

CRYSTAL STRUCTURE OF POWELLITE CaMoO_4

V. B. Aleksandrov, L. V. Gorbatyi, and V. V. Ilyukhin

Institute of Crystallography, Academy of Sciences of the USSR

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The crystal structures of molybdates and tungstates are of interest in relation to the use of these as laser materials. At the same time, the need for more accurate atom localization has increased. Structures have been determined for new compounds of ABO_4 type, and previous models have been refined [1-7]. In the cases of NiWO_4 and CaWO_4 [1-3] it has been shown that the earlier work [8-10] gave the structure pattern correctly but that the coordinates of the oxygen atoms were very inaccurate. The cation-oxygen distances calculated from the data of [8-10] agree poorly with the ionic radii [11]. The present structure refinement for CaMoO_4 is a continuation of the work of CdWO_4 and CdMoO_4 [4, 5].

The CaMoO_4 crystals were made by hydrothermal recrystallization and were kindly made available by L. N. Dem'yanets (Institute of Crystallography, AN SSSR). Irregular fragments were ground to nearly spherical form (diameter 0.40 ± 0.05 mm). The intensities of the h0l reflections were measured with Mo $K\alpha$ radiation on a URS-50I diffractometer

TABLE 1. Coordinates of the Basis Atoms in CaMoO_4 , Origin at 4

Atom	x/a	y/b	z/c
Ca	0	0	0
Mo	0	0	0.500
O	0.257	0.145	0.167

fitted with a MSTR-4 counter. The integral intensities were evaluated from the summed areas of the $K\alpha_1$ and $K\alpha_2$ peaks as recorded by an ÉPP-09 recorder. A correction was applied for counter non-linearity (determined by foil superposition) for peaks of maximum count rate over 200 sec^{-1} . Reflections of intensity over 1000 sec^{-1} were attenuated with a calibrated attenuator. The F_0^2 were calculated with allowance for absorption ($\mu R = 1.15$).

The cell parameters and space group of CaMoO_4 were as previously found [12]: $a = 5.224$, $c = 11.430 \text{ \AA}$, $Z = 4$, $C_{4h}^{6h} - I4_1/a$. This group demands that the Ca and Mo lie at the special points on the

TABLE 2. Interatomic Distances and Valence Angles in Mo(W) Polyhedra

ABO_4	Interatomic distance, Å				Valence angles	
	1st coord. sphere		2nd coord. sphere		1st coord. sphere	2nd coord. sphere
	B-O	O-O	B-O	O-O	O-B-O	O-B-O
CaMoO_4	B-O ₁₋₄ 1,76	O ₁ -O ₂ 2,94 (4) O ₃ -O ₄ 2,82 (2)	B-O ₅₋₈ 2,94	O ₅ -O ₆ = 4,50 O ₇ -O ₈ = 4,90	O ₁ -B-O ₂ = 114°50' O ₃ -B-O ₄ = 107°	O ₅ -B-O ₆ = 99° O ₇ -B-O ₈ = 114°55'
CaWO_4	B-O ₁₋₄ 1,78	O ₁ -O ₂ 2,98 (4) O ₃ -O ₄ 2,83 (2)	B-O ₅₋₈ 2,91	O ₅ -O ₆ = 4,45 O ₇ -O ₈ = 4,87	O ₁ -B-O ₂ = 113°30' O ₃ -B-O ₄ = 107°23'	O ₅ -B-O ₆ = 100°30' O ₇ -B-O ₈ = 114°35'
CdMoO_4	B-O ₁₋₄ 1,82	O ₁ -O ₂ 2,98 (4) O ₃ -O ₄ 2,96 (2)	B-O ₅₋₈ 2,86	O ₅ -O ₆ = 4,50 O ₇ -O ₈ = 4,80	O ₁ -B-O ₂ = 110°05' O ₃ -B-O ₄ = 108°35'	O ₅ -B-O ₆ = 104°20' O ₇ -B-O ₈ = 114°
CdWO_4	B-O _{1,2} 1,80 B-O _{3,4} 1,87 B-O _{5,6} 2,22	O ₁ -O ₂ 2,75 O ₁ -O ₃ 2,70 O ₁ -O ₄ 2,84 O ₂ -O ₃ 2,92 O ₂ -O ₄ 2,84 O ₃ -O ₄ 2,80 O ₁ -O ₅ 2,80 O ₃ -O ₆ 3,00	B-O _{7,8} 3,40		O ₁ -B-O ₂ = 99°40' O ₃ -B-O ₄ = 85°05' O ₁ -B-O ₃ = 91°15' O ₂ -B-O ₃ = 94°20' O ₁ -B-O ₄ = 90° O ₂ -B-O ₄ = 90° O ₃ -B-O ₅ = 91° O ₁ -B-O ₆ = 91°	

TABLE 3. Interatomic Distances (Å) in the Ca Polyhedron of CaMoO₄

Ca - O	O - O
2.45 (4)	3.19 (2)
2.48 (4)	3.75 (4)
	2.78 (4)
	2.96 (4)
	3.05 (4)

$\bar{4}$ axes. The oxygen atoms lie in general positions and were localized from P(xz) and σ (xz) syntheses. Table 1 gives the final values of the coordinates for R = 11% (for all reflections in the range $\sin\theta/\lambda \leq 1.4 \text{ \AA}^{-1}$, B = 0.5 \AA^2).

The structure pattern of CaMoO₄ is exactly as for scheelite [2, 3], with Ca in a polyhedron with eight vertices characteristic of scheelite [13] and with the Mo in a nearly regular tetrahedron. Tables 2 and 3 give the interatomic distances. CaMoO₄ resembles CdMoO₄ and CdWO₄ in that the Mo(W) atom has four nearest O atoms (1.76-1.82 Å) and four more O atoms (2.86-2.94 Å) forming a second coordination sphere. There is a tendency for the distances in the first sphere to increase (from 1.76 to 1.82 Å) from CaMoO₄ to CdMoO₄, which is accompanied by reduction of the Mo(W)-O distances in the second sphere (Table 2), which leads on further deformation not only to change in the coordination numbers of both cations but also to the CdWO₄ structure type [14, 15].

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