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**ALUMOKLYUCHEVSKITE, $K_3Cu_3AlO_2(SO_4)_4$, A NEW OXYSULPHATE
OF K, Cu AND AI FROM VOCANIC EXHALATIONS, KAMCHATKA,
RUSSIA¹**

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АЛЮМОКЛЮЧЕВСКИТ, $K_3Cu_3AlO_2(SO_4)_4$ — НОВЫЙ ОКСИСУЛЬФАТ К, Си И АІ ИЗ
ВУЛКАНИЧЕСКИХ ЭКСГАЛАЦИЙ КАМЧАТКИ, РОССИЯ

Alumoklyuchevskite $K_3Cu_3AlO_2(SO_4)_4$ is an aluminium analogue of klyuchevskite $K_3C_3Fe^{3+}O_2(SO_4)_4$ (Vergasova e. a., 1989) compare to which the Fe^{3+} position is predominantly occupied by Al. Like klyuchevskite it was found in the volcanic exhalation products on the 2-nd cone of the Northern Breakthrough of the Main Tolbachik fracture eruption (1975—1976) (Main Tolbachik fracture eruption, 1984).

¹ The mineral is approved by the Commission on New Mineral and Mineral Names of the Mineralogical Society of Russia on December 25, 1992, by the Commission on New Mineral and Mineral Names of the International Mineralogical Association on May 26, 1993.



The appearance of alumoklyuchevskite aggregates (a) and single crystals (b) (REM-photo; mlt: 70 (a) and 800 (b)).

Внешний вид агрегатов (а) и кристаллов (б) алюмоключевского (РЕМ-фото, увеличение 70 (а) и 800 (б)).

The appearance of the mineral is similar to that of klyuchevskite. The mineral differs from the latter by the chemical composition, colour, refractive indices, pleochroism and some other properties. It forms the shaftlike aggregates (fig. a) of the long prismatic to needle crystals (fig. b). The crystals are up to 1 mm long and < 0.1 mm thick. Alumoklyuchevskite occurs in aggregates with fedotovite, tenorite, in association with langbeinite, lammerite.

The chemical composition obtained as an average of 26 electron microprobe analyses (mas. %): K₂O 18.68 (16.81—19.61), CuO 31.19 (30.04—32.79), Al₂O₃ 4.65 (3.69—5.71), Fe₂O₃ 3.70 (2.31—5.23), SO₃ 40.70 (39.32—42.00), sum 98.92 (97.57—100.32) (table 1) yields the following formula calculated on the basis of O = 18: K_{3.07}Cu_{3.04}(Al_{0.71}Fe_{0.36})_{1.07}(S_{3.94}O₁₈). The empirical formula is K₃Cu₃AlO₂(SO₄)₄. Operation conditions were 20 kV, 35 nA sample current. Standards used were dolerophanite Cu₂OSO₄ (Cu,S), synthetic magnesiopherrite MgFe₂O₄ (Fe), hanite ZnAl₂O₄ (Al), synthetic orthoclase KAlSi₃O₈ (K).

In alumoklyuchevskite as compared to klyuchevskite Al predominates over Fe³⁺. The content of Fe₂O₃ is 7.86 while Al₂O₃ is 1.38 (mas. %) (Gorskaya e. a., 1992). For alumoklyuchevskite we have 3.61 and 4.78 respectively. Al/Fe ratio for the investigated samples of both minerals varies from 2.51 to 2.56 in klyuchevskite and from 0.26 to 0.88 in



Продолжение рисунка.

alumoklyuchevskite. So we can consider the existance of alumoklyuchevskite—klyuchevskite series.

The composition of the investigated samples doesn't permit to draw the conclusion about the extent of the isomorphic substitution of Fe^{3+} —Al. The members of this series with $\text{Fe}/\text{Al} < 1$ are considered as alumoklyuchevskite.

Chemical composition of alumoklyuchevskite calculated from the ideal formula is: K_2O 18.81, CuO 31.77, Al_2O_3 6.79, SO_3 42.63 (mas. %).

