

Dzierżanowskite, CaCu_2S_2 – a new natural thiocuprate from Jabel Harmun, Judean Desert, Palestine Autonomy, Israel

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ABSTRACT

Dzierżanowskite, CaCu_2S_2 ($P\bar{3}m1$, $a = 3.9400(4)$, $c = 6.523(1)$ Å, $V = 87.69(2)$ Å³, $Z = 1$), a thiocuprate, was found in larnite pseudoconglomerate rocks of the Hatrurim Complex at Jabel Harmun, Palestinian Autonomy, Israel. Dzierżanowskite occurs in larnite pebbles, which are embedded in a low-temperature mineral matrix. Associated minerals are larnite, brownmillerite, fluorellestadite, ye'elimite, gehlenite, periclase, ternesite, nabimusaitite, vorlanite, vapnikite, fluormayenite, fluorkyuygenite, oldhamite, jasmundite, covellite, chalcocite and pyrrhotite. Electron microprobe analyses yield an average composition of Cu 55.25, Fe 0.13, S 27.46 and Ca 16.99, total 99.83 wt.%. The empirical formula of dzierżanowskite, based on 5 atoms, is $\text{Ca}_{0.98}\text{Cu}_{2.02}\text{Fe}_{0.01}\text{S}_{1.99}$. Dzierżanowskite forms grains up to 15 µm in size or rims on oldhamite and laminar intergrowths with chalcocite and covellite. Dzierżanowskite is dark orange, has a cream streak and a submetallic lustre. In reflected light it is grey, with a cream tint and characteristic yellow-orange internal reflections. The calculated density of dzierżanowskite is 4.391 g cm⁻³. Three bands at 300, 103 and 86 cm⁻¹ are observed in the Raman spectrum. The strongest lines of the calculated powder diffraction pattern are [d , Å (I) hkl]: 2.358(100) 102, 1.970(93) 110, 3.023(78) 011, 6.523(36) 001, 3.412 (28) 100, 1.834(28) 103. Dzierżanowskite was also found in unusual jasmundite rocks, forming small 'paleofumaroles' within areas of low-temperature hydrothermal rocks bearing larnite pseudoconglomerates at Jabel Harmun. Dzierżanowskite is a superimposed phase of the high-temperature alteration of pyrometamorphic rocks subjected to by-products (melts/fluids and gases) of pyrometamorphism originating in the deeper levels of combustion.

KEYWORDS: dzierżanowskite, CaCu_2S_2 , new mineral, thiocuprate, Jabel Harmun, Hatrurim Complex.

Introduction

DZIERŻANOWSKITE (IMA2014-032), CaCu_2S_2 ($P\bar{3}m1$, $a = 3.9400(4)$, $c = 6.523(1)$ Å, $V = 87.69(2)$ Å³, $Z = 1$) was found in pyrometamorphic (combustion metamorphic) rocks of the Hatrurim Complex ("Mottled

Zone"; Bentor, 1960; Gross, 1977; Vapnik *et al.*, 2007; Sokol *et al.*, 2007, 2010; Novikov *et al.*, 2013) on Jabel Harmun (Jabel = mountain in Arabic). The type locality, near the Palestinian village of Nabi Musa, lies close to a historical place with an eponymous name (probably the Tomb of Moses), situated in the Judean Desert, West Bank, Palestinian Autonomy, Israel (31°46'N, 35°26'E).

Dzierżanowskite, CaCu_2S_2 , belongs to the thiocuprate group of minerals (Boller, 2007); its

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synthetic analogue (Purdy, 1998), related to the structural type CaAl_2Si_2 , is known (Condrón *et al.*, 2007). Chvilevaite $\text{Na}(\text{Cu}, \text{Fe}, \text{Zn})_2\text{S}_2$, is the only mineral of note among the natural phases (Kachalovskaya *et al.*, 1988), characterised by a structure similar to dzierżanowskite. The chvilevaite structure with disordered distribution of Cu with mixed valence, Fe^{2+} and Zn in tetrahedral sites, was refined in the non-centrosymmetric $P3m1$ space group ($a = 3.873 \text{ \AA}$, $c = 6.848 \text{ \AA}$, $V = 88.959 \text{ \AA}^3$; Kaplunnik *et al.*, 1990) as differing from CaCu_2S_2 (Purdy, 1998). Natural CaCu_2S_2 was detected in 2012 by Piotr Dzierżanowski during a study of the composition of sulfides in larnite rocks in probe no. MII-33 (M. Murashko's collection) on a CAMECA SX100 microprobe analyser. Three new minerals, vapnikite, Ca_3UO_6 (Galuskin *et al.*, 2014), nabimusaite, $\text{KCa}_{12}(\text{SiO}_4)_4(\text{SO}_4)_2\text{O}_2\text{F}$ (Galuskin *et al.*, 2015a) and its Ba-analogue dargaite, $\text{BaCa}_{12}(\text{SiO}_4)_4(\text{SO}_4)_2\text{O}_3$, were discovered in the same probe sample. Dargaite as new species has been studied from Nahal Darga, Judean Desert, Palestinian Autonomy (Gfeller *et al.*, 2015a).

The name is given in honour of Dr Piotr Dzierżanowski (1947–2015) from the Institute of Geochemistry, Mineralogy and Petrology, University of Warsaw, Poland. Piotr Dzierżanowski was a recognized expert in microprobe analyses and was connected intimately with the development of Polish mineralogy during the last 15 years. His premature death is a major loss for Polish science; he was a co-author of the descriptions of 31 new mineral species from the pyrometamorphic and metasomatic rocks of Israel and the Caucasus and Baikal regions, Russia.

Until recently, specimen no. MII-33 was the only sample in which dzierżanowskite, approved by CNMNC IMA in 2014, was found. A fragment of the holotype, specimen no. MII-33 is deposited in the collections of the Mineralogical Museum, University of Wrocław, Cybulskiego 30, 50-205 Wrocław, Poland, catalogue number MMUWr II-20464.

