

Article

The Structure of Ferroselite, FeSe_2 , at Pressures up to 46 GPa and Temperatures down to 50 K: A Single-Crystal Micro-Diffraction Analysis

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Abstract: We conducted an in situ crystal structure analysis of ferroselite at non-ambient conditions. The aim is to provide a solid ground to further the understanding of the properties of this material in a broad range of conditions. Ferroselite, marcasite-type FeSe_2 , was studied under high pressures up to 46 GPa and low temperatures, down to 50 K using single-crystal microdiffraction techniques. High pressures and low temperatures were generated using a diamond anvil cell and a cryostat respectively. We found no evidences of structural instability in the explored P-T space. The deformation of the orthorhombic lattice is slightly anisotropic. As expected, the compressibility of the Se-Se dumbbell, the longer bond in the structure, is larger than that of the Fe-Se bonds. There are two octahedral Fe-Se bonds, the short bond, with multiplicity two, is slightly more compressible than the long bond, with multiplicity four; as a consequence the octahedral tetragonal compression slightly increases under pressure. We also achieved a robust structural analysis of ferroselite at low temperature in the diamond anvil cell. Structural changes upon temperature decrease are small but qualitatively similar to those produced by pressure.

Keywords: FeSe_2 ; high pressure; low temperature; single-crystal diffraction

1. Introduction

Iron selenides form economically important ore deposits and are relevant to the geochemical cycle of chalcogenides. In material science, dichalcogenides are extensively explored for solar energy applications because of their suitable thermoelectric and optical properties along with their availability and low toxicity [1–3]. Compounds with the marcasite crystal structure display a variety of intriguing physical properties intimately related to their structural arrangements [4]. Furthermore, the marcasite structure type is adopted by several interesting high-pressure phases such as Fe, Rh and Os pernitrides [5–7].

Ferroselite is a mineral of the chalcogenide series with end-member composition FeSe_2 and with the marcasite-type crystal structure. Ferroselite is the stable phase of iron diselenide at ambient conditions. Upon heating at ambient pressure, FeSe_2 does not show phase transitions until its decomposition at 850 K [8]; upon heating at 1200 K under moderately high pressure (65 Kbar) iron diselenide adopts the pyrite structure type [9,10]. Iron is bonded to six selenium atoms in ferroselite, whereas selenium forms a monatomic bond and 3 bonds with iron (Figure 1). The FeSe_6 edge-sharing octahedra form chains along the *c*-direction, while the Se-Se dumbbell bond lies in the *ab* plane

connecting octahedral chains. In the marcasite-type structure, symmetry constraints impose some degree of distortion to the coordination geometries. The octahedron shows angular distortion and features two different bond lengths, in ferroselite those with multiplicity two are slightly shorter than the four equatorial bonds. Iron is located in $2a$ with all symmetry-constrained coordinated whereas selenium, located in $4g$, shows variable coordinates x and y .

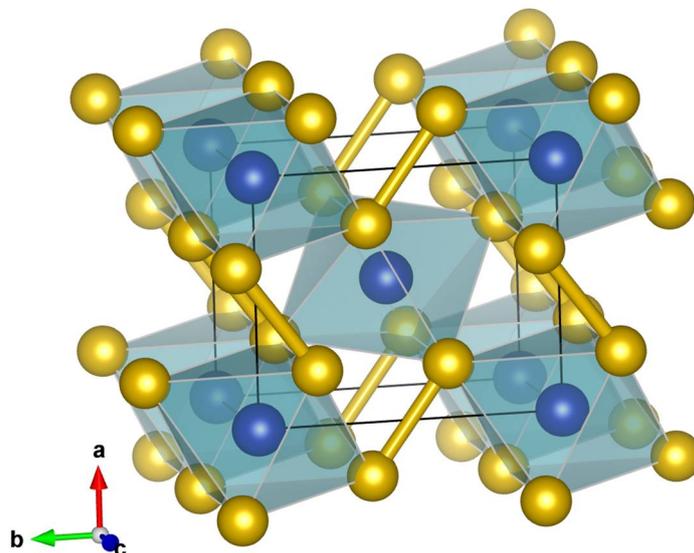


Figure 1. Representation of the crystal structure of marcasite-type FeSe_2 generated with VESTA [11]. Iron atoms (blue) are coordinated to 6 selenium atoms (yellow) defining a tetragonally compressed octahedron. Selenium dumbbells lie in the ab plane and connect the chains of octahedra running parallel to the c -axis.

The thermal expansion of the ferroselite lattice has been explored from ambient conditions up to decomposition temperatures [8], while magnetism and electrical properties were explored in a broad range of temperatures [2]. Mechanically alloyed ferroselite nanocrystalline materials were studied under pressure via absorption spectroscopy up to 19 GPa, showing no evidence of phase transitions [12]. The elastic properties of FeSe_2 were recently determined by means of first-principles calculations [13].

