PERMINGEATITE, Cu₃SbSe₄, FROM PŘÍBRAM (CZECH REPUBLIC): DESCRIPTION AND RAMAN SPECTROSCOPY INVESTIGATIONS OF THE LUZONITE-SUBGROUP OF MINERALS

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Abstract

Permingeatite from a new occurrence, the Příbram uranium-base metal ore district (central Bohemia, Czech Republic), has been studied in detail. Its occurrence is similar to the type occurrence in Předbořice (Czech Republic). Based on electron microprobe analysis, the empirical formula of the studied permingeatite (mean of 10 point analyses, based on Sb + Cu + Fe + Se + S = 8 *apfu*) is Cu_{3.00}(Sb_{0.99}Fe_{0.04})_{Σ 1.03}(Se_{3.74}S_{0.23})_{Σ 3.97}. The unit-cell parameters refined from powder X-ray diffraction data are *a* 5.6323(2) Å, *c* 11.2354(7) Å, with *V* 356.41(2) Å³ (for space group *I*42*m*). The Se-S substitution is characteristic for the permingeatite from Příbram, in contrast to permingeatife from Předbořice, where the As-Sb substitution dominates. Regarding substitution trends, only two, Sb-As and S-Se, play a role in the luzonite group minerals. The only other element (as documented by the present study) which enters the structure of the luzonite-group minerals. The dominant feature in the Raman spectroscopy was used to characterize vibrational properties of the luzonite group of minerals. The dominant feature in the Raman spectra of the studied minerals is a suite of spectral bands that corresponds to the stretching and bending vibrations of tetrahedral XY_4 (X = As/Sb; Y = S/Se) groups. Intrinsic shifts in observed energies for distinct studied minerals are connected with different elemental compositions of the tetrahedral groups. The reflectance data obtained from the studied permingeatite are close to the published data, but more significant differences were found, particularly in the areas of 420–500 nm and about 640 nm. The studied association was formed, based on the presence of umangite, at temperatures below 112 °C.

Keywords: permingeatite, luzonite subgroup, Raman spectroscopy, reflectance, selenide minerals, Příbram, Czech Republic

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INTRODUCTION

Permingeatite was originally described as a new mineral species from the Předbořice uranium deposit (Czech Republic) by Johan et al. (1971). It is a member of the luzonite subgroup of the stannite group, along with luzonite, Cu₃AsS₄ (Weisbach 1874), and famatinite, Cu₃SbS₄ (Stelzner 1873). All three are isostructural, crystallizing in the space group $I\overline{4}2m$ with a derivative of the sphalerite structure. Enargite is an orthorhombic analogue of luzonite with a wurtzite-type structure (Pauling & Weinbaum 1934). The Se-analogue of both luzonite and enargite is not known. The complete solid-solution series between luzonite and famatinite was discovered by Pfitzner & Bernert (2004). There are a few investigations of crystal structures and phases of the famatinite-luzonite series, primarily carried out on synthetic compounds (Pfitzner 1994, Pfitzner & Reiser 2002, Pósfai & Buseck 1998), but also of natural materials (Gaines 1957, Marumo & Nowacki 1967).

The current study aims to present new experimental data for natural permingeatite, obtained using specimens from the uranium-base-metal district near Příbram (central Bohemia, Czech Republic) (Fig. 1). Permingeatite is an extremely rare Se mineral found only in uranium deposits in the Czech Republic and mostly with microscopic dimensions. We used the opportunity to study rare macroscopic permingeatite by electron microprobe, reflectance, powder X-ray diffraction, and Raman spectroscopy. The newly obtained results are compared here with those previously published, and are further discussed in broader relation to the luzonite subgroup of minerals. This paper is part of our systematic study of selenides from various types of Czech occurrences (Litochleb et al. 2004, Sejkora et al. 2006, 2011, Škácha et al. 2009a,b, 2010, Topa et al. 2010) in order to better understand the evolution of Se-mineralization in similar camps.



FIG. 1. Simplified geological map of the Příbram area.

