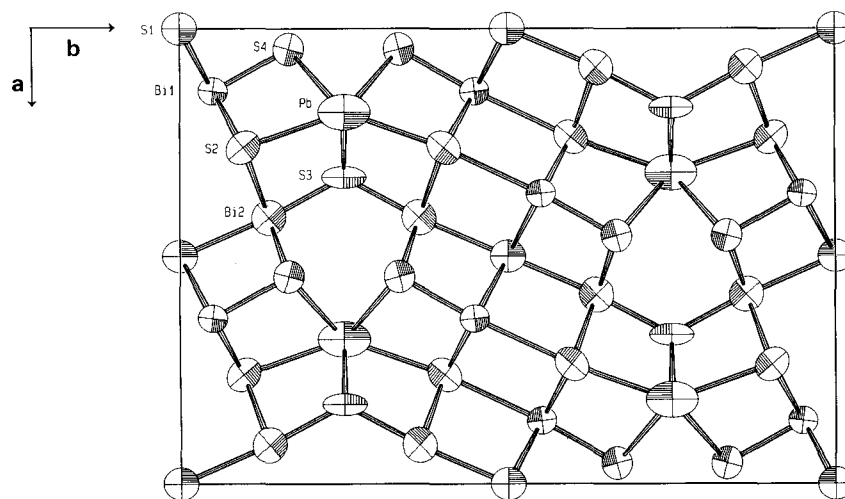


Crystal structure of lead silver tribismuth sulfide, $\text{PbAgBi}_3\text{S}_6$

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Synthesis: The sulfide was prepared by recrystallization in silica tubes at 770 K. Strong differences between weighted R-value and unit weighted R-value are caused by very small crystals and few observed reflections. Therefore the residual electron density was also calculated based on the observed reflections only. Natural gustavite show Ag-Bi-ordering (Steins et al., 1991), whereas synthetic gustavite show statistical distribution.

Orthorhombic, Cmc m (No. 63), $a = 4.077(2)$, $b = 13.477(7)$, $c = 19.88(2)$ Å, $V = 1092.3$ Å 3 , $Z = 4$, $R = 0.076$.

Table 1. Parameters used for the X-ray data collection

Diffractometer type:	Siemens-Stoe	Number of unique reflections:	1376
Wave length:	Mo K α radiation (0.7107 Å)	Criterion for unobserved reflections:	$F_o < 1.5\sigma(F_o)$
Crystal characteristics:	lamina, size 0.08 × 0.032 × 0.0005 mm	Number of refined parameters:	39
Temperature of measurement:	288 K	Scan mode:	ω/θ
$2\theta_{\max}$:	70°	μ :	670.315 cm $^{-1}$
		Structure solution program used:	ORXFLS3

Table 2. Final atomic coordinates and displacement parameters (in Å 2)

Atom	x	y	z	U $_{iso}$ /U $_{11}$	U $_{22}$	U $_{33}$	U $_{12}$	U $_{13}$	U $_{23}$
Pb	0.0	0.1835(2)	0.25	0.025(1)	0.031(2)	0.070(3)	0.0	0.0	0.0
Bi(1)	0.5	0.1377(1)	0.0502(1)	0.0188(7)	0.0186(8)	0.022(1)	0.0	0.0008(9)	0.0
Bi(2)	0.5	0.4150(2)	0.3645(1)	0.023(1)	0.030(1)	0.030(2)	0.0	0.003(1)	0.0
S(1)	0.0	0.0	0.0	0.016(6)	0.026(9)	0.03(1)	0.0	0.0	0.0
S(2)	0.0	0.2612(9)	0.0968(7)	0.023(4)	0.028(6)	0.030(6)	0.0	-0.011(6)	0.0
S(3)	0.5	0.327(1)	0.25	0.029(8)	0.012(7)	0.05(1)	0.0	0.0	0.0
S(4)	0.5	0.0450(9)	0.1659(6)	0.018(5)	0.026(6)	0.024(7)	0.0	-0.001(5)	0.0

Further details of the structure determination (e.g. structure factors) have been deposited within the relevant database and can be accessed as Collection No. 300265 or ordered from the Fachinformationszentrum Karlsruhe, D-7514 Eggenstein-Leopoldshafen.

References:

1. Karup-Moller, S.: Gustavite, a new sulphosalt mineral from greenland. *Can. Min.* **10** (1970) 173–190.
2. Harris, D.C., Chen, T.T.: Gustavite: two canadian occurrences. *Can. Min.* **13** (1975) 411–414.
3. Steins, M., Bente, K., Engel, M., Meier-Salimi, M.: Metallsubstitution und -ordnung in Sulfosalzen am Beispiel von Boulangerit (Pb $_5$ Sb $_4$ S $_{11}$) und Gustavit (PbBi $_3$ AgS $_6$). *Eur. J. Mineral.* **3** (1991) 258.
4. Engel, M.: Strukturuntersuchungen zur Metallverteilung in Sulfosalzen am Beispiel synthetischer Gustavite. Master thesis Göttingen (1992).