To study the bulk and atomic response of FeSe_2 to external high pressure and to low temperature we performed synchrotron single-crystal microdiffraction experiments using a diamond anvil cell and a cryostat to generate target conditions. We performed three different experiments: (i) compression at ambient temperature up to ~ 46 GPa; (ii) cooling down to 50 K at ambient pressure; (iii) a combined high-pressure low-temperature experiment down to 110 K at 3.8 GPa. Both low-temperature experiments were conducted with single crystals loaded in a diamond anvil cell (DAC) contained in a cryostat (hereafter DAC and cryostat).

2. Materials and Methods

The specimen investigated in this study is a mineral from Paradox Valley, Uravan District, Montrose County, Colorado, USA obtained from the RRUFF collection (RRUFF.info/R070461). The composition reported in the RRUFF database [14] was determined via electron microprobe analysis. Measured elements were Se, Fe, Pb, S, Zn, Cu, Ag; within experimental resolution, the sample is pure and stoichiometric.

Conditions of high pressure and of low temperature were generated with a 4-post diamond anvil cell (DAC) and with a liquid-flow helium cryostat (Figure 2). The DAC was equipped with conical diamonds anvils [15] of 85° aperture and 0.3 and 0.6 mm culet diameter for high pressure and low

temperature experiments respectively. Gaskets were fabricated from pure Re or W foils; 160 and 360 μm diameter holes in the center of 35 μm thick indentations provided the sample chambers for the high-pressure and the low-temperature experiments respectively. The sample chambers were filled with pre-pressurize neon in order to maintain quasi-hydrostatic stress on the crystals in the whole range of experimental conditions.

X-ray microdiffraction data (Figure S1) were collected at the insertion device station 16ID-B of HPCAT, Sector 16, Advanced Photon Source, Argonne National Laboratory. Experiments were performed using hard X-rays ($\lambda = 0.40662, 0.36793 \text{ \AA}$) focused to about $5 \times 5 \mu\text{m}$ FWHM at the sample position. The experimental station was equipped with a heavy-duty motorized sample stage suitable for measurements with a cryostat for high pressure studies. The cryostat provided wide angular X-ray access and was equipped with a gas membrane for pressure control. Diffracted X-rays were collected with the rotation method (vertical ω axis) using a MAR165-CCD area detector (marXperts, Norderstedt, Schleswig-Holstein, Germany) and a MAR345IP detector (marXperts, Norderstedt, Schleswig-Holstein, Germany), both were calibrated using powder patterns of CeO_2 standard and the GSASII software [16]. The rotation range was 72° for high-pressure data and 60° for low-temperature data, the X-ray access was reduced by the gas membrane and cryostat body in the latter case. For the samples in the cryostat diffraction images were collected with the detector (CCD) at three different locations in the horizontal direction perpendicular to the beam. The maximum resolution achieved was 0.6 \AA .

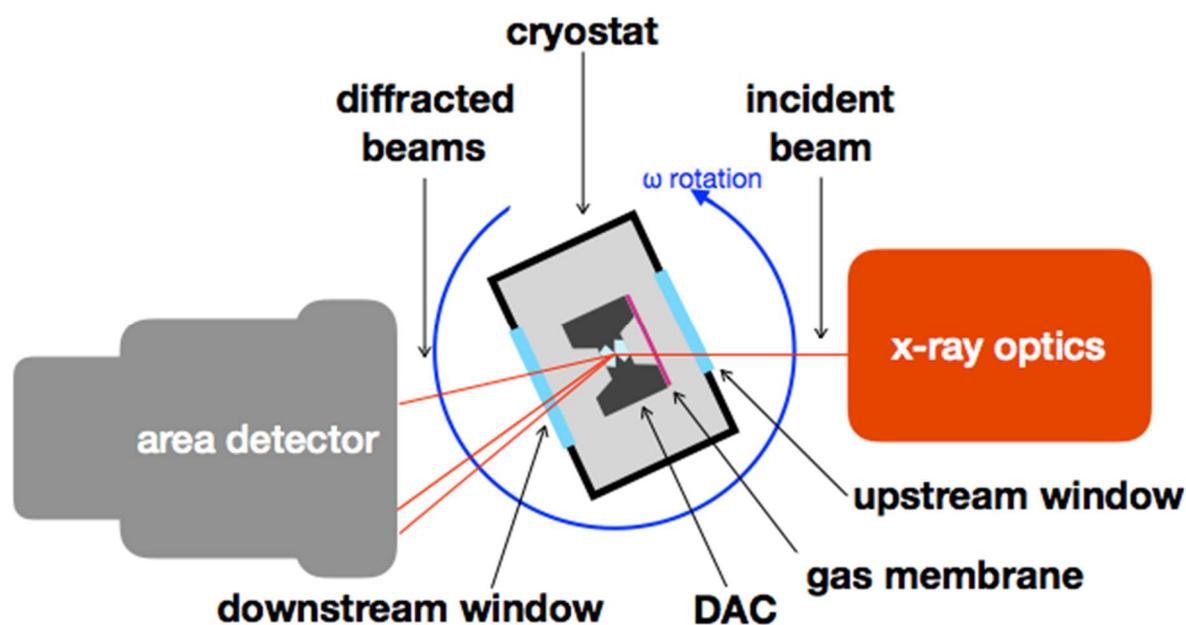


Figure 2. Schematic representation of the high-pressure low-temperature setup at beamline 16ID-B of the APS, ANL.