GEOLOGY AND MINERALOGY

A new occurrence of the rare species permingeatite, Cu₃SbSe₄, has been discovered in the uranium and basemetal district near Příbram, Central Bohemia, Czech Republic, where specimens with permingeatite were found in dump material from the "No. 11A" (Bytíz) uranium mine. This uranium and base-metal district is situated in an exocontact of the Central Bohemian plutonic complex with the Teplá-Barrandien sedimentary unit. Uranium mineralization occurs in a zone 25 km in length and up to 2 km wide. The hydrothermal ore veins are concentrated in so called "ore nodes". Shaft No. 11A is located in the most important "Bytíz ore node", in the central part of the district. In the "Bytíz ore node" significant uranium reserves were located in veins Bt4 and Bt4H, with totals of 3673 and 1810 t of mined U. This ore node overall provided 52% of the production of the entire Příbram uranium district. Shaft 11A exposed the upper parts of the veins from the surface to a depth of 1500 m. The underground mining of uranium in Příbram area took place from 1948 to 1991 (Ettler et al. 2010).

There are four main mineralization stages in the Příbram deposit: (1) siderite-sulphidic; (2) calcite; (3) calcite-uraninite; and (4) calcite-sulphidic. Selenides are closely related to uraninite in carbonate veins within the third mineralization stage. Uranium ore is represented by dominant uraninite (pitchblende form), coffinite, and "antraxolite". Uraninite dating by U-Pb gave an age of 265 ± 15 Ma (Legierski 1973). Uraninite from the nearby Příbram base-metal deposit has similar ages $(269.8 \pm 20.3 \text{ Ma and } 263.2 \pm 8.9 \text{ Ma})$ (Škácha *et al.* 2009a). Selenium mineralization is relatively rare at Příbram and was first mentioned by Růžička (1986) and more recently by Litochleb et al. (2004), who recognized an assemblage of selenides on a historical specimen from the 1950's in the National museum's collection (Prague, Czech Republic). Large amounts of selenides were recently found in dump material from shafts no. 11A Bytíz and no. 16 Háje (Škácha et al. 2009a,b, 2010). We suggest that the majority of this material originated from the Bt4 vein, because this vein was the most important vein in the deposit, accounting for the majority of mined material with uranium ores. The Příbram carbonate veins containing uraninite originated from low-temperature (80–140 °C) and low-salinity fluids, according to the fluid inclusion study (Žák & Dobeš 1991). The origin of selenides in carbonate veins follows the crystallization of uraninite, as the redox potential of the hydrothermal fluids fell and the fugacity of Se increased (Dymkov 1985).

DESCRIPTION OF THE SAMPLE

The permingeatite-bearing specimen analyzed in this study, originally about $15 \times 10 \times 10$ cm in size, is made up of a complicated system of calcite veins of 1

cm average thickness, maximally about 2 cm. Calcite is strongly hematitized at the edges, and is rarely milky white, but more often black due to numerous inclusions of selenides and sulphides, at the cores. Local country rock has been strongly hematitized.

Permingeatite is found rarely as large irregular grains up to 1 mm in size, growing in groups in brownish or blackish younger carbonate (mostly calcite) veins in association with chalcopyrite, hematite, uraninite, löllingite, clausthalite, hakite, berzelianite, umangite, and eskebornite. It forms compact intergrowths with hakite and eskebornite. Permingeatite grains frequently have thin rims of clausthalite (Fig. 2). Fine-grained aggregates of permingeatite are macroscopically bronze brown in color with a yellowish tint, showing strong metallic luster. Uraninite is locally abundant on the specimen occurring in kidney-shaped black aggregates with strong glassy luster.