In 2015 we found unusual jasmundite, apoflamite rocks at Jabel Harmun, in which dzierżanowskite is well represented. Jabel Harmun is the type locality (the outcrop of pyrometamorphic rocks of the Hatrurim Complex is a few km^2 in area) of several new minerals: harmunite CaFe_2O_4 (IMA2012-045, Galuskin *et al.*, 2014), nabimusaite $\text{KCa}_{12}(\text{SiO}_4)_4(\text{SO}_4)_2\text{O}_2\text{F}$ (IMA2012-057, Galuskin *et al.*, 2015a), fluormayenite $\text{Ca}_{12}\text{Al}_{14}\text{O}_{32}\text{F}_2$ (IMA2013-019, Galuskin *et al.*, 2015b), vapnikite Ca_3UO_6 (IMA2013-082, Galuskin *et al.*, 2014); and gazeveite,

$\text{BaCa}_6(\text{SiO}_4)_2(\text{SO}_4)_2\text{O}$ (IMA2015-37, Galuskin *et al.*, 2017). In the present paper we characterize dzierżanowskite from the holotype specimen no. MII-33, representing the larnite rock, and from recently found jasmundite-rich rocks. We also discuss the mechanism and conditions of dzierżanowskite formation.

Methods

The crystal morphology and composition of dzierżanowskite and associated minerals were examined using optical microscopy, a scanning electron microscope (Philips XL30 ESEM/EDAX, Phenom XL; Faculty of Earth Sciences, University of Silesia), and electron microprobes CAMECA SX100 and CAMECA SXFiveFE (Institute of Geochemistry, Mineralogy and Petrology, University of Warsaw). Chemical analyses were carried out on the electron microprobe (15 kV, 20 nA, 1–2 μm beam diameter) using the following lines and standard materials: $\text{AsL}\alpha$ – GaAs, $\text{ZnK}\alpha$ – sphalerite; $\text{CuK}\alpha$, $\text{FeK}\alpha$, $\text{SK}\alpha$ – CuFeS_2 ; $\text{PbM}\alpha$ – PbS; $\text{NiK}\alpha$ – NiO; $\text{CoK}\alpha$ – CoO; $\text{AgL}\alpha$ – Ag_2Te ; $\text{CaK}\alpha$ – diopside, $\text{SeL}\alpha$ – Bi_2Se_3 .

The Raman spectrum of dzierżanowskite was recorded on a WITec alpha 300 confocal Raman microscope (Jagiellonian Centre for Experimental Therapeutics, Krakow, Poland) equipped with an air-cooled solid state laser (488 nm) and a CCD camera (-62°C). Raman-scattered light was fed into a microscope through a 50 μm diameter optical fibre and monochromator with a 600 mm^{-1} grating. An air Olympus MPLAN (100x/0.90NA) objective was used. The power of the laser at the sample position was 12 mW. Some 15–20 scans with integration times of 10–15 s and resolution 3 cm^{-1} were collected and averaged. The monochromators of both spectrometers were calibrated using the Raman scattering line of a silicon plate (520.7 cm^{-1}).

Dzierżanowskite forms inhomogeneous, separate grains 10–15 μm in size and intergrowths with other sulfides. It is commonly partially oxidized, which effectively excludes the possibility of extracting a single crystal. Thus, the symmetry and cell parameters of dzierżanowskite were determined by electron back-scatter diffraction (EBSD) using a JEOL JSM-6480 scanning electron microscope (SEM) equipped with EBSD (University of Silesia, Poland). The EBSD images were recorded with a HKL Nordlys II camera using 30 kV beam energy. The EBSD study was performed on microprobe thin sections repolished with an Al_2O_3 suspension of

20 nm particle size. Calibration of the geometry of the SEM and EBSD system was carried out on Si for two detector distances, i.e., 177 mm (normal working position) and 150 mm (camera refracted position). The program *Channel5* (Day and Trimby, 2004) was used for the interpretation of the EBSD diffraction patterns.

Occurrence and description

The holotype specimen no. MII-33 is a fragment of a dark-brown larnite pebble from a pseudoconglomerate of the Hatrurim Complex, collected by M. Murashko in 2008 on the south slope of Jabel Harmun, Judean Desert, Palestinian Autonomy, Israel. Rocks of the Hatrurim Complex are distributed widely in the region surrounding the Dead Sea in the territory of Israel, Palestinian Autonomy and Jordan (Picard, 1931; Kolodny and Gross, 1974; Burg *et al.*, 1991, 1999). The larnite rocks were formed during pyrogenic metamorphism due to caustobiolith combustion at temperatures above 800°C (Kolodny and Gross, 1974; Sokol *et al.*, 2007, 2010). Their pebble-like shape can be explained by low-temperature hydrothermal alteration of larnite-rich rocks with subsequent weathering (Gross, 1977). A geological description of the Jabel Harmun locality and existing hypotheses for the genesis of pyrometamorphic rocks of the Hatrurim Complex are summarized by Galuskina *et al.* (2014 and references therein).

The holotype specimen of dzierżanowskite (no. MII-33) is a larnite-ye'elimite-fluorellestadite-brownmillerite rock with an inhomogeneous distribution of minerals of the fluormayenite-fluorkyuygenite series. This rock is partitioned by a network of light, bluish veins characterized by higher porosity, in which larger grains of larnite, fluorellestadite, brownmillerite and ye'elimite, and newly-formed ternesite, nabimusaite, jasmundite and Ca, Cu, Fe, Ni, Ag sulfides are noted (Fig. 1a, see also fig. 3 in Galuskin *et al.*, 2015a). Accessory minerals in the dark part of the rock are vapnikite, vorlanite, periclase, shulamite, magnesioferrite-spinel, gehlenite, pyrrhotite and baryte. Grains of dzierżanowskite, up to 15 µm in size, occurs in association with Cu sulfides, represented by chalcocite, covellite, anilite, djurleite and oldhamite (Figs 1, 2; Table 1). Dzierżanowskite forms rims on oldhamite and laminar intergrowths with chalcocite and covellite (Fig. 1b, c). Micrometre-sized sulfides of Ni, Zn, Pt, Pd, Ag

and selenides of Cu and Ag form inclusions in chalcocite; according to results from SEM energy-dispersive spectroscopy, heazlewoodite, polydymite, vysotskite, naumannite, braggite, millerite and sphalerite were determined. Submicrometre-sized grains of CaNi_2S_2 , BaCu_2S_2 , KCu_4S_3 and a murunskite-like phase, $\text{K}_2\text{Cu}_3\text{FeS}_4$, are also found in association with dzierżanowskite. Dzierżanowskite is characterized by a dark orange colour, a cream streak and submetallic lustre. In reflected light it is grey, with a cream tint and characteristic yellow-orange internal reflections (Fig. 2b). The density could not be measured because of the small grain size; the calculated density is 4.391 g cm^{-3} using the empirical formula $\text{Ca}_{0.98}\text{Cu}_{2.02}\text{Fe}_{0.01}\text{S}_{1.99}$ (Table 1, analysis 1). Dzierżanowskite shows strong orange cathodoluminescence; partially hydrated dzierżanowskite grains have weak yellowish cathodoluminescence.

