# CRYSTALLOGRAPHIC STUDIES OF SULPHOSALTS: BAUMHAUERITE, MENEGHINITE, JORDANITE, DIAPHORITE, FREIESLEBENITE

## CHARLES PALACHE, Harvard University, Cambridge, Mass.

# WITH X-RAY STUDIES BY W. E. Richmond and Horace Winchell

## BAUMHAUERITE

Information regarding this mineral is confined to the paper of Solly (1903). The author measured two crystals and from the gnomonic plot of one of them discovered a typographical error in Solly's original statement of the axial ratio. This error did not affect the values of his angles but it has gone into every printed description of the mineral. Since no complete two-circle angle table has been published for baumhauerite, one has been calculated and the angles for the more important forms are given below; 118 forms are listed by Solly whose position and elements (corrected) have been used. The mineral is characterized by the enormous development of the orthodome zone to which belong two-thirds of the observed forms.

BAUMHAUERITE-Pb4As6S7

Monoclinic; prismatic—2/m

 $\begin{aligned} a:b:c=1.3687:1:0.9472; \ \beta=97^{\circ}17'\\ p_0:q_0:r_0=0.6920:0.9396:1; \ \mu=82\ 43\\ r_2:p_2:q_2=1.0643:0.7365:1;\\ p_0'=0.6976, \ q_0'=0.9472; \ x_0'=0.1278 \end{aligned}$ 

Fo	orms	¢	ρ	$\phi_2$	$\rho_2 = B$	С	A
с	001	90°00′	7°17′	82°43′	90°00′	0°00′	82°43′
Ь	010	0 00	90 00		0 00	90 00	90 00
a	100	90 00	90 00	0 00	90 00	82 43	0 00
F	520	$61\ 29\frac{1}{2}$	90 00	0 00	$61\ 29\frac{1}{2}$	83 36	$28 \ 30\frac{1}{2}$
H	210	55 491	90 00	0 00	$55 \ 49\frac{1}{2}$	83.59	$34\ 10\frac{1}{2}$
K	320	47 51	90 00	0 00	47 51	$84 \ 36\frac{1}{2}$	42 09
m	110	$36\ 22\frac{1}{2}$	90 00	0 00	$36\ 22\frac{1}{2}$	$85 \ 41\frac{1}{2}$	$53 \ 37\frac{1}{2}$
0	120	20 13	90 00	0 00	20 13	87 29불	69 47
k	011	7 41	$43 \ 42\frac{1}{2}$	82 43	46 47	43 13	84 42
,	702	90.00	68 44	21 16	90 00	61 27	21 16
v	301	90.00	65 45	24 14 <sup>±</sup>	90 00	58 28 <sup>1</sup> / <sub>2</sub>	$24\ 14\frac{1}{2}$
μ	502	90 00	61 53	28 07	90 00	54 36	28 07

822

	Forms	$\phi$	ρ	$\phi_2$	$\rho_2 = B$	С	Α
ξ	201	90 00	56 42 <sup>1</sup> / <sub>2</sub>	$33\ 17\frac{1}{2}$	90 00	$49\ 25\frac{1}{2}$	$33\ 17\frac{1}{2}$
ρ	302	90 00	49 35	40 25	90 00	42 18	40 25
$\psi$	101	90 00	39 32	50 28	90 00	32 15	50 28
Δ	102	90 00	25 29	64 31	90 00	18 12	64 31
п	103	90 00	19 49	70 11	90 00	12 32	70 11
Σ	104	90 00	16 49	73 11	90 00	9 32	73 11
$\Phi$	106	90 00	13 43	76 17	90 00	6 26	76 17
g:	T04	-9000	2 40	92 40	90 00	9 57	92 40
1:	<b>T</b> 02	-90 00	$12 \ 27\frac{1}{2}$	$102 \ 27\frac{1}{2}$	90 00	$19 \ 44\frac{1}{2}$	$102 \ 27\frac{1}{2}$
q:	101	-90 00	$29 \ 40\frac{1}{2}$	$119 \ 40^{\frac{1}{2}}$	90 00	$36\ 57\frac{1}{2}$	$119 \ 40^{\frac{1}{2}}$
t:	302	-90 00	42 34	132 34	90 00	49 51	132 34
w	: 704	-90 00	$47 \ 32\frac{1}{2}$	$137 \ 32\frac{1}{2}$	90 00	$54 \ 49\frac{1}{2}$	$137 \ 32\frac{1}{2}$
2:	201	-90 00	$51 \ 43\frac{1}{2}$	141 $43\frac{1}{2}$	90 00	$59\ 00^{\frac{1}{2}}$	141 $43\frac{1}{2}$
C	502	-9000	58 15	148 15	90 00	65 32	148 15
E	: 301	-90 00	$63 \ 01\frac{1}{2}$	$153 \ 01\frac{1}{2}$	90 00	$70 \ 18\frac{1}{2}$	$153 \ 01\frac{1}{2}$
Þ	111	41 04	51 29	50 28	53 51	46 56	59 04
0	Ī11	-31 02	47 52	$119 \ 40^{\frac{1}{2}}$	50 33	51 54	$112\ 28\frac{1}{2}$
n	122	$26 \ 42\frac{1}{2}$	$46 \ 40^{\frac{1}{2}}$	64 31	49 28	$43 \ 46\frac{1}{2}$	70 55
N	<u>1</u> 22	-13 08	44 12북	$102\ 27\frac{1}{2}$	47 14	$46\ 17\frac{1}{2}$	99 07

BAUMHAUERITE-Continued

Errata: Solly (1903) for a=1.1368 read 1.3687

Dana (1909) for a=1.1368 read 1.3687

Goldschmidt (1928) for  $p_0' = 0.8402$  read 0.6976;

for e'=0.1305 read 0.1278

for  $p_0 = 0.8332$  read 0.6920;

for  $\mu = 82^{\circ}34' \text{ read } 82^{\circ}43'$ 

#### References

Dana, E. S., and Ford, W. E. (1909): Second Appendix to the Sixth Edition of Dana's System of Mineralogy, p. 13.

Goldschmidt, V., and Gordon, S. G. (1928): Crystallographic Tables for the Determination of Minerals—Special Publication No. 2, Acad. Nat. Sci. Philadelphia, 44; No. 1089.

Solly, R. H. (1903): Baumhauerite—Mineral, Mag., vol. 13, p. 151, and Zeits. Krist., vol. 37, p. 321.

