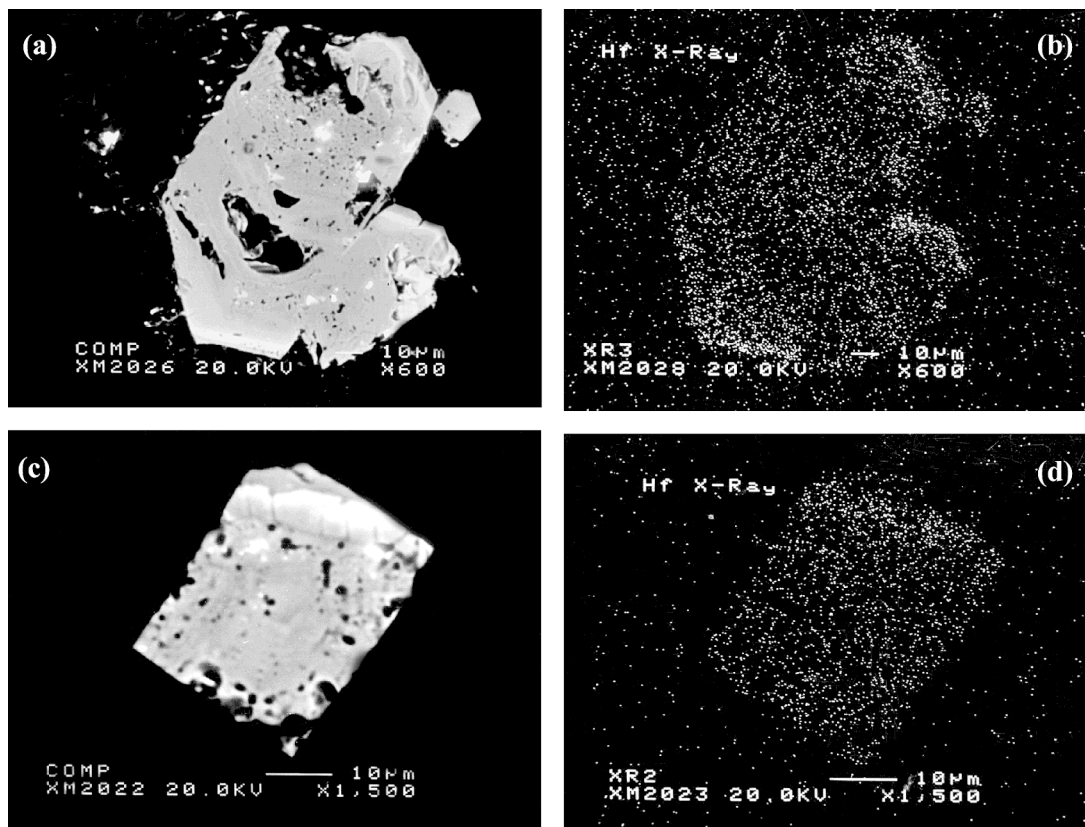


FIG. 6. Back-scattered electron images of zoned zircon (a, c). Images of distribution of Hf (b, d) reveal a pattern of Hf enrichment toward the margin.

TABLE 5. REPRESENTATIVE COMPOSITIONS OF THORITE FROM THE XIJIUASHAN GRANITIC COMPLEX, CHINA

	G-a								G-b								G-c	
	1 IZ	2 D	3 D	4 D	5 WZ	6 IZ	7 IZ	8 WZ	9 D	10 D	11 IZ	12 IZ	13 IZ	14 Sec	15 Sec	16 Sec	17 Sec	18 IZ
SiO_2 , wt%	16.53	14.38	14.82	14.82	14.41	16.56	17.41	15.64	15.90	15.50	16.82	16.02	14.97	16.52	16.64	18.22	16.72	15.48
UO_2	25.00	15.59	16.69	15.05	25.01	27.55	29.91	15.28	11.48	15.27	26.30	32.83	30.96	26.22	26.47	24.49	25.58	22.58
ThO_2	49.88	61.68	55.06	61.38	51.78	50.61	48.44	58.27	60.68	53.35	48.18	45.45	43.23	47.09	46.13	45.14	46.98	49.74
P_2O_5	1.89	1.49	2.72	1.35	2.98	1.71	1.19	1.29	1.42	2.11	2.32	1.46	3.32	1.28	1.37	0.25	1.18	0.93
Y_2O_3	3.10	2.90	4.17	2.82	4.41	2.19	1.59	3.76	2.62	5.30	3.54	1.79	4.03	1.02	1.29	1.84	1.02	1.03
PbO	1.08	0.21	0.28	0.16	0.20	1.10	1.10	0.18	0.22	0.17	0.33	1.08	0.80	0.13	0.21	0.14	0.18	0.30
La_2O_3	-	-	-	0.02	-	-	-	0.08	0.06	-	0.03	-	0.06	-	0.01	-	-	-
Ce_2O_3	-	-	0.24	0.19	0.09	0.04	0.03	0.24	0.01	0.01	-	-	0.04	0.04	-	0.04	-	0.03
CaO	0.17	0.38	0.36	0.37	0.31	-	0.07	0.39	0.59	0.40	0.78	0.56	0.45	0.24	0.25	0.15	0.23	2.09
Total	97.65	96.63	94.33	96.16	99.19	99.75	99.73	95.13	92.97	92.11	98.28	99.19	97.85	92.53	92.36	90.28	91.90	92.18
Si <i>apfu</i>	0.903	0.833	0.833	0.855	0.787	0.899	0.943	0.889	0.916	0.878	0.901	0.896	0.815	0.949	0.953	1.043	0.963	0.942
U	0.304	0.201	0.209	0.193	0.304	0.333	0.361	0.193	0.147	0.192	0.314	0.409	0.375	0.335	0.337	0.312	0.328	0.306
Th	0.620	0.812	0.703	0.805	0.643	0.624	0.597	0.753	0.795	0.687	0.587	0.578	0.535	0.616	0.600	0.588	0.615	0.688
P	0.087	0.073	0.129	0.066	0.138	0.079	0.054	0.062	0.069	0.101	0.105	0.069	0.153	0.062	0.066	0.012	0.057	0.048
Y	0.090	0.089	0.125	0.087	0.128	0.063	0.046	0.114	0.080	0.160	0.101	0.053	0.117	0.031	0.039	0.056	0.031	0.033
Pb	0.016	0.003	0.004	0.002	0.003	0.016	0.016	0.003	0.003	0.003	0.005	0.016	0.012	0.002	0.003	0.002	0.003	0.005
La	-	-	-	0.001	-	-	-	0.002	0.001	-	0.000	-	0.001	-	0.000	-	-	-
Ce	-	-	0.005	0.004	0.002	0.001	0.001	0.005	0.000	0.000	-	-	0.001	0.001	-	0.001	-	0.001
Ca	0.010	0.024	0.021	0.023	0.018	-	0.004	0.023	0.036	0.024	0.045	0.034	0.026	0.015	0.015	0.009	0.014	0.136
Total	2.030	2.035	2.029	2.036	2.023	2.015	2.022	2.044	2.047	2.045	2.058	2.055	2.035	2.011	2.013	2.023	2.011	2.159
USiO_4	31.06	18.46	20.36	17.92	28.42	33.63	36.94	18.33	14.65	18.73	31.63	40.23	37.16	34.26	34.73	32.77	33.84	30.08
ThSiO_4	60.18	73.44	67.27	73.73	59.53	60.21	58.51	70.40	77.38	65.93	58.31	54.72	51.42	62.49	61.25	61.29	62.95	66.61
YPO_4	8.76	8.10	12.37	8.35	12.05	6.16	4.55	11.27	7.97	15.35	10.06	5.05	11.42	3.24	4.03	5.94	3.21	3.31

Structural formula calculated on the basis of O = 4 atoms per formula unit (*apfu*). -: below detection limits. IZ: inclusion in zircon; D: discrete grain, WZ: associated with zircon, Sec: Secondary crystal. Electron-microprobe data.