Pressure was calibrated using the equation of state of platinum [17] for ambient temperature data and ruby [18] for low-temperature data. Temperature was measured using silicon diode temperature sensors positioned on the DAC body and on the copper block, the two differ by less than two degrees during data collection. Data reduction was performed GSE_ADA & RSV [19], WinGX [20], and DIOPTAS [21]. Structural refinements were carried out using Shelxl [22]. Standard powder patterns were analyzed with GSASII [16].

3. Results and Discussion

3.1. Ambient Conditions Structural Refinements

Measuring structure factors in the diamond anvil cell with a micron-sized X-ray beam is an established technique providing extremely valuable and robust results in spite of its challenges. In addition to high-pressure measurements, here we collected data from small single crystals loaded in the DAC which was then loaded in a cryostat (Figure 2). The cryostat we used for high pressure measurements is a relatively bulky device with several connections, the most cumbersome being the liquid helium supply line and the vacuum line. These connections caused an increase in the sphere of confusion of the rotation axis. Because crystals are roughly 20 μm in diameter and the beam is around 5 μm FWHM, we anticipated that the X-ray flux on the crystals would vary more dramatically during data collection compared with ambient temperature measurements. As a consequence, the set of observed structure factors would not be in scale. Furthermore, for relatively low-symmetry crystals, a reliable empirical correction calculated from comparing sets of equivalent reflections might be difficult to define due to the low data redundancy. To compensate for such effects we loaded three crystals with different crystallographic orientations in the diamond anvil cell for low temperature work (Table S1). Furthermore, we collected several rotation images in a small grid pattern around the crystal center for a few datapoints. In Table 1 literature data are compared with some of our results. All data collections were performed with samples in environmental cells, DAC and DAC and cryostat, before conditions were changed. In addition to the scale factor, we refined the two symmetry-unconstrained fractional coordinates of Se and isotropic displacement parameters for both atoms for a total of 5 variables. Overall results are in good agreement. We note that: (i) because we always measured a good number of reflections and we only had 2 positional parameters to refine for the heavier element, these were always reasonable, including high R factors refinements; (ii) merging grid diffraction patterns provided in most cases excellent results and low disagreements between equivalent reflections and better results than empirical corrections; (iii) datasets are not uniform because, we infer, the increase in the sphere of confusion caused by the cryostat is not fully reproducible. In conclusion, it appears beneficial to acquire redundant datasets, this allows for adopting different data reduction strategies and provides the best likelihood that robust refinements can be obtained.

Table 1. Ambient-conditions unit-cell parameters and atomic fractional coordinates from the literature and from this work (see text).

Sample	$N_{\text{all}}, N_{\text{ind}}$	a (Å)	b (Å)	c (Å)	R_{eq}, R_1 (%)	x (Se)	y (Se)	Ueq-Fe (Å ²)	Ueq-Se (Å ²)
Ref. [23]		4.8002(4)	5.7823(5)	3.5834(4)		0.2127(6)	0.3701(5)		
Ref. [24]		4.804(2)	5.784(3)	3.586(2)		0.2134(2)	0.3690(1)		
RRUFF ^(a)		4.795(3)	5.777(4)	3.584(1)					
Ref. [2]		4.8031(6)	5.7849(2)	3.5840(4)		0.2127(2)	0.3691(7)		
Measurement in the DAC									
DAC	204, 98	4.801(3)	5.787(2)	3.5859(7)	12, 4.9	0.213593)	0.3692(2)	0.0095(5)	0.0099(5)
Measurement in the DAC and cryostat									
C1 ^(b)	205, 82	4.804(4)	5.781(3)	3.5814(7)	8.4, 4.4	0.2133(3)	0.3692(1)	0.0056(5)	0.0059(4)
C2 ^(c)	253, 103	4.8016(9)	5.777(1)	3.5850(8)	9.6, 5.4	0.2137(2)	0.36921(13)	0.0090(5)	0.0090(4)
C3 ^(b)	214, 94	4.799(1)	5.785(2)	3.5820(4)	33, 12	0.2130(7)	0.3699(3)	0.0067(12)	0.0086(10)

^(a) same specimen of the present work RRUFF.info/R070461; ^(b) no grid scan, no empirical corrections;

^(c) three detector positions, hkl from merged grid scans, no empirical corrections.

3.2. High Pressure

Diffraction data of ferroselite were collected up to 46 GPa (Table 2). There are no indications of phase transitions but the highest pressure pattern shows moderate peak broadening that might be the result of non-hydrostatic stress as well as the manifestation of an incipient phase transition. The datapoint at 3.63 GPa was collected with the DAC and cryostat before decreasing temperature,