X-ray powder diffraction study of alumoklyuchevskite was carried out in $\text{Co}_{K\alpha}$ radiation with internal Ge standard. The indices for powder diffraction data were obtained using unit cell parameters from the X-ray single crystal study.

The X-ray powder diffraction patterns of klyuchevskite and alumoklyuchevskite are very similar, especially for the most strong reflections (table 2) of alumoklyuchevskite (I — d — hkl): 84—9.15—002, 100—9.04—200, 52—7.20—202, 37—3.781—204, 33—3.757—402, 21—2.786—604. The unit cell parameters calculated by LS method are appeared to be closely allied for the both minerals (table 3), although the volume of alumoklyuchevskite's unit cell is about 4% over that of klyuchevskite due to the parameter $b = 5.139 \text{ \AA}$ (table 3).

The mineral is transparent, dark-green in colour. Strongly pleochroic: Z —dark-green, Y —greyish-green, X —dirty-green. The refractive indices of

Table I

The chemical analyses data for alumoklyuchevskite (mas.%)

N	K ₂ O	CuO	Fe ₂ O ₃	Al ₂ O ₃	SO ₃	Sum	Fe/Al
1	16.81	32.22	3.95	4.71	41.42	99.11	0.54
2	17.57	31.32	4.14	4.13	40.41	97.57	0.64
3	16.78	32.52	3.46	5.28	41.91	99.95	0.42
4	16.85	32.79	3.97	4.79	41.92	100.32	0.53
5	17.50	31.58	2.31	5.64	42.26	99.29	0.26
6	17.17	31.84	3.03	5.71	41.40	99.15	0.34
7	18.23	31.56	3.00	5.15	42.00	99.94	0.37
8	18.92	31.10	4.11	4.42	40.57	99.13	0.59
9	19.20	30.94	5.10	3.69	41.11	100.04	0.88
10	18.50	30.73	3.81	4.74	40.15	97.94	0.51
11	19.49	30.81	3.14	4.76	39.59	97.79	0.42
12	19.41	30.04	3.74	4.46	40.58	98.23	0.54
13	19.46	30.50	3.42	4.65	40.05	98.08	0.47
14	19.05	30.70	3.93	4.47	40.31	98.47	0.56
15	19.15	30.67	4.33	4.38	40.58	99.11	0.63
16	19.16	30.42	3.80	4.57	40.95	98.90	0.53
17	19.61	31.13	3.68	4.45	40.51	99.38	0.53
18	19.24	30.68	3.20	4.63	40.57	98.33	0.44
19	19.29	30.76	3.67	4.59	40.18	99.48	0.51
20	19.44	30.99	2.89	4.80	40.57	98.69	0.38
21	19.51	31.24	2.92	4.75	40.54	98.95	0.39
22	18.98	30.67	5.23	4.01	40.41	99.31	0.83
23	19.06	31.20	4.36	4.19	40.09	98.89	0.66
24	19.05	31.32	4.50	4.42	40.56	99.86	0.65
25	19.06	31.98	3.21	4.88	40.30	99.34	0.42
26	19.16	31.44	3.28	4.70	39.32	97.90	0.47
Average:	18.68	31.19	3.70	4.65	40.70	98.92	0.51

alumoklyuchevskite are less than that of klyuchevskite (table 3) due to the low value of specific refraction for Al₂O₃ (0.242) as compared to that for Fe₂O₃ (0.315) (Mandarino J. A., 1976).

Z is parallel to b axis. Y and X are in the perpendicular plane. It was impossible to determine the orientation of the Y and X owing to the small size of the crystals. 2V angle was not measured for this reason too.

Alumoklyuchevskite's crystals are stretched along the b axis (up to 1 mm long) and practically isometric in the perpendicular section (< 0.1 mm thick). The crystals have perfect cleavage along (h0l).

Hardness is $H = 42$ kgs/mm² (Mohs—2.3). The density of alumoklyuchevskite measured in the hard liquid (Klerichy liquid + water) is 3.1 g/cm³, calculate — 2.95 (5) g/cm³.

The optical and physical constants, X-ray data, chemical composition of alumoklyuchevskite and klyuchevskite show that they are related minerals. Isovalent isomorphism Fe³⁺—Al in alumoklyuchevskite—klyuchevskite series occurs.

