Biphosphammite: a second occurrence

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SUMMARY. Biphosphammite, $NH_4H_2PO_4$, recorded once as secondary powder in 1870, occurs with bat guano in Murra-el-elevyn cave, $31^{\circ} 20' S.$, $126^{\circ} 0' E.$, Western Australia.

The mineral occurs as tapering tetragonal prisms with pyramidal terminations, which are colourless to light buff, very soft, water soluble, contain fine syngenite inclusions, and have D 2.04, ω 1.525, ϵ 1.480. Space group *I*42*d*, *a* 7.4935 Å, *c* 7.340 Å, six strongest powder X-ray lines 3.75 Å (10) 200; 5.24 Å (9) 101; 3.02 Å (9) 112; 1.993 Å (8) 312,213; 2.65 Å (7) 220; 2.368 Å (7) 310,301. Partial analysis of purest available material gave: P₂O₅ 51.1, K₂O 14.2, (NH₄)₂O 12.3, SO₃ 5.59, Na₂O 0.16, water-insolubles 0.81 per cent, remainder mainly CaO and H₂O. Calculated mineral content is 88 % biphosphammite (62 % NH₄H₂PO₄, 38% isomorphous KH₂PO₄), 11.5% syngenite, insolubles 0.81 % calc. total 100.3 per cent.

Material proposed to be neotype is preserved at the Government Chemical Laboratories, Perth, Western Australia.

IN 1967 while exploring Murra-el-elevyn cave $(31^{\circ} 20' \text{ S.}, 126^{\circ} 0' \text{ E.})$ on the Nullarbor Plain, Western Australia, P. J. Bridge collected some unusual crystals associated with bat guano. At the W.A. Government Chemical Laboratories, L. V. Bastian identified the crystals as $(NH_4, K)H_2PO_4$, the second record of the mineral biphosphammite and the first *in situ* occurrence.

The first record was by C. U. Shepard, Sr., (1870) who studied a bird guano shipment from Guañape Island ($8^{\circ} 25'$ S., $79^{\circ} 25'$ W.) at Charleston, South Carolina, U.S.A.

Shepard noted that lumps and crystal coatings of phosphammite, $(NH_4)_2HPO_4$, lost ammonia on exposure to become coated with a powder of secondary biphosphammite, $NH_4H_2PO_4$.

Occurrence.¹ Murra-el-elevyn cave is close to the Eyre Highway and approximately 130 miles east of Balladonia. The surface expression of the cave is a fresh collapse doline 60 ft in diameter with sheer 30 ft walls in the flat Nullarbor Plain. Underground the cave slopes steeply for 500 ft to a lake at a vertical depth of 270 ft, then continues horizontally for a further 400 ft as a wide low undulating chamber inhabited by a maternity colony of *Chalinolobus morio* GRAY (the Little Chocolate Bat) identified by Hamilton-Smith (1965).

The biphosphammite occurs as rounded crusts coating the walls and floor, apparently as a crystalline product of the liquid fraction of bat guano.

Physical properties. The biphosphammite is a finely granular crust with raised radiating groups of crystals, mainly tapering prisms as large as 2 mm long and 0.2 mm

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 - ¹ Described by P. J. Bridge, Government Chemical Laboratories, Perth, W.A.

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thick. Syngenite is closely intergrown and also enclosed in the biphosphammite crystals.

The crust has a dull earthy appearance and varies from white through to deep shades of brown. The crystals are very soft with a dull lustre and occasional vitreous clusters. The streak is between white and Ridgway (1912) Light Buff 17'f. Using the sink-float method in diluted bromoform some of the cleanest crystals gave D 2.04 ± 0.02 , which would be a little high due to inclusions of syngenite.

Optical properties. The crystals are colourless with many fine syngenite inclusions and have $\omega 1.525$, $\epsilon 1.480$. By the same method Analar grade NH₄H₂PO₄ and KH₂PO₄ gave $\omega 1.525$, $\epsilon 1.480$ and $\omega 1.510$, $\epsilon 1.468$ respectively, in good agreement with Winchell and Winchell (1964).

