

MONAZITE FROM THE MOUNT PLEASANT DEPOSIT, NEW BRUNSWICK

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Samples from the 750 adit of the Fire Tower area, Mount Pleasant deposit of Brunswick Tin Mines Limited in New Brunswick were studied at the Mines Branch (renamed CANMET) from 1970 to 1975 to determine ore characteristics and solve ore beneficiation problems. The geological setting and mineralogy of the deposit were described by Petruk (1973). Only a few monazite grains, most smaller than 5 microns in diameter, were found during the initial study and analyzed semi-quantitatively. These grains occurred as inclusions in arsenopyrite and as separate grains in gangue near rutile, zircon and xenotime (referred to by Petruk 1973 as a ytterbium-erbium phosphate). In 1974, a wolframite concentrate containing about 15% monazite and 2% xenotime was produced from a composite sample of ore from the 750 adit while conducting ore-dressing tests in CANMET laboratories. The monazite and xenotime grains in the concentrate were up to 100 microns in diameter (mean 20 microns) and unzoned. They were identified by x-ray diffraction using a 57.3 mm-diameter Debye-Scherrer powder camera. Seven monazite grains were analyzed with a Materials Analysis Company (MAC) electron microprobe operated at 25 kV using the following standards: apatite for Ca and P, uranophane for Th, uraninite for U, monazite for La, synthetic CeO₂ for Ce, synthetic YFeO₃ for Y and Fe, synthetic NdFeO₃ for Nd, synthetic SmFeO₃ for Sm, and synthetic GdFeO₃ for Gd (Table 1). Trace amounts of Dy (< 1.0 wt %)

were detected but not analyzed for because of lack of adequate standards. Other elements were looked for, but rare-earth elements (including praseodymium) could not be detected because suitable standards were not available and because of interference by the major rare-earth elements in the mineral. The data were processed with a computer program of Rucklidge & Gasparrini (1969).

The analyses for individual grains show some compositional variation for the mineral. In particular, it is noted that mineral grains enriched in ThO₂ and U₃O₈ have reduced Nd₂O₃ contents and grains enriched in Gd₂O₃ and Sm₂O₃ have reduced La₂O₃ contents (Table 1).

Previous workers have successfully correlated monazites on the basis of the ThO₂ contents and ratios of certain rare-earth elements. A summary of the geologic occurrences of monazite by Overstreet (1967) shows that the ThO₂ content is variable, but monazites in crystalline rocks of the epizone generally have low ThO₂ contents (less than 1 wt %), those of the mesozone have moderate ThO₂ contents, and those of the katazone have high ThO₂ contents (greater than 7 wt %). The ThO₂ content of the Mount Pleasant monazite (3.6 wt %), which occurs in a greisenized acid volcanic rock, corresponds with the ThO₂ contents of monazites from cassiterite and/or wolframite granites of the mesozone (Overstreet 1967).

The ratios (in atomic %) for some rare-earth-element pairs in an average Mount Pleasant monazite are Ce/La = 2.1, Ce/Nd = 2.4, La/Nd = 1.2 and Nd/Sm = 5.2. These values are nearly the same as the average for Malayan alluvial monazite from a tin placer deposit (Flinter *et al.* 1963), and are similar to values for monazites from granites and granite pegmatites (Murata *et al.* 1953, 1957, 1958).

The Mount Pleasant monazite has a fairly high and uniform Gd₂O₃ content. Richartz (1961) studied magnetic fractions of a monazite from Brazil and found that Gd₂O₃ and Dy₂O₃ are strongly paramagnetic, and monazite with high concentrations of these elements has a high magnetic susceptibility. It is concluded that the Mount Pleasant monazite was concentrated with wolframite in the magnetic fraction

TABLE 1: ELECTRON MICROPROBE ANALYSES OF MONAZITE GRAINS

Wt. %	1	2	3	4	5	6	7	Average
F ₂ O ₅	29.9	29.3	30.4	30.4	26.6	30.1	31.2	29.7
Ce ₂ O ₃	33.3	32.6	31.2	31.0	29.2	29.0	27.3	30.6
La ₂ O ₃	15.6	17.3	14.0	16.1	15.4	11.7	10.3	14.3
Nd ₂ O ₃	10.2	13.1	13.9	13.8	10.3	14.2	14.0	12.8
Gd ₂ O ₃	3.3	3.8	4.1	4.0	3.2	4.3	5.8	4.1
Y ₂ O ₃	0.0	0.0	1.4	0.7	0.8	3.4	3.1	1.4
Sm ₂ O ₃	1.8	1.8	2.2	2.0	1.5	3.3	4.8	2.5
Fe ₂ O ₃	0.0	0.1	0.2	0.2	0.2	0.1	0.0	0.1
U ₃ O ₈	1.7	0.6	0.5	0.4	1.3	0.0	0.4	0.7
ThO ₂	6.8	2.1	3.2	3.1	7.0	1.0	2.2	3.6
CaO	1.1	0.1	0.6	0.5	0.5	0.2	0.6	0.5
Total	103.7	100.8	101.7	102.2	96.0	97.3	99.7	100.3

because it has a high magnetic susceptibility due to its Gd_2O_3 content.

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