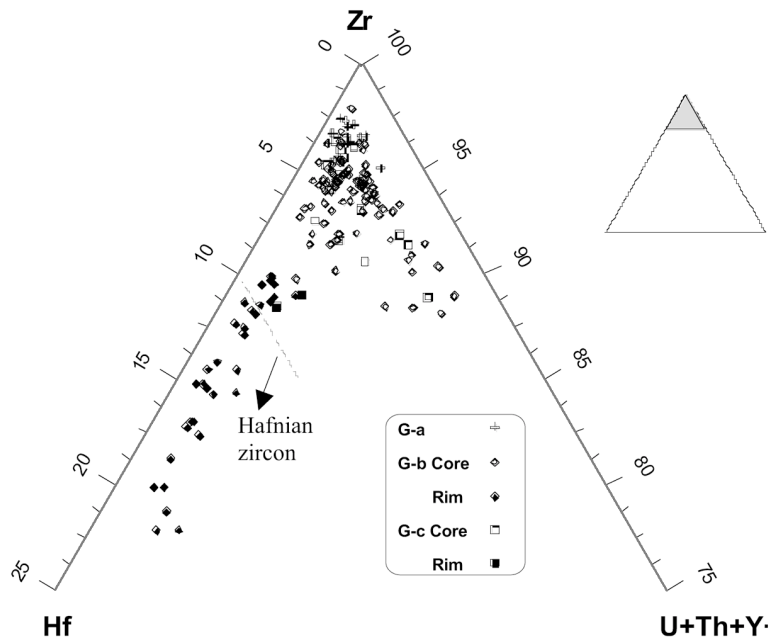


Fig. 7. Plot of compositions of zircon in the triangular diagram Zr – Hf – (U + Th + Y + REE) (atom %).

An accurate composition of the micro-inclusions of xenotime-(Y) could not be determined owing to their very small dimension (generally less than 2 μm in diameter). Representative electron-microprobe results on xenotime II are shown in Table 6. The *LREE* oxide concentrations are mostly below 1 wt%; however, the *HREE* concentrations are variable, ranging from 8.4 to 25.18 wt% of the respective oxides. Both of the actinide elements, U and Th, are present in concentrations less than 1 wt% UO_2 or ThO_2 , although one analysis did give a maximum Th content of 2.04 wt% ThO_2 (no. 7, Table 6), and several spots gave U contents of 1–1.5 wt% UO_2 . CaO concentrations are mostly less than 1 wt% CaO.

Monazite-(Ce)

Monazite-(Ce) is the typical carrier of the light rare-earth elements (*LREE*) in S-type granites (Broska *et al.* 2000), and in the Xihuashan complex. It is restricted to

the early stage of granite crystallization (the G–a granite), where it may form aggregates with xenotime-(Y) [much less abundant than monazite-(Ce)], zircon and thorite. In the G–b granite, monazite-(Ce) is only sporadically observed, with secondary thorite + bastnäsite-(Ce) + fluorapatite in aggregates (Fig. 9). Compositions of monazite-(Ce) are characterized by 0.65–2.06 wt% SiO_2 , 4.06–10.73% ThO_2 , 0.2–0.43% UO_2 , 0.69–1.74% Y_2O_3 and 0.12–0.83% CaO (Table 7). Monazite from S-type granites generally contains 6–12 wt% ThO_2 (Cuney & Friedrich 1987) or 4–12 wt% ThO_2 (Förster 1998a). In fact, concentrations of Th in monazite-(Ce) from the G–a granite fall near or within this range. The positive relationship between Th and Si indicates the existence of the huttonite component in solid solution up to 10 mol. % (No. 4, Table 7).

Synchysite-(Y)

This mineral is found only as bladed elongate crystals in an assemblage with fluorite III, in which it is dis-

TABLE 6. REPRESENTATIVE COMPOSITIONS OF XENOTIME-(Y) FROM THE XIHUASHAN GRANITIC COMPLEX, CHINA

	G-b							G-c		
	1	2	3	4	5	6	7	8	9	10
P_2O_5 wt%	34.3	33.75	34.87	34.45	33.68	35.42	33.17	34.62	34.87	35.06
SiO_2	-	-	-	-	0.45	-	0.29	0.51	-	0.02
ThO_2	0.08	0.39	-	0.14	0.69	0.04	2.04	0.15	-	0.47
UO_2	0.68	0.48	-	0.87	1.10	0.22	2.57	0.68	-	0.4
Y_2O_3	40.35	39.39	40.46	41.44	41.04	46.06	33.67	44.59	40.46	43.29
La_2O_3	0.13	-	0.32	0.09	-	0.02	-	0.08	0.32	-
Ce_2O_3	0.11	0.11	0.58	0.07	0.15	0.20	0.03	0.01	0.58	0.08
Nd_2O_3	0.66	0.48	-	0.17	-	-	0.31	-	-	0.28
Sm_2O_3	0.91	0.79	0.06	1.09	0.52	0.47	0.95	-	0.06	0.35
Gd_2O_3	5.60	5.43	4.5	4.645	4.63	2.24	3.00	0.91	4.41	4.01
Tb_2O_3	0.74	0.69	0.74	0.66	0.59	0.47	1.13	0.29	0.74	0.58
Dy_2O_3	6.95	7.15	6.25	5.61	5.52	3.56	6.60	2.74	5.625	4.70
Ho_2O_3	1.37	0.96	0.69	0.61	1.17	1.01	1.05	0.75	0.69	-
Er_2O_3	4.43	5.18	5.09	3.85	4.24	3.92	6.38	4.25	5.09	4.87
Tm_2O_3	0.44	0.44	0.45	0.53	0.40	0.81	0.34	0.84	0.45	0.64
Yb_2O_3	3.43	4.15	4.44	4.35	3.97	4.90	6.32	6.03	4.44	4.17
Lu_2O_3	0.2	0.91	0.08	0.35	-	0.16	0.35	0.08	0.08	0.48
CaO	-	0.06	0.01	-	0.01	0.06	0.05	0.07	0.01	0.04
PbO	0.35	0.32	0.46	0.32	0.67	0.30	0.46	1.81	0.46	0.38
Total	100.73	100.68	99.00	99.25	98.83	99.86	98.73	98.41	98.29	99.82
P apfu	1.000	1.000	1.000	1.000	0.984	1.000	0.990	0.983	1.000	0.999
Si	-	-	-	-	0.016	-	0.010	0.017	-	0.001
Th	0.001	0.003	-	0.001	0.005	0.000	0.016	0.001	-	0.004
U	0.005	0.004	-	0.007	0.008	0.002	0.020	0.005	-	0.003
Y	0.739	0.734	0.729	0.756	0.754	0.817	0.631	0.796	0.729	0.776
La	0.002	-	0.004	0.001	-	0.000	-	0.000	0.002	-
Ce	0.001	0.001	0.007	0.001	0.002	0.002	0.000	0.000	0.007	0.001
Nd	0.008	0.006	-	0.002	-	-	0.004	-	-	0.003
Sm	0.011	0.010	0.001	0.013	0.006	0.005	0.012	-	0.001	0.004
Gd	0.064	0.063	0.051	0.053	0.053	0.025	0.035	0.010	0.049	0.045
Dy	0.008	0.008	0.008	0.007	0.007	0.005	0.013	0.003	0.008	0.006
Tb	0.077	0.081	0.068	0.062	0.061	0.038	0.075	0.030	0.061	0.051
Ho	0.015	0.011	0.007	0.007	0.013	0.011	0.012	0.008	0.007	-
Er	0.048	0.057	0.054	0.041	0.046	0.041	0.071	0.045	0.054	0.051
Tm	0.000	0.000	0.000	0.006	0.004	0.008	0.000	0.009	0.005	0.007
Yb	0.036	0.044	0.046	0.045	0.042	0.050	0.068	0.062	0.046	0.043
Lu	0.000	0.000	0.000	0.004	-	0.002	0.000	0.001	0.001	0.005
Ca	-	0.002	0.000	-	0.000	0.002	0.002	0.003	0.000	0.001
Pb	0.003	0.003	0.004	0.003	0.006	0.003	0.004	0.016	0.004	0.003
Total	1.018	1.027	0.979	1.009	1.007	1.011	0.963	0.989	0.974	1.003