### MENEGHINITE

Meneghinite has been found in definite crystals at but one locality, Bottino, Italy. Our knowledge of its crystallography rests upon studies made simultaneously by Krenner (1883) and Miers (1883), which established its orthorhombic character and yielded substantially the same elements. The two authors differed, however, in one respect. Miers found a series of typical forms with simple indices and with them additional forms of equally good quality to which he could only assign very complex indices. He insisted that these forms were to be regarded as true members of the form series. Krenner also observed such forms but regarded them as vicinal and discarded them. He pointed out that such vicinal forms accounted for the earlier erroneous monoclinic interpretation of the crystals by vom Rath (1867).

No further observations seem to have been made on meneghinite; but Ungemach (1923) discussed the form series, suggested a new choice of unit form and concluded that the aberrant forms might be best explained by regarding the mineral as monoclinic with concealed twinning, analogous to jordanite with which isomorphism had been suspected by several authors.

The author tested this theory by measuring crystals, and Mr. W. E. Richmond made an x-ray study which is reported below. The results of these studies are positive as to the orthorhombic character of meneghinite; a new unit cell is imperative which differs from that of any previous observer; the aberrant forms are confirmed but wholly un-explained; and the fact is established that it is not isomorphic with jordanite.

The crystals are slender needles with minute terminal facets. The acicular direction is taken as c by all observers. The new elements required by the x-ray measurements have the same directions as before, but the new unit (111) is the form (414) of Miers and (214) of Krenner. Transformations:—

Miers to Palache  $\frac{1}{4}00/010/00\frac{1}{4}$ Krenner to Palache  $\frac{1}{2}00/010/00\frac{1}{4}$ 

As the basis of the angle table, the author has employed the elements of Goldschmidt (1897), which are the mean of those of Miers and Krenner. Table 1 is therefore a restatement of Goldschmidt's angles with new indices for the forms; the letters have been preserved unchanged except for two prisms.

The author measured three crystals from the type locality. They show a prism zone so deeply grooved by striations that but a few typical faces could be recognized except the pinacoid parallel to which there is perfect cleavage. This face, always good, was taken as (010). The presence of basal cleavage was also verified, but both cleavages are obtained only with considerable difficulty. Table 2 shows the terminal faces found on two of the measured crystals.