The mineral is covered by the white powder if it is exposed to the open air about a week, that is probably connected with the hidratation process.

The mineral is named alumoklyuchevskite due to the chemical and structural relation to klyuchevskite. Type mineral is preserved in Mineralogical museum of Saint-Petersburg Mining Institute.

Table 2

X-ray powder diffraction data for alumoklyuchevskite and klyuchevskite

Alumoklyuchevskite			Klyuchevskite (Vergasova e. a., 1989)		
I/I_0	$d_{\text{meas.}}$, Å	$d_{\text{calc.}}$, Å	hkl	I/I_0	$d_{\text{meas.}}$, Å
12	11.72	11.71	101	10	11.73
84	9.15	9.15	002	95	9.17
100	9.04	9.02	200	100	9.03
				6	8.53*
				2	7.63*
52	7.20	7.20	202	35	7.20
				2	6.60*
6	6.16	6.17	103	5	6.17
10	6.09	6.09	301	8	6.09
8	5.85	5.86	202	7	5.86
12	5.40	5.40	301	10	5.40
1	4.568	4.575	004	6	4.569
15	4.509	4.510	400	25	4.502
8	4.469	4.462	204		
				3	4.398*
8	3.908	3.906	310		
		3.904	303	8	3.905
3	3.874*				
37	3.781	3.782	204		
33	3.757	3.754	402	55	3.762
			501	23	3.678
4	3.592	3.597	404	3	3.594
2	3.560*				
4	3.495*				
10	3.447	3.453	105	20	3.453
15	3.416	3.411	501	28	3.409
2	3.336*				
5	3.176	3.176	413		
4	3.127*				
8	3.077	3.077	318		
		3.086	206	6	3.084
8	3.046	3.050	006	6	3.046
			602		
			600	2	3.010
			510	3	2.918
11	2.813	2.805	406		
21	2.786	2.786	604	15	2.780
6	2.740	2.726	206		
2	2.552	2.555	613		
			512	8	2.651
			701	7	2.624
			116	2	2.509
			022	2	2.470
6	2.340	2.343	505	5	2.344
4	2.308	2.307	604	3	2.311
		2.312	514		
4	2.288	2.288	008	3	2.287
			800	2	2.253
			613	2	2.234
			703	3	2.215
3	2.126	2.125	224		
		2.122	118		
4	2.122	2.120	422	3	2.062
		2.118	125		
8	2.028	2.030	608	2	2.028
		2.028	617		

* — probably the oxysulphate of Cu, K, Fe³⁺.

Table 3

Comparison of some properties for alumoklyuchevskite and klyuchevskite

Character	Alumoklyuchevskite	Klyuchevskite (Vergasova e. a., 1989)	Character	Alumoklyuchevskite	Klyuchevskite (Vergasova e. a., 1989)
Crystal chemical formula	$K_3Cu_3AlO_2(SO_4)_4$	$K_3Cu_3Fe^{3+}O_2(SO_4)_4$	N_g	1.641 (1)	1.680 (1)
System	monoclinic	monoclinic	N_m	1.548 (1)	1.550 (1)
Space group	12	12	N_p	1.542 (1)	1.549 (1)
$a, \text{ \AA}$	18.423 (5)	18.405 (7)	$N_g - N_p$	0.099	0.131
$b, \text{ \AA}$	5.139 (1)	4.940 (2)	$2V^0_{\text{calc}}$	30	11
$c, \text{ \AA}$	18.690 (7)	18.667 (7)	Pleochroism		
$\beta, ^\circ$	101.72 (2)	101.50 (5)	Z	dark-green	dark-olives
$V, \text{ \AA}^3$	1732 (1)	1663 (2)	Y	greyish-green	green
Z	4	4	X	light-green	olives
$D_x, \text{ g/cm}^3$	2.95 (5)	2.98	$D_{\text{meas}}, \text{ g/cm}^3$	3.1 (1)	3.10 (5)
			H (Mohs)	2.3	2.7
			Colour	dark-green (bright-green in thin sections)	dirty-green (with brown shadow)

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