structural parameters at this pressure are in the same trend of data collected in the DAC, confirming our ability to collect full datasets in these conditions without introducing systematic errors in our analysis. The anisotropy of the deformation of the orthorhombic cell of ferroselite under quasi-hydrostatic compression is clear after 17 GPa. As shown in Figure 3 the lattice is more compressible in the direction of the *a*-axis and is stiffer along the *b*-axis. This observation is consistent with the recently predicted behavior [13]. The bulk compression of marcasite-FeSe₂ can be modeled by a second-order Birch–Murnaghan EoS with an ambient bulk modulus, K_0 , of 121.6 GPa (blue curve in Figure 4). Although we collected too few datasets to reliably fit a third order EoS, it can be inferred from the plot in Figure 4 that a third order EoS might be more appropriate to describe the compressibility of this material, data in the low-pressure range fall below the blue curve whereas the highest pressure datapoint falls above the curve. Also, values significantly larger than 4 for the first derivative of the bulk modulus, K_0' , have been suggested for FeS₂ marcasite [25] and several other marcasite-type chalcogenides. Fixing K_0' to 4.6, a value that both leads to best fitting and is close to the value in marcasite, and using the third-order Birch–Murnaghan EoS results in a K_0 of 114.1 GPa. More data ought to be collected in order to better constrain the bulk modulus and its first derivative. We could not however satisfactorily fit our data with the same EoS equation and using the recently proposed bulk modulus of 74.7 GPa [13].

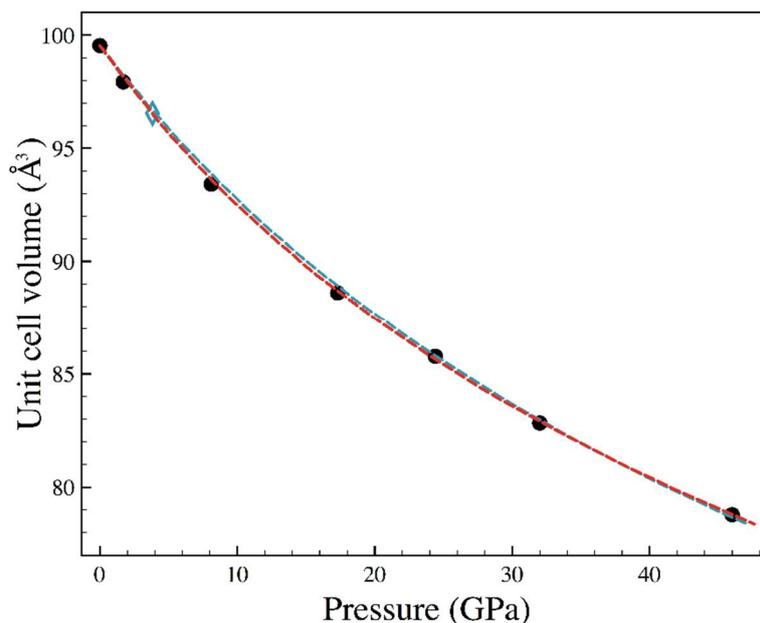


Figure 3. Relative compression of the orthorhombic unit-cell edges.

Table 2. Unit-cell parameters and atomic fractional coordinates of ferroselite at high pressure.

P (GPa)	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>x</i> (Se)	<i>y</i> (Se)	Ueq-Fe (Å ²)	Ueq-Se (Å ²)
1.7	4.771(2)	5.756(1)	3.5663(6)	0.2129(4)	0.36884(13)	0.0079(6)	0.0083(5)
3.83 ^(a)	4.751(2)	5.729(2)	3.5467(11)	0.2125(6)	0.3687(7)	0.014(2)	0.013(2)
8.1	4.698(3)	5.667(2)	3.5088(7)	0.2123(6)	0.3688(2)	0.0083(8)	0.0090(7)
17.3	4.603(3)	5.581(2)	3.4492(9)	0.2123(3)	0.36854(9)	0.0085(5)	0.0088(5)
24.4	4.555(2)	5.524(12)	3.4090(5)	0.2114(4)	0.3683(2)	0.0066(9)	0.0057(7)
32	4.488(4)	5.471(3)	3.3734(11)	0.2111(6)	0.3676(2)	0.0063(13)	0.0064(12)
46	4.404(3)	5.391(2)	3.3185(9)	0.2103(4)	0.36689(13)	0.0066(9)	0.0070(8)

^(a) DAC and cryostat measurement, refinement from merged grid scan images.

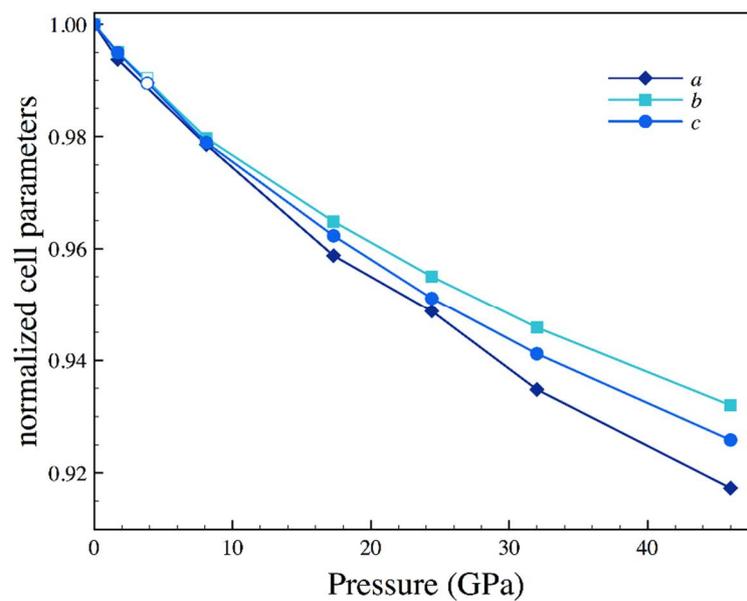


Figure 4. Bulk compressibility of FeSe₂. The blue symbol shows data collected at ambient pressure in DAC and cryostat. The blue line shows a second order Birch–Murnaghan EoS fit, the red line a third order Birch–Murnaghan EoS fit (see text).

