

(Science Reports of the Yokohama National University, Sec. II. No. 1, 1952)

“Yugawaralite”, A New Zeolite

By

K. SAKURAI and A. HAYASHI

Preface

While studying zeolites from Yugawara Hot Spring area, Kanagawa Prefecture, in 1930, SAKURAI found a zeolite which was quite different from Epistilbite, Desmine, Laumontite, Ptilolite, Chabazite etc. By the morphological study and chemical analysis, I assumed this to be a new unknown zeolite. Later in 1948, I was able from detailed study with HAYASHI's co-operation, to certify this be an unknown new mineral.

Locality and Occurrence

This unknown new mineral is found in rocks near “Fudo-no-taki” Fall (Fig. 1 Atami 1/25,000 Ac) on the right side of the road to Okuyugawara Hot

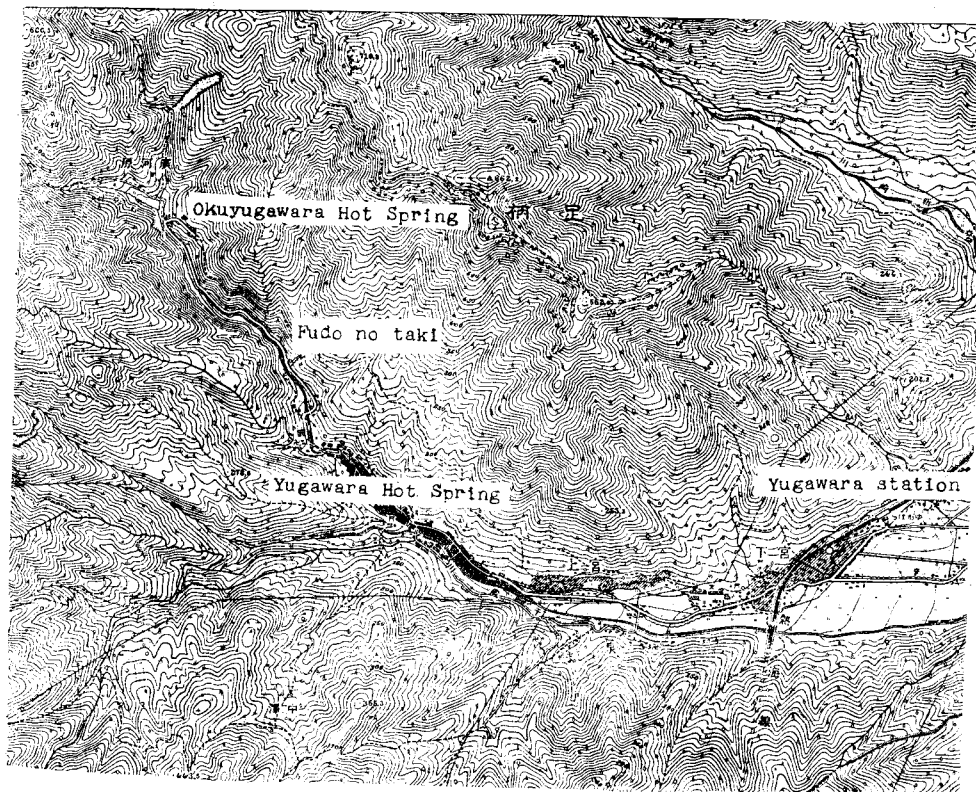


Fig. 1

Spring leading along the Chitose River, this place is about 300 m from Yugawara Hot Spring. Along the Chitose River, "Yugashima" series (lower miocene, tertiary) develops, and "Tenshozan" Andesite, develops over it. Yugawaralite occurs, forming net work, in the Andesite tuff of this Yugashima series, which, being metamorphosed very much by hot spring action, is grayish green near the outcrop. The net work vein is 5~15 mm wide and has many pretty crystals in its cavity. Co-existing mineral is Rock crystal, and a few amount of Laumontite, Chabazite, Ptilolite.

