Clinotobermorite, $Ca_5Si_6(O,OH)_{18}$ ·5H₂O, a new mineral from Fuka, Okayama Prefecture, Japan

C. Henmi

Department of Earth Sciences, Faculty of Science, Okayama University, Okayama 700 Japan

AND

I. KUSACHI

Department of Earth Sciences, Faculty of Education, Okayama University, Okayama 700 Japan

Abstract

Clinotobermorite, $Ca_5Si_6(O,OH)_{18}$ ·5H₂O, has been found as a vein-forming mineral in gehlenitespurrite skarns at Fuka, Okayama Prefecture. It is associated with tobermorite, plombierite, apophyllite, and calcite. The clinotobermorite is colourless or white and occurs as tabular or acicular crystals. It is monoclinic with the space group *Cc* or *C2/c*. The unit cell dimensions are *a* 11.331, *b* 7.353, *c* 22.67 Å, β 96.59°. Microtwinning and stacking disorder on (001) are observed. On heating the clinotobermorite at 300 °C, the 002 spacing is reduced from 11.3 to 9.3 Å. Its refractive indices are α 1.575, β 1.580, γ 1.585, and the density 2.58 g/cm³ (meas.), 2.69 g/cm³ (calc). The Moh's hardness is 4.5. Calculation of the analytical data on the basis of six tetrahedral cations shows that this mineral has a simplified chemical formula Ca_{5.3}Si₆(O,OH,F)₁₈·5H₂O. The chemical composition and the unit cell are closely related to those of tobermorite. It is most likely that clinotobermorite is a low-temperature polymorph of tobermorite.

KEYWORDS: clinotobermorite, new mineral, tobermorite, Okayama Prefecture, Japan.

Introduction

TOBERMORITE (11.3 Å tobermorite), plombierite (14 Å tobermorite), tacharanite (12.6 Å tobermorite), riversideite (9.3 Å tobermorite) and oyelite (10 Å tobermorite) have been known as tobermorite group minerals (McConnell, 1954; Heller and Taylor, 1956; and Kusachi *et al.*, 1984).

Several calcium silicate hydrate minerals occur in gehlentie–spurrite skarns at Fuka, Okayama. Three members of the tobermorite group minerals have been reported for this locality; these are tobermorite and plombierite (Mitsuda *et al.*, 1972) and oyelite (Kusachi *et al.*, 1984). During a mineralogical study of tobermorite, we found that some samples which were considered to consist of tobermorite crystals show an X-ray powder diffraction pattern distinctly different from that of usual tobermorite. This mineral is not orthorhombic but monoclinic, and has a chemical composition almost identical to that of tobermorite. Our study revealed that it is a monoclinic polymorph of tobermorite.

We have named this new mineral clinotober-

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morite. The mineral species and its name have been approved by the Commission on New Minerals and Mineral Names, International Mineralogical Association, in May 1990. The type material of clinotobermorite is deposited at the National Science Museum, Tokyo, Japan.

The present paper deals with the mineralogical properties and mode of occurrence of the clinotobermorite found at Fuka. This mineral has been previously reported as 'monoclinic tobermorite' by the authors in an article in Japanese (Henmi and Kusachi, 1989).

Occurrence

Skarns were formed on both sides of the quartz monzonite dykes that penetrate limestones at Fuka. The skarns consist mainly of gehlenite and spurrite. Some parts of the skarns were retrogressively altered and were cut by numerous vcins consisting of such minerals as tobermorite (Mitsuda *et al.*, 1972), oyelite, bultfonteinite, scawtite, xonotlite (Kusachi *et al.*, 1984), afwillite and jennite (Kusachi *et al.*, 1989).

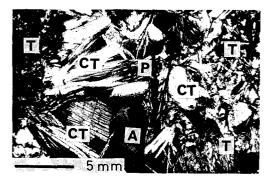


FIG. 1. Photomicrograph of clinotobermorite. Crossed nicols. Abbreviations: T, tobermorite; CT, clinotobermorite; P, plombierite; and A, apophyllite.