Structural formula calculated based on P + Si = 1 atom per formula unit (apfu). - : below detection limits. Electron-microprobe data.

TABLE 7. REPRESENTATIVE COMPOSITIONS OF MONAZITE-(Ce) FROM THE G-a GRANITE AT XIHUASHAN, CHINA

	G-a					
	1	2	3	4	5	6
P_2O_5 wt%	30.55	29.23	28.33	26.83	28.11	28.53
SiO_2	0.65	1.03	1.40	2.06	1.73	0.74
ThO_2	4.06	6.95	7.87	10.73	6.87	6.57
UO_2	0.26	0.43	0.31	0.40	0.20	0.30
Y_2O_3	0.95	0.98	0.80	0.99	1.74	0.69
La_2O_3	13.26	12.79	14.53	9.97	12.35	12.22
Ce_2O_3	28.83	25.30	28.18	23.06	27.15	25.31
Pr_2O_3	0.78	0.65	0.58	-	-	-
Nd_2O_3	12.85	11.99	10.68	14.20	12.68	11.70
Sm_2O_3	2.82	3.06	2.23	4.72	2.51	4.78
Gd_2O_3	2.12	2.26	1.76	2.34	1.43	2.69
Tb_2O_3	0.12	0.00	0.01	0.09	0.02	0.19
Dy_2O_3	0.54	0.63	0.50	0.84	0.69	0.74
Ho_2O_3	0.28	0.00	0.07	-	0.11	-
Er_2O_3	0.15	0.07	0.01	0.13	0.30	-
Tm_2O_3	0.32	0.04	0.12	0.21	0.04	0.40
Yb_2O_3	0.11	0.04	0.21	0.68	0.02	-
Lu_2O_3	0.01	-	0.02	-	-	-
CaO	0.24	0.32	0.31	0.16	0.12	0.83
PbO	0.08	-	0.11	0.10	0.04	-
Total	98.98	95.77	98.02	97.50	96.11	95.69
P apfu	0.976	0.960	0.945	0.917	0.932	0.970
Si	0.024	0.040	0.055	0.083	0.068	0.030
Th	0.035	0.061	0.071	0.099	0.061	0.060
U	0.002	0.004	0.003	0.004	0.002	0.003
Y	0.019	0.020	0.017	0.021	0.036	0.015
La	0.184	0.183	0.211	0.148	0.178	0.181
Ce	0.398	0.359	0.406	0.341	0.389	0.372
Pr	0.011	0.009	0.008	-	-	-
Nd	0.173	0.166	0.150	0.205	0.177	0.168
Sm	0.037	0.041	0.030	0.066	0.034	0.066
Gd	0.027	0.029	0.023	0.031	0.019	0.036
Tb	0.001	0.000	0.000	0.001	0.000	0.003
Dy	0.007	0.008	0.006	0.011	0.009	0.010
Ho	0.003	0.000	0.001	-	0.001	-
Er	0.002	0.001	0.000	0.002	0.004	-
Tm	0.075	0.010	0.030	0.053	0.010	0.100
Yb	0.001	0.000	0.003	0.008	0.000	-
Lu	0.000	-	0.009	-	-	-
Ca	0.009	0.013	0.013	0.007	0.005	0.036
Pb	0.001	-	0.001	0.001	0.000	-
Total	0.985	0.904	0.973	0.998	0.925	1.05

Structural formula calculated based on P + Si = 1 atoms per formula unit (apfu). - : below detection limits. Electron-microprobe data.