Crystal

Monoclinic, flat crystal. Reaches to 5 mm (a-axis), 2 mm (b-axis) and 10 mm (c-axis), Crystal faces are as follows:

$$a (100), b (010), c (001), m (110), l (120), u (011), p (\bar{1}\bar{1}1).$$

b is considerably developed. The crystal is slightly longer along *c*-axis. Normally, the crystal consists of *b*, *m*, *a*, *c*, *u* and *p*, (Fig. 2) but sometimes, of *b*, *m*, *l*, *u* and *p*. (Fig. 3). Stereo projection is shown on Fig. 4. Often, two or more crystals are in parallel growth along (010). Irregular mass of many crystals are found in the cavity of the vein. Data measured by GOLDSCHMIDT's double circle goniometer are on Table I.

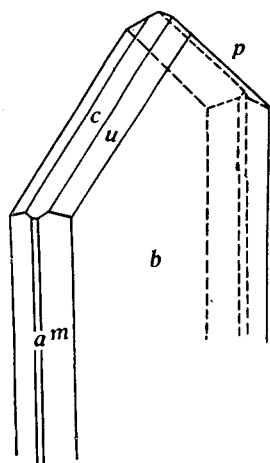


Fig. 2

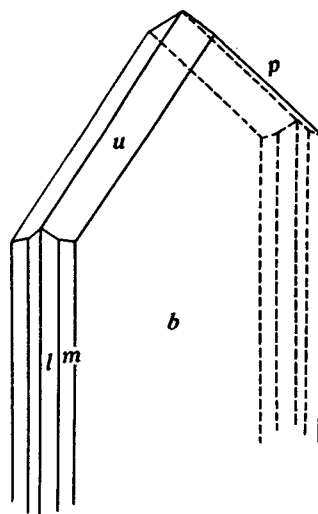


Fig. 3

c (001)
b (010)
a (100)
m (110)
l (210)
p ($\bar{1}\bar{1}1$)
u (011)

On the surface
these data, a
different from

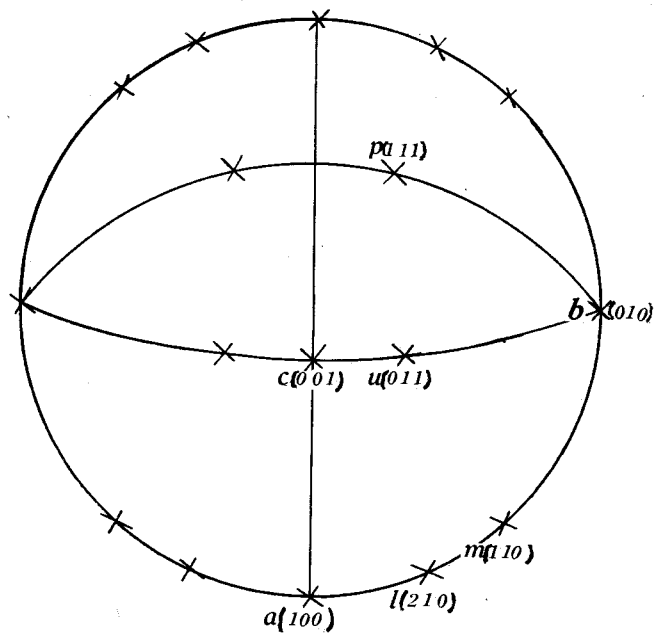


Fig. 4

Table I

	Observed		Calculated ¹⁾		Refraction ²⁾ Strength	Size of ³⁾ Face
	ρ	φ	ρ	φ		
c (001)	21°32'*	89°56'*	21°32'	90°00'	A	M
b (010)	90°00'	0°00'	90°00'	0°00'	A	V. L.
a (100)	90°00'	90°00'	90°00'	90°00'	A	S
m (110)	90°00'	47°48'*	90°00'	47°49½'	B	M
l (210)	90°00'	65°33'	90°00'	65°36'	B	M
p (111)	54°30'*	-59°25'*	54°23'	-59°20'	A	M
u (011)	39°30'	28°58'	39°10½'	29°05'	A	M

1) Calculated from marked * by formula.