Clinotobermorite occurs in veins filling fissures of the contaminated rock which consists mainly of feldspar and pyroxene and may be considered to be a kind of endoskarn. The thickness of the vein varies from one to two centimetres. In these veins, minerals develop from the wall to the centre of the vein in the following order: calcite, tobermorite, clinotobermorite, plombierite and apophyllite. Fig. 1 shows the occurrence of clinotobermorite in one of these veins.

Physical and optical properties

Clinotobermorite occurs as tabular crystals parallel to the c-face with up to 5 mm width and as aggregates of acicular crystals parallel to the baxis up to 2 mm in length. The mineral is colourless or white with a vitreous luster in handspecimen and colourless in thin section. Cleavage is perfect on (001) and poor on (100). The mineral is optically biaxial with refractive indices $\alpha = 1.575$, $\beta = 1.580$ and $\gamma = 1.585$. We were unable to obtain its optical axial angles (2V) because of well-developed microtwinning. The 2V angle calculated from the refractive indices is 89.8°. Vickers microhardness is 251-174 kg/mm² (25 g load) and the Mohs hardness is 4.5. The density measured with heavy liquids is 2.58 g/cm³. The calculated density is 2.69 g/cm³.

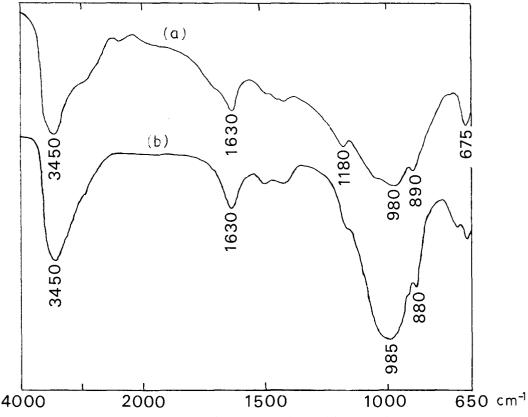


FIG. 2. Infrared spectra for (a) clinotobermorite and (b) tobermorite from Fuka.

	clino- tobermorite	tobermorite			plombierite	oyelite	
	1	1	2	3	1	3	
sio ₂	46.55	48.98	45.12	46.7	43.34	35.3	
ті02	0.01	0.03	-	0.0	-	0.0	
A1203	0.36	1.03	3.31	2.2	0.39	0.3	
^B 2 ⁰ 3	0.23	0.03	-	0.0	-	4.8	
Fe_20_3	0.01	0.03	0.23	0.0	-	0.0	
MnO	0.06	0.03	-	0.0	-	0.0	
MgO	0.11	0.04	0.04	0.0	0.00	0.0	
Ca0	39.04	37.90	35.24	37.3	34.64	41.2	
Na ₂ 0	0.02	0.01	0.24	0.1	0.05	0.1	
к ₂ 0	0.10	0.02	0.23	0.1	0.32	0.0	
11 ₂ 0	13.75	11.46	15.52	13.2	19.69	16.7	
F	0.18	0.00	-	-	-	-	
- 0 = F	0.08	0.00	-	-	-	-	
Total	100.34	99.56	99.93	99.6	99.51*	99.5**	

Table 1. Chemical analyses of tobermorite group minerals from $\ensuremath{\mathrm{Fuka}}$.

1. Present study.

* includes H₂O(-) 1.08 %.

2. Mitsuda and Taylor (1978). ** includes $\rm H_{2}O(\mbox{--})$ 0.7% and $\rm CO_{2}$ 0.4%.

3. Kusachi et al. (1980 and 1984)

Clinotobermorite loses molecular water on heating at 300 °C in air, resulting in shrinkage of the d_{002} reflection from 11.3 to 9.3 Å.

The IR spectra of clinotobermorite are measured using the conventional KBr method for the region 4000 to 650 cm^{-1} . Fig. 2 shows the spectra of clinotobermorite and tobermorite for comparison. They are very similar. The absorption bands at 3450 cm^{-1} are attributed to the OH stretching vibration and the weak bands at 1630 cm^{-1} to the OH bending vibration. Numer-

ous bands below 1200 cm^{-1} are characteristic of both tobermorite (Mitsuda and Taylor, 1978) and clinotobermorite.