$\begin{array}{c c c c c c c c c c c c c c c c c c c $				Orthorhom	oic; dipyram	idal— <i>m m m</i>		
$\begin{array}{c c c c c c c c c c c c c c c c c c c $			a:b:c =	0.4736:1:0.	1715; po:q	$r_0: r_0 = 0.3621$	:0.1715:1	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $			$q_1:r_1:p_1 =$	0.4736:2.76	17:1; $r_2:p$	$q_2: q_2 = 5.8309$	:2.1114:1	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	F	orms	φ	$\rho = C$	$\phi_1$	$\rho_1 = A$	$\phi_2$	$\rho_2 = B$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	с	001		0°00′	0°00′	90°00′	90°00′	90°00′
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	b	010	0°00′	90 00	90 00	90 00	-	0 00
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	a	100	90 00	90 00		0 00	0 00	90 00
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	е	160	19 23	90 00	90 00	70 37	0 00	19 23
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	R	140	$27 \ 49\frac{1}{2}$	90 00	90 00	$62\ 10^{\frac{1}{2}}$	0 00	27 491
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	S	130	35 08	90 00	90 00	54 52	0 00	35 08
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	l	380	38 22	90 00	90 00	51 38	0 00	38 22
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	f	5.12.0	41 20	90 00	90 00	48 40	0 00	41 20
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Т	120	46 33	90 00	90 00	43 27	0 00	46 33
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	g	340	57 43	90 00	90 00	32 17	0 00	57 43
m       110       64 $39\frac{1}{2}$ 90       90       90 $25$ $20\frac{1}{2}$ 000 $64$ $39\frac{1}{2}$ k       520       79 $16\frac{1}{2}$ 90       00       90       00 $10$ $43\frac{1}{2}$ 000       79 $16\frac{1}{2}$ k       310       81 $01\frac{1}{2}$ 90       00       90       00       8 $88\frac{1}{2}$ 000       79 $16\frac{1}{2}$ y       032       0       00       14 $25\frac{1}{2}$ 14 $25\frac{1}{2}$ 90       00       90       00       71 $04$ o       083       0.00       24       35       24       35       90       00       90       00       71 $04$ o       083       0.00       24       35       24       35       90       00       90       00       71 $04$ o       00       19 $54\frac{1}{2}$ 0.00       70 $05\frac{1}{2}$ 90       90       90       90       90       90       90       90       90       90       90       90       90       9	i	780	$61 \ 34\frac{1}{2}$	90 00	90 00	$28\ 25\frac{1}{2}$	0 00	61 341
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	т	110	$64 \ 39\frac{1}{2}$	90 00	90 00	$25 \ 20^{\frac{1}{1}}_{\frac{1}{2}}$	0 00	$64 \ 39\frac{1}{2}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	h	520	79 16 <u>1</u>	90 00	90 00	$10\ 43\frac{1}{2}$	0 00	79 16 <del>1</del>
y032014 $25\frac{1}{2}$ 14 $25\frac{1}{2}$ 9000900075 $35\frac{1}{2}$ d02100018561856900090007104o08300024352435900090006525v04100034 $27\frac{1}{2}$ 34 $27\frac{1}{2}$ 90009000655321n101900019 $54\frac{1}{2}$ 00070 $05\frac{1}{2}$ 70 $05\frac{1}{2}$ 9000W403900025 $46\frac{1}{2}$ 0007005\frac{1}{2}900000W403900025 $46\frac{1}{2}$ 00070 $05\frac{1}{2}$ 9000W403900025 $46\frac{1}{2}$ 00070 $05\frac{1}{2}$ 70 $05\frac{1}{2}$ 9000W403900025 $54\frac{1}{2}$ 000540554059000u111 $64$ $39\frac{1}{2}$ 215094470 $21\frac{1}{2}$ 80 $05\frac{1}{2}$ 80 $50\frac{1}{2}$ $\beta$ 221 $64$ $33$ 26 $30\frac{1}{2}$ 185671 $05\frac{1}{2}$ 80 $05\frac{1}{2}$ 71 $0\frac{1}{2}$ s13135083211<	k	310	$81 \ 01\frac{1}{2}$	90 00	90 00	$858\frac{1}{2}$	0 00	81 011
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	У	032	0 00	$14\ 25\frac{1}{2}$	$14\ 25\frac{1}{2}$	90 00	90 00	$75 \ 35\frac{1}{2}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	d	021	0 00	18 56	18 56	90 00	90 00	71 04
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	0	083	0 00	24 35	24 35	90 00	90 00	65 25
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	v	041	0 00	$34\ 27\frac{1}{2}$	$34\ 27\frac{1}{2}$	90 00	90 00	$55 \ 32\frac{1}{2}$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	n	101	90 00	$19 54\frac{1}{2}$	0 00	$70\ 05\frac{1}{2}$	70 $05\frac{1}{2}$	90 00
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	W	403	90 00	$25 \ 46\frac{1}{2}$	0 00	$64\ 13\frac{1}{2}$	$64\ 13\frac{1}{2}$	90 00
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	V	201	90 00	35 55	0 00	54 05	54 05	90 00
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	u	111	$64 \ 39\frac{1}{2}$	21 50	9 44	$70\ 21\frac{1}{2}$	$80\ 05\frac{1}{2}$	80 501
t       121       46 33       26 $30\frac{1}{2}$ 18 56       71 $05\frac{1}{2}$ 80 $05\frac{1}{2}$ 72 $07\frac{1}{2}$ s       131       35 08       32 11       27 14       72 09       80 $05\frac{1}{2}$ 64 11         r       141       27 $49\frac{1}{2}$ 37 $48\frac{1}{2}$ 34 $27\frac{1}{2}$ 73 $22\frac{1}{2}$ 80 $05\frac{1}{2}$ 64 11 $\mu$ 211       76 $40\frac{1}{2}$ 36 $39\frac{1}{2}$ 9 44       54 29       54 05       82 $05\frac{1}{2}$ $\phi$ 241       46 33       44 56       34 $27\frac{1}{2}$ 59 09       54 05       60 $56\frac{1}{2}$ $\phi$ 0.24.13       0 00       17 34       17 34       90 00       90 00       72 26 $\phi$ 0.24.11       0 00       20 $31\frac{1}{2}$ 20 $31\frac{1}{2}$ 90 00       90 00       69 $28\frac{1}{2}$ $q$ 24.0.11       90 00       38 19       0 00       51 41       51 41       90 00 $\lambda$ 24.24.13       64 $39\frac{1}{2}$ 36 $29\frac{1}{2}$ 17 34       57 $29\frac{1}{2}$ 56 14       75 15 $\sigma$ 24.24.11       64 $38\frac{1}{2}$ 41 $09\frac{1}{2}$ 20 31       53 30       51 41       73 38	β	221	$64 \ 39\frac{1}{2}$	$38 \ 42\frac{1}{2}$	18 56	55 35	54 05	74 281
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	t	121	46 33	$26 \ 30\frac{1}{2}$	18 56	$71 \ 05\frac{1}{2}$	$80 \ 05\frac{1}{2}$	72 $07\frac{1}{2}$
r       141       27 $49\frac{1}{2}$ 37 $48\frac{1}{2}$ $34$ $27\frac{1}{2}$ $73$ $22\frac{1}{2}$ $80$ $05\frac{1}{2}$ $57$ $10\frac{1}{2}$ $\mu$ 211       76 $40\frac{1}{2}$ $36$ $39\frac{1}{2}$ $9$ $44$ $54$ $29$ $54$ $05$ $82$ $05\frac{1}{2}$ $\rho$ 241       46       33       44 $56$ $34$ $27\frac{1}{2}$ $59$ $09$ $54$ $05$ $82$ $05\frac{1}{2}$ $\phi$ $0.24.13$ $0$ $00$ $17$ $34$ $17$ $34$ $90$ $00$ $90$ $00$ $72$ $26$ $\phi$ $0.24.11$ $0$ $00$ $20$ $31\frac{1}{2}$ $90$ $00$ $90$ $00$ $72$ $26$ $\phi$ $0.24.11$ $90$ $00$ $38$ $19$ $0$ $00$ $51$ $41$ $51$ $41$ $90$ $00$ $\chi$ $24.24.13$ $64$ $39\frac{1}{2}$ $36$ $29\frac{1}{2}$ $17$ $34$ $57$	s	131	35 08	32 11	27 14	72 09	$80\ 05\frac{1}{2}$	64 11
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	r	141	$27 \ 49\frac{1}{2}$	$37 \ 48\frac{1}{2}$	$34\ 27\frac{1}{2}$	$73\ 22\frac{1}{2}$	$80\ 05\frac{1}{2}$	57 101
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	μ	211	$76 \ 40\frac{1}{2}$	$36 \ 39\frac{1}{2}$	9 44	54 29	54 05	82 $05\frac{1}{2}$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Þ	241	46 33	44 56	34 27 <del>1</del>	59 09	54.05	60 561
$ \phi  0.24.11  0  00  20  31\frac{1}{2}  20  31\frac{1}{2}  90  00  90  00  69  28\frac{1}{2} \\ q  24.0.11  90  00  38  19  0  00  51  41  51  41  90  00 \\ \lambda  24.24.13  64  39\frac{1}{2}  36  29\frac{1}{2}  17  34  57  29\frac{1}{2}  56  14  75  15 \\ \sigma  24.24.11  64  38\frac{1}{2}  41  09\frac{1}{2}  20  31  53  30  51  41  73  38 \\ \end{array} $	δ	0.24.13	0 00	17 34	17 34	90 00	90.00	72 26
q24.0.1190 0038 190 0051 4151 4190 00 $\lambda$ 24.24.1364 39 $\frac{1}{2}$ 36 29 $\frac{1}{2}$ 17 3457 29 $\frac{1}{2}$ 56 1475 15 $\sigma$ 24.24.1164 38 $\frac{1}{2}$ 41 09 $\frac{1}{2}$ 20 3153 3051 4173 38	φ	0.24.11	0 00	20 31 <sup>1</sup> / <sub>2</sub>	$20 \ 31\frac{1}{2}$	90 00	90 00	$69\ 28\frac{1}{2}$
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	q	24.0.11	90 00	38 19	0 00	51 41	51 41	90.00
$\sigma$ 24.24.11 64 38 <sup>1</sup> / <sub>2</sub> 41 09 <sup>1</sup> / <sub>2</sub> 20 31 53 30 51 41 73 38	λ	24.24.13	64 39 <u>1</u>	36 291	17 34	57 294	56 14	75 15
10.00	σ	24.24.11	$64\ 38\frac{1}{2}$	$41 \ 09\frac{1}{2}$	20 31	53 30	51 41	73 38

Table 1. Meneghinite— $Pb_{13}Sb_7S_{23}$ 

			TABL	E 1.—Contin	ued		
1	Forms	φ	$\rho = C$	$\phi_1$	$\rho_1 = A$	$\phi_2$	$\rho_2 = B$
ρ	24.48.11	46 33	47 25 <sup>1</sup> / <sub>2</sub>	36 49	57 41	51 41	59 34 <u>1</u>
$\psi$	24.48.13	46 33	$42\ 38\frac{1}{2}$	32 21	$60\ 32\frac{1}{2}$	56 14	62 14
Х	24.72.13	35 08	49 16 <sup>1</sup> / <sub>2</sub>	43 32	$64 \ 08\frac{1}{2}$	56 14	51 42
π	24.96.13	$27 \ 49\frac{1}{2}$	$55\ 04\frac{1}{2}$	51 42 <sup>1</sup> / <sub>2</sub>	67 30	56 14	43 31
ω	7.21.1	35 08	$77\ 12\frac{1}{2}$	74 29	55 52	21 32	37 06