2) A: strong. B: middle. C: weak.

3) V.L: very large, M: middle, S: small.

On the surface *b* are often found streaks parallel to principal axis. From these data, axial ratio and axial angle are obtained as follows, which are different from those of other zeolites.

$$a : b : c = 0.9758 : 1 : 0.7129$$

$$\beta = 68^\circ 28'$$

Physical Properties

Colorless transparent or white semi-transparent, strong vitreous lustre. On *b* has often iridescence. Streak: white. Cleavage: incomplete on *b*. Hardness: $4\frac{1}{2}$. Specific gravity: 2.201 (SAKURAI), 2.198 (HAYASHI).

Optical Properties

Optically positive, optic axial plane: perpendicular to (010). $2v$: 56° , 68° , 76° , 89° etc. (different by the measured specimens) Optically negative is not observed.

Refraction index: $\alpha=1.495$ $\beta=1.497$ $\gamma=1.504$ $\gamma-\alpha=0.009$

Shows straight extinction, elongation is negative.

Absorbtion: $\rho < \nu$

Chemical Properties

Results of chemical analysis by Messrs. A. HAYASHI and N. SIMODA are shown on Table II (1, 2).

	1	2
SiO ₂	57.94%	58.44%
Al ₂ O ₃	17.65	17.31
Fe ₂ O ₃	0.35	0.36
CaO	9.79	9.75
MgO	0.86	0.42
MnO	—	—
Na ₂ O	0.38	0.36
K ₂ O	0.41	0.12
H ₂ O (+)	10.70	10.55
H ₂ O (-)	1.80	2.03
	99.88	99.34
	by A. Hayashi (1948)	by N. Shimoda (1951)

The following atomic ratio of the elements in unit cell of Yugawaralite is calculated from volume of unit cell, specific gravity, etc.

Si 19.60, Al 6.99, Fe 0.08, Mg 0.42, Ca 3.56, Na 0.24, K 0.16,

H₂O 14.11, O 54.00.

From these data is obtained molecular formula Ca₄Al₇Si₂₀O₅₄ · 14H₂O. Furthermore, this zeolite has a specific characteristic that is not attacked by cold or hot hydrochloric acid and does not produce gelatine, the latter is decided from non-coloration by methylene-blue.

Nam
group w

The
graph :
(MoK α , 7

a
Axial ra
that fro
The
with tho
lattice d

Yugaw	
CoK α	
D=7	
d (Å)	
	6.4
	5.6
	5.2
	4.7
	4.2
	3.9
	3.7
	3.5
	3.2
	3.0
	2.9

Namely, Yugawaralite, like Mordenite and Ptilolite, belongs to zeolite group which is not attacked by hydrochloric acid.

Study By X-Ray

The following lattice constant is determined from X-ray oscillation photograph around every axis. Camera radius: 3.41 mm, Anticathode: Mo (MoK α , $\lambda=0.71\text{\AA}$)

$$a_0=13.26\text{\AA}, \quad b_0=13.65\text{\AA}, \quad c_0=9.73\text{\AA}, \quad \beta=68^\circ 30'$$

Axial ratio from these data a : b : c = 0.9715 : 1 : 0.7136 coincides very well with that from goniometric data.

The lines of Debye-Scherrer photograph of Yugawaralite do not coincide with those of Mordenite, Heulandite, Epistilbite and Desmine. Table III shows lattice distance and intensity I of these zeolites.