EXPERIMENTAL

Quantitative chemical data for the studied samples were obtained using a Cameca SX100 electron microprobe (at the Laboratory of Electron Microscopy and Microanalysis of Masaryk University and Czech Geological Survey, Brno) in the wavelength-dispersive mode with an accelerating voltage of 25 kV, a specimen current of 20 nA, and a beam diameter of about 1 μ m. The following standards and X-ray lines were used: Cu (CuK α), Ag (AgL α), PbSe (SeL β), CuFeS₂ (SK α), CdTe (CdL β), PbS (PbM α), Sb (SbL β), and Bi (BiM β). Peak counting times (CT) were 20 s for all elements, and CT for the background was one half that of the peak time. Raw intensities were converted to concentrations using PAP (Pouchou & Pichoir 1985) matrix-correction software.

X-ray powder diffraction data were collected with a PANalytical X'Pert Pro powder diffractometer equipped with an X'Celerator solid-state detector and a secondary graphite monochromator using CuK $\alpha_{1,2}$ radiation (40 kV and 30 mA). Data were measured over the range between 10 and 100° 20 with 0.02° step size and a counting time of 10 s per step. Peak positions and intensities were determined using the XFIT program using the Pearson VII profile function (Coelho & Cheary 1997). Relative intensities were obtained with the High-Score program (PANalytical 2003). The unit-cell parameters of permingeatite were refined from the powder X-ray diffraction data using the UnitCell software (Holland & Redfern 1997).

A Renishaw RM-1000 Raman microscope was used to collect Raman spectra from separated permingeatite grains mounted in a polished section. Raman scattering was excited with an Ar laser of wavelength 514.5 nm operating at a power of 6 mW (about 1 mW on the sample) and spectra were recorded in backscattering geometry. The frequency calibration was done on Si wafer. The estimated diameter of the laser spot used was 2–3 μ m. A grating of 2400 lines/mm, providing a spectral resolution of better than 1.5 cm⁻¹, and a NEXT grating filter enabled good stray light rejection in the 15–550 cm⁻¹ range. The spectra were acquired from different spots on the permingeatite grain in order to obtain spectra from different crystal orientations. Spectral manipulations were performed using the Omnic Spectral Tools software. Gaussian/Lorentzian profile functions of the band-shape were used to obtain decomposed band components of the spectra. The decomposition was based on the minimization of the difference in the observed and calculated profiles, until the squared correlation coefficient (r^2) was greater than 0.995.

Quantitative reflectance values were obtained with a Leica microscope (objective 100×) equipped with a J & M Tidas MSP 400 spectrophotometer. After careful levelling of the specimen and standard (Zeiss, WTiC), measurements were undertaken at intervals of 5 nm from 400 to 700 nm in air. Weak anisotropy made it difficult to orient the measured grains precisely at their extinction positions, but 14 grains were checked photometrically for maximal and minimal values.

CHEMICAL COMPOSITION

According to previously published data on minerals of the luzonite-group, the types of substitution trends are very limited, except for those primarily involving Sb-Se-As. The only other element found to be entering the structures of the luzonite-group minerals in higher concentrations is Fe. We assume that the Fe is trivalent, similar to other minerals of the stannite group (Evstigneeva et al. 2003). Electron-microprobe data of permingeatite from Příbram (Table 1) indicates significant Se-S substitution, corresponding to the permingeatite-famatinite solid solution series, and minor contents of Fe, which does not clearly correlate with Cu, but Cu-Fe substitution, such as seen in stannite, is possible. The empirical formula (mean of 10 point analyses; on the basis of Sb + Cu + Fe + Se + S = 8*apfu*) is $(Cu_{3,00}Fe_{0,04})_{\Sigma_{3,04}}Sb_{0,99}(Se_{3,74}S_{0,23})_{\Sigma_{3,97}}$. The relation between As and Sb apfu contents is shown in Figure 3. The observed Se-S substitution, according to the X-ray map, does not show any evolutionary trends or evidence of zoning.

