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67. Parasymplesite, a New Mineral Polymorphous with Symplesite

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Symplesite from Felsöbánya was first described by Krenner as a monoclinic crystal.¹⁾ Wolfe investigated with X-rays the same mineral from Lobenstein and demonstrated the triclinic nature of its lattice.²⁾ Unaware of Wolfe's work we analysed the crystal structure of symplesite using the fine vivianite-like specimens from Kiura, Japan.³³⁴⁾ As revealed by this study the crystals from Kiura are not triclinic but monoclinic in symmetry, being closely related to vivianite in structure. On the other hand, our recent re-examination of the original symplesite from Felsöbánya and from Lobenstein, which were kindly put at our disposal by Professor Cliford Frondel of Harvard University, has confirmed in every detail the findings of Wolfe and has placed the dimorphism of symplesite beyond question. We therefore propose here the new name parasymplesite (in analogy with parawollastonite,⁵⁾ paracelsian⁶⁾ etc.) for the monoclinic variety of symplesite whose structure we worked out, retaining, following Frondel,⁷ the name symplesite for the triclinic variety originally described by Krenner and revised by Wolfe.

Occurrence

Parasymplesite from Kiura, Ohita (Bungo), Japan occurs in the contact zone of limestone with granitic diorite, forming usually radiating aggregates of fine crystals grown on massive limonite, an alteration product of various ore minerals (scorodite, arsenopyrite etc.) constituting the deposits.⁴⁾ It is these crystals we describe below.

Crystallography

The crystals of parasymplesite are about $3 \times 1.5 \times 1 \text{ mm}$ in dimensions and dark to light greenish blue in colour. Crystals are thick tabular after (010), with large b(010) and $w(\overline{2}01)$ (Fig. 1a) or stout-prismatic with m(110) and other prisms nearly equally well developed and with relatively large E(502) (Fig. 1b). Numerous No. 4]

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 $\rho_c \text{ and } \varphi_c \mathbf{c}$ deduced from

The unit $\beta = 103^{\circ}50^{\circ}$ Ka, $\lambda = 0.71^{\circ}$

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prismatic faces are observed in both types of crystals, other kinds

a) Fig. 1. Parasymplesite from Kiura

of faces being only occasionally noticed.

The results measurement of by means of a Goldschmidt twocircle reflexion goniometer are given in Table I. The axial ratio has been deduced from the structural data (given below) as follows: a:b:c=0.760:1:0.350,

 $\beta = 103^{\circ}50'.$

Table I. Two-circle goniometry of parasymplesite

Form	ρ _o	φo	ρ _c	φc	Nos. of faces measured
b (010)	89°43′	0°00′	90°00′	0°00′	5
z (1, 20, 0)	90 3	3 7	••	3 58	1
N (190)	90 5	8 19		8 20	1
h (170)	90-5	$11 \ 15$		10 56	1
f (120)	90 2	33 0		$34 \ 10$	1
1 (580)	90 4	40 18	••	$40 \ 15$	$\overline{2}$
k (570)	89 40	$45 \ 37$.,	44 03	$\overline{4}$
H(560)	90 1	49 32		48 28	3
m(110)	90 3	$51 \ 37$		$53 \ 34$	6
M(430)	90 2	$60 \ 12$,,	$61 \ 02$	1
Q (530)	90 8	65 51	.,	66 07	$\overline{2}$
n (210)	90 0	$69 \ 41$	••	$68 \ 47$	1
y (310)	90 1	$76\ 13$		$76 \ 10$	3
o (810)	89 47	83 20	,,	84 43	3
a (100)	89 53	89 20		90 00	4
w(201)	35 3	-90 23	34 58	-90.00	2
E (502)	$42 \ 36$	- 88.17	43 07		1
t (401)	$59 \ 5$	-93 42	58 43	,,	1
v (221)	$44 \ 44$	-48 12	45 27	~ 47 40	3
r (111)	22 0	-32 35	22 35	-3252	$\overset{\circ}{2}$

 ρ_e and φ_e calculated on the basis of the axial ratio: a:b:c=0.760:1:0.350, $\beta=103^{\circ}50'$, deduced from the structural data

The unit cell has the dimensions: $a=10.25\text{\AA}$, $b=13.48\text{\AA}$, $c=4.71\text{\AA}$, $\beta=103^{\circ}50'$,³⁾ containing two molecules of Fe₃(AsO₄)₂ · 8H₂O (Mo Ka, $\lambda=0.710\text{\AA}$). The space group is C 2/m.

