

A RHOMBOHEDRAL POLYTYPE OF MOLYBDENITE*

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Rhombohedral molybdenite (molybdenite-3R) has been identified as an accessory mineral in quartz-feldspar porphyry occurring at the Con mine, Yellowknife, District of Mackenzie. The following observations on the occurrence and origin of the quartz-feldspar porphyry have been taken from Boyle (1961).

Irregular dyke-like bodies of quartz-feldspar porphyry are found in the underground workings of the Con, Negus and Rycon mines. The porphyries contain phenocrysts of quartz and zoned plagioclase in a groundmass of fine-grained plagioclase, quartz, microcline, muscovite, carbonates and chloritized biotite. Pyrite and pyrrhotite are particularly abundant in the porphyries in the vicinity of the Con mine, and molybdenite is present as disseminated flakes in a few of the bodies. Other accessory minerals include small amounts of apatite, zircon, rutile and epidote. The sulphides appear to be primary constituents, and probably crystallized with the quartz and feldspar, because they fill interstices between these minerals and form along the cleavage planes of biotite.

The origin of the porphyry dykes and masses is an enigma requiring further research. On the one hand the porphyries present metasomatic features and may have formed from processes of metamorphic differentiation. This is borne out by the facts that some contain an abundance of sulphides and carbonates and are enriched in chromium, features foreign to granitic rocks. On the other hand they display sharp contacts, are dyke-like, and as regards their major components, have a similar composition to the western granodiorite. This suggests that they may have originated either by magmatic or granitization processes. The latter seems to fit the facts best and the author is of the opinion that the porphyries represent concentrations of various elements mobilized during granitization at depth.

The present writer has examined x -ray powder diffraction patterns of molybdenites from three different types of occurrences in the Yellowknife district. Molybdenites from a pegmatite associated with the Prosperous Lake granite and from a quartz lens in a shear zone of the Crestaurum system give x -ray patterns in close agreement with the standard pattern of common hexagonal molybdenite. The x -ray pattern of molybdenite from the Con quartz-feldspar porphyry shows very distinct differences, most noticeable of which are the absence of the (10.2) reflection and presence of strong doublets at the reflecting positions of the (10.3) and (10.5) planes of normal molybdenite. These differences have been noted

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by Bell & Herfert (1957) and Zelikman *et al.* (1961) on patterns of synthetic rhombohedral molybdenum disulphide.

Bell & Herfert (1957) have shown that the synthetic material has rhombohedral symmetry, space group $R\bar{3}m-C_3^2$, with the a axis identical to that of hexagonal MoS_2 and the c axis 1.5 times larger. The CdCl_2 -type structure proposed by Bell & Herfert has been refuted by Jellinek *et al.* (1960) on the grounds that it is incompatible with the x -ray diffraction data. The latter authors show that the structure can be described as: 3 Mo in $3(a)$, with $z = 0$; 3 S in $3(a)$, with $z \approx \frac{1}{4}$; and 3 S in $3(a)$ with $z \approx \frac{5}{12}$. Zelikman *et al.* have reported the possible existence of several structural modifications of synthetic MoS_2 which may result from alternating combinations of hexagonal and rhombohedral packing.

X -ray diffraction data for rhombohedral molybdenite from the Con mine are listed in Table 1 and compared with hexagonal molybdenite from the Yellowknife area. The patterns were taken with a 57.3 mm.