CRYSTAL STRUCTURE

The powder diffraction data for permingeatite from Příbram is presented in Table 2. The refined unit-cell parameters are a 5.6323(2) Å, c 11.2354(7) Å, with V 356.41(2) Å³ (for space group $I\overline{4}2m$). The unit-cell parameters of permingeatite from Příbram are similar to those reported by Johan *et al.* (1971) for permingeatite from the type-locality Předbořice deposit (Table 3), and are smaller compared to those of synthetic perm-

ingeatite as reported by Pfitzner (1994). While there is an obvious Se-S substitution in permingeatite from Příbram, permingeatite from Předbořice is characterized by As-Sb substitution. Those data are consistent with



FIG. 2. Permingeatite grain in reflected light (grey with purple tint) with a rim of clausthalite (silver white) and thin purple veins of hakite.

the radii of the ions involved in the substitution in all cases. So it is obvious, according to our and published data, that As and S substitution in permingeatite leads to lower the cell parameters.

There are no mentions of the valence of Sb/As in minerals of the luzonite group in literature, but we presume that the valence of Sb/As in the permingeatite structure is 3+.

RAMAN SPECTROSCOPY

In the structure of permingeatite (Pfitzner 1994) SbSe₄ and CuSe₄ tetrahedra are linked through shared corners (Fig. 4). In the primitive spectroscopic unitcell three Cu, one Sb, and four Se atoms are localized (Table 4). The irreducible representations, representing the total number of the normal modes (k = 0), are Γ_{red} = 2A₁(R) + A₂ (-) + 2B₁ (R) + 5B₂ (IR,R) + 7E (IR,R), where IR and Raman activities of the modes are given in parentheses. The total number of degrees of freedom is half (24 instead of 48), because in *I*42*m* symmetry the content of the primitive unit cell is half with regard to *Z*. One B₂ and one E mode are translations (corresponding to the three acoustic branches), and the A₂ (inactive)



FIG. 3. Plot *apfu* Sb *versus apfu* As for permingeatite and the famatinite-luzonite series. Famatinite (Sejkora unpublished data) occurrences: Quiruvillca, La Libertad (Peru), Julcani, Huancavelica (Peru), Laurani, La Paz (Bolivia); famatinite (RRUFF) occurrences: Quiruvillca, La Libertad (Peru), Mohawk mine, Nevada (USA). Luzonite (Sejkora unpublished data) occurrences: Nevados de Famatina, La Rioja (Argentina), Castrovirreyna, Huanvelica (Peru); luzonite (RRUFF) occurrences: Mohawk mine, Nevada (USA) and Huranon, Pasco (Peru). Dashed line is the ideal correlation Sb:As = 1:1. Correlation relation for the whole set of data in the plot is y = 0.9679 - 0.9535x and r^2 value for this relation and data is 0.989.

and one E mode are rotations. Thus, the number of the optical normal modes (without the acoustic ones) is $\Gamma_{vib} = 2A_1 + A_2 + 2B_1 + 4B_2 + 6E$, and as the E modes are doubly degenerated, the total number of degrees of freedom remains 21. In total, 14 Raman active vibrations can be detected, corresponding to vibrations of one SbSe₄ and one CuSe₄ tetrahedra. The orientations of the corresponding Raman tensors are also listed in Table 4.

Results of the band-component analysis of permingeatite, famatinite, and luzonite spectra (Fig. 5) are listed in Table 5, which gives the position of each vibration, its relative intensity, and the full width at half maximum (FWHM). The dominant feature of all permingeatite spectra measured from different spots on the mineral grain is the presence of intense vibration bands in the 250–150 cm⁻¹ region. Lower frequency vibrations (below 100 cm⁻¹) should correspond to external modes, or translations of Cu and Sb atoms, and the middle-frequency bands to internal modes of the Se-tetrahedra (Fig. 6).

According to the literature (Nakamoto 1986), the main middle-frequency bands can be assigned to stretching vibrations of the (Sb/Cu)Se₄ tetrahedra. Copper atoms occupy two different sites in the lattice, forming two types of tetrahedra with different internal bond-lengths. Two different crystallographic Cu sites could also affect the number of external modes. The internal vibrations of isolated tetrahedra have been studied and measured for some compounds (Naka-

moto 1986). In the case of the permingeatite structure, there is a network of linked tetrahedra which are not isolated; nevertheless, some comparison can be done. The normal tetrahedral vibrations in cubic symmetry can be divided into v_1 (A₁, symmetric stretching), the doublet v_2 (E, bending), and the triplets v_3 (F₂, antisymmetric stretching) and v_4 (F₂, antisymetric bending). In tetragonal symmetry, the doublet E splits into A₁ + B₁ and triplets F₂ split into B₂ + E modes.