Table III

Yugawaralite		Mordenite		Epistilbite		Heulandite		Desmine	
CoK α D=2.865cm		CuK α D=5.50cm		FeK α D=2.865cm		CuK α D=5.019cm		CuK α D=2.865cm	
d (Å)	I	d (Å)	I	d (Å)	I	d (Å)	I	d (Å)	I
		10.00	8	7.65	8	9.73	1	5.82	5
		8.26	1	6.11	10	8.18	8	5.09	6
		7.26	8	5.25	10	4.89	1	4.57	6
								4.03~	10
6.48	6	6.41	6	4.53	9	4.45	2	4.22	
		5.48	5	3.78	8	3.88	10	3.83	6
5.69	10	4.95	8	3.59	2	3.25	1	3.45	6
5.23	5	4.48	10	3.50	2	3.07	4	3.04	10
4.71	9			3.18	3	2.92	10	2.79	4
		4.30	6	2.96	4	2.77	5	2.61	4
4.27	4	4.14	5	2.77	6	2.68	4	2.36	3
3.90	4	3.74	10	2.60	2	2.37	2	2.08	5
3.72	10	3.51	10	2.31	4	1.94	6	1.84	3
3.52	3	3.33	5	2.17	5	1.77	5	1.68	2
3.20	2	3.13	5	2.11	2	1.71	1	1.60	6
3.07	10	2.98	3	2.02	2	1.64	1	1.56	5
2.91	4	2.72	4	1.92	1	1.48	3	1.46	3
		2.48	3	1.85	4	1.35	2	1.37	2
		2.21	2	1.79	2	1.27	2	1.31	5
		2.05	2	1.71	1	1.23	1	1.24	5

g vitreous lustre.
incomplete on b.
AYASHI).

10). 2v: 56°, 68°,
ly negative is not

$\gamma=0.009$

and N. SIMODA are

4%

1

6

5

12

36

12

55

33

34

I. Shimoda

1)

of Yugawaralite is

0.24, K 0.16,

$\cdot 14\text{H}_2\text{O}$. Further-
attacked by cold or
atter is decided from

We obtain $n=1.04$ by calculation from specific gravity (2.20), molecular formula $(\text{Ca, Mg, Na, K})_{4.83} (\text{Al, Fe})_{7.07} \text{Si}_{19.60} \text{O}_{54} \cdot 14.11 \text{H}_2\text{O}$ (Table II, 1) and volume of unit cell which is obtained from the above mentioned axial length and axial angle of unit cell. Therefore the number n of molecules in unit cell of Yugawaralite is 1. Crystal constants and space groups of zeolites now known, are found in Table IV.

Table IV

	Axial length. (unit Å)			Axial angle.	Space group.	
	a_0	b_0	c_0			
Mordenite	18.25	20.35	7.50	—	D_{2h}^{17} or C_{2v}^{12}	W. H. Taylor (1938)
Heulandite.	7.45	18.70	15.85	$88^\circ 36'$	C_{2h}^3	T. Wyart (1933)
Epistilbite.	14.90	17.87	10.32	$89^\circ 20'$		A. Hayashi (1949)
Yugawaralite.	13.26	13.65	9.73	$68^\circ 28'$		K. Sakurai A. Hayashi (1949)
Desmine.	13.60	18.13	11.29	$52^\circ \pm \frac{1}{2}'$	C_{2h}^3	T. Sekania T. Wyart (1937)
Harmotome.	9.80	14.10	8.66	$55^\circ 10'$	C_{2h}^2 or C_2^2	T. Sekania T. Wyart (1937)
Phillipsite.	10.00	14.25	8.62	$54^\circ 20'$	C_{2h}^2 or C_2^2	P. Chatelain T. Wyart (1938)
Gismondite.	13.68	14.28	10.60	—	D_{2h}^{19} or D_{2h}^{22}	O. Kraus (1939)
Stellerite.	17.52	18.34	13.43	—	V_h^{23} , V	K. Takane (1937)
Chabazite.	13.75	—	14.95	Rhombohedral $a=9.37$ $\alpha=94^\circ 34'$	D_{3d}^5	T. Wyart (1933)
Thomsonite.	13.04	13.06	13.22	—	C_{2v}^{10}	T. Wyart (1933)
Gonnardite.	13.35	13.35	6.65 (?)	—		M. H. Hey F. A. Banister (1932)
Natrolite.	18.3	18.6	6.57	—	C_{2v}^{19}	W. H. Taylor C. A. Meek (1933)
Mesolite.	5.67	6.54	18.44	$90^\circ \pm 20'$	C_2^3 or C_{2v}^{19}	T. Wyart (1933)
Scolesite.	18.44	18.90	6.53	—	$C_s^4 = C_c$	T. Wyart (1933)
Edingtonite.	9.59	9.59	6.53	—	V_d^3	M. H. Hey F. A. Banister (1934)
Analcime.	13.68	—	—	—	O_h^{10}	W. Haritng (1931)