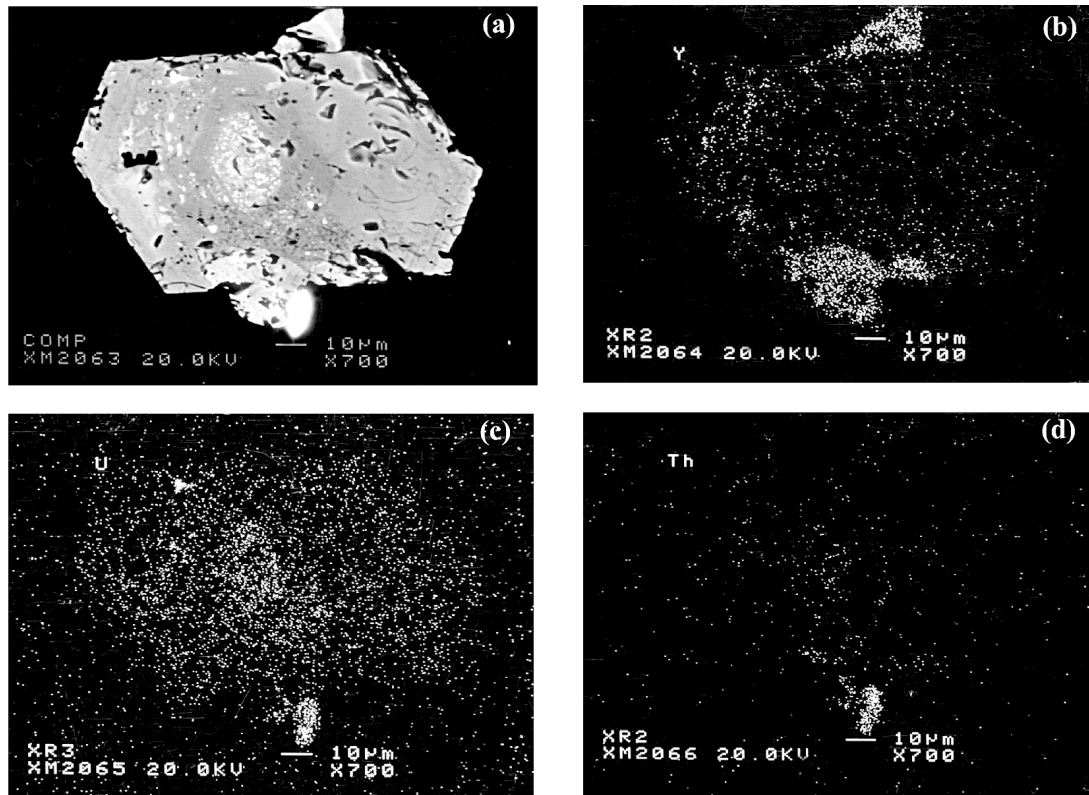


FIG. 8. Zoned crystal of zircon. (a). Back-scattered electron image. (b–d). Distribution maps of Y, U and Th. Note that the porous core has up to 4.81 wt% UO_2 , and contains micro-inclusions of uraninite. The rim is relatively rich in Hf. Around the zircon are the associated xenotime II and thorite II. e. Map of the crystal, with results of analyses.

TABLE 8. REPRESENTATIVE COMPOSITIONS OF URANINITE FROM THE XIHUASHAN GRANITIC COMPLEX, CHINA

	G-b								G-c			
	WF	WN	WF	WZ	WF	WF	WF	IZ	WZ	WZ	WZ	
UO_2 , wt%	79.90	83.69	83.27	80.77	83.81	80.64	84.18	75.00	83.51	86.62	84.01	
ThO_2	8.33	7.43	5.45	6.93	5.58	6.99	6.89	6.37	5.66	7.27	5.96	
P_2O_5	-	0.01	-	-	-	-	-	-	-	-	-	
Y_2O_3	3.40	0.97	2.83	3.87	2.25	3.13	1.49	6.33	1.24	0.41	1.74	
PbO	2.11	1.99	2.04	1.90	2.22	2.12	2.21	1.94	2.02	1.96	2.10	
Ce_2O_3	0.36	0.07	0.03	0.00	0.14	0.03	0.06	0.13	0.37	0.15	0.27	
Total	94.10	94.17	93.63	93.53	94.00	92.91	94.82	89.76	92.80	96.43	94.08	
U <i>apfu</i>	0.844	0.901	0.889	0.852	0.897	0.866	0.899	0.809	0.911	0.915	0.901	
Th	0.090	0.082	0.059	0.075	0.061	0.077	0.075	0.070	0.063	0.078	0.065	
P	-	0.001	-	-	-	-	-	-	-	-	-	
Y	0.086	0.025	0.072	0.098	0.058	0.080	0.038	0.163	0.032	0.010	0.045	
Pb	0.027	0.026	0.026	0.024	0.029	0.027	0.028	0.025	0.027	0.025	0.027	
Ce	0.006	0.001	0.001	0.000	0.002	0.001	0.001	0.002	0.007	0.003	0.005	
Total	1.053	1.036	1.047	1.049	1.047	1.051	1.041	1.069	1.04	1.031	1.043	

Structural formula calculated on the basis of O = 2 atoms per formula unit (*apfu*). - : below detection limits. WF: intergrown with fergusonite-(Y); WZ: intergrown with zircon; IZ: inclusion in zircon. Electron-microprobe data.

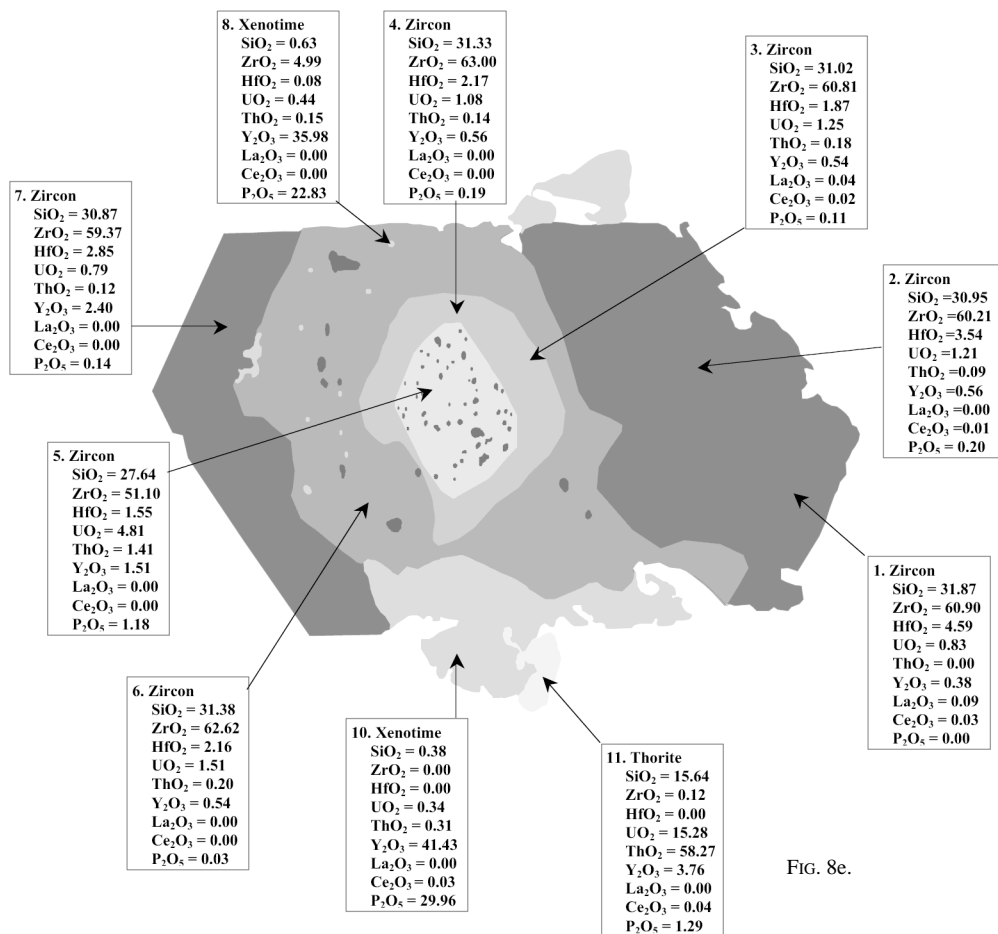


FIG. 8e.

tributed along cleavage planes (Fig. 2c). Representative electron-microprobe data for synchysite-(Y) are shown in Table 2. Owing to small dimensions of the grains, it is difficult to assess the compositional homogeneity on back-scattered electron images. However, electron-microprobe analyses demonstrate compositional variability from one spot to another with respect to the proportion of the Y, LREE and HREE. Concentrations of LREE oxide vary from 0.72 to 13.99 wt%, and that of the HREE oxide, from 4.08 to 16.74 wt%. Uranium and thorium are very low in synchysite-(Y), generally less than 0.3 wt% in terms of the oxides.