Robust crystal structure analysis allows for exploring the changes in atomic arrangement with pressure. Figure 5 shows the pressure dependence of interatomic distances and their relative variations. As could be expected, the longest bond, the Se–Se dumbbell, shows the greatest compressibility. The octahedral bonds however show a less obvious behavior, with the shorter bond being more compressible than the longer bond, hence the octahedral distortion increases with pressure.

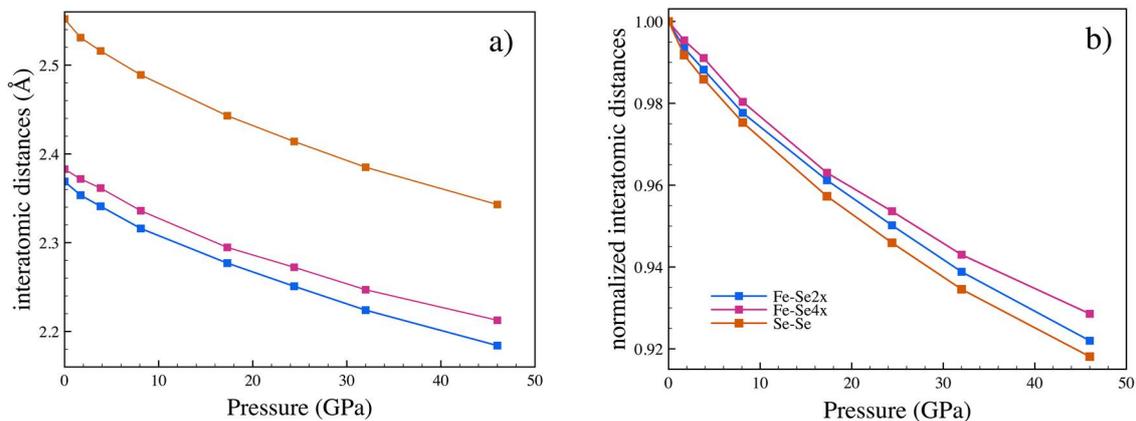


Figure 5. Absolute (a) and relative (b) interatomic distances as a function of pressure. Uncertainties are smaller than the symbols' sizes.

3.3. Low Temperature

Structural data of FeSe₂ were collected ambient pressure at three temperatures 198.2, 148.4, and 50.4 K and at about 3.7 GPa at 197 and 116 K (Table 3). Upon attempting to maintain a constant pressure in the sample chamber during further cooling by increasing the gas membrane pressure, the experiment failed abruptly when the load on the DAC applied with the gas membrane was rapidly transferred to the diamond anvils upon overcoming the DAC friction.

Table 3. Unit-cell parameters and atomic fractional coordinates at low temperature and both ambient pressure and high pressure. For ambient pressure, weighed averages of three crystals are reported.

T (K)	P (GPa)	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>x</i> (Se)	<i>y</i> (Se)
198.2	10 ^{−4}	4.7885	5.7782	3.5821	0.2148(4)	0.3692(2)
148.4	10 ^{−4}	4.7893	5.7775	3.5795	0.2135(2)	0.3691(2)
50.4	10 ^{−4}	4.7834	5.7767	3.5789	0.2138(3)	0.3694(2)
197	3.68	4.751(2)	5.723(2)	3.5501(11)	0.2131(7)	0.3699(6)
116	3.64	4.7446(11)	5.772(2)	3.5512(8)	0.2130(6)	0.3695(7)

We expected the effect on structural parameters of lowering temperature from 300 to 50 K to be small, close to the resolution of our experiment. Hence we loaded three crystals with different orientations for this experiment in order to obtain a more complete sampling of the reciprocal space and increase data redundancy. Because of the limited access to the reciprocal space different crystallographic directions can be probed with different precision in differently oriented crystals (Table S1) as can be inferred inspecting error bars in Figure 6A–C. As a consequence, unit cell variations of individual crystals hardly show discernible patterns; however trends are appreciable when weighted average are considered (Figure 6, black symbols). As for the high pressure behavior and for the high temperature behavior [8], the greatest lattice parameters variations are observed in the direction of the *a*-axis (Figure 6D). Unit-cell edges variations along the other principal axes are close to uncertainties.