Chemical composition

The chemical composition of clinotobermorite was obtained using an electron microprobe analyser except for F, B and H_2O . The concentrations of F, B and H_2O were determined by

Table 2. Unit cell dimensions of tobermorite group minerals from Fuka.

_	clinotobermorite	tobermorite	plombierite	oyelite*
a(A)	11.331(9)	11.233(3)	11.250(6)	11.25(2)
b(A)	7.353(7)	7.372(3)	7.344(4)	7.25(2)
e(A)	22.67(2)	22.56(1)	27.99(1)	20.46(3)
β(°)	96.59(7)			

*Kusachi et al. (1980 and 1984).

Table 3. X-ray powder diffraction data for clinotobermolte and tobermorite from Fuka. The calculated values are based on a=11.331, b=7.353, c=22.67 Å and β =96.59°. After Henmi and Kusachi (1989).

clinotobermorite			tobermorite						
h	k 1	d(obs.)	1	d(calc.)	d(obs.)	Ι	h	k	1
$0 \\ 0 \\ 2 \\ 1$	$\begin{pmatrix} 0 & 2 \\ 0 & 4 \\ 0 & 0 \\ 1 & -2 \end{pmatrix}$	11.25 5.60	100 20	$11.262 \\ 5.631 \\ 5.628 \\ 5.550$	$\substack{11.3\\5.63}$	100 9	$(\begin{array}{c} 0 \\ 0 \\ 2 \end{array})$	0 0 0	2 4 0
2 1 2 2 0 1	$ \begin{pmatrix} 0 & -2 \\ 1 & 2 \\ 0 & 2 \\ 0 & 4 \\ 0 & 6 \\ 1 & -5 \end{pmatrix} $	5.27 4.782	20 10	5.283 5.264 4.818 3.770 3.754 3.749	5.45	38	2	0	1
0	$\begin{pmatrix} 0 & 4 \\ 0 & 6 \\ 1 & 6 \end{pmatrix}$	3.745	36	3.770	3.754	7	0	0	6
		3.304	51		$3.642 \\ 3.514 \\ 3.309$	$30 \\ 16$	$\begin{array}{c} 1 \\ 2 \\ 0 \end{array}$	$^{1}_{0}_{2}$	5 5 3
02223	$ \begin{array}{ccc} 2 & 3 \\ 2 & -1 \\ 2 & 1 \\ 1 & -4 \end{array} $	$3.068 \\ 3.034 \\ 3.012 \\ 2.950$	45 60 37 25	$3.302 \\ 3.079 \\ 3.024 \\ 3.012$	3.080	92	2	2	0
3232	$\begin{pmatrix} 1 & -6 \\ 1 & -3 \\ 2 & -3 \end{pmatrix}$	2.950	25 23	2.970 2.944 2.913	2.972	70	2	2	2
2022323204421	$\begin{array}{c} 0 & -6 \\ 2 & -1 \\ 2 & -1 \\ 1 & -4 \\ 0 & 3 \\ 2 & -8 \\ 0 & -2 \\ 1 \\ 1 \\ \end{array}$	2.811 2.794	41 60	3,303 3,302 3,079 3,024 2,970 2,944 2,913 2,814 2,807 2,787 2,781	2.806	68b	(0	8 0
1 2 4	$ \begin{bmatrix} 1 & 7 \\ 0 & -8 \\ 0 & -4 \end{bmatrix} $	2.641	27	2.781 2.642 2.641	2.723	6	4	0	2
		2.495	11		2.513	20	4	0	4
$ \begin{array}{c} 1 \\ 2 \\ 4 \\ 1 \\ 4 \\ 3 \end{array} $	$ \begin{array}{ccc} 1 & 8 \\ 2 & 7 \\ 0 & 8 \\ 0 & 4 \\ 3 & 0 \\ 3 & -1 \\ 0 & -6 \\ 1 & 6 \end{array} $	2.400 2.414 2.394 2.380	12 20 15	$\begin{array}{c} 2.502\\ 2.421\\ 2.410\\ 2.395\\ 2.388\\ 2.388\\ 2.387\\ 2.378\end{array}$	2.424	9	0	2	7
0		2.000	10		2.282	18b	(]	$ \begin{array}{c} 3 \\ 1 \\ 0 \end{array} $	10^{3}
$^{0}_{4}_{4}_{1}$	$\begin{pmatrix} 0 & 10 \\ 2 & -1 \\ 2 & -3 \\ 3 & 4 \end{pmatrix}$	$2.248 \\ 2.190$	$27 \\ 13$	2.252 2.244 2.197 2.185	2.250	18	3	ò	Ğ
		2.117	8		2.141	18	$(\frac{4}{5})$	$\frac{2}{1}$	3_1
5441 5	$^{1}_{\begin{array}{c} 0 \\ 2 \\ 1 \\ 1 \\ 1 \end{array}}^{1} \stackrel{1}{} \stackrel{)}{} \stackrel{)}{} \stackrel{)}{} {} {} \stackrel{1}{} \stackrel{)}{} \stackrel{)}{} \stackrel{)}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{)}{} \stackrel{)}{} \stackrel{)}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{)}{} \stackrel{)}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{)}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{)}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{1}{} \stackrel{)}{} \stackrel{1}{} \stackrel$	2.075	20b	2.121 2.115 2.079 2.074 2.073	$2.079 \\ 2.071$	$13 \\ 13$	$\begin{pmatrix} 2\\ 4\\ (0\\ 5 \end{pmatrix}$	$\frac{2}{2}$	8 4 9 3
		$2.005 \\ 1.917 \\ 1.875$	$^{10b}_{16b}_{13}$		2.002	24	١5	1	3
		1.837 1.748 1.712 1.670	14 14b 10		$1.843 \\ 1.836$	19 22b			
		1.712 1.670 1.647 1.604 1.593	10 17 12b 10 17b		1.669 1.618	21 9			