TABLE 1	2.	MEASUREMENTS	OF	MENEGHINITE
---------	----	--------------	----	-------------

0 1 1 1	Meas	ured	Calcul	lated	0.1
Crystal I	$\phi$	ρ	$\phi$	ρ	Quality
010	0°00′	90°00′	0°00′	90°00′	excellent
0.24.11	0 00	20 07	0 00	$20 31\frac{1}{2}$	good
121	46 37	26 35	46 33	$26 \ 30\frac{1}{2}$	excellent
24.24.13	64 32	36 40	$64 \ 39\frac{1}{2}$	$36\ 29\frac{1}{2}$	poor—end of chain
Crystal 2					
010	0 00	90 00	0 00	90 00	excellent
041	0 02	34 36	0 00	34 271	excellent
0.24.11	0 02	20 35	0 00	$20 \ 31\frac{1}{2}$	good
111	65 16	21 36	64 39 <del>1</del>	21 50	very poor
121	46 22	26 52	46 33	$26 \ 30\frac{1}{2}$	Door
131	35 09	32 11	35 08	32 11	excellent
141	28 04	38 00	$27 \ 49\frac{1}{2}$	$37 \ 48\frac{1}{2}$	poor
T41	-28 00	37 55	$-27 49\frac{1}{2}$	37 481	good
211	76 45	37 10	$76 \ 40\frac{1}{2}$	$36 \ 39\frac{1}{2}$	very poor
514	76.20	17 00	54 101	26 201	
211	- 10 32	37 00	- /0 40 <sup>±</sup>	30 392	very poor
241	40 22	43 35	40 33	44 50	very poor
24.24.13	05 10	30 44	04 39 <u>5</u>	30 29 <u>5</u>	poor
24.24.13	-6432	36 34	$-64\ 39\frac{1}{2}$	36 29 <sup>1</sup> / <sub>2</sub>	good
24.24.11	65 16	41 31	64 39 <sup>1</sup> / <sub>2</sub>	$41 \ 09\frac{1}{2}$	poor
24.24.11	-64 32	41 04	$-64\ 39\frac{1}{2}$	$41 \ 09\frac{1}{2}$	poor
24.48.13	46 32	42 50	46 33	42 $38\frac{1}{2}$	excellent
7.21.1	35 32	77 12	35 08	$77\ 12\frac{1}{2}$	poor
7.21.1	35 09	76 43	-3508	$77\ 12\frac{1}{2}$	poor

This table shows clearly that each of these crystals has faces of both normal and aberrant forms, intermingled and all in good position. There is no difference observable in quality between them and in no way could one say that one set was more typical than the other. All the faces are so minute that it is difficult to observe the actual crystal surfaces. It is noteworthy that the aberrant forms are displaced by small angular distances, from faces of simple indices but always in a radial relation, the  $\phi$  angles being alike. It is difficult to picture this relation as being due to any type of twinning.

Two new forms were noted as shown in the table, (0.24.11) and (7.21.1), each with two faces. They are simply to be added to the list of aberrant forms. Q and w, listed in Miers from vom Rath are omitted as very uncertain.

The third crystal measured showed no trace of aberrant forms but yielded a characteristic series in good position, including (010), (100), (140), (380), (120), (110), (021), (041), (101), (111), (121), (131), and (141).

### STRUCTURAL LATTICE OF MENEGHINITE

### by W. E. RICHMOND

The structural lattice was determined from rotation and Weissenberg photographs about the needle axis [001]. The lattice constants computed from the x-ray photographs are:

$$a_0 = 11.29; b_0 = 23.78; c_0 = 4.12$$

giving the ratio:-

a:b:c=0.4750:1:0.1733 in close agreement with the morphological ratio:—

$$a:b:c=0.4736:1:0.1715.$$

The volume of the unit cell  $V_0 = 1103$ ; with the specific gravity 6.358 (vom Rath) this gives a molecular weight for the unit cell of  $M_0 = 4162$ . The content of the unit cell. The analysis of meneghinite by vom Rath has the smallest amount of impurity, so is made the basis of the calculation as shown in the following table.

	1	2	3	4	5	6
Pb	61.47	63.75	.308	13.08	63.92	62.88
Sb	18.37	19.05	.156	6.68	18.77	19.91
S	16.97	17.20	.537	22.85	17.31	17.21
Cu	0.39					
Fe	0.23					
Ins.	0.82					
						1.000
	98.25	100.00			100.00	100.00

TABLE 3. ANALYSIS OF MENEGHINITE

1. Meneghinite from Bottino, Italy. Vom Rath, analyst.

2. Recalculated to 100% after deducting CuFeS2 and insoluble.

3. Atomic proportions of 2.

4. Atomic content of unit cell.

5. Calculated composition of formula Pb<sub>4</sub>Sb<sub>2</sub>S<sub>7</sub>.

6. Calculated composition of formula Pb13Sb7S23.

The figures of column 4 yield the formula  $Pb_{13}Sb_7S_{23}$ , replacing the old accepted formula  $Pb_4Sb_2S_7$ . The density calculated for this formula is 6.391, which compares favorably with the value of vom Rath, 6.358.

#### References

Goldschmidt, V. (1897): Winkeltabellen, p. 238. Krenner, J. A. (1883): *Folt. Közl.*, vol. 13, pp. 297 and 350. Miers, H. A. (1883): *Mineral. Mag*, vol. 5, p. 325. vom Rath, G. (1867): *Ann. Phy. & Chem.*, vol. 132, p. 372. Ungemach, H. (1923): *Zeits. Krist.*, vol. 58, p. 158.

### JORDANITE

The latest account of jordanite is contained in the paper by Solly (1900), in which he traces the history of the mineral to that time and adds new forms to the already long lists of Baumhauer. Solly employs the position and elements of the latter, namely:

# a:b:c=0.4945:1:0.2655 $\beta=90^{\circ}33\frac{1}{2}'$

The author measured two crystals from the type locality, confirming the angles and many of the forms of earlier observers and adding five new forms. The crystals were measured with (010) as pole, and the gnomonic projection at once suggested both to the author and to Dr. Peacock a possible better choice of orientation of the axes. The choice

finally made by Peacock on a morphological basis was confirmed, as is shown below, by *x*-ray study and is accepted as the proper setting.