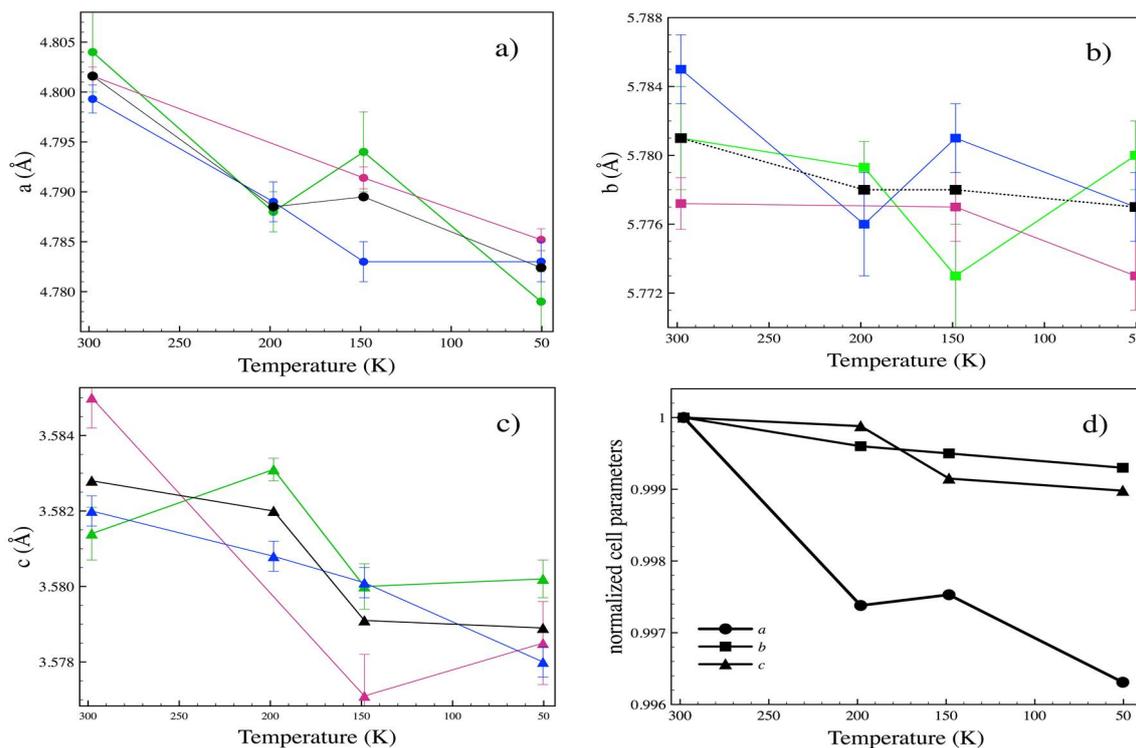


Figure 6. Unit-cell parameters as a function of temperature at ambient pressure. Absolute variations of lattice vectors *a*, *b* and *c* are shown in (a), (b) and (c) respectively; relative variations of the three vectors are shown in (d). Red, Blue and green symbols: different crystals in the same sample chamber; black: weighted average of the three crystals.

The anisotropy of the lattice variations observed with temperature decrease is qualitatively similar to that observed with pressure. The unit-cell volume decreases by ~0.6% upon cooling from ambient temperatures down to 50 K (Figure 7, Table 3), a change that corresponds to a pressure of 0.8–0.9 GPa

depending on the EoS adopted (see above). The volume-pressure trend is appreciably non linear, and it might in part reflect changes in the physical properties of the material upon cooling [2].

Variations of the Fe-Se bond lengths are also within uncertainties as shown in Figure 8. The Se-Se bond shrinkage is clearer, so it could be argued that variations in the unit-cell length along the a -axis and volume are mostly attributed to the monoatomic bond.

We were able to collect just two pressure points at combined low temperature and high pressure (Table 3). Variations on the structure of ferroselite induced by temperature at this pressure are within uncertainties.

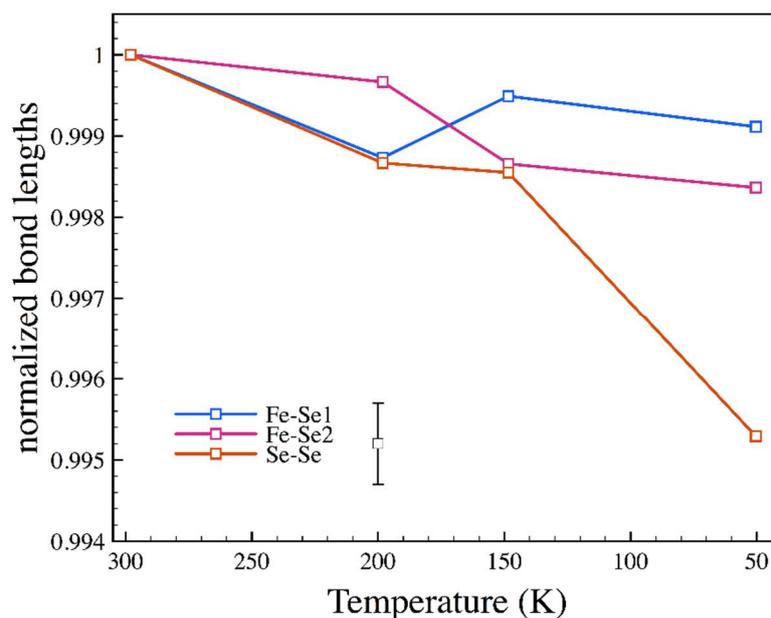


Figure 7. Variation of the unit-cell volume as a function of temperature at ambient pressure. Red, green and blue symbols show values of three different crystals whereas black symbols refer to weighted average values.