means of wet chemical analyses for the specimen purified through heavy liquids and hand picking separation. The result is given in Table 1 and is compared with the chemical compositions of tobermorite group minerals from Fuka. According to the formula of tobermorite reported by Megaw and Kelsey (1956), the empirical formula calculated on the basis of Si + Al + B = 6 is $(Si_{5.90}Al_{0.05}B_{0.05})_{6.00}$ $(Ca_{5,29}Mg_{0,02}K_{0,02})_{5,33}$ $(O_{16.54}OH_{1.39}F_{0.07})_{18.00}$ · 5.1H₂O. This result shows that the ideal structural formula of clinotobermorite is Ca₅Si₆(O,OH)₁₈·5H₂O. The chemical composition of clinotobermorite is very close to that of tobermorite. The aluminium content is lower than that of tobermorite and the calcium content is slightly higher. The low aluminium and high calcium contents found in the Fuka clinotobermorite are possible not essential for clinotobermorite in general.

X-ray studies

Precession and Weissenberg photographs show that clinotobermorite is monoclinic with the possible space group Cc or C2/c. The detailed crystallography of this mineral was already reported by Henmi and Kusachi (1989). It is summarised as follows. All the crystals are twinned on the *c*-plane or along the *a*-axis. The reflections with h = 2n and k = 2n show streaks parallel to c^* . This indicates the presence of stacking disorder on the *c*-plane. No single crystals suitable for crystal structure analysis have been found.

The unit cell dimensions of clinotobermorite, refined from the X-ray powder data, are given in Table 2, and are compared with those of the tobermorite group minerals from Fuka. The unit cells of tobermorite group minerals have the *a*-

