

**Crystal Data:** Orthorhombic. *Point Group:*  $2/m\ 2/m\ 2/m$ . Tiny crystals and microcrystalline clusters, in earthy masses.

**Physical Properties:** Hardness = n.d.  $D(\text{meas.}) = 2.2$   $D(\text{calc.}) = 2.134$

**Optical Properties:** Semitransparent. *Color:* White to yellow, yellow-orange.  
*Optical Class:* Biaxial (-) (synthetic). *Orientation:*  $Y = b$ ;  $Z \wedge a = 6.8(1)^\circ$  [*sic*].  $\alpha = 1.494$   
 $\beta = 1.512$   $\gamma = 1.524$   $2V(\text{meas.}) = 66.8(7)^\circ$   $2V(\text{calc.}) = 77.4^\circ$

**Cell Data:** *Space Group:*  $Pna2_1$  (synthetic).  $a = 12.015(2)$   $b = 6.518(1)$   $c = 11.173(1)$   
 $Z = 4$

**X-ray Powder Pattern:** Synthetic.  
6.00 (100), 5.29 (92), 2.897 (65), 2.999 (45), 2.648 (26), 2.531 (18), 2.207 (18)

**Chemistry:** (1) Larderello, Italy; identity depends on the chemical analysis of a mixture with biringuccite, and the correspondence of lines in the mixture's X-ray powder pattern with those of synthetic material.

**Occurrence:** As scales on piping in a geothermal field.

**Association:** Biringuccite, thénardite, orpiment, quartz.

**Distribution:** From Larderello, Val di Cecina, Tuscany, Italy.

**Name:** To honor Raffaello Nasini (1854–1931), Italian chemist.

**Type Material:** University of Florence, Florence, Italy, 16803/G; National Museum of Natural History, Washington, D.C., USA, 163785.

**References:** (1) Cipriani, C. and P. Vannuccini (1961) Hoferite [= biringuccite] e nasinite: due nuovi borati fra i prodotti di Lardarello. Pt. I. Atti Rend. Accad. Lincei, 30, 74–83; Pt. II. 235–245 (in Italian). (2) (1963) Amer. Mineral., 48, 709–711 (abs. ref. 1). (3) Corazza, E., Menchetti, S., C. Sabelli, and A. Stoppioni (1977) Hydrothermal synthesis at 150 °C and X-ray study of resulting products in the  $\text{NaOH}-\text{B}_2\text{O}_3-\text{H}_2\text{O}$  system. Neues Jahrb. Mineral., Abh., 131, 208–223. (4) Corazza, E., S. Menchetti, and C. Sabelli (1975) The crystal structure of nasinite,  $\text{Na}_2[\text{B}_5\text{O}_8(\text{OH})] \cdot 2\text{H}_2\text{O}$ . Acta Cryst., 31, 2405–2410.