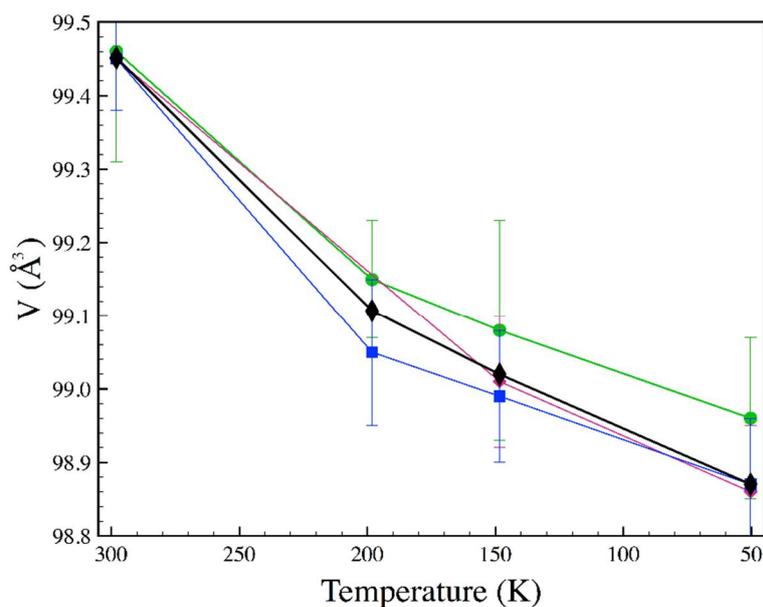


Figure 8. Normalized bond lengths vs temperature at ambient pressure. A representative error bar is shown in black.

4. Conclusions

The physical properties of materials important for critical technologies such as solar energy ought to be defined in great detail. Studying a material's behavior in a broad range of conditions allows more stringent constraints to modeling hence a better general understanding of the material. We conducted a detailed examination of the crystal structure of marcasite-type iron diselenide at pressures up to 46 GPa, temperatures down to 50.4 K and combined high pressure and low temperature conditions of ~3.8 GPa and 197 and 116 K. The phase shows no clear signs of phase transitions in this range, even though it is probably metastable at the highest pressures, considering that it transforms to the pyrite structure above ~6.5 GPa upon heating [9]. We described in details the anisotropy of the lattice response to external conditions and changes in the atomic arrangement.

Maintaining sufficient centering of microcrystals while performing rotation data collections of samples in bulky environmental cells such as combined DAC and cryostat is challenging and not necessarily reproducible. Hence we find that collecting redundant datasets, in this case grid scans, is the safest way to gather datasets from which reliable structure factors can be extracted. Multiple crystal orientations allow obtaining a uniform precision in lattice parameters determinations in addition to better coverage of the reciprocal space.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2073-4352/8/7/289/s1>, Figure S1: Representative diffraction pattern of ferroselite collected in the DAC, Table S1: Orientation matrices of the three crystals (C1, C2, C3) studied at low temperature.

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Conflicts of Interest: The authors declare no conflict of interest.

References

1. Gudelli, V.K.; Kanchana, V.; Vaitheeswaran, G.; Valsakumar, M.C.; Mahanti, S.D. Thermoelectric properties of marcasite and pyrite FeX_2 (X = Se, Te): A first principle study. *RSC Adv.* **2014**, *4*, 9424–9431. [CrossRef]
2. Li, G.; Zhang, B.; Rao, J.; Gonzalez, D.H.; Blake, G.R.; de Groot, R.A.; Palstra, T.T.M. Effect of vacancies on magnetism, electrical transport, and thermoelectric performance of marcasite $\text{FeSe}_{2-\delta}$ ($\delta = 0.05$). *Chem. Mater.* **2015**, *27*, 8220–8229. [CrossRef]
3. Ghosh, A.; Thangavel, R. Electronic structure and optical properties of iron-based chalcogenide FeX_2 (X = S, Se, Te) for photovoltaic applications: A first principle study. *Indian J. Phys.* **2017**, *91*, 1339–1344. [CrossRef]
4. Goodenough, J.B. Energy bands in TX_2 compounds with pyrite, marcasite, and arsenopyrite structures. *J. Solid State Chem.* **1972**, *5*, 144–152. [CrossRef]
5. Wang, Y.X.; Arai, M.; Sasaki, T. Marcasite osmium nitride with high bulk modulus: First-principles calculations. *Appl. Phys. Lett.* **2007**, *90*, 061922. [CrossRef]
6. Niwa, K.; Dzivenko, D.; Suzuki, K.; Riedel, R.; Troyan, I.; Eremets, M.; Hasegawa, M. High pressure synthesis of marcasite-type rhodium pernitride. *Inorg. Chem.* **2014**, *53*, 697–699. [CrossRef] [PubMed]
7. Wang, Z.; Li, Y.; Li, H.; Harran, I.; Jia, M.; Wang, H.; Chen, Y.; Wang, H.; Wu, N. Prediction and characterization of the marcasite phase of iron pernitride under high pressure. *J. Alloys Compd.* **2017**, *702*, 132–137. [CrossRef]
8. Kjekshus, A.; Rakke, T. Compounds with the marcasite type crystal structure. XI. High temperature studies of chalcogenides. *Acta Chem. Scand.* **1975**, *A29*, 443–452. [CrossRef]