Chemistry. A sample was prepared for analysis from the small amount of relatively pure crystals by centrifuging with diluted bromoform. The final concentrate still contained fine inseparable syngenite inclusions.

A partial analysis by P. Hewson using standard methods gave $P_2O_5 51 \cdot I$, $K_2O 14 \cdot 2$, $(NH_4)_2O 12 \cdot 3$, $SO_3 5 \cdot 59$, $Na_2O 0 \cdot 16$, water-insoluble fraction 0.81 %, remainder mainly CaO and H_2O . Calculation to constituent minerals gave $88 \cdot 0$ % biphosphammite of composition 62 % ammonium salt $NH_4H_2PO_4$ with 38 % isomorphous potassium salt KH_2PO_4 . The remainder consisted of $11 \cdot 5$ % syngenite, insolubles 0.81 %, calculated total $100 \cdot 3$ %.

X-ray data. Weissenberg photographs confirmed that the biphosphammite conforms to space group $I\bar{4}2d$. Some disorder along c in the crystals is indicated by the oo8 and 208 reflections being recorded on the films as unresolved streaks.

A unit-cell refinement gave the values $a 7.4935\pm0.0005$ Å, $c 7.340\pm0.003$ Å, c/a 0.9795, V 412.2 Å³ for biphosphammite. The length of c corresponds to a composition 64 % NH₄H₂PO₄ and 36 % KH₂PO₄.

The powder X-ray data, Table I, were measured on films taken with a Guinier-Hagg focusing camera with Cu- $K_{\alpha 1}$ radiation and ThO₂ internal standard. Coates and Woodard (1963) produced powder data with similar *d* spacings from artificial (0.73 NH₄H₂PO₄). (0.27 KH₂PO₄) formed in fertilizer.

Unit cell refinements of Analar grade chemicals gave $a 7.4987\pm0.0005$ Å, $c 7.550\pm0.005$ Å for NH₄H₂PO₄ and $a 7.451\pm0.001$ Å $c 6.972\pm0.0015$ Å for KH₂PO₄, in good agreement with the results of Swanson and Fuyat (1953) and Swanson, Fuyat, and Ugrinic (1953). The refinements were all done by the method described by Pryce (1970). All X-ray exposures were made in the range 15 to 20 °C.

Conclusions. The density and *c*-axis measurements of biphosphammite correspond closely to the chemical composition on analysis but the refractive indices are close to those of pure $NH_4H_2PO_4$. There is no obvious reason, although minerals are occasionally found to have different refractive indices from the more-pure artificial equivalents.

Neotype material is preserved at the Government Chemical Laboratories, Perth, Western Australia.

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BIPHOSPHAMMITE

hkl	Ι	$d_{ m obs}$	$d_{ m cale}$	hkl	Ι	$d_{ m obs}$	$d_{ m calc}$
101	9	5·24 Å	5·244 Å	431)	4	1·470 Å	∫1·468 Å
200	10	3.75	3.747	413/			1.459
211	4	3.05	3.048	521)	2	1.365	(1.367
112	9b	3.02	3.012	512)			1.364
220	7	2.650	2.649	440	3	1.324	1.325
202	4	2.623	2.622	404	3b	1.310	1.311
310)	7	2.368	∫2·369	530)	2	1.285	(1.285
301)			2.365	503)			1.278
321	5	2.002	2.000	433)	I	1.220	(1.278
312)	8	1.993	∫1∙991	334)			1.273
213)			1.976	442	3b	1.250	1.246
400	I	1.875	1.873	424	3b	1.240	1.537
303	3b	1.754	1.748	532)	3b	1.514	(I·2I3
420)	4	1.676	∫1·675	523)	30	1 214	1.210
402)	4	10/0	1.669	620	4	1.186	1.182
332	6	1.203	1.201	602	2	1.182	1.185
323	3b	1.286	1.284	541	I	1.156	1.126
422	3b	1.515	1.524	514	I	1.120	1.147
224	3b	1.210	1.209	631	4	1.102	1.104
				613	2	I.100	1.100

TABLE I. X-ray powder diffraction data for biphosphammite (NH₄:K = 61:39). Guinier focusing camera, Cu- $K_{\alpha 1}$, ThO₂ internal standard.

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