A note on the occurrence of geocronite and boulangerite in the Rajpura belt, Udaipur district, Rajasthan, India¹

Two rare sulphosalts have recently been identified in the drill core samples of a Precambrian base metal deposit located at Rajpura $(24^{\circ} 58' \text{ N.}, 74^{\circ} 07' \text{ E.})$ in the Udaipur district, Rajasthan.

Geocronite occurs as a massive granular aggregate associated with minor galena and disseminated arsenopyrite in a matrix of pyroxene, tremolite, quartz, and calcite. The mineral is lead grey with a pale brownish shade and shining metallic lustre. H. about 3 (Moh's scale), sp. gr. > 6. The mineral takes excellent polish; in reflected light it is white with a pale purplish tint. Reflectance pleochroism is weak: yellowish-white to greenish-white. Anisotropism is strong with polarization colours of yellowish-grey to greenish-grey. Lamellar twinning defined by lamellae of uniform thickness is characteristic. Extinction is oblique with respect to the twin lamellae. Reflectivity 40.9 (in air, white light). Etch reactions HNO₃-effervesces feebly, stains black; HCl-stains brown; KCN, FeCl₃, KOH, and HgCl₂ give negative results. Microchemical tests indicate lead, antimony, arsenic, and sulphur. The X-ray powder pattern is an excellent match with that recorded by Douglass et al. (1954),² and indexed well with a 8.94 Å, b 31·92 Å, c 8·48 Å, β 118° 28'.

Boulangerite occurs as needles and fine prismatic grains within galena in a medium-grained assemblage of galena, tetrahedrite, chalcopyrite, and pyrite associated with gangue minerals like pyroxene, tremolite, and calcite. The mineral is lead grey with a pale greenish shade and shining metallic lustre. It occurs in intimate association with galena and a pure fraction of the mineral could not be separated. The mineral takes excellent polish and is distinctly pleochroic; galena white to greenishwhite. Anisotropism is strong: light tan, grey, yellowish-white. Twinning is absent. Reflectivity 36 (minimum) and 40-5 (maximum) (in air, photo-cell, white light). Etch reactions are as follows: HNO_3 —effervesces, stains black; HCl—stains faintly grey; KCN, FeCl₃, and HgCl₂ negative. Micro-chemical tests are positive for lead, antimony, and sulphur.

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 $^{^{2}}$ The names geocronite and jordanite on A.S.T.M. cards 8–94 and 11–100 were interchanged. Both sets of data were from Douglass *et al.* (1954).

The X-ray pattern agreed well with that of boulangerite given in A.S.T.M. card (9-470) and the cell dimensions for boulangerite were calculated as a = 21.54 Å, b = 23.50 Å, c = 8.10 Å, $\beta = 101^{\circ} 15'$.

A chemical analysis of the geocronite was carried out as follows:

The specimen was decomposed with a mixture of $HCl-H_2SO_4-HNO_3$. Pb was separated as $PbSO_4$ along with other insolubles. This was filtered and the residue was treated with ammonium acetate acetic acid mixture to remove $PbSO_4$, and the insolubles ignited and weighed. Pb was estimated as chromate in the usual manner. From the filtrate As, after reduction with SO_2 , was removed by H_2S at 10 N acidity. As_2S_3 was dissolved in NaOH containing H_2O_2 and As was determined as magnesium pyroarsenate. From the filtrate after removing arsenic Sb was precipitated as Sb_2S_3 by passing H_2S at 1 N acidity. The Sb_2S_3 was treated with H_2SO_4 -KHSO₄ and fumed to copious fumes to obtain a solution of Sb^{3+} . Sb was then estimated by titration with KBrO₃ using methyl orange as an indicator. Sulphur was estimated as usual by fusion with Na_2O_2 .

The results were Pb 63.34, As 3.01, Sb 10.98, S 16.15, insolubles 6.10, total 99.58 %.

After deducting the insoluble, which was found to be silica, present as impurity in the mineral, the other constituents were recalculated to 100 %, giving Pb 67.77, As 3.21, Sb 11.74, S 17.28.

This agrees well with geocronite, $Pb_9(As,Sb)_4S_{15}$ (Douglass, Murphy, and Pabst, 1954). A chemical analysis of the boulangerite could not be carried out as a pure fraction was not available.

This appears to be the first reported occurrence of geocronite in India.

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Geological Survey of India,	D. R. DASGUPTA
29 Chowringhee, Calcutta-16,	B. C. Poddar
India.	N. R. SEN GUPTA

Reference

DOUGLASS (R. M.), MURPHY (M. J.), and PABST (A.), 1954. Amer. Min , vol. 39, p. 908.

Fortran IV programme for molecular norm calculation

A FORTRAN IV programme has been written for the calculation of molecular norms of igneous rocks according to the method of Barth (1962). Input to the programme consists of N (the number of analyses to be