In 2015, the second discovery of dzierżanowskite was connected with the finding of unusual jasmundite (apo-flamite) rocks, forming, as a rule, rounded bodies (forms), so-called 'paleofumaroles', 1–1.5 m across in low-temperature hydrothermal rocks bearing larnite pseudoconglomerates (Fig. 3a). There are four such 'fumaroles' on the south-west slope of Jabel Harmun. Jasmundite forms chains of crystals along cracks in these rocks (Fig. 3b) and spot-like aggregates of xenomorphic grains (Fig. 4a), the size of which is much bigger than the size of the mineral grains in the primary pyrometamorphic association. The primary pyrometamorphic association is represented by flamite-brownmillerite rocks (Fig. 3c). Flamite is partially hydrated and contains inclusions of nabimusaite, ternesite and fluorellestadite (Fig. 3c, 4a; see also fig. 8 of Galuskin *et al.*, 2016). Eutectic intergrowths of fluormayenite and flamite, on the rims of which thin zones of brownmillerite have crystallized (Fig. 4a, b), are a characteristic feature of flamite rocks. Fluormayenite is almost completely replaced by fluorkyuygenite.

Studies of unaltered fractures in the rocks showed that oval inclusions of flamite with dissolved surfaces and abundant micrometre-sized Cu-sulfides are widespread in the jasmundite (Fig. 3d). Sometimes, rounded empty channels with smooth, unaltered walls are noted in the jasmundite. More often these channels are filled with calcite, ettringite or Cl-bearing Ca-hydrosilicates. In such cases the jasmundite forming the walls of the channels is intensively hydrated (Fig. 3d). In specimen no. JBB4 dzierżanowskite

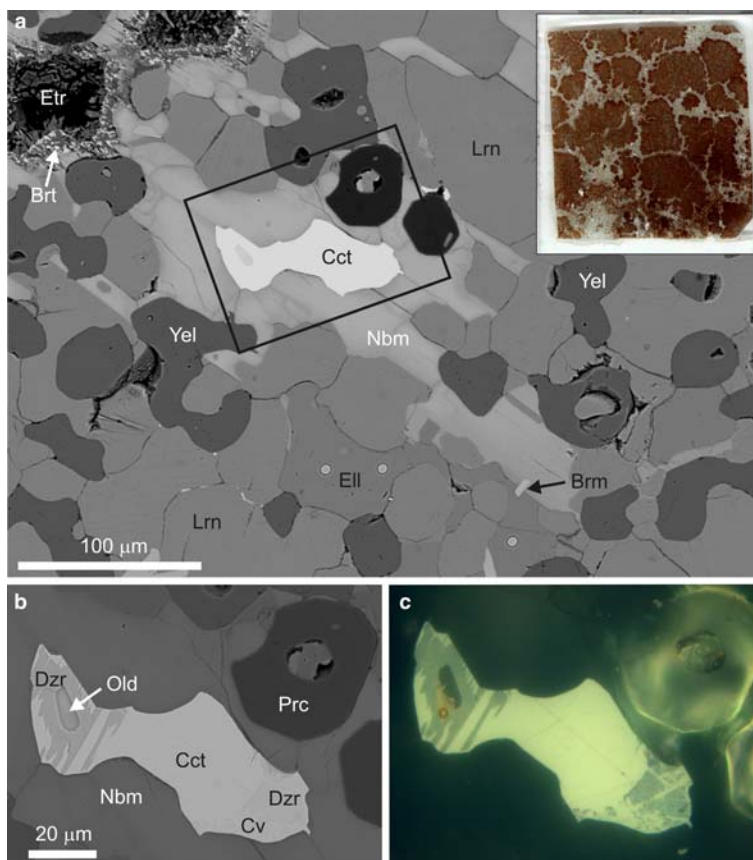


FIG. 1. (a) Back-scattered electron (BSE) image of a chalcocite grain with oldhamite inclusions in altered lamite rock (holotype specimen MII-33), Jabel Harmun, Palestinian Autonomy; the black frame marks the area magnified in Fig. 1b and c; an image of the thin section in normal light is inset top right (field of view = 2 cm). (b) BSE image of dzierżanowskite overgrowing an oldhamite inclusion on the contact with chalcocite. (c) Reflected light image of (b); dzierżanowskite is grey; the brown circle is a crater formed after measurement in a Raman microscope at high power of laser. DZR = dzierżanowskite, Cct = chalcocite, Cv = covellite, Lrn = larnite, Ell = fluorellestadite, Brm = brownmillerite, Nbm = nabimusaite, Yel = ye'elimite, Brt = baryte, Etr = ettringite, Prc = periclase.

usually forms thin rims on rounded oldhamite grains, sometimes almost completely replacing it (Fig. 4c, d). Rarer oldhamite and dzierżanowskite build complex mineral aggregates together with newly-formed brownmillerite, fluorellestadite and Cu-sulfides (Fig. 4e). Between dzierżanowskite and jasmundite small zones of chalcocite or covellite occur, where sulfides form subtle worm-like inclusions in jasmundite (Fig. 4d). Other Ca, Cu, Fe, Ni, Ag sulfides and, rarely, exotic K and Na sulfides, for example the Na-analogue of murunskite (Fig. 4d; Table 1, analysis 9), are found in specimen no. JBB4 together with jasmundite as well as in the holotype specimen.