Mr. M. Ko
a). Curve of
is clearly diffe
or Ptilolite (F
what similar t
rather rapidly
400°C, next ra
molecule.

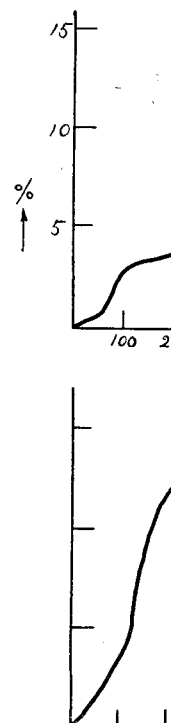


Fig. 5 Cur
A. Y
B. L.
C. D
D. H
E. P

molecular formula
(Table II, I) and
tioned axial length
molecules in unit
ips of zeolites now

Thermal Properties

Mr. M. Koidzumi gives thermal data about several zeolites as follows:
a). Curve of Heating Loss (Fig. 5). The curve of Yugawaralite (Fig. 5, A) is clearly different from that of Heulandite (Fig. 5, C), Deimine (Fig. 5, D) or Ptilolite (Fig. 5, E), and generally de-hydrates in four steps and is somewhat similar to that of Laumontite (Fig. 5, B). Namely, Yugawaralite loses rather rapidly more than 3 molecule of water till 100°C, then slowly 6 till 400°C, next rapidly 4 between 400°C and 450°C, later very slowly 1~1.5 molecule.

W. H. Taylor (1938)
F. Wyart (1933)
A. Hayashi (1949)
K. Sakurai A. Hayashi (1949)
S. Sekania F. Wyart (1937)
S. Sekania F. Wyart (1937)
P. Chatelain F. Wyart (1938)
D. Kraus (1939)
K. Takane (1937)
F. Wyart (1933)
F. Wyart (1933)
M. H. Hey F. A. Banister (1932)
W. H. Taylor C. A. Meek (1933)
F. Wyart (1933)
F. Wyart (1933)
M. H. Hey F. A. Banister (1934)
W. Haritng (1931)