Uraninite

In the Xihuashan granitic complex, uraninite is present as an important accessory mineral in the G-b and G-c granites. Uraninite did not crystallize in the G-a granite, because uranium was incorporated in the uranoan thorite (25 wt% UO₂), which is more stable than

uraninite at temperatures above 350°C (Cuney & Friedrich 1987). It occurs in three different modes in the G-b and G-c granites: (1) intergrown with fergusonite (Fig. 11), (2) as micro-inclusions in zircon or spessartine (Fig. 8), and (3) associated with zircon.

Representative compositions of uraninite are reported in Table 8 and Figure 10. Thorium is the major trace element in the structure; irrespective of the mineral association, uraninite has commonly incorporated Th in the range of 2.71 to 8.38 wt%, mostly more than 5 wt%. Th-rich uraninite may crystallize in equilibrium with uranoan thorite, as demonstrated in phase assemblages in the system SiO₂-UO₂-ThO₂ (Zimmer 1986). Yttrium is invariably present in uraninite, with from 0.4 to 6.33 wt% Y₂O₃, but generally in the range of 1-3 wt% Y₂O₃. The P content is generally much less than 0.1 wt% P₂O₅, and in some cases even below the level of detection. Uraninite contains around 2 wt% PbO (1.85 < PbO < 2.27 wt%).

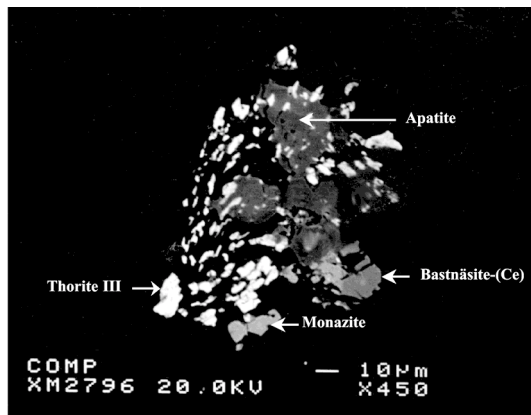


FIG. 9. Back-scattered electron image showing association of thorite III, apatite and bastnäsite-(Ce), derived from the breakdown of monazite.

Pyrochlore-group mineral

A pyrochlore-group mineral is only found in association with thorite in the G–b granite. Its composition is characterized by (wt%): 27.39–31.46 Nb₂O₅, 2.86–5.25 Ta₂O₅, 23.62–30.33 PbO, 2.75–3.37 WO₃, 4.49–5.68 Y₂O₃, 4.95–5.61 (REE)₂O₃, 2.57–2.89 wt% ThO₂ (Table 9). The elevated content of Pb allows us to define it as plumbopyrochlore. This mineral has been recently noted in the Rutherford #2 pegmatite, in Virginia (Lumpkin 1998) and in the Laoshan A-type granite, China (Wang *et al.* 2001). In rare-metal-enriched granites, uranmicrolite may be present where the granite has a high Ta:U ratio, *e.g.*, in the Beauvoir granite (Cuney *et al.* 1992) or in the Yichun granite (Huang *et al.* 2002). In contrast, when Ta/U is low, a pyrochlore-group mineral forms instead, as in the case of the Xihuashan granites (0.27 < Ta/U < 0.57).

Fergusonite-(Y)

Fergusonite-(Y) is the major depository for Nb in the Xihuashan granites. It occurs commonly as discrete crystals in the G–b granite, but also as micro-inclusions in spessartine from the G–b and G–c granites. In both cases, it is ubiquitous in its association with uraninite (Fig. 11). Representative compositions of fergusonite-(Y) are given in Table 10. Analytical data indicate that fergusonite-(Y) from the Xihuashan granite contains as much as 1.2–5.53% WO₃, 1.32–4.19% Ta₂O₅, 0.28–3.17% ThO₂, and 0.55–2.23% UO₂.

Ferberite–hübnerite

Ferberite–hübnerite is observed in the G–a granite and also in the G–b granite. It is included in scheelite in

the G–b granite. That in the G–a granite exhibits a progressive core-to-rim zoning characterized by increasing amounts of Fe together with decreasing amounts of Mn. Nb content reaches up to 1.18 wt% Nb₂O₅ (No. 5, Table 11).

Unidentified Nb–Ta–W mineral

An unidentified niobotantalate mineral (~30 µm in size) has been found in association with uraninite and zircon in the G–b granite (Fig. 12). A representative composition is given in Table 9 (No. 4). Nb, Ta, W, Y, U and Th are major components of this mineral. It contains exceptionally up to 15 wt% WO₃. This mineral is most likely a pyrochlore-group mineral, but because of the lack of X-ray data, this hypothesis cannot be confirmed.

Scheelite

Scheelite appears in the G–b granite, where it is associated with ferberite. It is also found as one of the assemblage of alteration products of yttrian fluorite in the G–c granite (Fig. 13). Apart from low contents of Fe and Mn, the electron-microprobe results correspond well to the stoichiometry of scheelite (Table 11). A high tungsten-carrying capacity in the granitic melt or

TABLE 9. REPRESENTATIVE COMPOSITIONS OF PLUMBOPYROCHLORE AND UNDETERMINED Nb–Ta–W MINERAL FROM THE XIHUASHAN GRANITIC COMPLEX, CHINA

	Plumbo-pyrochlore				Nb-Ta-W phase				
	1	2	3	4	1	2	3	4	
Y ₂ O ₃ wt%	4.49	5.22	5.68	6.52	Y <i>apfu</i>	0.308	0.356	0.428	0.410
La ₂ O ₃	-	-	0.06	0.15	La	0.079	0.043	0.049	0.007
Ce ₂ O ₃	1.76	0.97	1.00	0.13	Ce	0.000	0.000	0.000	0.006
Nd ₂ O ₃	0.47	0.68	0.32	0.20	Nd	0.020	0.030	0.015	0.008
Sm ₂ O ₃	-	-	-	0.40	Sm	0.000	0.000	0.000	0.016
Gd ₂ O ₃	1.05	1.42	1.67	2.12	Gd	0.043	0.057	0.074	0.083
Tb ₂ O ₃	0.10	0.02	0.06	0.27	Tb	0.004	0.001	0.003	0.010
Dy ₂ O ₃	0.34	0.70	0.93	2.38	Dy	0.013	0.027	0.040	0.091
Ho ₂ O ₃	0.14	0.27	0.31	0.95	Ho	0.005	0.010	0.013	0.036
Er ₂ O ₃	0.03	0.22	-	2.90	Er	0.001	0.009	0.000	0.108
Tm ₂ O ₃	0.18	-	-	0.56	Tm	0.007	0.000	0.000	0.021
Yb ₂ O ₃	0.83	1.11	1.11	2.78	Yb	0.031	0.041	0.046	0.100
Lu ₂ O ₃	0.05	0.21	0.15	-	Lu	0.002	0.008	0.006	-
ThO ₂	2.64	2.57	2.89	1.26	Th	0.074	0.071	0.088	0.034
UO ₂	6.03	5.65	7.80	7.23	U	0.165	0.153	0.233	0.190
CaO	0.41	0.35	0.48	0.42	Ca	0.054	0.045	0.069	0.053
FeO	6.05	5.56	10.90	5.09	Fe	0.623	0.565	1.222	0.503
MnO	0.01	0.05	0.05	1.34	Mn	0.001	0.005	0.006	0.134
PbO	30.33	29.08	23.62	0.50	Pb	1.006	0.951	0.853	0.016
SiO ₂	3.81	3.36	3.76	0.76	Si	0.469	0.408	0.504	0.090
P ₂ O ₅	0.07	0.05	0.10	-	P	0.007	0.005	0.011	-
Nb ₂ O ₅	31.46	31.92	27.39	26.13	Total	2.912	2.785	3.66	1.916
Ta ₂ O ₅	3.54	2.86	5.25	18.57					
TiO ₂	0.45	0.51	0.40	0.09	Nb	1.752	1.753	1.660	1.395
WO ₃	2.75	3.37	3.11	14.22	Ta	0.118	0.094	0.191	0.597
F	1.25	0.21	0.61	0.08	Ti	0.000	0.000	0.003	0.008
-F=O	0.52	0.09	0.26	0.03	W	0.485	0.081	0.259	0.435
Total	97.68	96.26	97.38	95.02	F	0.042	0.046	0.040	0.030