EBSD and Raman data for dzierżanowskite

Single-crystal X-ray studies could not be performed because of the small crystal size. Structural data were obtained using EBSD techniques (Fig. 5) and by fitting data to the structural model of synthetic CaCu_2S_2 (Purdy, 1998). The preparation of the dzierżanowskite surface for EBSD study was critical because of the limited amount of the holotype specimen no. MII-33, so the quality of the Kikuchi images obtained for dzierżanowskite and the associated oldhamite was rather low (Fig. 5). The EBSD pattern for dzierżanowskite was obtained at working distances 155 and 177 mm. Fitting of the

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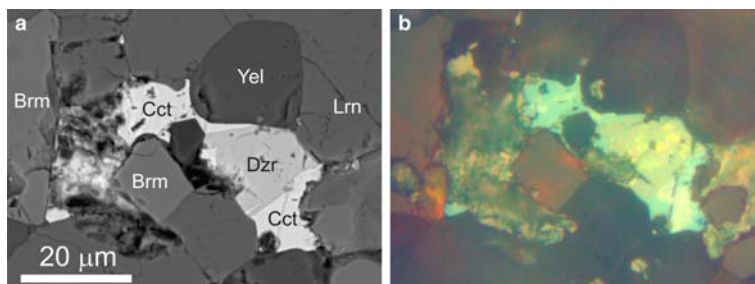


FIG. 2. One of the largest segregations of dzierzanowskite in holotype specimen MII-33: (a) BSE image. (b) Reflected light, poor polishing of chalcocite has resulted in its blue hue (covellite); dzierzanowskite is grey with cream hue, light-orange internal reflexes are visible in it; brownmillerite shows red internal reflexes. DZR = dzierzanowskite, Cct = chalcocite, Lrn = lamite, Brm = brownmillerite, Yel = ye'elimite.

EBSD pattern at 177 mm for a CaCu_2S_2 model with $a = 3.9400(4)$, $c = 6.523(1)$ Å, $V = 87.69(2)$ Å³, $Z = 1$, space group: $P\bar{3}m1$ (Purdy, 1998) resulted in the parameter $\text{MAD} = 0.46^\circ$ (good fit). More qualitative Kikuchi images for the working distance 177 mm

were obtained for dzierzanowskite and oldhamite from specimen no. JJB4 (Fig. 4e; Fig. 6a,b). We also tried to determine the Cu-sulfides intergrowing with dzierzanowskite and oldhamite using the EBSD method. Cu-sulfides with stoichiometry close to

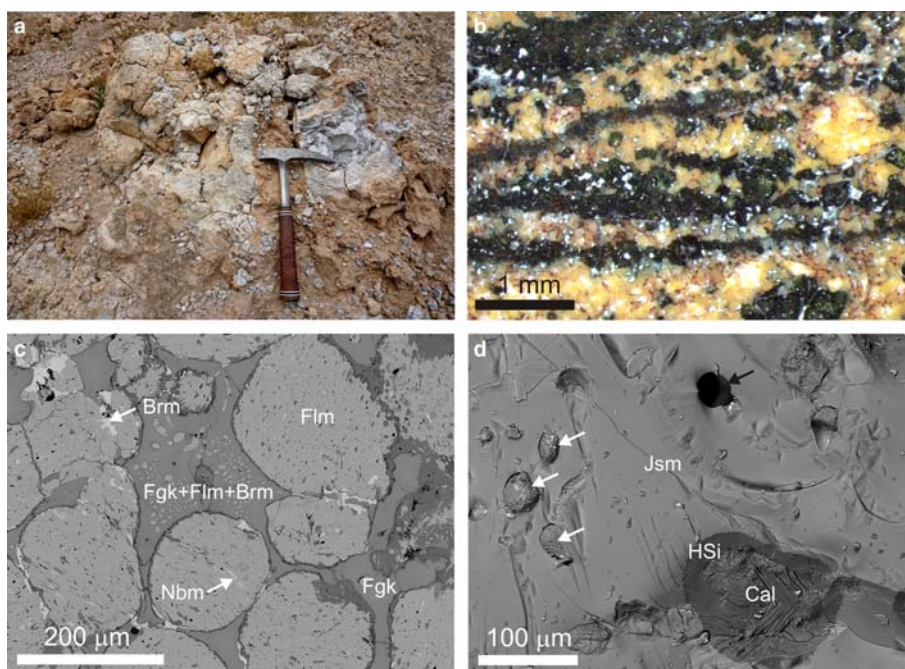


FIG. 3. (a) Jasmundite rock outcrop on the south slope of the Jabel Harmun, where dzierzanowskite was found. (b) Dark-green jasmundite crystals form chain-like aggregates along fissures in altered flamite rock; brownmillerite rims (red) are visible on flamite grains (yellow). (c) Altered flamite rock: BSE image of a yellow fragment of the rock. (d) Fresh fracture of a jasmundite crystal; oval inclusions of flamite relics are shown by white arrows; and a hollow channel is shown by a black arrow. A channel with hydrated walls filled by calcite is visible at bottom right corner. DZR = dzierzanowskite, Flm = flamite, Brm = brownmillerite, Nbm = nabimusaita, Fgk = fluorquygenite, Cal = calcite, Jsm = jasmundite, HSi = hydrosilicate.

TABLE 1 Representative compositions of dzierzanowskite and associated sulfide minerals.