The new axial ratio, calculated from the old, is:

a:b:c=0.2354:1:0.1397  $\beta=93^{\circ}53'$ 

and its position is related by the transformations

Baumhauer to Peacock $\overline{103}/0\overline{40}/101$ Peacock to Baumhauer $\overline{103}/010/101$ 

This is equivalent to taking (100) Baumhauer as (I01) (010) Baumhauer as (0I0) (001) Baumhauer as (301)

Twinning referred to the new axes is most common on  $\{100\}$  and is often lamellar, yielding a surface of parting; it is common on  $\{001\}$ , rare on  $\{101\}$  and vary rare on  $\{\overline{3}01\}$ .

Cleavage is parallel to {010}, which is the direction of dominant tabular development.

The following forms of Solly's list are omitted, being regarded as vicinal to closely neighboring forms:

Form Solly	Palache	Vicinal to	Diff. in angle to (010)
(12.49.0)	$(\overline{3}.49.3)$	(1.16.1)	28 minutes
(9.32.0)	$(\overline{9}.128.9)$	(1.14.1)	23 minutes
(7.24.0)	(7.96.7)	$(\bar{1}.14.1)$	31 minutes
(047)	(21.16.7)	(321)	84 minutes
(28.3.28)	(28.3.0)	(910)	3 minutes

The following forms are added to the list on the basis of the author's observations:

Syn	nbol				
Palache	Solly	Meas	ured	Calcu	lated
		$\phi$	ρ	φ	ρ
(091)	(391)	∫85°57′	38°30′	86°07′	38°3312′
		86 09	38 32		
(183)	(221)	74 52	70 32	$75\ 05\frac{1}{2}$	70 12
(1.54.1)	(1.28.1)	56 37	9 42	56 28	9 02
$(\overline{3}.14.1)$	(371)	150 04	45 19	$149 \ 46\frac{1}{2}$	45 27
(3.16.1)	(381)	150 00	41 37	$149 \ 46\frac{1}{2}$	41 38

Because no complete angle table has been calculated for jordanite since it was determined to be monoclinic, the author has calculated such a table and gives herewith some of the more important forms. There are 115 forms known.

TABLE	1.	JORDANITE-	-Pb14As7S24
-------	----	------------	-------------

		1	Monoclin a:b:c=0.2 $p_0:q_0:r_0=0.5$	nic; prismation 354:1:0.139 935:0.1394:	c - 2/m 7; $\beta = 93°53'$ 1; $\mu = 86°07'$		
		1	$p_2: p_2: q_2 = 7.1$	747:4.2579:	1; $r_0' = 0.0679$		
	Forms	¢	ρ	$\phi_2 = 0.1357,$	$\rho_2 = B$	С	A
b	010	0°00′	90°00′		0°00′	90°00′	90°00′
G	100	90 00	90 00	0°00′	90 00	86 07	0 00
J	180	$28 \ 01\frac{1}{2}$	90 00	0 00	$28 \ 01\frac{1}{2}$	$88\ 10^{1}_{2}$	$61 58\frac{1}{2}$
Ì	L 160	35 21 <u>1</u>	90 00	0 00	35 21 <sup>1</sup> / <sub>2</sub>	87 45	54 38 <sup>1</sup> / <sub>2</sub>
1	M 150	40 25	90 00	0 00	40 25	87 29	49 35
Ç	2 130	54 50	90 00	0 00	54 50	86 50	35 10
	5 120	$64 \ 50\frac{1}{2}$	90 00	0 00	$64  50\frac{1}{2}$	86 29	25 09 <sup>1</sup> / <sub>2</sub>
7	ı: 101	90 00	33 32	56 28	90 00	29 39	56 28
9	: 101	-90 00	27 47	117 47	90 00	31 40	117 47
s	: 301	-90 00	59 46 <u>1</u>	$149 \ 46\frac{1}{2}$	90 00	63 39 <u>1</u>	149 46 <u>1</u>
1	7 123	70 43	15 45	$75 \ 05\frac{1}{2}$	84 51 <sup>1</sup> / <sub>2</sub>	13 59	$75 \ 09\frac{1}{2}$
ļ	W 163	43 37	21 06	$75 \ 05\frac{1}{2}$	$74 \ 53\frac{1}{2}$	19 51	75 37
1	5 111	78 06	$34\ 06\frac{1}{2}$	56 28	$83\ 21\frac{1}{2}$	30 19	$56\ 43\frac{1}{2}$
i	141	$49\ 51\frac{1}{2}$	40 55	56 28	$65 \ 01\frac{1}{2}$	38 01	59 57
l	161	38 20	46 54	56 28	$55\ 03\frac{1}{2}$	44 34	$63 \ 04\frac{1}{2}$
7	ı 181	30 40	52 25	56 28	$47 \ 01\frac{1}{2}$	50 31	66 09 <u>1</u>
t	1.12.1	21 34	60 59	56 28	35 351	59 37	71 15
e	Ĩ21	-6204	$30\ 48\frac{1}{2}$	117 47	76 07	34 17	$116\ 54\frac{1}{2}$
1	<b>I</b> 41	-43 19	37 31 <sup>1</sup> / <sub>2</sub>	117 47	$63 \ 41\frac{1}{2}$	$40\ 16\frac{1}{2}$	114 42
2	161	-3209	44 43	117 47	$53\ 26\frac{1}{2}$	$46\ 52\frac{1}{2}$	111 59 <sup>1</sup> / <sub>2</sub>
1	181	$-25 \ 14\frac{1}{2}$	51 01	117 47	$45 \ 19\frac{1}{2}$	52 45 <sup>1</sup> / <sub>2</sub>	109 2112
1	5 I.10.1	-20 40	56 11	117 47	$38\ 58\frac{1}{2}$	57 38	107 03
7	· T.12.1	-17 27	$60\ 21\frac{1}{2}$	117 47	$33 59\frac{1}{2}$	$61 \ 35\frac{1}{2}$	$105 \ 06\frac{1}{2}$

# STRUCTURAL LATTICE OF JORDANITE

# by W. E. RICHMOND

The structural lattice was determined from rotation and zero-layer Weissenberg photographs about the axis [010]. The lattice constants computed from the *x*-ray photographs are:

 $a_0 = 7.529$ Å,  $b_0 = 31.87$ Å,  $c_0 = 4.421$ Å;  $\beta = 93^{\circ}59'$ 

giving the axial ratio:

$$a_0: b_0: c_0 = 0.2362: 1: 0.1387; \beta = 93°59'$$

in close agreement with the morphological ratio:

$$a:b:c=0.2354:1:0.1397; \beta=93°53'$$

The volume of the unit cell,  $V_0$ , is 1058.1 cubic Ångstroms; with the specific gravity 6.413 (Jackson) this gives a molecular weight for the unit cell of  $M_0=4103.2$ ; with specific gravity 6.32 (new determination)  $M_0=4053$ .