TABLE 1. ELECTRON MICROPROBE COMPOSITIONS FOR PERMINGEATITE FROM PŘÍBRAM

	Mean	Range					
wt.%		min	max				
Cu	30.76	30.11	31.45				
Sb	19.48	18.98	19.89				
Fe	0.35	0.18	0.86				
Se	47.62	45.84	48.60				
S	1.17	0.83	2.26				
Sum	99.51						
Cu*	2.999	2.926	3.059				
Sb*	0.992	0.961	1.017				
Fe*	0.039	0.020	0.095				
Se*	3.737	3.546	3.830				
S*	0.226	0.160	0.431				

*based on 8 *apfu*. (*n* = 10)



FIG. 4. Crystal structure of permingeatite (after Pfitzner 1994) in general projection, showing mutually interconnected CuSe₄ (grey) and SbSe₄ (azure) tetrahedra. Cu atoms are green, Se atoms are red. Unit-cell edges are outlined by a solid line.

From the literature it is known that famatinite spectra show the (Sb/Cu)S₄ modes v_1 (A₁), v_3 (B₂ + E), with frequencies around 366 and 380 cm⁻¹ respectively (Nakamoto 1986). The replacement of S by Se in permingeatite should result in a substantial decrease of these frequencies [more than 100 cm⁻¹, as observed in Nb(S/Se)₄ or Mo(S/Se)₄ tetrahedra; see Nakamoto 1986]. This is confirmed in our spectra: $v_1(A_1) \sim 180$, v_3 (B₂ + E) ~ 220 cm⁻¹. The A₁ mode frequency should not depend

TABLE 2. X-RAY POWDER DIFFRACTION PATTERN OF PERMINGEATITE FROM PŘÍBRAM

h	k	Ι	d _{obs}	$d_{\rm calc}$	/// ₀
0	1	1	5.033	5.035	6
1	1	0	3.985	3.983	1
1	1	2	3.248	3.249	100
0	1	3	3.121	3.119	8
0	2	2	2.5227	2.5175	1
1	2	1	2.4602	2.4578	2
1	1	4	2.2942	2.2954	<1
1	0	5	2.0887	2.0871	<1
2	0	4	1.9893	1.9887	22
0	0	6	1.8731	1.8726	<1
0	3	1	1.8543	1.8518	<1
3	1	2	1.6978	1.6978	10
1	3	4	1.5061	1.5042	<1
2	3	3	1.4438	1.4417	<1
0	4	0	1.4088	1.4081	1
3	1	6	1.2919	1.2905	2
2	4	4	1.1497	1.1492	2
3	3	6	1.0842	1.0830	<1

much on the central atom; however, the antisymmetric $(B_2 + E)$ vibration could be doubled due to the presence of Cu and Sb. As the Cu atom is almost half as light as Sb, we could expect higher frequencies for vibrations where Cu is involved. Further assignment concerning bending modes would be rather speculative due to the mixing of the normal modes in the case of non-isolated tetrahedra. Modes with A₁ symmetry arise from the symmetric stretching movement of the Se atoms. They have the highest Raman intensity in famatinite and luzonite (see Fig. 5), therefore the highest intensity band at ~185 cm⁻¹ is attributable to this symmetric stretching vibration v₁ (A₁) (Cu/Sb)Se₄. The band at ~130 cm⁻¹ was ascribed to the bending mode v₄ because of its lower intensity and lower frequency.