Structural formula are calculated on the basis of Nb + Ta + Ti + W = 2 atoms per formula unit (*apfu*). -; below detection limits. Electron-microprobe data.

TABLE 10. REPRESENTATIVE COMPOSITIONS OF FERGUSONITE-(Y) FROM THE G-b AND G-c UNITS OF THE XIHUASHAN GRANITIC COMPLEX, CHINA

	G-b						G-c			G-b						G-c			
	1	2	3	4	5	6	7	8	9	1	2	3	4	5	6	7	8	9	
Nb ₂ O ₅ wt%	42.20	42.36	41.60	41.14	39.95	41.81	42.05	40.25	41.20	Nb <i>apfu</i>	0.890	0.881	0.868	0.876	0.841	0.886	0.872	0.856	0.861
Ta ₂ O ₅	4.05	3.93	4.19	1.57	1.64	2.52	2.93	1.32	1.61	Ta	0.051	0.049	0.053	0.020	0.021	0.032	0.037	0.017	0.020
Y ₂ O ₃	27.07	27.31	27.65	31.54	32.83	27.98	26.68	28.51	29.75	Y	0.672	0.668	0.679	0.790	0.813	0.698	0.651	0.714	0.732
La ₂ O ₃	-	0.11	0.02	-	-	0.05	0.03	-	-	La	-	0.002	0.000	-	-	0.001	0.001	-	-
Ce ₂ O ₃	0.57	-	0.02	0.52	0.53	0.63	0.73	0.80	0.96	Ce	0.010	0.000	0.000	0.009	0.009	0.011	0.012	0.014	0.016
Nd ₂ O ₃	0.14	-	0.57	0.74	0.79	1.75	0.67	1.07	0.85	Nd	0.002	0.000	0.009	0.012	0.013	0.029	0.011	0.018	0.014
Sm ₂ O ₃	0.09	1.15	0.37	0.65	0.56	0.97	0.12	0.25	0.29	Sm	0.001	0.018	0.006	0.011	0.009	0.016	0.002	0.004	0.005
Gd ₂ O ₃	4.41	3.15	3.82	3.04	3.00	5.17	4.79	3.73	4.12	Gd	0.068	0.048	0.058	0.047	0.046	0.080	0.073	0.058	0.063
Tb ₂ O ₃	0.16	0.78	0.64	0.30	0.32	0.38	0.37	0.37	0.33	Tb	0.002	0.012	0.010	0.005	0.005	0.006	0.006	0.006	0.005
Dy ₂ O ₃	3.55	4.66	4.77	3.66	3.61	5.86	5.21	3.19	3.30	Dy	0.053	0.069	0.071	0.056	0.054	0.088	0.077	0.048	0.049
Ho ₂ O ₃	0.68	1.09	0.85	0.76	0.83	0.87	0.76	0.87	0.97	Ho	0.010	0.016	0.012	0.011	0.012	0.013	0.011	0.013	0.014
Er ₂ O ₃	3.68	3.48	3.79	3.24	3.50	2.10	3.39	3.99	2.70	Er	0.054	0.050	0.055	0.048	0.051	0.031	0.049	0.059	0.039
Tm ₂ O ₃	0.24	0.41	0.51	0.31	0.39	0.45	0.54	0.38	0.65	Tm	0.003	0.006	0.007	0.005	0.006	0.007	0.008	0.006	0.009
Yb ₂ O ₃	2.17	2.44	2.70	2.08	2.00	2.28	2.33	2.87	2.32	Yb	0.031	0.034	0.038	0.030	0.028	0.033	0.033	0.041	0.033
Lu ₂ O ₃	0.93	0.93	0.43	0.72	0.49	0.89	0.23	0.35	0.42	Lu	0.013	0.013	0.006	0.010	0.007	0.013	0.003	0.005	0.006
ThO ₂	2.14	0.64	0.29	2.47	2.63	1.36	0.89	3.17	0.52	Th	0.023	0.007	0.003	0.026	0.028	0.014	0.009	0.034	0.005
UO ₂	1.14	2.23	2.04	1.22	1.30	0.78	0.55	1.61	0.59	U	0.012	0.023	0.021	0.013	0.013	0.008	0.006	0.017	0.006
CaO	0.55	0.42	0.31	0.93	0.63	0.42	0.44	0.50	0.35	Ca	0.027	0.021	0.015	0.047	0.031	0.021	0.022	0.025	0.017
FeO	0.17	-	-	0.69	0.63	0.44	1.02	0.64	1.28	Fe	0.007	0.000	0.000	0.027	0.025	0.017	0.039	0.025	0.050
MnO	0.02	0.03	0.04	0.12	0.14	0.04	1.86	0.95	1.74	Mn	0.001	0.001	0.002	0.005	0.005	0.002	0.072	0.038	0.068
PbO	0.50	0.44	0.62	0.67	0.77	0.64	0.56	0.66	0.58	Pb	0.006	0.005	0.008	0.009	0.010	0.008	0.007	0.008	0.007
P ₂ O ₅	0.09	-	-	0.01	0.09	0.03	0.01	0.09	-	P	0.004	0.000	0.000	0.001	0.004	0.001	0.000	0.004	-
WO ₃	4.01	5.14	5.53	1.20	2.87	1.98	4.31	3.58	4.24	W	0.048	0.061	0.066	0.015	0.035	0.024	0.051	0.044	0.051
Total	98.56	100.70	100.76	97.57	99.47	99.40	100.48	99.14	98.74	Total	1.988	1.984	1.987	2.073	2.066	2.039	2.052	2.054	2.07

Structural formula calculated on the basis of O = 4 atoms per formula unit (*apfu*). -: below detection limits. Electron-microprobe data.