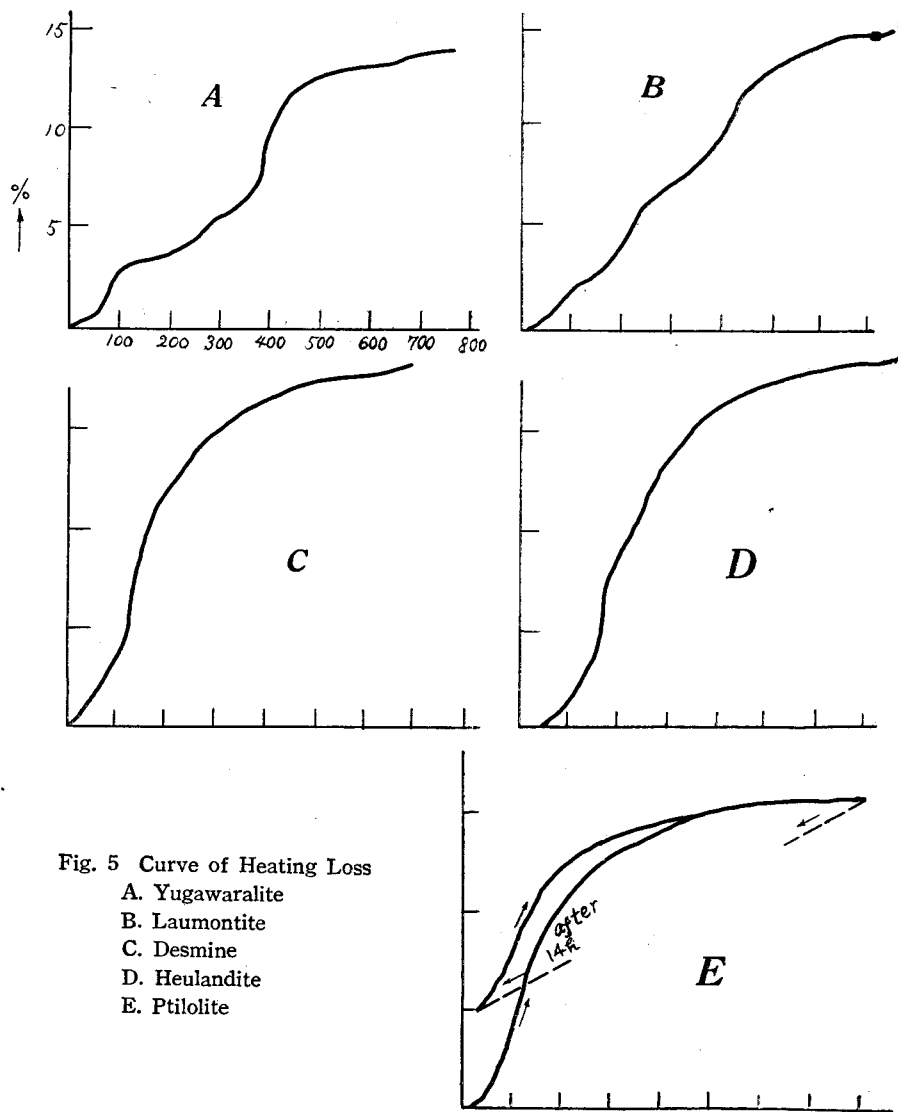


Fig. 5 Curve of Heating Loss
A. Yugawaralite
B. Laumontite
C. Desmine
D. Heulandite
E. Ptilolite

b). Curve of Differential Thermal Analysis (Fig. 6). This curve of Yugawaralite (Fig. 6, C) is very characteristic and quite different from that of other zeolites.