Anal. no.* wt. %	MII- 33			JBB4 [2] mean n = 3	MII-33						JBB4		
	[1] Mean n = 9	s.d.	range		[3]	[4]	[5]	[6]	[7]	[8]	[9]	[10]	
Ag	n.d.			n.d.	n.d.	1.80	n.d.	1.00	n.d.	25.47	n.d.	n.d.	
Zn	n.d.			n.d.	n.d.	n.d.	n.d.	n.d.	0.19	n.d.	n.d.	n.d.	
Cu	55.25	0.55	54.26–55.93	55.78	34.69	77.62	79.79	78.42	65.16	52.67	48.26	n.d.	
Fe	0.13	0.10	0–0.30	0.13	30.37	0.00	0.12	0.00	0.00	0.00	12.04	n.d.	
S	27.46	0.17	27.24–27.64	27.81	35.28	19.67	23.01	20.58	32.50	14.04	30.18	43.78	
Se	n.d.			n.d.		0.26	n.d.	0.27	0.47	7.97	n.d.	n.d.	
Ca	16.99	0.21	16.73–17.37	16.52	0.00	0.20	0.25	0.23	0.60	0.33	0.43	55.36	
Na	n.d.			n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	9.71	n.d.	
Total	99.83			100.24	100.34	99.55	103.17	100.50	98.92	100.48	100.62	99.14	
					Calculated on atoms per formula unit								
	5			5	4	3	10	47	2	3	10	2	
Ca	0.98			0.95		0.01	0.03	0.14	0.01	0.02	0.05	1.01	
Na											1.80		
Cu	2.02			2.03	1.00	1.97	6.97	30.62	0.99	1.54	3.23		
Ag						0.03		0.23		0.44			
Fe	0.01			0.01	0.99		0.01				0.92		
S	1.99			2.01	2.01	0.99	3.98	15.93	0.98	0.81	4.01	0.99	
Se						0.01		0.09	0.02	0.19			

*Key: [1], [2] – dzierzanowskite, $\sim \text{CaCu}_2\text{S}_2$; [3] – chalcopyrite, $\sim \text{CuFeS}_2$; [4] – chalcocite, $\sim \text{Cu}_2\text{S}$; [5] – anilite, $\sim \text{Cu}_7\text{S}_4$; [6] – djurleite, $\sim \text{Cu}_{31}\text{S}_{16}$; [7] – covellite, CuS , [8] – covellite-like mineral, $\sim (\text{Cu,Ag})_2(\text{S,Se})_1$; [9] – Na-analogue of murunskite, $\sim \text{Na}_2\text{Cu}_3\text{FeS}_4$; [10] – oldhamite, $\sim \text{CaS}$.

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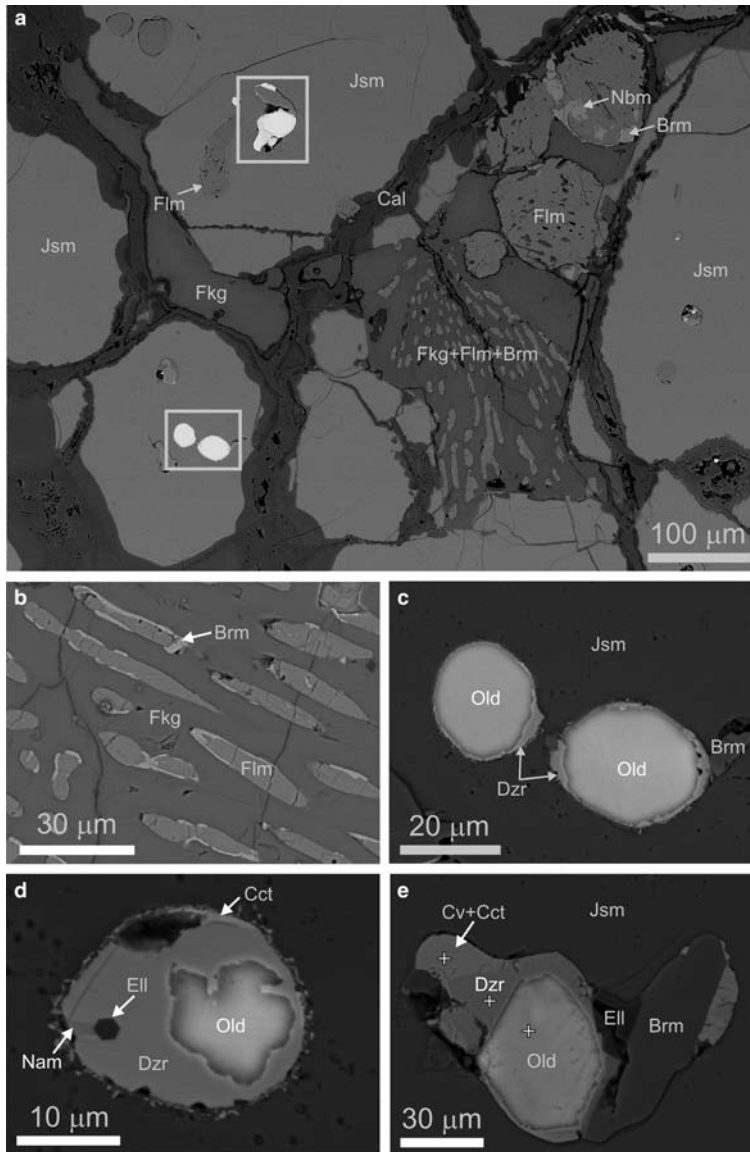


FIG. 4. (a) Typical view of altered flamite rock, in which oval inclusions of oldhamite and Cu-sulfides are hosted, mainly, in large jasmundite grains. Eutectic intergrowths of flamite and fluorkyuyenite (primary, fluormayenite), fragments in the frame are magnified in Fig. 4e. (b) brownmillerite crystallized in eutectic intergrowths on the boundary of flamite and fluormayenite. (c) Thin rims of dzierżanowskite on oldhamite. (d) Oval segregation of dzierżanowskite with oldhamite relics, chalcocite external thin rim, fluorellestadite and the Na-analogue of murunskite inclusions. (e) Oldhamite inclusion in jasmundite with dzierżanowskite rims intergrowing with chalcocite–covellite and brownmillerite. Kikuchi (EBSD) images were obtained at points marked by crosses. Dzr = dzierżanowskite, Cct = chalcocite, Cv = covellite, Lm = larnite, Flm = flamite, Ell = fluorellestadite, Jsm = jasmundite, Fkg = fluorkyuyenite, Brm = brownmillerite, Nbm = nabimusaita, Cal = calcite, Nam = Na-analogue of murunskite.