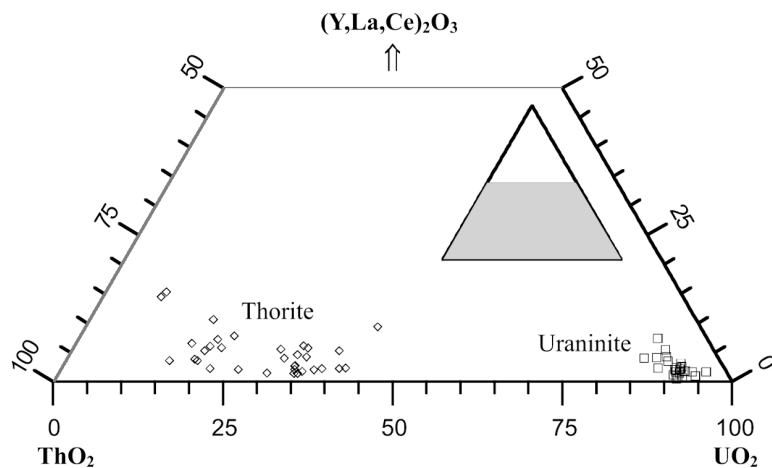


FIG. 10. Plot of compositions of thorite and uraninite in the triangular diagram ThO₂ – UO₂ – (Y,La,Ce)₂O₃ (wt%).

postmagmatic fluid at Xihuashan is also indicated by the presence of W-bearing niobotantalates and associated ferberite–hübnerite.

DISCUSSION

From trace-element and isotopic data, Maruéjól *et al.* (1990) have suggested that the different units of the Xihuashan complex appear to be cogenetic, and further

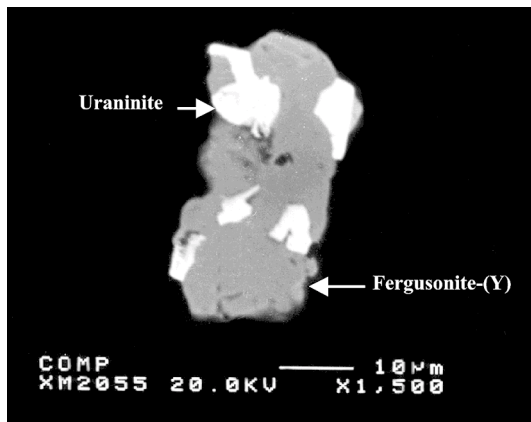


FIG. 11. Intergrowth of fergusonite-(Y) and uraninite. Back-scattered electron image.

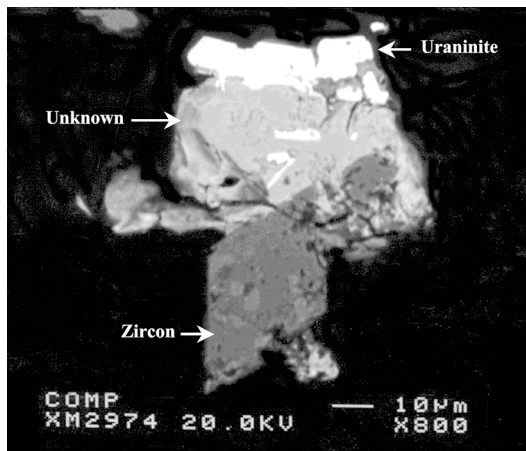


FIG. 12. Unidentified niobotantalate. The back-scattered electron image illustrates its intimate association with zircon and uraninite.

pointed out a magmatic evolution from the G-a granite to the G-b and G-c granites by simple fractionation, with a later overprint of hydrothermal alteration in the G-b granite. In this study, we give mineralogical evidence for the magmatic-hydrothermal evolution of the Xihuashan granitic complex.

The accessory-mineral associations of the Xihuashan granitic complex are summarized in Table 12. Early accessory phases crystallizing at the magmatic stage are principally yttrian fluorite, spessartine, zircon, monazite-(Ce), xenotime-(Y), fergusonite-(Y), gadolinite-(Y), uraninite and uranoan thorite. However, these minerals experienced further evolution at the late-magmatic and hydrothermal stage.

The G-a granite is mostly composed of quartz, feldspars and biotite. The REE possess very low partition coefficients for these minerals. Monazite-(Ce) is a stable mineral phase that fractionates very early because of its very low solubility in peraluminous melts, which leads to a magma rapidly depleted in LREE, but enriched in HREE (Rapp & Watson 1986, Wark & Miller 1993). Crystallization of abundant monazite in the G-a granite caused the depletion in LREE in the G-b and G-c granites, as demonstrated by the scarcity of monazite-(Ce). This mineral may be surrounded by bastnäsite-(Ce), thorite and fluorapatite in the G-b and G-c granites. The work of Ayers & Watson (1991) and of Williams-Jones & Wood (1992) indicates that monazite is significantly soluble at very low pH and high fluoride activity, and the stability of bastnäsite increases with increasing activity of CO_3^{2-} . Thus, the textural relationship between monazite-(Ce) and bastnäsite-(Ce) may be attributed to hydrothermal alteration of magmatic Th-bearing monazite in presence of fluid enriched in F and CO_2 via the following reaction:

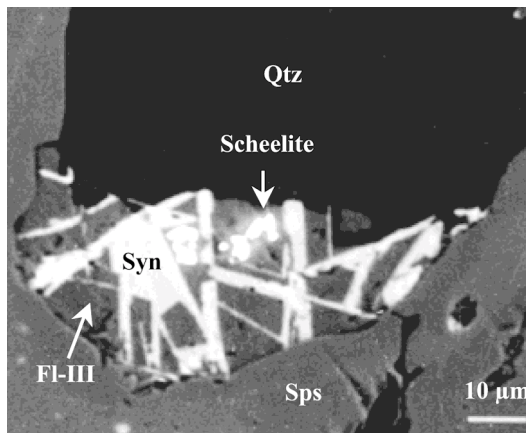
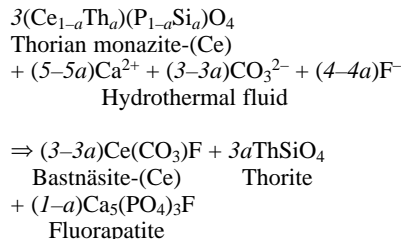


FIG. 13. Back-scattered electron image showing small crystals of scheelite associated with fluorite III (Fl-III) and synchysite-(Y) (Syn). Qtz: quartz, Sps: spessartine.