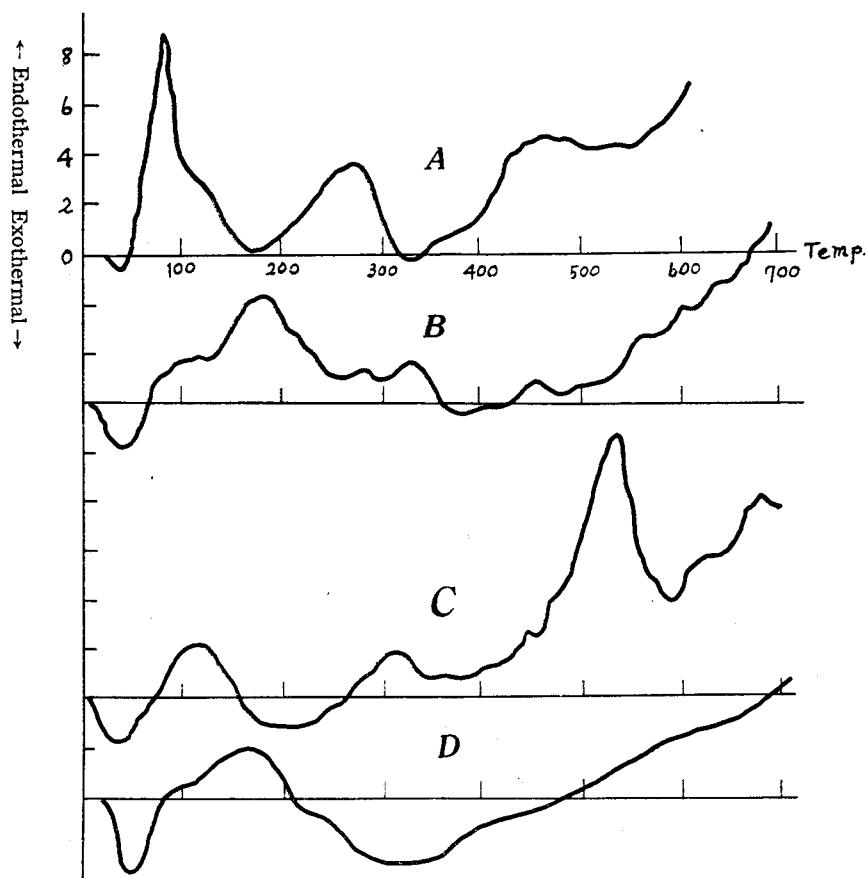


Fig. 6 Curve of Differential Thermal Analysis.
A Laumontite, B Desmine, C Yugawaralite,
D Ptilolite.

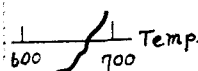
Summary

- 1). Morphological: Monoclinic. $a : b : c = 0.9748 : 1 : 0.7129$ $\beta = 68^\circ 28'$. Crystal faces: c (001), b (001), a (100), m (110), l (120), u (011), p ($\bar{1}11$). b (010) is large. Flat.
- 2). Physical: Transparent, colorless or semitransparent, white. Strong vitreous lustre. Cleavage: imperfect on (010). Hardness: 4.5 Specific gravity: 2.20.
- 3). Optical: Optically positive, Optic axial plane \perp (010), $2V : 56^\circ \sim 89^\circ$
 $\alpha = 1.495$, $\beta = 1.497$, $\gamma = 1.504$


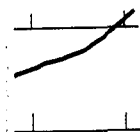
- 4). Chemical: C
- 5). X-ray: Lattice cell contains Debye-Sch
- 6). Thermal: Y differential attention to

By the data a other zeolites, and it "Yugawaralite" Finally, the a his long kind g SHIMODA, T. YOKO

is curve of Yuga-
erent from that of



Temp.
600 700

$\beta=68^{\circ}28'$. Crystal
(11), $p(\bar{1}11)$, $b^{\circ}(010)$

white. Strong vitre-
rdness: 4.5 Specific

, 2V: $56^{\circ}\sim 89^{\circ}$

- 4). Chemical: $\text{Ca}_4\text{Al}_7\text{Si}_{20}\text{O}_{54} \cdot 14\text{H}_2\text{O}$. Not attacked by HCl.
- 5). X-ray: Lattice constant $a_0=13.26$, $b_0=13.69$, $c_0=9.73$, $\beta=68^{\circ}28'$. Unit cell contains 1 molecule.
Debye-Scherrer photograph does not coincide with that of other zeolites.
- 6). Thermal: Yugawaralite dehydrates in 4 steps like Laumontite. Curve of differential thermal analysis is characteristic. Necessary to pay attention to endothermal reaction at 540° .

Conclusion

By the data above mentioned, we can not identify this mineral with the other zeolites, and we may consider this to be a new zeolite. We wish to it "Yugawaralite" after its locality.

Finally, the authors hope to express their hearly thanks to Dr. T. ITO for his long kind guidance, and also to Messrs. E. SORIDA, M. KOIDZUMI, N. SHIMODA, T. YOKOTA, K. ISHIOKA, T. KAWAI and K. NAGASHIMA.