Using the analysis of Jackson (Solly, 1900), we obtain the figures of Table 2 for the probable content of the unit cell.

	1	2	2	4	5	6
	T	2	3	4	3	0
Pb	68.61	69.22	.334	13.78	13.53	14 or 13
S	18.19	18.36	.577	23.67	23.38	24 or 23
As	12.32	12.42	.166	6.81	6.73	7
	99.12	100.00				

TABLE 2. ANALYSIS OF JORDANITE

1. Jordanite from Binn. Jackson, analyst.

2. Recalculated to 100%.

3. Atomic proportions of 2.

4. Atomic content of unit cell using G=6.413.

5. Atomic content of unit cell using G = 6.32.

6. Rounded out atomic numbers.

The figures show little choice between the formulae  $Pb_{14}As_7S_{24}$  and  $Pb_{13}As_7S_{23}$ . We are inclined to select the former. Table 3 shows the calculated composition and density of both formulae as well as of the generally accepted one of  $Pb_4As_2S_7$ .

TABLE 3. CALCULATED COMPOSITION ANI	DENSITY OF	VARIOUS I	ORMULAE OF	ORDANITE
-------------------------------------	------------	-----------	------------	----------

	$\mathrm{Pb_{14}As_7S_{24}}$	$\mathrm{Pb}_{13}\mathrm{As}_7\mathrm{S}_{23}$	$\mathrm{Pb}_4\mathrm{As}_2\mathrm{S}_7$	
 Pb	69.20	68.13	68.90	
S	18.34	18.61	18.65	
As	12.46	13.26	12.45	
	100.00	100.00	100.00	
G	6.54	6.17	5.63	

Our inclination is to assign to jordanite the formula  $Pb_{14}As_7S_{24}$ . If, however, the alternative formula were selected, the chemical identity with meneghinite would be preserved and the two minerals would be dimorphous. New chemical work will be needed before the final decision on this matter can be made.

#### Reference

Solly, R. H. (1900): Jordanite, *Mineral. Mag.*, vol. 12, p. 290. Analysis by Jackson, same page 289.

### DIAPHORITE

Diaphorite was described by Zepharovich in 1871 as an orthorhombic mineral with the same composition as freieslebenite and nothing of importance has been added since the original description. A specimen of diaphorite from Freiberg in the Karabachek collection yielded a wealth of new data and the results of its study are here presented. Eight crystals were measured in all, four from the new specimen and four from older specimens in the Harvard collection. The latter were similar to the type description and one of them was a twin on the recognized law, twin plane  $\{120\}$ . The crystals from the Karabachek specimen were highly complex in development; one presented one hundred and fourteen faces representing fifty six forms; and by their study more than fifty new forms were added to the twenty four previously known.

On morphological grounds a new fundamental pyramid was selected which was the pyramid  $\{114\}$  of Zepharovich and this choice was confirmed by Winchell's *x*-ray study presented on a later page. The transformation Zepharovich to Palache is  $100/010/00\frac{1}{4}$ .

New elements were calculated from the measurements of fifty faces of twenty-five forms on eight crystals.

 $a:b:c=0.4953:1:0.1840 \\ p_0:q_0:r_0=0.3715:0.1840:1 \\ This ratio is closely comparable with that of Zepharovich, \\ a:b:\frac{1}{4}c=0.4919:1:0.1838$ 

The table following presents the observations made on these crystals in condensed form, the calculated angles being based on the new elements. The known forms were all found with the exception of the following five:—

		$\phi$	ρ	
α	1.11.0	10°24′	90°00′	
k	5.12.0	40 04	90 00	
q	0.20.3	0 00	50 48 <sup>1</sup> / <sub>2</sub>	(probably vicinal to {071}, a form missing in the series of domes with $\rho = 52^{\circ}10\frac{1}{2}'$ )
d	141	26 47	39 30	
Ś	241	45 16	46 17	