As we are not able to establish the crystal grain orientation in the matrix of the polished-section, we are not able to determine and assign the precise character (symmetry) of a given vibration mode. Therefore the assignments of the low frequency bands are only tentative. The external modes comprise vibrations of translational character $B_2 + E$ (Raman active) and librations $A_2 + E$ (where A_2 is only IR active). We can deduce that librations arise from movements of Se atoms coordinating a central atom represented by Cu or/and Sb. In Figure 5 several bands are observed; those at ~60 cm⁻¹ and $\sim 20 \text{ cm}^{-1}$ correspond to emission lines of the laser. and the bands at the lowest wavenumbers should not be considered, due to the absorption edge of the NEXT filter below 20 cm⁻¹. Raman bands at 75, 78, 45, and 40 cm⁻¹ should be connected with external modes translations and librations. The presence and origin of the weak intensity band at 51 cm^{-1} is questionable, but could be assigned to the translational B₂ mode.

	permingeatite this work	permingeatite Johan <i>et al</i> . (1971)	permingeatite Pfitzner (1994)	luzonite Downs (2006) RRUFF ID R060390	luzonite Downs (2006) RRUFF ID R060390
a [Å]	$\begin{array}{r} {}^{(1)}Cu_{3.00}(Sb_{0.99}\ Fe_{0.04})_{\Sigma 1.03}\\ (Se_{3.74}S_{0.23})_{\Sigma 3.97}\\ 5.6323(2)\end{array}$	Cu _{3.01} (Sb _{0.89} As _{0.12}) _{Σ1.01} Se _{3.98} 5.631(2)	Cu ₃ SbS ₄ 5.6609(8)	Cu _{3.01} (As _{0.85} Sb _{0.15}) _{Σ1.00} S ₄ 5.339(6)	⁽²⁾ Cu _{3.01} (As _{0.75} Sb _{0.28}) _{Σ1.03} S ₄ 5.3193(9)
c [Å]	11.2354(7)	11.230(5)	11.280(5)	10.60(1)	10.543(2)
V [Å ³]	356.41(2)	356(1)	361.5	302.2(2)	298.3(1)
a/c	0.5013	0.4726	0.5037 0.5045		
luzonite-famatinite series Pfitzner & Bernert (2004)			famatinite Garin & Parthé (1972)	famatinite Pfitzner & Reiser (2001)	famatinite Downs (2006) RRUFF ID R110021
	$Cu_3(As_{0.33}Sb_{0.67})_{\Sigma1.00}S_4$	Cu ₃ (As _{0.736} Sb _{0.264}) _{51.00} S ₄	Cu_3SbS_4	Cu_3SbS_4	Cu _{3.02} (Sb _{0.53} As _{0.48}) _{Σ1.01} S ₄
a [Å]	5.353(1)	5.315(1)	5.294	5.391(1)	5.345(2)
c [Å]	10.652(2)	10.536(2)	10.653	10.764(1)	10.630(6)
V [Å ³]	305.2(1)	297.7(1)	298.5656	312.83(9)	303.7(1)
a/c	0.5025	0.5045	0.4969	0.5008	0.5028

TABLE 3. THE REFINED UNIT-CELL PARAMETERS OF MEMBERS OF THE LUZONITE GROUP

⁽¹⁾ Mean from 10 electron microprobe analyses

(2) Mean of three electron microprobe analyses



FIG. 5. Raman spectra of permingeatite (red), famatinite (blue, violet), and luzonite (black). Spectra are vertically offset for clarity.



FIG. 6. Raman spectra of permingeatite acquired from three different spots (indicated by different line colors) within a grain in the polished section.

A comparison and comments on the spectra of famatinite and luzonite

As the crystal structures of luzonite, famatinite, and permingeatite are similar, we can expect similar Raman spectra, and that is what was observed (Figs. 5 and 6). The differences are only in the positions of the bands, because the energies of the X–Y (X = As/Sb; Y = S/Se) vibrations differ, according to Hook's rule. An intrinsic shift is observable comparing Sb-Se and Sb-S stretching modes, corresponding to permingeatite and famatinite, respectively.