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We reproduce in Figs. 2 and 3, the 0-layer Weissenberg photographs of parasymplesite and symplesite (twinned on $(1\overline{10})$), taken with a Wiebenga integrating X-ray goniometer both about the *c* axis (Fe Ka). The X-ray powder photographs, and 'Norelco' spectrograms (Co Ka) of symplesite and parasymplesite are compared



b) Symplesite

Fig. 2. Weissenberg photographs of parasymplesite and symplesite. c-axis, 0-layer. Fe radiations, no filter. Camera diameter 57.3 mm. Coupling 1mm to 2°

in Figs. 4 and 5. The difference as well as similarity of the two minerals in symmetry and texture is well evidenced on these diagrams. (Also see Table II.)

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b) Symplesite

Fig. 4. Powder photographs of parasymplesite and symplesite. Camera diameter 57.3 mm. $CoK\alpha$. 3.5 KV, 9 MA, 3 hrs.

Parasymplesite from Kiura			Symplesite from Felsöbánya		
Ι	20	d (Å)	I	20	d (Å)
18	11.40°	9.006	15	11.45°	8.971
12	12.65	8.119			5 S
17	13.70	7.499	16	13.70	7.499
38	14.55	7.063			
100	15.05	6.830	100	15.15	6.785
9	20.40	5.051	- 7	20.50 ·	5.027
7	23.30	4.429			
8	25.55	4.051	7	25.40	4.069
9	25.90	3.991	8	25.80	4.007
10	27.80	3.723	- 8	27.65 '	3.743
r 9	(30.30	[3.423	8	30.50	3.401
<u>ز 10</u>	130.55	13.395	1		
10	31.95	3.250			
10	32.70	3.177	7	32.70	3.177
9	34.25	3.038	ſ 6	(33.20	(3.131)
9	34.70	2.999	16	33.60	13.095°
11	36.65	2.845	7	35.05	2.970
9	38.80	2.693	7	36.75	2.838
9	40.70	2.572	5	40.60	2.578
• 7	42.00	2.496	$\lceil 2 \rangle$	(42.05)	∫2.493
6	43.55	2.411	12	142.50	12.468
ſ9	ſ45.05	$\int 2.335$	ſ2	$\hat{44.25}$	(2.375)
(թ	45.15	12.330	13.	44.65	12.355
6	45.65	2.306			
ſ6	∫46.55	[2.264]	2	46.15	2.282
16	\ 46.80	2.252	,		
8	50.45	2.099	1	55.25	1.929
6	51.50	2.059	1	57.50	1.860
8ع	63.75	1.694م	1	57.70	1.854
{8	$\{63.90$	$\{1.690$	1	63.10	1.709
l8	164.00	1.688	2	64.20	1.683

Table II. X-ray powder data of parasymplesite and symplesite

'Norelco' X-ray spectrometer. Co K α , $\lambda = 1.78890 \text{Å}$. No filter 16-1-8 slit 1°, 0.006″, 1°. Speed 1′/min. Scale factor 16′. Multiplier 1. Time constant 8 sec.

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Fig. 5. 4 CoKa, no filte [Vol. 30,



1 diameter

.ya d (Ä) 8.971 7.499 6.7855.0274.0694.007 3.7433.401 3.1773.131 3.095 2.970 2.838 2.5782.4932.468 2.375 2.355 2.282.929 .854 ..709 .683 006/, 1°.

Chemical Composition

The chemical analysis of parasymplesite has been carried out by one of the present writers (H. M.) on the specimens which were separated from scorodite and other associated minerals with the result: As_2O_5 38.43%, Fe_2O_3 0.81%, FeO 37.70%, MgO non, CaO non, P_2O_5 non, $H_2O(+)$ 12.70%, $H_2O(-)$ 10.67%, Total 100.33%. This may be represented by the formula, $Fe_{3.03}As_{1.99}O_8 \cdot (H_2O)_{4.08+3.43'}$ or very closely $Fe_3(AsO_4)_2 \cdot 8H_2O$.





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Table III. of par	Optical asympl	properties esite
e∧Z	31°20/	

UN:		01 20
2V		large, negative (?)
Indices	of re	fraction *
æ		1.628
₿		1.660
r		1.705
Pleochr	oism	
X		bluish green
Y		yellowish
Z		brownish yellow

Optical and Other Physical Properties

The optical data are given in Table III. Cleavage is very perfect after (010). Hardness 2. The density has been determined with a pychnometer to be 3.07 gr/cm³ at 20°C against 3.097 gr/cm³ calculated on the basis of the X-ray data above given.

* Determined by the immersion method using dispersion liquids and a prism monochromator

References

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