		Cala	ulotod	Oharm		Range		No	o. of	
		φ	ρ	φ	eα, mean ρ	φ	ρ	faces	Cryst	s. Qual
*с	001	o/	0°00′	o/	0°00′	0 / 0 */		3	3	fair
b	010	0 00	90 00	0 06	90 00	$0 \ 00 - 0 \ 16$		6	4	poor
a	100	90 00	90 00	89 50	90 00	89 27 90 00		7	7	good
*β	170	$16 \ \theta 5\frac{1}{2}$	90 00	16 39	90 00	16 08 -17 10	-	2	2	poor
$*\gamma$	160	18 36	90 00	$18 \ 36\frac{1}{2}$	90 00	$18\ 25\ -18\ 48$	÷	2	2	good
ρ	150	21 59	90 00	22 07	90 00	22 00 -22 12	i	3	3	fair
* <sub>σ</sub>	140	26 47	90 00	26 32	90 00	25 55 -27 10	222	2	2	poor
π	130	33 56	90 00	33 57	90 00	33 45 -34 08		11	5	fair
n	120	45 16	90 00	45 12	90 00	44 43 -45 34	5.55	13	8	good
* \epsilon	230	53 23	90 00	53 43	90 00	—	-	1	1	good
m	110	63 39	90 00	63 40	90 00	63 05 -64 00	-	14	8	good
*x	320	71 431	90 00	71 40	90 00	71 30 -71 49	<del>)</del>	2	1	good
*δ	210	$76\ 05\frac{1}{2}$	90 00	76 271	90 00	76 25 -76 30		2	2	poor
t	310	80 371	90 00	81 52	90 00	81 17 -82 27		2	1	poor
*h	011	0 00	$10\ 25\frac{1}{2}$	0 00	9 01	_		1	1	fair
u	021	0 00	20 12	0 00	20 18		20°00′-21°00′	8	5	poor
r	041	0 00	36 21	0 00	36 32	_	36 15 -36 41	6	4	good
v	061	0 00	47 50	0 00	47 36		47 32 -47 42	2	2	fair
w	081	0 00	55 48 <del>1</del>	0 00	55 50		55 28 - 56 09	12	8	good
*f	0:10.1	0 00	61 28 <del>1</del>	0 00	61 28	—	61 15 -61 43	5	3	fair
*g	0.12.1	0 00	65 38	0 00	$65 38\frac{1}{2}$	_	65 15 -66 47	3	2	fair
*j	0.14.1	0 00	68 47	0 00	68 38			1	1	good
$\psi$	201	90 00	$36 \ 36\frac{1}{2}$	90 00	36 30 <sup>1</sup> / <sub>2</sub>		36 11 - 36 47	7	6	good
x	401	90 00	56 03 <sup>1</sup> / <sub>2</sub>	90 00	56 07	_	55 52 -56 15	7	6	good
*A	112	63 39	11 42 <sup>1</sup> / <sub>2</sub>	63 57	11 47	63 32 -64 22	11 34 -12 00	2	2	poor
i	111	63 39	22 31	63 54	22 34	63 32 -64 27	22 06 -23 03	8	4	good
*B	332	63 39	31 52 <del>1</del>	63 45	31 43	63 38 - 63 53	31 41 -31 45	3	2	fair
У	221	63 39	39 39 <u>1</u>	63 37	39 42	63 04 -64 22	39 28 -40 02	13	8	good
*C	331	63 39	51 12	$63\ 35\frac{1}{2}$	51 21	63 22 -63 53	51 00 -51 53	5	4	fair
*D	441	63 39	58 54 <u>1</u>	63 50	58 48	63 47 -63 53	58 47 -58 49	2	2	poor
*E	551	63 39	$64 \ 14\frac{1}{2}$	63 58	$64 \ 11\frac{1}{2}$	63 53 -64 01	64 00 -64 41	4	3	poor
*F	133	33 56	$12 \ 30\frac{1}{2}$	33 45	12 44	33 10 -34 20	12 42 -12 46	2	1	poor
*G	173	16 05	$24 \ 04\frac{1}{2}$	16 40	24 27	16 25 -16 55	24 12 -24 42	2	2	very good
*П	132	33 56	18 24	33 52	18 34	33 11 -34 20	18 20 -18 48	3	2	fair
*J	172	16 05	33 50	$16\ 17\frac{1}{2}$	33 52	15 47 -16 50	33 43 - 34 05	5	2	fair
*K	192	12 38 <sup>1</sup> / <sub>2</sub>	40 19	12 45	$40\ 26\frac{1}{2}$	12 23 -13 07	40 07 -40 33	4	2	good
$^{*L}$	283	26 47	$28 \ 47\frac{1}{2}$	26 31	28 46	26 18 - 26 56	28 38 - 28 55	3	3	poor
M	2.22.3	10 24	$53 54\frac{1}{2}$	10 00	54 00		0.00	1	1	good
0	131	33 56	$33 \ 38\frac{1}{2}$	34 00	33 38	33 55 -34 03	33 30 -33 50	4	4	very good
*N	151	21 59	$44 \ 46\frac{1}{2}$	22 00	44 39	21 50 -22 16	44 16 -44 56	3	2	poor
*0	171	16 05	$53\ 16\frac{1}{2}$	$16\ 04\frac{1}{2}$	53 19	$15 \ 40 \ -16 \ 50$	53 00 -53 39	8	3	fair
*P	191	12 381	59 29 <del>1</del>	12 38	59 29	12 16 -12 59	59 20 -59 26	3	2	good

Diaphorite Table of Calculated and Observed Angles

		Calculated		Observe	d mean	Ra	unge	No	. of	Oual
		φ	ρ	φ	ρ	φ	p	faces	Crysts	
*0	1.11.1	10 24	64 05	10 29	64 071	10 20 -10 38	64 05 -64 10	2	2	good
$R^*$	1.13.1	8 491	67 33	8 35	67 56	8 30 - 8 40	67 52 -68 00	2	2	poor
*S	312	$80 \ 37\frac{1}{2}$	29 27 $\frac{1}{2}$	80 331	29 57	80 27 -80 40	29 35 -30 19	2	2	poor
*T	352	50 271	35 51	50 34	35 48 <del>1</del>	50 12 -50 56	35 40 -35 57	2	2	poor
$^{*}U$	392	33 56	$44  56\frac{1}{2}$	34 03	44 53	34 02 -34 05	44 28 - 45 05	3	3	poor
*V	211	$76 \ 05\frac{1}{2}$	37 251	76 02	37 30	_	3 <del>11</del> 3	1	1	poor
*\eta	251	38 55 <u>1</u>	49 47	38 55	49 40	_	-	1	1	poor
*0	261	33 56	53 041	33 59 <sup>1</sup> / <sub>2</sub>	$53 \ 04\frac{1}{2}$	33 38 -34 20	52 33 -53 18	7	3	good
* i	281	26 47	58 46	26 46	58 44	26 38 - 26 50	58 40 -58 50	3	2	good
*к	2.10.1	21 59	63 15	21 58	63 29	21 50 -22 16	63 06 -63 47	6	3	poor
*λ	2.12.1	18 36	66 46	18 471	66 37 <del>1</del>	$18 \ 40 \ -18 \ 48$	66 22 -66 53	3	2	poor
*μ	2.14.1	16 05	69 32 <u>1</u>	16 04	69 25	16 00 -16 08	69 00 -69 50	2	2	good
ω	311	80 37 <sup>1</sup> / <sub>2</sub>	48 28 <sup>1</sup> / <sub>2</sub>	80 321	48 30	80 16 -80 43	48 17 - 48 42	5	4	poor
*v	351	50 27불	55 19	50 26	55 22	50 12 -50 35	55 00 -55 42	3	3	fair
*ξ	391	33 56	$63\ 23\frac{1}{2}$	34 071	63 21	34 02 -34 20	62 52 -63 45	4	3	good
* <sub>T</sub>	3.13.1	24 59	69 14 <u>1</u>	24 54	69 07	24 54 -25 02	69 00 -69 14	2	2	good
$^{*}\phi$	712	85 57	52 30	86 13	52 50	85 55 -86 32	52 36 - 52 57	3	2	poor
z	421	76 051	56 50 <sup>1</sup> / <sub>2</sub>	76 08	56 54	75 50 76 20	56 30 -57 00	5	4	very good
*W	431	69 37	57 45	68 40	57 58	68 31 -68 50	57 56 -58 00	2	2	fair
*X	4.16.1	26 47	73 08	26 56	$72 56\frac{1}{2}$	26 46 -27 02	72 35 -73 18	4	2	poor
*Z	511	84 201	61 49	84 12	62 05	83 5684 31	61 55 -62 13	3	2	very good
е	531	73 27	62 42	73 30½	$62 \ 48\frac{1}{2}$	73 20 -73 45	62 30 -63 12	12	6	good
*Δ	621	80 371	66 071	80 42	66 07	80 34 -80 52	66 00 66 15	4	3	very good
*Λ	641	71 431	$66\ 55\frac{1}{2}$	71 44	67 03	71 4071 48	66 46 -67 20	2	2	good
*Ξ	711	85 57	69 00 <u>1</u>	86 08	69 30	86 06 - 86 09	69 21 -69 40	2	2	good
*Σ	731	78 01	69 23	$78 \ 01\frac{1}{2}$	69 38 <u>1</u>	77 55 -78 22	69 30 -69 47	3	3	good
*Θ	841	76 051	71 541	76 24	72 05	76 21 -76 26	71 45 -72 19	3	2	fair
*Y	971	68 56	74 24	68 57	74 32			1	1	good
$^{*\Omega}$	10.2.1	84 201	75 00	84 40	75 18	5 <del></del>	<del></del>	1	1	very good
*Ψ	16.2.1	86 271	80 28	86 12	80 26			3	1	very good

DIAPHORITE—Continued

\* Denotes new form.