TABLE 1. X -RAY POWDER DATA FOR MOLYBDENITE POLYTYPES

Molybdenite-2H Yellowknife district $P6_3/mmc$ $a = 3.16, c = 12.28\text{\AA}$				Molybdenite-3R Con Mine $R\bar{3}m$ $a = 3.16, c = 18.33\text{\AA}$			
I	$d(\text{meas.})$	$d(\text{calc.})$	hkl	I	$d(\text{meas.})$	$d(\text{calc.})$	hkl
10	6.01	6.14	002	10	6.09	6.11	003
$\frac{1}{2}$	3.04	3.07	004	$\frac{1}{2}$	3.04	3.06	006
2	2.73	2.74	100	7	2.71	2.71	101
1	2.66	2.67	101	6	2.63	2.62	012
1	2.50	2.50	102				
8	2.27	2.275	103	6	2.344	2.349	104
1	2.04	2.045	104, 006	6	2.194	2.193	015
				3	2.034	2.037	009
5	1.82	1.828	105	4	1.889	1.892	107
$\frac{1}{2}$	1.64	1.639	106	3	1.755	1.757	018
2	1.58	1.580	110	7	1.581	1.580	110
3	1.53	1.532	008, 112	7	1.529	1.529	00.12, 113
$\frac{1}{2}$	1.48	1.477	107	$\frac{1}{2}$	1.426	1.423	01.11
				$\frac{1}{2}$	1.403	1.403	116
$\frac{1}{2}$	1.36	1.360	201	2	1.363	1.364	021
				1	1.357	1.353	202
$\frac{1}{2}$	1.337	1.337	202, 108	1	1.313	1.311	024
1	1.297	1.298	203	1	1.283	1.282	205
1	1.251	1.251	204, 116	4	1.249	1.248	119
$\frac{1}{2}$	1.226	1.228	00.10	$\frac{1}{2}$	1.225	1.222	00.15
$\frac{1}{2}$	1.220	1.221	109	1	1.214	1.213	027
1	1.195	1.195	205	$\frac{1}{2}$	1.179	1.181	01.14
1	1.101	1.101	118	5	1.099	1.098	11.12, 02.10
				$\frac{1}{2}$	1.057	1.057	10.16, 20.11

TABLE 1 (Concluded)

Molybdenite—2H Yellowknife district <i>P6₃/mmc</i> <i>a</i> = 3.16, <i>c</i> = 12.28Å				Molybdenite—3R Con Mine <i>R3m</i> <i>a</i> = 3.16, <i>c</i> = 18.33Å			
<i>I</i>	<i>d</i> (meas.)	<i>d</i> (calc.)	<i>hkl</i>	<i>I</i>	<i>d</i> (meas.)	<i>d</i> (calc.)	<i>hkl</i>
1	1.034	1.034	10, 11, 210	$\frac{1}{2}$	1.031	1.030	211, 122
$\frac{1}{2}$	1.021	1.021	208, 212	2	1.017	1.018	00.18
1	1.002	1.003	213	2	1.009	1.009	214
$\frac{1}{2}$	0.969	0.970	11.10	2	0.996	0.995	125
1	0.953	0.953	215	$\frac{1}{2}$	0.981	0.982	02.13
$\frac{1}{2}$	0.913	0.913	300, 20.10	2	0.968	0.967	11.15
$\frac{1}{2}$	0.902	0.902	302	1	0.962	0.962	217
1	0.894	0.893	10.13	1	0.945	0.946	20.14
$\frac{1}{2}$	0.877	0.877	00.14	3	0.913	0.912	300
1	0.865	0.865	20.11	2	0.902	0.902	303, 21.10
$\frac{1}{2}$	0.836	0.835	10.14	$\frac{1}{2}$	0.879	0.879	02.16, 12.11
$\frac{1}{2}$	0.834	0.833	306	$\frac{1}{2}$	0.874	0.874	00.21, 306
1	0.790	0.790	220, 21.10	$\frac{1}{2}$	0.855	0.856	11.18
3	0.784	0.784	222, 308	$\frac{1}{2}$	0.848	0.847	20.17
2	0.778	0.777	20.13	2	0.834	0.833	309, 21.13
				2	0.791	0.790	220
				5	0.784	0.783	223, 30.12

diameter Norelco camera and nickel-filtered copper radiation. The indexing based on unit cell dimensions $a = 3.16$ and $c = 18.33$ is consistent with space group *R3m*. A spectrographic analysis of the mineral showed Mo as the only major constituent; minor amounts of Si, Fe and Al, and a trace of Mg, were reported.

Because of the possibility that other polytypes may be discovered it is proposed that the system of nomenclature introduced by Ramsdell (1947) for SiC, and later adopted for wurtzite, also be used for molybdenite. Following this system, the common hexagonal polytype is designated molybdenite-2H and the new polytype from the Con mine is designated molybdenite-3R.

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