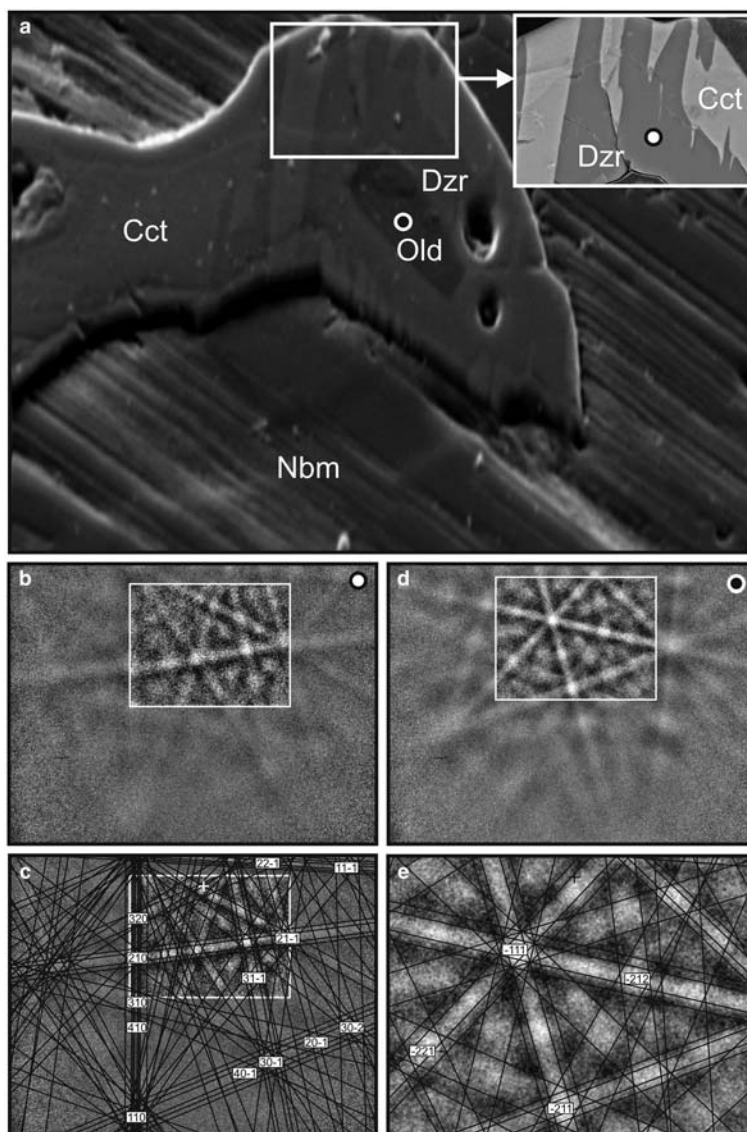


FIG 5. EBSD patterns of dzierzanowskite and oldhamite. (a) High relief image of minerals in thin-section (Fig. 1) after colloidal silica treatment; a fragment limited by the frame is shown in the inset as a BSE image (circles show points at which EBSD patterns of dzierzanowskite and oldhamite were obtained). (b) EBSD patterns of dzierzanowskite for working distances of 177 mm and 155 mm (shown in frame). (c) Fitted result of the EBSD pattern with the CaCu_2S_2 model for a working distance of 177 mm, $\text{MAD} = 0.46^\circ$ (good). (d) EBSD patterns of oldhamite for working distances of 177 mm and 155 mm (in frame). (e) Fitted result of the EBSD pattern for a working distance of 155 mm, $\text{MAD} = 0.72^\circ$. DZR = dzierzanowskite, Cct = chalcocite, Old = oldhamite, Nbm = nabimusaite.

chalcocite after preparation of the surface using colloidal Al_2O_3 usually did not provide a Kikuchi image or showed a fuzzy Kikuchi image, fitting of which pointed to covellite (Fig. 6c).

Because micrometre-sized dzierzanowskite occurs only in small concentrations, powder X-ray diffraction data could not be collected. Powder X-ray diffraction data (in \AA for $\text{CuK}\alpha$) were

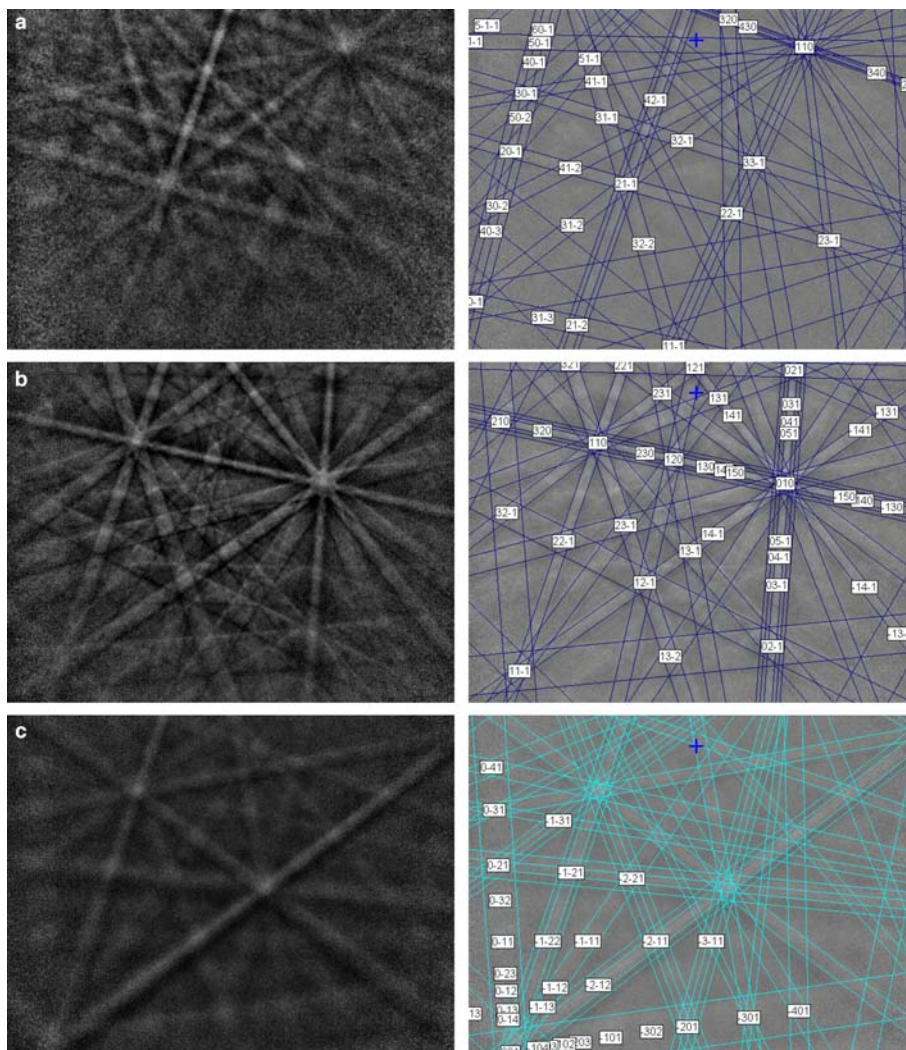


FIG. 6. EBSD patterns obtained at the points shown in Fig. 4e by crosses and the fitted result of the EBSD pattern for a working distance of 177 nm: (a) dzierżanowskite; (b) oldhamite; (c) covellite.