In peraluminous magmas, Zr, REE, Th, U, Y, P possess a very high solubility, and therefore have strong

TABLE 11. COMPOSITION OF TUNGSTEN MINERALS FROM THE XIHUASHAN GRANITIC COMPLEX, CHINA

	Scheelite				Ferberite-hübnerite								
	1	2	3	4	5	6	7	8C	8R	9C	9R	10	11
Nb ₂ O ₅ wt%	-	-	0.02	-	1.18	0.61	1.01	0.22	0.51	0.18	0.38	0.58	0.51
Ta ₂ O ₅	0.02	0.03	-	0.05	0.16	0.15	0.30	0.11	0.13	-	0.11	-	0.11
W ₂ O ₆	80.52	80.59	80.53	80.30	74.64	75.09	74.60	75.66	75.88	77.06	75.20	75.93	76.12
FeO	0.17	0.26	0.09	0.03	14.25	5.73	10.74	15.22	16.18	15.71	16.44	15.58	15.57
MnO	0.21	0.26	0.09	0.07	9.22	17.64	12.48	7.62	6.83	7.93	6.95	8.10	7.89
CaO	19.31	19.23	18.75	18.82	0.02	0.04	0.03	-	-	-	0.04	0.05	0.07
Total	100.22	100.37	99.48	99.27	99.46	99.26	99.16	98.82	99.53	100.89	99.12	100.24	100.26
Nb <i>apfu</i>	-	-	0.000	-	0.027	0.014	0.023	0.005	0.012	0.004	0.009	0.013	0.012
Ta	0.000	0.000	-	0.001	0.002	0.002	0.004	0.002	0.002	-	0.002	-	0.001
W	0.998	0.998	1.007	1.006	0.977	0.986	0.980	1.001	0.996	0.999	0.991	0.988	0.991
Fe	0.007	0.010	0.003	0.001	0.602	0.243	0.455	0.650	0.686	0.657	0.699	0.655	0.654
Mn	0.009	0.011	0.004	0.003	0.394	0.757	0.536	0.330	0.293	0.336	0.299	0.344	0.336
Ca	0.990	0.985	0.970	0.975	0.001	0.002	0.002	-	-	-	0.002	0.003	0.004
Total	2.004	2.004	1.984	1.986	1.976	1.99	1.977	1.983	1.977	1.992	1.993	1.99	1.986
Fbr	0.66	1.04	0.35	0.14	60.35	24.24	45.87	66.35	70.06	66.16	69.88	65.34	65.85
Hbn	0.85	1.05	0.39	0.30	39.54	75.53	53.96	33.65	29.94	33.84	29.92	34.38	33.79
Scl	98.49	97.91	99.25	99.57	0.11	0.24	0.17	0.00	0.00	0.00	0.20	0.28	0.36

C: core, R: rim; 10 and 11: inclusions in scheelite. Structural formula calculated on the basis of O = 4 atoms per formula unit (*apfu*). All Fe is expressed as Fe²⁺; - below detection limits. Symbols: Fbr: ferberite, Hbn: hübnerite, Scl: scheelite. Electron-microprobe data.

TABLE 12. ASSOCIATIONS OF ACCESSORY MINERALS IN THE XIHUASHAN GRANITIC COMPLEX, CHINA

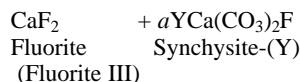
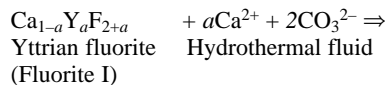
Unit	G-a	G-b	G-c
Fluorite I		+	+
Fluorite II		+	+
Fluorite III		+	+
Spessartine		+	+
Gadolinite-(Y)		+	+
Zircon	+	+	+
Thorite I		+	+
Thorite II	+	+	+
Xenotime-(Y) I		+	+
Xenotime-(Y) II	+	+	+
Monazite-(Ce)	+	+(?)	+(?)
Synchysite-(Y)		+	+
Bastnäsite-(Ce)		+	+
Uraninite		+	+
Fergusonite-(Y)		+	+
Pyrochlore (<i>s.l.</i>)		+	+
Ferberite-hübnerite	+	+	+
W-Nb-Ta mineral		+	+
Scheelite		+	+

lithophile behavior (Linnen 1998). Thus large amounts of a wide variety of accessory minerals crystallize more or less simultaneously in most differentiated magma, as in the case of the G-b and G-c granites at Xihuashan. Magmatic zircon crystallized with minute inclusions of xenotime and thorite, which are commonly associated. These phases, with fergusonite-(Y) and gadolinite-(Y), may also occur as inclusions in spessartine. Apart from

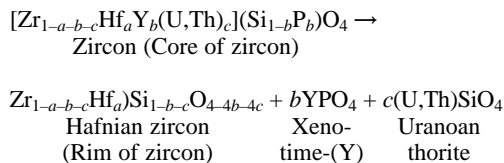
uranian thorite, uranium formed significant amounts of uraninite. Commonly, the elevated content of Th determined in uraninite is attributed to equilibrium crystallization of uranoan thorite.

In addition to minute inclusions of Y-bearing minerals, spessartine contains a striking amount of Y and HREE in its structure, particularly in its central area. This fact suggests that the melt was enriched in Y and HREE at an early stage during crystallization of the G-b and G-c granites, as a result of LREE depletion by crystallization of monazite in the G-a granite. The consumption of Y and HREE leads to a decrease in these elements in the melt, the crystal being impoverished in (or even free of) Y and HREE at the rim.

Fluorite I (yttrian fluorite) is one of the earliest depositories of Y and REE in the G-b and G-c granites; it contains up to 15 wt% Y₂O₃. However, this phase seems to be unstable in the later fluid-rich environment. Evidence for subsequent alteration of yttrian fluorite is seen as a peripheral association of Y-poor fluorite III and synchysite-(Y) in the G-b and G-c granites (Figs. 2a, c). This process may be possibly described by the following reaction:



Zircon is a common accessory mineral present throughout the Xihuashan granites, but it varies in composition. In the G–a granite, zircon contains on average 2.5 wt% HfO₂, 0.38% UO₂, 0.1% ThO₂ and 0.49% Y₂O₃, and does not exhibit any textural evidence of zoning. Zircon from the G–b and G–c granites exhibits zoning, represented by the presence of polyminerally micro-inclusions of xenotime-(Y), thorite and uraninite in the core, and enrichment in Hf at the rim. Significant mobility of Hf was postulated as resulting from hydrothermal transfer induced by melt degassing (Raimbault *et al.* 1995). This phenomenon, already cited in studies of other granites (Suzhou: Wang *et al.* 1996, Laoshan: Wang *et al.* 2000, 2001), may be described as follows:



In summary, observations based on accessory minerals point to a magmatic environment for the G–a granite. In contrast, the G–b and G–c granites experienced a degassing process during their crystallization. Primary minerals such as yttrian fluorite and monazite-(Ce) may have broken down to secondary phases in fluid-rich environments. This type of late- to postmagmatic fluid must have been enriched in F and CO₂, and derived from the final oversaturation of the granitic melt (Maruéjol *et al.* 1990). On the basis of results from mine geology, the mineralization in tungsten at Xihuashan is considered to be related to the granites G–b and G–c (Hu *et al.* 1984, Li *et al.* 1986). Similarities regarding the nature of the ore-forming fluids in the Xihuashan tungsten deposit have been investigated by Giuliani *et al.* (1988). These authors have identified the earliest fluid phases as CO₂-bearing aqueous solutions circulating at temperature up to 420°C. The CO₂-rich fluids played important role in the transport of the tungsten. The formation of the extensive tungsten mineralization in Xihuashan may be therefore interpreted to be caused by a stable and long-lasting flow of hydrothermal fluid through the granites.

Finally, we conclude that the pattern of evolution of Y, REE, Zr, U, Th and Nb in the Xihuashan pluton was a complex, multi-stage process, and involved primary magmatic crystallization and late-stage hydrothermal alteration.

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