Reflectance

Due to the relatively significant differences between the published reflectance curves for permingeatite from Předbořice (Johan *et al.* 1971, Picot & Johan 1982), the reflectance values were redetermined for permingeatite from Příbram (Table 6). The obtained data are close (Fig. 7) to the data published by Picot & Johan (1982). More significant differences were found when compared to the data published by Johan *et al.* (1971), particularly in the areas of 420–500 nm and about 640 nm (Fig. 8). It is possible that the material studied by Johan *et al.* (1971) contains inclusions of other minerals (*e.g.*, hakite, clausthalite, berzelianite) and their results were affected.

CONCLUSIONS

Permingeatite has recently been identified in material originating from the hydrothermal uranium deposit near Příbram by means of X-ray powder diffraction, reflectance, electron microprobe analysis, and Raman spectroscopy. Chemical compositions show greater Se-S substitution. No As, even trace amounts, was found by microprobe analysis in the permingeatite



FIG. 7. Reflectance curves (R_{min}, R_{max}) for permingeatite from Příbram (red) in comparison with published data for permingeatite from Předbořice (Picot & Johan 1982).



FIG. 8. Reflectance curves (R_{min}, R_{max}) for permingeatite from Příbram (red) in comparison with published data for permingeatite from Předbořice (Johan *et al.* 1971).

TABLE 4. FACTOR GROUP ANALYSIS FOR PERMINGEATITE, Cu₃SbSe₄ (SPACE GROUP NO. 121, *I*42*m*)

Atom	Wyckoff	Site-symmetry	$\Gamma_{red} G_F$
Sb	a(2)	42 <i>m</i>	B ₂ +E
Cu	b(2)	<u>4</u> 2 <i>m</i>	B ₂ +E
Cu	d(4)	$\overline{4}$	B ₁ +B ₂ +2E
Se	<i>i</i> (8)	т	2A ₁ +A ₂ +B ₁ +2B ₂ +3E
TOTAL		- (B ₂ + E)	= 2A ₁ + A ₂ + 2B ₁ + 4B ₂ + 6E
	IR	RA	
A ₁		a _{xx} +a _{vv} , a _{zz}	
A ₂	Rz		
B₁		avv-avv	
B ₂	T ₂	a ^z w	
Ē	$(T_x, T_y); (R_x, R_y)$	(a^{y}_{xz}, a^{x}_{yz})	
	· · · · · · · · · · · · · · · · · · ·	· · _ · ,_ ·	

Perm ¹	Perm ²	Perm ³	Fam ^{R070247}	Fam ^{R070628}	Luz ^{R060390}
	n	osition / relative inter	nsity / FW/HM (cm ⁻¹ / %	/ cm ⁻¹)	
			695 /4/37		693/1/21
				669 /2/16	003/1/21
			676 /1/8	000/2/10	
			646 /9/26		
				636/6/21	639/5/25
			585 /1/73	584/1/21	590 /<1/32
				492 /1/32	495 /<1/40
			470/4/04		460 /1/23
			4/0/4/01 4/8/2/40	AAG/3/16	113/1/16
			440/2/40	440/3/10	443/1/10 427/1/17
				395 /1/7	421/1/17
374 /5/16	379 /2/10		379 /6/9		
366 /4/10	367 /3/10	368 /4/18	365 /43/16		362 /4/14
357 /4/15	358 /2/8		357 /2/1	359 /41/19	
		349 /2/18	352 /46/17	346 /61/12	351 /10/12
			205/400/44	334 /25/8	200/400/40
319 /5/10	216/3/7	310/6/12	325/100/11 310/87/10	321/100/13	322/100/12 315/41/13
310/3/12	310/3/7	313/0/12	313/07/19	309/26/10	305/7/14
				302 /8/7	000///14
276 /2/3			275 /36/15	277/15/11	272 /11/17
				269 /15/9	
			263 /5/3	261 /20/11	
251 /4/14	252 /2/10		257 /32/23		254 /5/19
		246 /4/19		249 /44/15	
237 /19/12	240 /5/12			237 /11/11	
202 /00/7	000/11/10	000/10/14			
229/60/7	229/41/13	230/46/14	222 1A12		
214 /7/13	215 /5/10	215 /4/8			
205/20/8	203/8/12	204 /7/13			202 /1/9
193/26/12	191 /29/13	192 /12/10		190 /1/5	
184 /100/7	184 /100/6	184 /100/6		184 /3/6	
		183 */31/25			
176 /20/8	177 /19/9	100/10/7			
167/36/8	167 /17/9	166/12/7			
159/1/4	156/3/7				
140 /4/12	147/2/0				
	136 /10/12	136 /9/15			
127 /7/22	125 /7/12	124 /6/17			
	113 /3/9				
	91 /3/3				
	80 /3/2	80 /3/4			
78/10/4	77/9/4	75/04/4			
15/32/3	75/28/3 70/4/4	15/24/4			
63 /21/2	63 /23/2				
59 /26/5	59 /29/5				
51 /3/2	51 /6/6				
45 /40/4	45 /33/5	45 /43/6			
40 /9/3	40 /14/3				
	36 /5/1				
26 /60/2	26 /65/3	26 /18/2			
21/90/2	21/88/2	21/28/2			
11/14/2	11/10/3				
1-12010	17/24/0				

