GEORGIUS AGRICOLA DE RE METALLICA

TRANSLATED FROM THE FIRST LATIN EDITION OF 1556

with

Biographical Introduction, Annotations and Appendices upon the Development of Mining Methods, Metallurgical Processes, Geology, Mineralogy & Mining Law from the earliest times to the 16th Century

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schist, marble, and also in stone which easily melts in fire of the second degree, and which is sometimes so porous that it seems completely decomposed. Lastly, gold is found in pyrites, though rarely in large quantities.

When considering silver ores other than native silver, those ores are

ARSENICAL MINERALS. Metallic arsenic was unknown, although it has been maintained that a substance mentioned by Albertus Magnus (De Rebus Metallicis) was the metallic form. Agricola, who was familiar with all Albertus's writings, makes no mention of it, and it appears to us that the statement of Albertus referred only to the oxide from sublimation. Our word "arsenic" obviously takes root in the Greek for orpiment, which was also used by Pliny (XXXIV, 56) as arrhenicum, and later was modified to arsenicum by the Alchemists, who applied it to the oxide. Agricola gives the following in Bermannus (p. 448), who has been previously discussing realgar and orpiment:—"Ancon: Avicenna also has a white variety. Bermannus: I cannot at all believe in a mineral of a white "colour; perhaps he was thinking of an artificial product; there are two which the Alchemists "make, one yellow and the other white, and they are accounted the most powerful poisons "to-day, and are called only by the name arsenicum." In De Natura Fossilium (p. 219) is described the making of "the white variety" by sublimating orpiment, and also it is noted that realgar can be made from orpiment by heating the latter for five horns in a sealed crucible. In De Re Metallica (Book X.), he refers to auripigmentum facticum, and no doubt means the realgar made from orpiment. The four minerals of arsenic base mentioned by Agricola were:—

Auripigmentum . Operment . Orpiment (As:S₃) . Orpiment Sandaraca . Rosgeel . Realgar (As S) . Realgar Arsenicum . Arsenik . Artificial arsenical oxide White arsenic Lapis subrutilus atque

.. splendens Mistpuckel .. Arsenopyrite (Fe As S) *Mispickel We are somewhat uncertain as to the identification of the last. The yellow and red sulphides, however, were well known to the Ancients, and are described by Aristotle, Theophrastus (71 and 80). Dioscorides (v, 81). Pliny (xxxIII, 22, etc.); and Strabo (xII, 3, 40) mentions a mine of them near Pompeiopolis, where, because of its poisonous character none but slaves were employed. The Ancients believed that the yellow sulphide contained gold—hence the name auripigmentum, and Pliny describes the attempt of the Emperor Caligula to extract the gold from it, and states that he did obtain a small amount, but unprofitably. So late a mineralogist as Hill (1750) held this view, which seemed to be general. Both realgar and orpiment were important for pigments, medicinal purposes, and poisons among the Ancients. In addition to the above, some arsenic-cobalt minerals are included under cadmia.

Iron Minerals.			
Ferrum purum	Gedigen eisen	Native iron	*Native iron
Terra ferria	Eisen ertz		
Ferri vena	Eisen ertz		
Galenae genus tertium	1		1
omnis metalli	Į.	Various soft and hard	1
inanissimi	Eisen glantz }	iron ores, probably	- Ironstone
Schistos	Glasköpfe oder	mostly hematite	
	blütstein	,)
Ferri vena jecoris			
colore	Leber ertz		
Ferrugo	Rüst	Part limonite	Iron rust
Magnes	Siegelstein oder		
	magnet	Magnetite	Lodestone
Ochra nativa	Berg geel	Limonite	Yellow ochre or ironstone
		Deet besette	Bloodstone or
Haematites	Blüt stein	Part hematite	
6 1 1 1		Part jasper	ironstone
Schistos	Glas köpfe	Part limonite	Ironstone
Pyrites	Kis	Pyrites	Pyrites
Pyrites argenti coloris	wasser oder	35 %	ATT : : :
241	weisser kis	Marcasite	*White iron pyrites
Misy	Gel atrament	Part copiapite	Misy (see note on p. 573)
Sory	Graw und	Partly a decomposed	Sory (see note on
	schwartz atramen		p. 573)
Melanteria	Schwartz und	Melanterite (native	Melanteria (see
	grau atrament		note on p. 573)
The classification of iron			