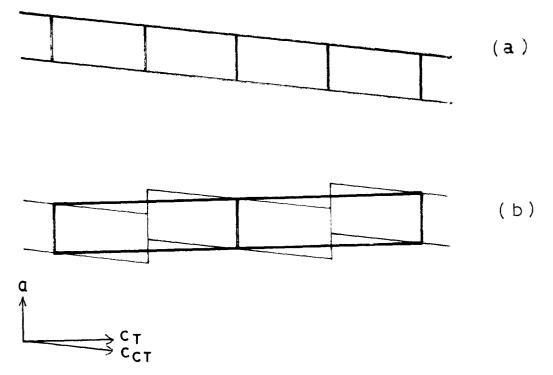


FIG. 3. Schematic diagram showing the geometrical relation of unit cells between clinotobermorite and tobermorite. (a) Clinotobermorite. (b) Stacking of the cells of clinotobermorite on (001) with an alternative displacement of 0 and 1/2a results in the cell of tobermorite with the doubled c-axis. Thick lines indicate the unit cells. The a- and baxes are in common for both cells.

and *b*-axis in common with one another. The unit cell of clinotobermorite is closely related to that of tobermorite. The stacking of the (001) layers of the clinotobermorite with an alternative displacement of 0 and 1/2a results in the cell similar to the tobermorite cell with the doubled *c*-axis and the *a*-and *b*-axes in common (Fig. 3).

The X-ray powder diffration data of clinotobermorite are listed in Table 3 together with those of tobermorite for comparison (after Henmi and Kusachi, 1989). They are considerably different from each other although there are some similar reflections with odd numbers of h and k.

Discussion

McConnell (1954) reported that tobermorite is orthorhombic with the space group $C222_1$. Because of the presence of stacking disorder, the precise structure of tobermorite has not been determined. Megaw and Kelsey (1956) presented essential features of the tobermorite structure and suggested that the true cell is triclinic. Hamid (1981) determined an average structure of tobermorite in the orthorhombic subcell with a 5.586, b 3.696, and c 22.779 Å and reported one of the possible ordered structures which is monoclinic with the space group $P2_1$. The cell of clinotobermorite presented in this paper is different from the monoclinic cell suggested by Hamid (1981). The stacking disorder of clinotobermorite on the (001) plane is similar to that of tobermorite reported by McConnell (1954). The cell dimensions, the relation shown in Fig. 3, and the IR spectra indicate that the crystal structure of clinotobermorite is closely related to that of tobermorite. It is noteworthy, however, that not only the single crystal data but also the powder data of clinotobermorite are distinctly different from those of tobermorite (Table 3).

The relation between the clinotobermorite and tobermorite unit cells (Fig. 3) resembles that between clinopyroxene and orthopyroxene described by Morimoto (1959). Orthopyroxene is also considered to be a twinned clinopyroxene (Ito, 1950). The twinning of clinotobermorite, in the same scheme as clinopyroxene, results in the cell of tobermorite with the doubled c-axis, which is the same as that formed by the alternative glide shown in Fig. 3.

The relation between the clinotobermorite and tobermorite unit cells is also similar to that between wollastonite-1T and wollastonite-2M reported by Trojer (1968). The stacking disorder in clinotobermorite may be similar to that in wollastonite, which was reported by Jefferson and Bown (1973), and also to that in foshagite, which was reported by Gard and Taylor (1960).

Although the chemical composition of the Fuka clinotobermorite falls within the variation range expected for normal tobermorite along with Ca/ (Si + Al) ratio (Mitsuda and Taylor, 1978), the high Ca content, exceeding that given by the ideal formula $Ca_5H_2(Si_3O_9)_2 \cdot 5H_2O$, is noteworthy together with the lower aluminium content. The chemical formula referred to by Hamid (1981) has smaller Ca content than that of the formula given above, Ca/Si = 4.5/6. The structure model permits the variable content of Ca. Meyer and Jaunarajs (1961) suggest that the Ca/Si ratio of synthetic tobermorite reaches unity. Also, an non-stoichiometric Ca/Si ratio is confirmed for killalaite (Taylor, 1977). Therefore, the variable Ca/Si ratios do not impede a prefix clino being applied to this case.

The mode of occurrence indicates that clinotobermorite precipitated after tobermorite in fissures. The aluminium content of clinotobermorite is lower than that in tobermorite, and the symmetry of clinotobermorite is lower than that of tobermorite. Therefore, it may be inferred that clinotobermorite is a low-temperature polymorph of tobermorite.

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