calculated from the results of the single-crystal structure refinement of synthetic CaCu_2S_2 and are listed in Table 2. The Raman spectrum of dzierżanowskite was recorded at low laser power. At standard power (~ 44 mW on sample) a crater was burnt in the mineral, and a Ca sulfate spectrum was recorded with a strong S–O stretching band at 1007 cm^{-1} (Fig. 7). Three bands at 300 (Cu–S stretching), 103 and 86 cm^{-1} (Ca–S stretching) are observed in the dzierżanowskite spectrum.

The composition, EBSD pattern and Raman spectrum of dzierżanowskite confirm that it is the

natural analogue of synthetic CaCu_2S_2 (Purdy, 1998), which belongs to a group of thio cuprates characterized by the formula $[\text{M}_a]^{z+}[\text{Cu}_b\text{S}_c]^{z-}$ (Boller, 2007). These thio cuprates possess a CaAl_2Si_2 structure type (Condon *et al.*, 2007). In dzierżanowskite a double tetrahedral layer is formed by CuS_4 edge-linked tetrahedra. There is a layer of Ca octahedra between the layers (Fig. 8a). The tetrahedra are arranged one above the other along the c axis in such a way that the specific channels containing Ca are visible (Fig. 8b).

TABLE 2. Calculated powder X-ray diffraction data for dzierżanowskite ($\text{CuK}\alpha_1 = 1.540598 \text{ \AA}$, Debye-Scherrer geometry, no anomalous dispersion, $I_{\text{calc}} > 1$).

<i>h</i>	<i>k</i>	<i>l</i>	d_{calc}	I_{calc}
0	0	1	6.523	36
1	0	0	3.412	28
0	0	2	3.262	1
1	0	1	3.023	25
0	1	1	3.023	78
1	0	2	2.358	100
0	1	2		5
0	0	3	2.174	14
1	1	0	1.970	93
1	1	1	1.886	10
0	1	3	1.834	22
1	0	3	1.834	28
2	0	0	1.706	3
0	2	1		3
2	0	1	1.651	13
2	0	2	1.512	1
0	2	2	1.512	22
0	1	4		3
1	0	4	1.471	4
1	1	3	1.460	22
0	2	3		9
2	0	3	1.342	7
0	0	5	1.305	2
2	1	0	1.290	3
1	2	1		9
2	1	1	1.265	3
1	0	5		5
0	1	5	1.219	6
2	1	2		19
1	2	2	1.199	1
0	2	4		2
2	0	4	1.179	1
3	0	0	1.137	14
1	2	3		8
2	1	3	1.109	10

The strongest lines are given in bold.

Genesis and growth mechanism of dzierżanowskite

Dzierżanowskite was found in larnite-fluorellestadite-brownmillerite-ye'elimite rocks (Fig. 1, holotype specimen no. MMII-33) and in flamite-fluormayenite-brownmillerite rocks (Fig. 3, specimen no. JBB4), where the primary pyrometamorphic mineral assemblage is similar to the phase content of Portland clinker (Taylor, 1997; Sokol *et al.*, 2014). Larnite ($\beta\text{-Ca}_2\text{SiO}_4$), flamite ($\alpha\text{-Ca}_2\text{SiO}_4$),

ye'elimite, brownmillerite and mayenite are typical components of clinkers. Flamite is a new mineral, described recently from pyrometamorphic gehlenite hornfels of the Hatrumim Complex in the Negev Desert, Israel (Sokol *et al.*, 2015; Gfeller *et al.*, 2015b).

A distinctive feature of rocks containing dzierżanowskite is the presence of S-bearing minerals of the nabimusaite–dargaite and ternesite–silicocarnotite series (Galuskin *et al.*, 2015a, 2016), jasmundite, oldhamite and various sulfides, which are confined to microporous fragments of the rocks. Nabimusaite–dargaite and ternesite–silicocarnotite are products of the high-temperature alteration of primary clinker mineral associations affected by sulfate-halogen-bearing melts (fluids, gases) at temperatures $>900^\circ\text{C}$, which are generated during pyrometamorphism close to the combustion foci (Galuskin *et al.*, 2015a,b; 2016).

Superimposed melts, fluids, or gases are probably originated as by-products of the combustion of a sedimentary protolith bearing organic fuel and/or combustion of external flows of combustible gases. Under conditions of low lithostatic pressure these gases migrate to the surface along channels and cracks in metamorphosed rocks. Pyrometamorphic gases are heterogeneous and contain not only sulfurous compounds, but also Cl and F, CO_2 , water gas and others. These compounds are similar to volcanic gases and to the gaseous products of burning coal seams and waste coal deposits (Mizutani, 1970; Symonds, 1993; Suárez-Ruiz *et al.*, 2012). We have not yet found nitrogen compounds that are typical of burning coals (Suárez-Ruiz *et al.*, 2012). Thus, superimposed mineralogenesis in this case was similar to an ultrahigh-temperature fumarole system.