9. Bither, T.A.; Prewitt, C.T.; Gillson, J.L.; Bierstedt, P.E.; Flippen, R.B.; Young, H.S. New transition metal dichalcogenides formed at high pressure. *Solid State Commun.* **1966**, *4*, 533–535. [[CrossRef](#)]
10. Bither, T.A.; Bouchard, R.J.; Cloud, W.H.; Donohue, P.C.; Siemons, W.J. Transition metal pyrite dichalcogenides. High-pressure synthesis and correlation of properties. *Inorg. Chem.* **1968**, *7*, 2208–2220. [[CrossRef](#)]
11. Momma, K.; Izumi, F. VESTA3 for three-dimensional visualization of crystal, volumetric and morphology data. *J. Appl. Crystallogr.* **2011**, *44*, 1272–1276. [[CrossRef](#)]
12. Campos, C.E.M.; de Lima, J.C.; Grandi, T.A.; Machado, K.D.; Itié, J.P.; Polian, A. Pressure-induced effects on the structural properties of iron selenides produced by mechano-synthesis. *J. Phys. Condens. Matter* **2004**, *16*, 8485–8490. [[CrossRef](#)]
13. Tian, X.-H.; Zhang, J.-M. The structural, elastic, electronic and optical properties of orthorhombic FeX_2 (X = S, Se, Te). *Superlattices Microstruct.* **2018**, *119*, 201–211. [[CrossRef](#)]
14. Lafuente, B.; Downs, R.T.; Yang, H.; Stone, N. The power of databases: The RRUFF project. In *Highlights in Mineralogical Crystallography*; Armbruster, T., Danisi, R.M., Eds.; W. De Gruyter: Berlin, Germany, 2015; pp. 1–30.
15. Boehler, R.; De Hantsetters, K. New anvil designs in diamond-cells. *High Press. Res.* **2004**, *24*, 391–396. [[CrossRef](#)]
16. Toby, B.H.; Von Dreele, R.B. GSAS-II: The genesis of a modern open-source all purpose crystallography software package. *J. Appl. Crystallogr.* **2013**, *46*, 544–549. [[CrossRef](#)]
17. Matsui, M.; Ito, E.; Katsura, T.; Yamazaki, D.; Yoshino, T.; Yokoyama, A.; Funakoshi, K.I. The temperature-pressure-volume equation of state of platinum. *J. Appl. Phys.* **2009**, *105*, 013505. [[CrossRef](#)]
18. Mao, H.; Xu, J.; Bell, P. Calibration of the ruby pressure gauge to 800-Kbar under quasi-hydrostatic conditions. *J. Geophys. Res. Solid Earth Planets* **1986**, *91*, 4673–4676. [[CrossRef](#)]
19. Dera, P.; Zhuravlev, K.; Prakapenka, V.; Rivers, M.L.; Finkelstein, G.J.; Grubor-Urosevic, O.; Tschauer, O.; Clark, S.M.; Downs, R.T. High pressure single-crystal micro X-ray diffraction analysis with GSE_ADA and RSV software. *High Press. Res.* **2013**, *33*, 466–484. [[CrossRef](#)]
20. Farrugia, L.J. WinGX and ORTEP for Windows: An update. *J. Appl. Crystallogr.* **2012**, *45*, 849–854. [[CrossRef](#)]
21. Prescher, C.; Prakapenka, V.B. DIOPTAS: A program for reduction of two-dimensional X-ray diffraction data and data exploration. *High Press. Res.* **2015**, *35*, 223–230. [[CrossRef](#)]
22. Sheldrick, G.M. A short history of SHELX. *Acta Crystallogr. Sect. A* **2008**, *64*, 112–122. [[CrossRef](#)] [[PubMed](#)]
23. Kjekshus, A.; Rakke, T.; Andresen, A.F. Compounds with the marcasite type crystal structure. IX. Structural data for FeAs_2 , FeSe_2 , NiAs_2 , NiSb_2 , and CuSe_2 . *Acta Chem. Scand.* **1974**, *A28*, 996–1000. [[CrossRef](#)]
24. Pickardt, J.; Reuter, B.; Riedel, E.; Söchtig, J. On the formation of FeSe_2 single crystals by chemical transport reactions. *J. Solid State Chem.* **1975**, *15*, 366–368. [[CrossRef](#)]
25. Tian, X.-H.; Zhang, J.-M. The structural, elastic and electronic properties of marcasite FeS_2 under pressure. *J. Phys. Chem. Solids* **2018**, *118*, 88–94. [[CrossRef](#)]

