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Crystal data for $Cd(NO_3)_2 \cdot 2H_2O$ and $Ca(NO_3)_2 \cdot 2H_2O$

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Both crystals were prepared by adding an equal volume of fuming nitric acid to a saturated aqueous solution of the nitrate and seeding the solutions with a crystal of the anhydrous salt. The cell parameters were obtained with a SINTEX $P\bar{1}$ automatic four-circle x-ray diffractometer using Mo$K\alpha$ radiation ($\lambda = 0.7107$ Å). The crystallographic data are given in Table 1 along with the other dihydrate nitrates of bivalent metals. The density was determined by flotation method. Futher work on the complete determination of the crystal structures of $Cd(NO_3)_2 \cdot 2H_2O$ and $Ca(NO_3)_2 \cdot 2H_2O$ are in progress.

Table 1. *Crystal data*

	$Cd(NO_3)_2 \cdot 2H_2O$	$Ca(NO_3)_2 \cdot 2H_2O$	$Zn(NO_3)_2 \cdot 2H_2O$	$Mg(NO_3)_2 \cdot 2H_2O$
a	5.972 (2)	11.838 (4)	5.754 (6)	5.81 (1) Å
b	8.996 (4)	16.510 (5)	5.978 (5)	5.99 (1) Å
c	12.129 (6)	12.600 (3)	8.557 (5)	8.65 (1) Å
β	102.19 (4)		91.0 (3)	90.7 (2) °
Z	4	16	2	2
D_m	2.72	2.12	2.50	1.95 g/cm³
D_x	2.84	2.14	2.54	2.02 g/cm³
Space group	$P2_1/c$	$Ccca$	$P2_1/c$	$P2_1/c$
Reference	This work	This work	Ribár *et al.* (1969)	Ribár *et al.* (1973)

References

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