The prism zone is strongly developed and is striated but distinct faces of  $\{100\}$ ,  $\{130\}$ ,  $\{120\}$ , and  $\{110\}$  are nearly always present. The termination is generally dominated by some or all of the domes  $\{021\}$ ,  $\{041\}$ ,  $\{081\}$ ,  $\{201\}$ , and  $\{401\}$ . The only pyramids commonly present are  $\{221\}$  and  $\{531\}$ . The crystals are minute and the faces are in most cases not sharply outlined. Nevertheless the angles, as shown in the table, are very consistent and the signals were good for such small faces.

## X-RAY STUDY OF DIAPHORITE AND FREIESLEBENITE by Horace Winchell

In order to better establish the relationship between the two minerals diaphorite and freieslebenite, long held to be dimorphous, an x-ray study was undertaken, employing crystals studied goniometrically, and including a discussion of their chemical character in view of the structural results.

Diaphorite. Weissenberg photographs about the b and c axes, on the zero and the first layers were studied, as well as rotation photographs about all three axes, using  $Cu_{\kappa\alpha}$  radiation. The axial lengths as given below were derived from weighted averages in which the higher order values received the greatest weight.

 $a_0 = 15.83$  Å,  $b_0 = 32.23$  Å,  $c_0 = 5.89$  Å  $a_0: b_0: c_0 = 0.491: 1:0.183$ 

This ratio agrees well with the morphological value given above and confirms the choice of the unit form.

The volume of the unit cell,  $V_0 = 3007$  cubic Ångstroms, with the mean observed value of the density, 5.97, gives for the molecular weight of the unit cell  $M_0 = 10879$ .

The following space group criteria were derived from the zero and first layer Weissenberg photographs about the c axis, and the zero layer about the a axis:—

hkl present only for k even hk0 present only for h and k even h0l present only for h even 0kl present only for k even

which defines the space group as  $D_{2h}^{21}(Cmma)$ .

There are two authentic analyses of diaphorite, both on material from Pribram.

	1	2	3	4	5	6	7
Cu	0.73	0.011					
Fe	0.67	0.012	26.11	23.36	0.216	23.52	23.80
Ag	23.44	0.217					
Pb	28.67	0.138	15.01	31.56	0.152	16.56	30.48
Sb	26.43	0.217	23.60	25.92	0.213	23.20	26.86
S	20.18	0.629	68.40	18.51	0.578	62.90	18.87
	100.12			99.35			100.01

1. Analysis by Helmhacker, 1864.

2. Atomic ratios.

3. Number of atoms in unit cell calculated from  $M_0$ .

4. Analysis by Moranski, 1878.

5. Atomic ratio.

6. Atoms in unit cell.

7. Calculated composition for Ag<sub>3</sub>Pb<sub>2</sub>Sb<sub>3</sub>S<sub>8</sub>.

The mean of the numbers in columns 3 and 6 when rounded out may be taken as 24:16:24:64 or 8 (3:2:3:8) which lead to the formula  $Ag_3Pb_2Sb_3S_8$ , with 8 molecules in the unit cell.

Freieslebenite. No new data on the morphology of this mineral were obtained. A single measurable crystal was found on a specimen from Hiendelencina, Spain, which confirmed the published angles. This crystal was used for the x-ray study. Rotation photographs about b and c, and Weissenberg photographs of the zero and first layers about c were obtained. The cell dimensions obtained are:—

 $a_0 = 7.53$  Å,  $b_0 = 12.79$  Å,  $c_0 = 5.88$  Å,  $\beta = 92^{\circ}14'$  (morphologic)  $a_0: b_0: c_0 = 0.589: 1:0.460$ a: b: c = 0.5871: 1:0.9277  $\beta = 92^{\circ}14'$  (Miller)

It follows that c must be halved and the transformation formula, Miller to Winchell reads  $100/010/00\frac{1}{2}$ 

The volume of the unit cell,  $V_0 = 567$  cubic Ångstroms; the specific gravity is 6.23 (Payr), 6.20 (Winchell); these values give as the molecular weight of the unit cell,  $M_0 = 2145$ .

The space group of freieslebenite is derived from the following systemic criteria:—

hkl present in all orders h0l present only for h even 0k0 present only for k even

Assuming that the crystal class is holohedral the space group is  $C_{2h}^{5}(P2_1/n)$ .

The only chemical analysis of freieslebenite which is accompanied by a density determination consistent with our data is that of Payr (1860) made on material from Pribram.

	1	2	3
Fe	0.63	0.11	4 83
Ag	23.08	.214	1.00
Pb	30.77	.148	3.11
Sb	27.11	.223	4.78
S	18.41	.574	12.31
	100.00		25.03

1. Analysis by Payr; density 6.23.

2. Atomic ratios.

3. Number of atoms in the unit cell.

The numbers in column 3 approximate a total of 25 atoms and justify the formula  $Ag_5Pb_3Sb_5S_{12}$ , with one molecule in the unit cell.

	SUMMARY TABLE	
	Diaphorite	Freieslebenite
Formula	$8(Ag_3Pb_2Sb_3S_8)$	$Ag_5Pb_3Sb_5S_{12}$
Symmetry	orthorhombic	monoclinic
	a <sub>0</sub> 15.83 Å	a <sub>0</sub> 7.53 Å
X-ray elements	b <sub>0</sub> 32.23 Å	<i>b</i> <sub>0</sub> 12.79 Å
	c <sub>0</sub> 5.89 Å	$c_0$ 5.88 Å
Axial ratio	a:b:c=.4953:1:.1840	a:b:c=.5871:1:.4638
		$\beta = 92^{\circ}14'$
Specific gravity measured	5.90-6.04	6.20-6.23
Specific gravity calculated	5.97	6.27

The crystallographic and other physical properties of the two minerals considered in this study are shown to be in greater contrast than was before evident; the chemical differences are believed to be real and seem to disprove the supposed dimorphism. Need for new analyses on material physically studied is evident before a final decision can be reached.

## SUMMARY TABLE