In the holotype specimen no. MII-33 dzierżanowskite is confined to light, porous fragments of larnite rocks (Fig. 1a). Here, recrystallization of clinker minerals and the appearance of high-temperature minerals, nabimusaite–dargaite, jasmundite and diverse sulfides, are notable (Fig. 1a, b). In specimen no. JBB4 grains of newly-formed jasmundite show sizes up to 1 cm. Jasmundite contains rounded oldhamite inclusions replaced peripherally by Cu-sulfides and also rare micrometre-sized channels (Fig. 3d). We consider that jasmundite, which is an S^{2-} bearing phase, is a product of the pneumatolytic (fluid-gaseous) metasomatism of larnite, flamite and fluorellestadite. Oldhamite also formed at that stage from the gas phase as a result of reaction between sulfur-bearing gases (for example, H_2S) and Ca-minerals:

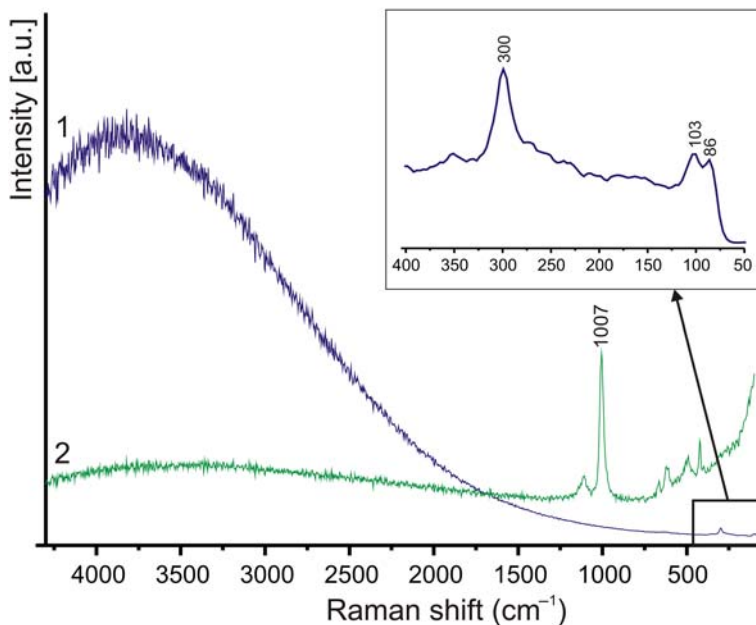


FIG. 7. Raman spectra of dzierżanowskite (1) and products of its alteration, calcium sulfate, formed under the laser beam (2).

$\text{CaO} + \text{H}_2\text{S} = \text{CaS} + \text{H}_2\text{O}$, at temperatures $\approx 800^\circ\text{C}$ (Westmoreland *et al.*, 1977). The rounded shapes of oldhamite are consistent with the channel cross-section geometry or these are relic forms of Ca-silicate replacement (Fig. 3d). At decreasing temperatures to $350\text{--}400^\circ\text{C}$, Cu sulfides formed, for example according to the reaction: $2\text{CuCl} +$

$\text{H}_2\text{S} = \text{Cu}_2\text{S} + 2\text{HCl}$ (Mizutani, 1970). Newly-formed Cu-sulfides were not in equilibrium with previously formed oldhamite, so between them there is a reaction zone of dzierżanowskite formed according to the reaction: $\text{Cu}_2\text{S} + \text{CaS} = \text{CaCu}_2\text{S}_2$. The same reaction was used for formation of the synthetic analogue of dzierżanowskite, which is

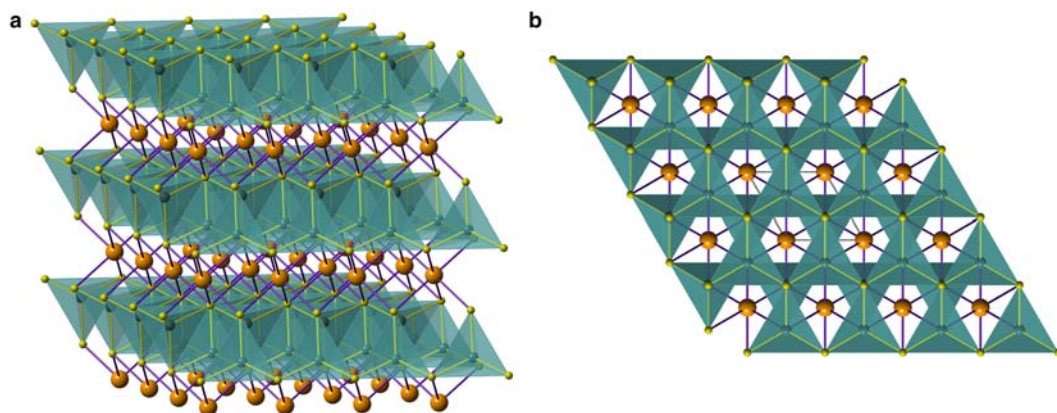


FIG 8. (a) The dzierżanowskite structure (CaAl_2Si_2 structure type) formed by two tetrahedral layers (CuS_4), where tetrahedra are linked by edges. Calcium ions with octahedral coordination are arranged between tetrahedral layers. (b) The packing mode of tetrahedral layers defines channels along the c axis, where calcium is located. Cu – green-blue balls, Ca – orange balls, S – yellow balls.

synthesized at temperatures $\approx 400^\circ\text{C}$ (Purdy, 1998). Dzierzanowskite is preserved only as inclusions in jasmundite, and more rarely inside other silicates. Fluormayenite associated with dzierzanowskite was replaced by fluorkyuygenite by the reaction $\text{Ca}_{12}\text{Al}_{14}\text{O}_{32}[\square_4\text{F}_2] + 4\text{H}_2\text{O} = \text{Ca}_{12}\text{Al}_{14}\text{O}_{32}[(\text{H}_2\text{O})_4\text{F}_2]$, as a result of overheated water vapour action. Fluorkyuygenite defects in the fluormayenite structure are stable at temperatures below 400°C (Galuskin *et al.*, 2015b). Cuprite and malachite developed after dzierzanowskite in the fluorkyuygenite bearing zone. Dzierzanowskite, along with nabimusaite, gazeevite, silicocarnotite and fluorkyuygenite in the pyrometamorphic rocks of the Hatrurim Complex, testifies to a superimposed new stage of the high-temperature alteration of pyrometamorphic rocks (natural clinker of the sanidinite facies). Such alteration was determined by by-products (melts/fluids and gases) of pyrometamorphism which originated in the deeper levels of combustion (Galuskin *et al.*, 2014; Galuskin *et al.*, 2015a,b, 2016).

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