TABLE 5. BAND-COMPONENTS OF THE RAMAN SPECTRA OF STUDIED MINERALS

λ [nm] -	Příbram this paper		Předbořice Picot & Johan (1982)		Předbořice Johan <i>et al.</i> (1971)			
	R _{min.}	R _{max.}	R _{min.}	R _{max.}	R _{min.}	R _{max.}	R _{min.}	R _{max.}
400	30.0	31.8	-	-	-	-	-	-
420	29.7	31.5	29.8	31.0	29.4	30.4	28.8	29.2
440	29.3	31.2	29.0	31.1	30.2	32.0	29.4	30.0
460	28.5	30.6	28.2	30.5	30.4	32.8	29.8	30.3
480	27.9	29.6	27.6	29.4	29.4	31.6	28.8	29.2
500	27.3	28.8	27.0	28.2	28.1	29.7	27.9	28.3
520	26.8	28.2	26.6	27.1	26.8	27.8	26.8	27.3
540	26.3	27.5	26.2	26.3	25.7	25.9	26.0	26.4
560	25.7	26.7	25.6	25.8	25.0	25.2	25.4	25.6
580	25.2	26.1	25.0	25.5	24.7	25.0	25.0	25.2
600	25.0	25.8	24.5	25.6	24.5	25.2	25.1	25.2
620	24.9	25.6	24.5	25.9	24.5	25.6	25.2	25.2
640	25.1	25.7	24.7	26.5	23.8	25.4	24.6	24.6
660	25.3	25.9	25.3	27.4	-	-	-	-
680	25.9	26.6	26.2	28.4	-	-	-	-
700	26.4	27.2	27.6	29.3	-	-	-	-

TABLE 6. REFLECTANCE DATA FOR PERMINGEATITE, MEASURED IN AIR

studied. This is probably due to the fact that, in the calcite-uraninite stage, nearly all of the As is in other mineral phases, such as native arsenic, arsenolamprite, rare löllingite, and arsenopyrite, which are older than permingeatite in the paragenetic sequence.

Comparison of the Raman spectra of the luzonite mineral group shows obvious shifts in the wavenumbers of the stretching vibrations corresponding to the appropriate X-Y bonding-pair (where X = As/Sb; Y = S/Se).

The reflectance of permingeatite from Příbram is similar to the data published by Picot & Johan (1982), but there are significant differences from the original data published by Johan *et al.* (1971), which could be evidence of inclusions of other ore minerals in the type material.

Permingeatite was formed together with other known selenides from the Příbram uranium and basemetal deposit in the calcite-uraninite stage at relatively low temperatures, about 140–80 °C and from lowsalinity solutions, probably of meteoric origin (Žák & Dobeš 1991). The presence of umangite in association with permingeatite supports the temperature of origin of the studied mineral assemblage being under 112 °C (Simon & Essene 1996).

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